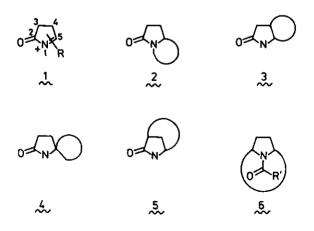
AN N-ACYLIMINIUM ROUTE TO THE 8-AZABICYCLO[3.2.1]OCTANE (TROPANE) AND THE 9-AZABICYCLO[4.2.1]NONANE RING SYSTEM SYNTHESIS OF (\pm) -ANATOXIN-A

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Abstract - Propargyl and allyl silanes 20-22, readily prepared from succinimide, cyclize on dissolution in formid acid to azabicycles 23-25 in excellent yields.

Intramolecular reactions of N-acyliminium intermediates 1, readily obtainable from succinimide, have proven to be eminently useful for the synthesis of azabicyclic compounds 1 . If the substituent R is a chain, containing a suitable and properly located nucleophilic carbon atom, either of the molecules 2-5 can be prepared, dependent on the site of attachment of R. Linearly fused systems 2^2 and 3^3 are available from 1- and 4-substituted 1, respectively. Spiro system 4 arises, if R is located at position 5^4 , and bridged system 5 is obtained, if R is at position 3^5 . In this communication we show that bridged system 6 is also easily accessible from succinimide by using N-acyliminium ion chemistry 6 . Azabicyclic 6 is the basic skeleton of pharmacologically important compounds like the tropane alkaloids 7 and anatoxin-a 8 .

Our synthetic route to 6 began with the addition reaction of the Grignard reagents, derived from $8-10^{9}$, 10, to succinimide, leading to hydroxy lactams 11-13. Best



yields were obtained by using 3 eq of Grignard reagent 11 . Alternatively, one can first employ 1 eq of MeMgCl to make the succinimide salt, followed by 2 eq of the more expensive Grignard reagent. Bromide $7^{9,10}$ was useless for our purposes, since all attempts to prepare its Grignard reagent failed. The crude hydroxy lactams 11-13 were not purified, but immediately reduced with NaBH₃CN in acetic acid 12 to pyrrolidones $14-16^{10}$. The overall yield of pure pyrrolidone was about $^{60\%}$ from succinimide.

Having established the first carbon-carbon bond by a Grignard reaction, the second carbon-carbon bond was thought to arise from an N-acyliminium cyclization reaction. We have found earlier that the allyl- and oropargylsilane moieties are excellent nucleophiles for this reaction type^{2c,5b}. To arrive at the required N-acyliminium ion, the methoxycarbonyl group was attached to the nitrogen. Best results were obtained by reaction of the lithium salts 14-16 (generated by using 1.1 eq of lithium dissopropylamide in THF at -78°C) with methyl cyanoformate¹³, furnishing 17-19¹⁰ in about 90% yield. Ethyl chloroformate appeared to be a very poor reagent for this purpose, giving a substantial amount of 0-acylation. Reduction of 17-19, by using the pH-controlled NaBH₄ method¹⁴ in ethanol at -20°C cleanly gave reaction of the ring carbonyl group to furnish hydroxy carbamates 20-22 in nearly quantitative yield. Higher reduction temperatures led to by-products resulting from ring opening and overreduction.

Ring closures of 20-22 were readily effected by dissolution in formic acid at room temperature. Propargylsilane 20 led to allene 23¹⁰ in 70% overall yield from 17.

The allenic structure was immediately apparent from the typical IR (1955 cm $^{-1}$) and 13 C NMR absorptions (6204.5 and 204.1, 106.9 and 106.3, 75.0; most carbon atoms showed two peaks due to hindered rotation). Allylsilane 21 afforded olefin $^{24}_{-}^{10,15}$ as a single stereoisomer in 75% overall yield from 18. The preference for formation of a six-membered ring with an equatorial vinyl group has been observed previously in related cyclization reactions 2c,5b . Allylsilane $^{22}_{-}$ gave olefin $^{25}_{-}^{10,15}$ in addition to a small amount of its stereoisomer (ratio 19:1) in 73% overall yield from 19.

The cyclization product 23-25 were obtained in clean, fast and irreversible reactions. These reaction characteristics are on the one side a consequence of the favorable nucleophilic properties of allyl- and propargylsilanes, and on the other hand due to the high electrophilicity of N-acyliminium ions 16 . The methodology presented here may well lead to various other ω -aza[x.y.1]bicycloalkanes, e.g. by using glutarimide and/or other nucleophilic chains as starting materials. Research in this direction is in progress.

Anatoxin-a (26) is a potent neurotoxin, produced by certain strains of the fresh water blue green alga Anabaena flos-aquae (Lyngb.) de Bréb⁸. This rather simple alkaloid with interesting pharmacological properties ¹⁷ has been synthesized both as racemate ¹⁸ and as pure enantiomer ^{17,19} by a number of research groups.

We herewith add a formal synthesis of racemic 26 starting from 25. Wacker oxidation ²⁰ of the inseparable 19:1 mixture of 25 (CuCl (1 eq), PdCl₂ (0.2 eq), 0₂, DMF,

H₂0) led to ketone 27 as a 1:1 mixture of isomers in 64% in addition to 5% of aldehyde 28. Treatment of 27 with in situ generated iodotrimethylsilane (NaI, Me₃SiCl) in refluxing acetonitrile²¹ furnished (±)-dihydro-anatoxin-a as a 4:1 mixture²² of 29 and its stereoisomer²³. This completed a formal synthesis of racemic anatoxin-a, since Rapoport et al. have published¹⁷ the conversion of 29 into 26. Our synthesis of the isomer mixture of 29 comprises 6 steps (23% overall) from succinimide and 10 steps (7% overall) from the THP-ether of 4-pentyn-1-ol^{8,2c}.

ACKNOWLEDGEMENT

This investigation was supported by the Netherlands Foundation for Chemical Research (SON) with financial aid from the Netherlands Organization for Advancement of Pure Research (ZWO).

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- 23. The yield of crude but virtually pure (NMR) dihydroanatoxin-a was 95%.

 Spectral data were identical to those published by Rapoport et al 19b.

Received, 22nd September, 1986