A NEW APPROACH TO C-9 HYDROXYLATION OF N(6)-SUBSTITUTED ELLIPTICINE

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<u>Abstract</u> - 6-Methylellipticine is converted to the corresponding 9-hydroxy derivative via a reaction sequence involving selective C-9 formylation followed by a Baeyer-Villiger rearrangement.

Several reports^{1a-e} suggest that the mechanism of antitumour activity of ellipticine (<u>1a</u>) is associated with an initial hydroxylation of the C-9 position of the pyridocarbazole nucleus.

9-Hydroxyellipticine (<u>1b</u>) is a metabolite which displays higher activity² than ellipticine.

The clinically employed anticancer drug elliptinium^R is 9-hydroxy-N-2-methylellipticinium acetate (2)^{3a,b}. Many of the chemical syntheses of 9-hydroxyellipticine utilize an appropriately

protected hydroxylated aromatic starting material ^{4a-b}. To our knowledge there is only one report in the literature which describes the synthesis of 9-hydroxyellipticine (1b) from ellipticine (1a) itself, albeit in low overall yield. The development of a facile procedure for the introduction of a hydroxyl group at the C-9 position of ellipticine and its derivatives is of practical value in the synthesis of potentially active ellipticine derivatives. In this communication we describe such a method.

The strategy for introduction of the hydroxyl group in $\underline{1a}$ recognized that the C-9 and possibly the C-7 positions of the molecule are potential sites of electrophilic attack. It was projected that a selective C-9 acylation of suitably protected $\underline{1a}$ followed by a Baeyer-Villiger rearrangement, should constitute a simple sequence for an overall hydroxylation process. To test the feasability of this procedure, a model study was carried out using N(9)-methylcarbazole ($\underline{3a}$). The formylation of $\underline{3a}$, employing Vilsmeyer conditions (DMF, POCl₃, 100-110°C) resulted in a selective reaction at C-3 and formation of $\underline{3b}^6$ in practical yields ($\frac{1}{2}$ 60%). The latter compound could be conveniently converted into the corresponding hydroxy derivative $\underline{3c}^7$ (90%) by treatment with $\underline{4a}_2\underline{0a}_2$ (aq. 35%) in the presence of acid⁸.

Having achieved the hydroxylation of <u>3a</u> we turned our attention to the ellipticine system and chose as the substrate 6-methylellipticine <u>1c</u>, which has been described by us ^{9a} and others ^{9b,c}. Compound <u>1c</u> underwent electrophilic substitution at C-9, exclusively, upon reaction with a variety of reagents ¹⁰. In the context of the present study, the formylation of <u>1c</u>, by reaction with ClCH₂OCHCl₂ (AlCl₃, CH₂Cl₂, 0°), followed by hydrolysis, proceeded in excellent overall yield (> 90%). The structure of 9-formyl-6-methylellipticine <u>1d</u> followed from its spectral data ¹¹. The rearrangement of <u>1d</u> to 9-hydroxy-6-methylellipticine <u>1e</u> ¹² proceeded smoothly (> 90%) upon reaction with hydrogen peroxide (aq. H₂O₂, 35%, H₂SO₄, MeOH). It is noteworthy that under these conditions the pyridine nitrogen (N-2) is not oxidized.

The hydroxylation procedure described in this communication is potentially applicable to the synthesis of a variety of 9-hydroxyellipticine derivatives.

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REFERENCES

- 1. (a) For a recent review see V.K. Kansal and P. Potier, Tetrahedron, 1986, 42, 2389;
 - (b) V.N. Reinhold, L. Bittman, R. Bruni, K. Thrun and D. Silveira, Proc.Am. Assoc.

 Cancer Research, 1975, 16, 135;
 - (c) P. Lesca, P. Lecointe, C. Paoletti and D. Mansuy, <u>C.R.Acad.Sc</u>., 1976, 282D, 1457; <u>Biochem.Pharmacol.</u>, 1977, 26, 2169;
 - (d) J.Y. Lallemand, P. Lemaitre, L. Beeley, P. Lesca and D. Mansuy, <u>Tetrahedron Letters</u>, 1978, 15, 1261;
 - (e) A.R. Branfman, R.J. Bruni, V.N. Reinhold, D.M. Silveria, M. Chadwick and D.W. Yessair, Drug, Metab.Disp., 1978, 6542.
- 2. C. Auclair, B. Meunier and C.Paoletti, Devel.Oncol., 1984, 15, 159.
- 3. (a) C. Paoletti, J.B. LePecq, N. Dat-Xuong, P. Lesca and P. Lecointe, <u>Curr.Chemother</u>., 1978, 1195:
 - (b) P. Jurek, J.E. Heron, J.E. Couette, T. Delozier and J.Y. Le Talaer, Cancer Treat.Rep., 1982, 66, 1909.
- (a) A. Gonyette, R. Reynaud, J. Sadet, M. Baillagé, C. Gansser, S. Cross, F. le Goffic,
 J.B. Le Pecq, C. Paoletti and C. Viel, <u>Eur.J.Med.Chem.-Chim.Ther.</u>, 1980, 15, 503;
 - (b) R.S. Miller and J.G. Stowell, <u>J.Org.Chem.</u>, 1983, 48, 886;
 - (c) P.A. Cranwell and J.E. Saxton, <u>J.Chem.Soc.</u>, 1962, 3482, also see: M. Sainsbury, <u>Synthesis</u>, 1977, 437;
 - (d) E. Bisagni, C. Ducrocq, J.-M. Lhoste, C. Rıvalle and A. Civier, <u>J.Chem.Soc. Perkin Trans. I</u>, 1979, 1706;
 - (e) D. Rousselle, J. Gilbert and C. Viel, C.R.Acad.Sc., 1977, 284C, 377;
 - (f) J. Gilbert, D. Rousselle, C. Gansser and C. Viel, J.Het.Chem., 1979, 16, 7;
 - (g) M.G. Saulnier and G.W. Gribble, <u>J.Org.Chem.</u>, 1983, 48, 2690; <u>J.Org.Chem.</u>, 1982, 47, 2810;
 - (h) For review see: G.W. Gribble and M.G. Saulnier, Heterocycles, 1985, 23, 1277.
- 5. Nguyen Dat-Xuong, M.-T. Adeline, P. Lecointe, M.-M. Janot, C.R.Acad.Sc., 1975, 281C, 623.
- 6. 3b: mp 79-82°C, lit. 74°C, (Ng.Ph.Buu-Hoi and Ng.Hoan, J.Am.Chem.Soc., 1951, 73, 98).

 IR (CHCl₃): 1685, 1620, 1590. NMR (CDCl₃)*: 6 3.7 (s, N-CH₃), 10.0 (s, -CHO).

- 7. 3c.: mp 145°C, NMR (DMSO-d₆): 6 3.75 (s, N-CH₃), 5.03 (dd, J_{1,2}=8.5, J_{2,4}=2, H-2), 5.66 (d, J_{2,4}=2, H-4). The compound, without its melting point, has been described in the literature, R.F. Novak, D.R. Koop and P.F. Hollenberg, Mol.Pharmacol., 1980, 17, 128.
- 8. M. Matsumoto, H. Kobayashı and Y. Hotta, <u>J.Org.Chem.</u>, 1984, 49, 4740.
- 9. (a) M.J. Wanner, G.J. Koomen and U.K. Pandit, Tetrahedron, 1983, 39, 3673.
 - (b) M. Watanabe and V. Snieckus, J.Am.Chem.Soc., 1980, 102, 1457;
 - (c) L.M. Werbel, M. Angelo, D.W. Fry and D.F. Worth, J.Med.Chem., 1986, 29, 1321.
- 10. Nitration (HNO $_3$ /AC $_2$ O) and acetylation (CH $_3$ COC1, AlC1 $_3$, CH $_2$ C1 $_2$, 0°C) of $\underline{\text{1c}}$ gave the corresponding 9-substituted derivatives.
- 11. $\underline{1d}$: mp 215°C (91%); IR (CHCl₃): 1677, 1590; NMR (CDCl₃)*: δ 2.96 (s, 5-CH₃), 3.14 (s, 11-CH₃), 4.06 (s, N-CH₃), 10.07 (s, -CH₀).
- 12. <u>1e</u>: mp > 350°C, (93%), IR (KBr): 1590, 1390, 1475. NMR (DMSO- d_6)*: 6 3.00 (s, 5-CH₃), 3.14 (s, 11-CH₃), 4.11 (s, N-CH₃), 9.49 (s, -OH).

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^{*} selected chemical shifts