ON THE SYNTHESIS OF N-FORMYL-7-METHYLDEHYDRONORAPORPHINES

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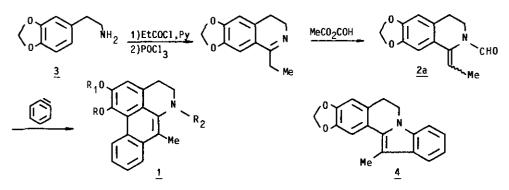
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<u>Abstract</u> - Intermolecular benzyne cycloaddition (IBC) was used for the synthesis of the aporphinoid <u>1a</u>, whose structure was corroborated by an independent photochemical route. Compound <u>1a</u> is not trichoguattine.

Structure <u>1a</u> and <u>1b</u> have recently been ascribed to trichoguattine¹ (isolated from **Guatteria trichostemon**) and duguespixine² (isolated from **Duguetia spixiana**), respectively. However, on the basis of our long experience with dehydroaporphines we felt that these structure assignments had been made on very weak spectroscopic grounds: the UV spectra of both trichoguattine³ and duguespixine² (λ_{max} : 430 nm) are quite different from those of N-ethoxycarbonylanonaine⁴ (λ_{max} : 376 nm) and closely related compounds, and the assignment of a three-proton singlet at 3.3 ppm to a C-7 methyl group on an N-formylnordehydroaporphine⁵ appeared highly speculative. We accordingly planned to check these structure assignments by synthesizing <u>1a</u> by intermolecular benzyne cycloaddition (IBC), a powerful method for the synthesis of isoquinoline alkaloids⁶⁻⁹.

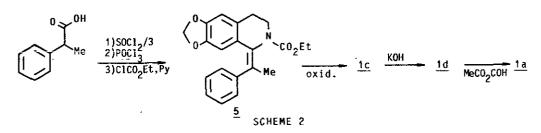
The dehydroaporphine <u>1a</u> was synthesized by reacting 1-ethylidene-2-formyl-6,7methylenedioxy-1,2,3,4-tetrahydroisoquinoline <u>2a</u> with benzyne, as described in our previous reports⁶⁻⁹. The required precursor <u>2a</u> was prepared uneventfully from 3,4methylenedioxyphenethylamine <u>3</u> (Scheme 1). The key reaction of <u>2a</u> with benzyne (generated by thermal decomposition of benzenediazonium-2-carboxylate¹⁰) yielded <u>1a</u> in an unexpectedly low yield(11%), the major product being dibenzopyrrocoline <u>4</u>¹¹. The spectroscopic properties of synthetic <u>1a</u> were those expected for this structure but differed considerably from those of trichoguattine. Particularly significant are the data: MS:m/z 305(M⁺, 100%); IR(KBr) 1680 cm⁻¹; NMR(CDCl₃): 9.09 (d,J=9 Hz,1H,H-11), 8.37(s,1H,CHO), 2.61(s,3H,Me); UV, λ_{max} : 253, 288, 325, 357, 376. Furthermore, the chromatographic behaviour of our synthetic <u>1a</u> was found to be significantly different from that of an authentic sample of natural trichoguattine kindly provided by Prof. A. Cavé.

The structure of synthetic <u>1a</u> was further confirmed by an independent synthesis along the lines of the classic photochemical approach (Scheme 2): N-carbethoxyben-zylideneisoquinoline 5^{12} prepared by conventional chemistry was irradiated (Hanovia lamp, 450 watts, vycor filter, I₂-copper acetate, 18 h) to yield <u>1c</u> (50%), which was then hydrolysed under the conditions described by Lenz⁴ (Scheme 2). The nor-dehydroaporphine <u>1d</u> so obtained was finally converted into <u>1a</u> (acetic-formic mixed anhydride), which was shown to be totally identical (MS,UV,PMR,tlc) with the product obtained by IBC.



a) $R+R_1=CH_2$: $R_2=CHO$. b) R=Me; $R_1=H$; $R_2=CHO$ c) $R+R_1=CH_2$; $R_2=CO_2Et$. d) $R+R_1=CH_2$; $R_2=H$

SCHEME 1



In conclusion, trichoguattine does not have the structure originally assigned to it. In our opinion, the structure proposed for the closely related duguespixine must also be wrong.

ACKNOWLEDGEMENTS We thank the Comisión Asesora (Spain) for its financial support and Professor A. Cavé for sending us an authentic sample of trichoguattine as well as copies of its UV, IR, NMR and MS spectra.

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All new compounds gave correct elemental analysis and/or spectroscopic data.

Received, 7th January, 1987