

THE REACTION OF 1-METHYL-2-VINYLIINDOLES WITH 4-PHENYL-1,2,4-  
TRIAZOLINE-3,5-DIONE

Rosa Teresa Sanchis-Llorca,<sup>a</sup> José Sepúlveda-Arques,<sup>a\*</sup> Elena  
Zaballos-García,<sup>a</sup> and R. Alan Jones<sup>b\*</sup>

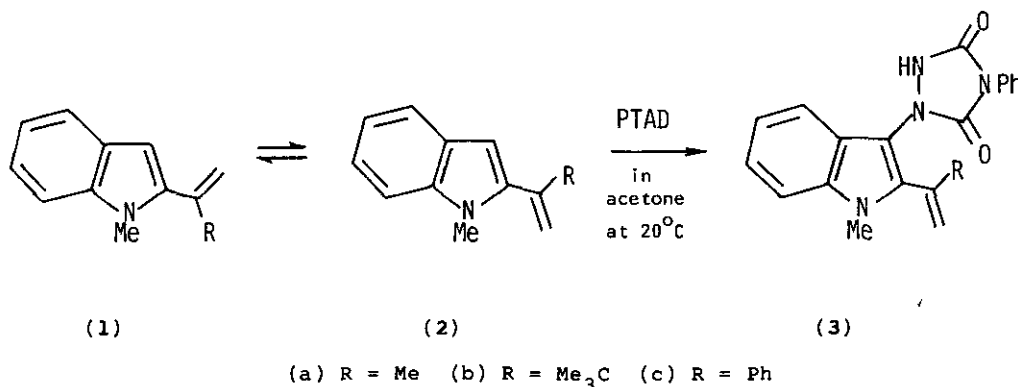
<sup>a</sup> Facultad de Farmacia, Departamento de Química Orgánica,  
Universidad de Valencia, 46010-Valencia, Spain

<sup>b</sup> School of Chemical Sciences, University of East Anglia,  
Norwich NR4 7TJ, UK

**Abstract** - 1-Methyl-2-vinylindoles ( $1 \rightleftharpoons 2$ ) react readily with  
4-phenyl-1,2,4-triazoline-3,5-dione at 20°C to yield the  
Michael-type 1:1 adducts (3). No evidence was found for the  
formation of any cycloadducts.

2-Vinylpyrroles react cleanly and in high yield at the free 5-position of the pyrrole ring with diethyl azodicarboxylate (DEAD) to yield Michael-type adducts<sup>1</sup> in a manner analogous to that reported for the corresponding reaction with simple 1-alkylpyrroles<sup>2</sup> or, when the 5-position is blocked or sterically hindered, they produce pyrrolo[3,2-c]pyridazine derivatives or dihydropyrrolyl-1,3,4-oxadiazines.<sup>1</sup> In contrast, the reactions of simple alkylindoles<sup>3</sup> and of 2- and 3-vinylindoles<sup>4</sup> with DEAD are extremely complex and produce many unstable products in low yield. We now report that 1-methyl-2-vinylindoles ( $1 \rightleftharpoons 2$ ) generally react readily at 20°C with 4-phenyl-1,2,4-triazoline-3,5-dione (PTAD) to produce stable crystalline 1-(1-methyl-2-vinylindol-3-yl)-4-phenyl-1,2,4-triazolidine-3,5-diones (3) in high yield. However, the instability of 1-methyl-2-vinylindole<sup>5</sup> is such that only polymeric material was obtained from the reaction with PTAD, and the relatively low yield of (3a) from 2-(1-methylindol-2-yl)propene can be rationalised in terms of its susceptibility to dimerise readily at relatively low temperatures.<sup>5</sup> No evidence was found for the formation of  $\pi 2 + \pi 4$  cycloadducts, which is in keeping with the reported Michael-type addition of 1-methyl-2-vinylindoles with dimethyl acetylene-dicarboxylate and their failure to yield dihydrocarbazoles.<sup>6</sup> These observations are

in accord with the analogous reaction of 1-methylindole with PTAD<sup>7</sup> and suggest that the "transoid" conformation for the diene system (2) predominates at ambient temperatures. Unfortunately, elevated temperatures, which should promote the conversion into the "cisoid" conformation, also promotes polymerisation of the vinylindoles ( $1 \rightleftharpoons 2$ ) and of the Michael adducts (3).



The Michael adducts (3) were fully characterised by elemental analysis, mass spectral data, and by <sup>1</sup>H and <sup>13</sup>C n.m.r. spectroscopy. In all of the products, n.m.r. signals, characteristic of the vinyl group, were observed near 5.8 (d, J = ca. 2Hz), 5.5 (d, J = ca. 2Hz), and 121 p.p.m. (t).

#### EXPERIMENTAL

<sup>1</sup>H and <sup>13</sup>C n.m.r. spectra were measured for ca. 40% solutions in CDCl<sub>3</sub> at 100 MHz and 25 MHz, respectively, using either a JEOL-JNM-100-FT or a Varian XL-100 spectrometer. All chemical shifts are given relative to the internal standard (Me<sub>4</sub>Si).

General procedure for the reaction of the vinylindoles with PTAD:— The vinylindole (0.0064 mol) in acetone (10 ml) was added with stirring to PTAD, obtained by the addition of *t*-butyl hypochlorite (0.069 g, 0.0064 mol) to *N*-phenylurazole (0.115 g, 0.00064 mol) in acetone (10 ml) at -60°C. The red solution of the PTAD was immediately decolourised. Evaporation of the solvent under reduced pressure and purification of the crude product by chromatography from Merck silica gel 60, using hexane:diethyl ether (1:1) as the eluant, gave the 1:1 adducts.

2-(1-Methylindol-2-yl)propene<sup>6</sup> gave 1-(1-methyl-2-(propen-2-yl)indol-3-yl)-4-phenyl-1,2,4-triazolidine-3,5-dione (0.70 g, 50%), m p 125 - 128°C, Rf 0.12 (Found: C, 69.1; H, 5.2, N, 16.2 C<sub>20</sub>H<sub>18</sub>N<sub>4</sub>O<sub>2</sub> requires C, 69.4, H, 5.2, N, 16.2%).  $\delta_{\text{H}}$  2.05 (3H, s), 3.60 (3H, s), 5.20 (1H, brs), 5.40 (1H, brs), and 7.10 - 7.40 (10H, m);  $\delta_{\text{C}}$  22.8 (q), 30.6 (q), 107.1 (s), 109.8 (d), 117.6 (d), 120.8 (d), 121.6 (t), 122.7 (d), 123.9 (s), 125.6 (d), 127.9 (d), 128.9 (d), 131.7 (s), 133.8 (s), 135.6 (s), 141.4 (s), 151.6 (s), and 153.1 (s).

3,3-Dimethyl-2-(1-methylindol-2-yl)but-1-ene<sup>6</sup> gave 1-(1-methyl-2-(3,3-dimethylbut-1-en-2-yl)indol-3-yl)-4-phenyl-1,2,4-triazolidine-3,5-dione (0.207 g, 82%), m p 104 - 107°C, Rf 0.12 (Found: C, 70.4; H, 6.2; N, 14.0, M<sup>+</sup> 388.1869 C<sub>23</sub>H<sub>24</sub>N<sub>4</sub>O<sub>2</sub> requires C, 71.1; H, 6.2; N, 14.4%, M<sup>+</sup> 388.1898).  $\delta_{\text{H}}$  1.10 (9H, s), 3.50 (3H, s), 5.18 (1H, d, J = 1Hz), 5.54 (1H, d, J = 1Hz), and 7.00 - 7.55 (9H, m);  $\delta_{\text{C}}$  31.3 (q), 37.3 (s), 108.4 (s), 109.9 (d), 117.8 (s), 120.6 (t), 120.9 (d), 122.6 (d), 123.5 (s), 125.6 (d), 128.0 (d), 129.0 (d), 131.8 (s), 135.6 (s), 140.5 (s), 148.0 (s), 151.2 (s), and 153.2 (s).

1-(1-Methylindol-2-yl)-1-phenylethene<sup>6</sup> gave 1-(1-methyl-2-(1-phenylethen-1-yl)indol-3-yl)-4-phenyl-1,2,4-triazolidine-3,5-dione (0.236 g, 90%), m p 169 - 171°C, Rf 0.16 (Found: C, 73.1; H, 4.9; N, 13.5 C<sub>25</sub>H<sub>20</sub>N<sub>4</sub>O<sub>2</sub> requires C, 73.5; H, 4.9; N, 13.7%).  $\delta_{\text{H}}$  3.23 (3H, s), 5.55 (1H, d, J = 1Hz), 5.83 (1H, d, J = 1Hz) and 7.00 - 7.50 (14H, m);  $\delta_{\text{C}}$  30.9 (q), 109.4 (s), 109.9 (d), 117.9 (d), 121.0 (d), 121.1 (t), 123.9 (s), 125.6 (d), 126.6 (d), 127.9 (d), 128.4 (d), 128.6 (d), 128.9 (d), 131.6 (s), 135.8 (s), 137.6 (s), 138.7 (s), 139.4 (s), 151.5 (s), and 153.1 (s).

#### ACKNOWLEDGEMENTS

We are grateful for financial support from C.A.I.C.Y.T. (project 1499/82) and the British Council. We also thank Mr John Egles of the Institute for Food Research Norwich for the measurement of the high resolution mass spectra.

#### REFERENCES

1. T. Aznar-Saliente, R.A. Jones, R.T. Sanchis-Llorca, and J. Sepúlveda Arques, J. Chem. Research (S), 1985, 12; (M) 0232 - 0240.
2. W. Flitsch and J. Heinrich, Tetrahedron Lett., 1980, 21, 3673; C.K. Lee, S.J. Lim, and C.S. Hahan, J. Org. Chem., 1980, 45, 1692.

3. M. Collona and A. Monte, Gazz. Chim. Ital., 1962, **92**, 1401; H. Pleininger and D. Wild, Chem. Ber., 1966, **99**, 3063.
4. (a) E. Zaballos-Garcia, Ph.D. Thesis, University of Valencia, 1986;  
(b) J.M. Weinmann, Ph.D. Thesis, University of Minnesota, 1964.
5. F.E. Ziegler, E.B. Spitzner, and C.K. Wilkins, J. Org. Chem., 1971, **36**, 1759.
6. R.A. Jones, P. Martinez Fresneda, T. Aznar-Saliente, and J. Sepúlveda Arques, Tetrahedron, 1984, **40**, 4837.
7. J.H. Hall, L. Kaler, and R. Herring, J. Org. Chem., 1984, **49**, 2579.

Received, 11th August, 1986