A SYNTHESIS OF A SPIROCYCLOPROPANE-1,1'-CARBAPENEM Choung Un Kim^{*}, Peter F. Misco, and Bing Y. Luh Bristol-Myers Company, Pharmaceutical Research and Development Division, P. O. Box 5100, Wallingford, CT 06492-7660, USA <u>Abstract</u> - The total synthesis of the spirocyclopropane 1,1'-

carbapenem <u>13</u> has been achieved through the Wittig cyclization of the phosphorane <u>11</u>. A modified Simmons-Smith reaction on the olefinic ketone <u>6</u> gave the key cyclopropane derivative <u>7</u> in good yield.

A recent report¹ by Shih and co-workers on the synthesis of $1-\beta$ -methylcarbapenems <u>1</u> which possess much improved chemical and metabolic stability than 1-unsubstituted carbapenems has created considerable interest in the chemical modification of the carbapenem ring at the C-1 position of carbapenems.² Specifically, this paper describes an efficient synthesis of a new carbapenem derivative having the cyclopropane ring attached at the C-1 position as depicted in the structure 2.



The chiral acetoxyazetidinone 3, readily available by several routes³ was chosen as the starting material for the construction of the cyclopropane intermediate 7. Nucleophilic displacement of the acetoxy group of 3^4 with the lithium anion 4 (THF, -20°, 2h) gave the trans product 5 in 75% yield as a mixture of diastereomers. Thermolysis of 5 (toluene, reflux 4h) generated cleanly the olefinic ketone 6 in 80% yield after chromatographic purification on silica gel. When the enone 6 was treated with Zn-Cu/CH₂I₂ (<u>in situ</u> generation of Zn-Cu from zinc dust and CuCl)⁵ in anhydrous ether, the desired cyclopropane intermediate 7 was obtained in 60-70% yield. This modified Simmons-Smith procedure was quite reliable for reproducibility and scale up. Assignment of the structure 7 is based upon the presence of four cyclopropane ring protons at 1.0-1.4 in the ¹H NMR.¹⁰ Condensation of 7 with allylglyoxylate (benzene, reflux) gave the hemiaminal 8, which upon treatment with thionyl chloride followed by triphenylphosphine and 2,6-lutidine afforded the phosphorane <u>10</u> in 63% overall yield. Exposure of <u>10</u> to TFA-anisole (0°, 1h) provided the desilylated product <u>11</u> in 85% yield. The phosphorane <u>11</u> was transformed into the desired carbapenem <u>12</u> in 82% yield by the intramolecular Wittig procedure (xylene, reflux, 14h) developed by Woodward.⁶ Finally removal of the allyl group with tetrakis(triphenyl-phosphine) palladium⁷ in the presence of potassium 2-ethylhexanoate gave the potassium salt <u>13</u> in 35% yield after HP-20 column purification. The product showed a UV absorption maximum at 275 nm (£6,700) and a β -lactam carbonyl IR absorption at 1780 cm⁻¹ characteristic of the bicyclic β -lactam.

In a similar manner as described above the olefinic ketone <u>6</u> was converted into the phosphorane <u>16</u> via the intermediates <u>14</u> and <u>15</u>. Wittig cyclization of <u>16</u> was achieved by heating in refluxing toluene (24h) to give the 1-methylene carbapenem <u>18</u> in 79% yield. Unfortunately, when the desilylated phosphonate <u>17</u> was subjected to the Wittig conditions, only decomposed material was obtained. Thus, it appears that decomposition of <u>19</u> is faster than the cylization of <u>17</u> in this case. Several other attempts to obtain the unprotected 1-methylene carbapenem free acid have not been successful so far. Recently the synthesis of Δ^3 -1-methylene-1-carbacephems has been achieved by the similar chemistry described above.⁸ The new carbapenem <u>13</u> showed a broad activity against gram positive and negative bacteria. However, to our surprise, compound <u>13</u> was hydrolyzed in vitro by dehydropeptidase-I about 100 times faster than the corresponding 1- β -methylcarbapenem analog.⁹







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- 9. We are indebted to Dr. M. Hitchcock of the Screening and Bio-chemical Research Department of Bristol-Myers Company for this test. The full biological activity of $\frac{1}{2}$ (R = CH₃) and $\frac{13}{2}$ will be published elsewhere.
- 10. Selected NMR Data:

 $\frac{6}{2} (CDC1_3 - D_2 O): \delta 2.41 (s, 3H), 2.94 (dd, J = 2.5, 8.6 Hz, 1H), 4.26 (m, 1H), 4.61 (d, J = 2.5 Hz, 1H), 6.14 (s, 1H), 6.22 (s, 1H).$ $<math display="block">\frac{7}{2} (CDC1_3): \delta 1.0 - 1.38 (m, 4H), 1.21 (d, J = 6.5 Hz, 3H), 1.89 (s, 3H), 2.26 (d, J = 2.1 Hz, 1H), 4.08 (m, 1H), 4.32 (d, J = 2.1 Hz, 1H).$ $\frac{12}{2} (CDC1_3): \delta 1.0 - 1.40 (m, 4H), 1.32 (d, J = 6.2 Hz, 3H), 1.70 (s, 3H), 2.95 (dd, J = 3.0, 7.8 Hz), 3.92 (d, J = 3.0 Hz), 4.10 (m, 1H).$ $\frac{13}{2} (D_2 O): \delta 1.1 - 1.5 (m, 4H), 1.22 (d, J = 6.2 Hz, 3H), 1.77 (s, 3H), 3.31 (dd, J = 3.3, 7.8 Hz), 4.06 (d, J = 3.3 Hz, 1H), 4.21 (m, 1H).$ $\frac{18}{2} (CDC1_3): \delta 2.15 (s, 3H), 3.14 (dd, J = 2.4, 7.5 Hz, 1H), 4.28 (m, 1H), 4.59 (d, J = 2.4 Hz, 1H), 5.21 (d, J = 2.0 Hz, 1H), 5.28 (d, J = 11.9 Hz, 1H), 5.39 (d, J = 2.0 Hz, 1H), 5.43 (d, J = 11.9 Hz, 1H).$

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