SYNTHESIS AN0 DETERMINATION OF THE ABSOLUTE STEREOCHEWISTRY OF THE ENANTIOMERS OF 3-SUBSTITUTED **1.2.3.4-TETRAHYDROISOQUINOLINES** RELATED TO THE CALCIUM ANTAGONIST VERAPAMIL'

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- The four stereoisomers of 3-[4-cyano-4-(3.4-dimethoxypheny1)- **5mthylhex-l-yll-1.2.3.4-tetrahydro-6.7dimethoxy-2mthy1isoquinoline** (2) were prepared. The stereochemistry of the side chain quaternary carbon was derived from iodide 4 of known absolute configuration. The absolute stereochemistry at the 3-position of the tetrahydroisoquinoline was determined by correlation with L-DOPA. These compounds are related to the calcium antagonist verapamil and the calcium antagonist activity was greatest for the (3S,4S) isomer.

Structure-activity relationship (SAR) studies to date on the Group 11 calcium antagonist verapamil $(1)^{2-4}$ have dealt with substitutions in the aromatic rings^{5,6} or modifications at the quaternary carbon atom.⁷⁻⁹ Our interest in the SAR of verapamil prompted us to prepare and evaluate a number of rigid phenethylamine analogs and preliminary results of these investigations indicated that compound **2** had interesting pharmacological activity. In this structure. the **[4-cyano-(3.4dimethoxyphenyl)-5-methyllhexan-l-yl** cwonent of verapamil is attached to the 3-position of **6.7dimethoxy-1.2.3.4-tetrahydroisoquinoline** (THIQ). Additional interest in **2** arose from the fact that remarkably few 3-substituted THlQ derivatives have been reported. This is in contrast to numerous examples of pharmacologically active 1, 2, and 4 substituted THlQ compounds.1° Since compound **2** contains two asynnetric centers, it was of interest to prepare the four isomers inherent in this structure. Reported herein are the synthesis and establishment of the absolute stereochemistry of these four stereoisomers.

The preparation of 2 from 3-(3,4-dimethoxyphenyl)propionic acid (3) is outlined in Scheme I. Alkylation of the dianion of **3** with iodide **4** (prepared from the corresponding chloride1' uith NaI-acetone) gave a mixture of diastereomeric acids 5. Treatment of 5 with diphenylphosphoryl azide (DPPA)¹² gave isocyanate <u>6</u> which, without isolation, was reduced with NaBH₄ to an

approximately 1:l mixture of formamides **1** as determined by tlc and 'H NWR analysis. Eischler-Napieralski cyclization of **Z** follwed by NaBH4 reduction gave THIQ @ which was subjected to Eschweiler-Clarke methylation to afford **2.** Compound **Z** thus obtained was an approximately 1:1 mixture of diastereomers as determined by $\frac{1}{H}$ NMR and hplc analysis.

OMe \sum COOH MeO $\diagdown\diagdown\diagdown\diagdown$ MeC ۱M۵ Ý ̈́CΝ **5 X=** COOH 6 **Xz** NCO 7 **X=** NHCHO OMe MeC $d.e.f$ **OMA** CΝ Met 8 R=H 2 R=CH.

> a) LDA, THF; b) (PhO)₂PON₃. TEA, toluene; c) NaBH_A, DME; d) POC1₃, AcN; e) NaBH_A, EtOH; f) CH₂O, HCO₂H.

Scheme I

The four isomers of **2** were obtained by the same basic route as was used for the parent compound (Schemes II and III). The (S) -iodide $\overline{4}$ was obtained from the known optically active chloride which had been used to prepare the enantiomers of verapamil.¹¹ Aklylation of 3 with (S)-4 followed by DPPA treatment and NaBH_A reduction as before gave diastereomeric formamides (-)-9 and (-)-10 which were separated by chromatography on silica gel. The pure diastereomers thus obtained were converted to the THIQ final products as previously. described to afford $(-)$ -(3 $8,45$)-11 and (+)-(3 $5,45$)-12. The stereochemical assignments for position 3 of the THIQ are made in a following section.

The other two isomers, $(+)-(35,48)-11$ and $(-)-(3R,4R)-12$, were obtained by an analogous series of reactions (and chromatographic separation) from $(R) -4$ which was obtained from the corresponding known (R) -chloride¹¹ (Scheme III). The physical data obtained for these four isomers (Table I) is in accord with the assignment of 11 and **¹²**as enantiomeric pairs. Compounds $(-)-11$ and $(+)-12$, both of which were derived from $(S)-4$, are assigned enantiomer excesses (ee) of >88% based on the optical purity (>88% ee) of $(S) - 4$ (experimental section). Compounds $(+)-11$ and $(-)-12$ are assigned ee of $>96\%$ based on the purity of $(\frac{R}{2})-4$ $(\geq 96\%$ ee).

Scheme II

TABLE I

PHYSICAL DATA FOR THlO DERIVATIVES

The assignment of the absolute stereochemistry of the 3 position of the THIQ was made by synthesis of $(-)-(3R,4S)-11$ from $(S)-13$ which was prepared from $(-)-(S)-3,4$ dihydroxyphenylalanine (L-DOPA)¹³ (Scheme IV). Amino alcohol 13 was converted to the BOC derivative **1Q** and then to tosylate **15.** Treatment of **15** with vinylmagnesium bromide and cuprous iodide afforded **16.** Precedent for this useful conversion comes from the reported reaction of BOC-serine derivatives with organocuprates.¹⁴ Conversion of 16 to alcohol 17 and mesylate 18 proceeded without incident. However, attempted alkylation of 18 with the lithium salt of 2-(3,4-dimethoxyphenyl)-3-methylbutyronitrile (19) gave none of the desired product (21). The major side reaction which occurred under these conditions was intramolecular alkylation of the BOC nitrogen to form a BOC protected pyrollidine. Fortunately, use of freshly prepared iodide
20, while an unstable material, did afford a modest yield of 21 as a mixture of two
... diastereomers. Deprotection of 21 followed by formylation gave the diastereomeric formamides which were separated by chromatography as before to afford $(-)$ -(3S,7R)-9 and $(+)$ -(3R,7R)-10.

The major formamide thus obtained, $(-)-9$, was cyclized, reduced, and N-methylated to afford (-)-11 identical in all respects to material obtained by the previous route (Scheme II). This sequence established the 3R configuration of the THIQ (-)-11 and, coupled with the known $4S$ configuration, established the complete absolute stereochemistry of $(-)-11$ as $3R,4S$). The complete absolute stereochemistries of the other three isomers follow from the enantlomeric and diastereomeric relationships already established.

Scheme IV

a) (\underline{t} -BuOCO)₂O, THF; b) TsCl, pyridine; c) (CH₂=CH)MgBr, CuI, THF; d) (Disiamyl)₂BH, THF; NaOH, H₂O₂; e) MsCl, TEA, CH₂Cl₂; f) Nal, acetone; g) 19. THF; h) HCO₂H, HCO₂Ac; i) POCl₃. AcN; j) NaBH₄, EtOH; **k)** CH20, HC02H.

These isomers were tested for calcium antagonist activity as determined by their ability to relax the barium chloride contracted rat aortic strip (Table 11). It was found that the majority of the calcium antagonist activity of the parent compound 2 resided in the (3S,4S)-12 enantiomer. This is consistent with the finding that the S enantiomer of verapamil is responsible for most of the parent compounds activity.¹⁵ In the THIQ derivatives, constraining the phenylethylamine moiety in the THIQ framework apparently introduces an additional stereochemical requirement in the binding of these compounds to the receptor which mediates the calcium antagonism in this series. **¹⁶**

Compound	(mM) E_{50}	Relative Potency
$(+)$ –(35,4R)–11	4.0	0.08
$(-)$ – (3R, 4S) – 11	0.7	0.4
$(+)-(35,45)-12$	0.1	з
$(-) - (3R, 4R) - 12$	2.0	0.15
2	0.3	٦
Verapamil (1)	0.2	

TABLE 11 CALCIUM ANTAGONIST PROPERTIES OF THIO DERIVATIVES[&]

^aDetermined by measuring the relaxation of BaCl₂ contracted rat aortic strips.

EXPERIMENTAL

'H NMR spectra were recorded with a Bruker **WI1** 300 instrument and are reported in ppm 6 downfield from an internal standard of TWS. Wedium pressure (flash) chromatography was performed using 230400 mesh Merck Kieselgel. Melting points are uncorrected. Elemental analyses were done by the Syntex analytical department.

(S) and $(R)-3-Cyano-3-(3,4-dimethoxyphenyl)-6-iodo-2-methylhexane (S)-4 and (R)-4)$

Resolution of **4-(3.44imethoxyphenyl)-4-(2-propyl)4-pentenoic** acid using (-)-cinchonidine was carried out as described in the literature.¹¹ The optical purities of the (S) - $(-)$ -acid and $(R)-(+)$ -acid thus obtained were determined by 1 H NMR analysis of the diastereomeric amides derived from the acid chlorides (oxalyl chloride, benzene) and $(\underline{S})-(-)$ -a-methylbenzylamine. In the 1 H NMR spectrum of the amide from the (S)-(-)-acid, resonances for the methoxy groups were at δ 3.63 and δ 3.87 while in the amide from the (R)-(-)-acid, resonances at δ 3.74 and *&* 3.86 were observed. In the amide from the racemic acid, these four resonances appeared with equal intensities (thus ruling out any differences in the rates of formation of the diastereomeric amides). The amide from the (S) -(-)-acid was determined to have a diastereomric ratio of >94:<6 by integration of the methoxy resonances at *b* 3.63 and *6* 3.74 respectively. Thus, an ee of >88% was assigned to the (S) -(-)-acid. The amide from the $(\underline{R})-(+)$ -acid was pure (>98%) by ¹H NMR analysis and the $(\underline{R})-(+)$ -acid was therefore assigned an ee of >96%.

These acids were converted into the (\underline{S}) -iodide \underline{A} and (\underline{R}) -iodide \underline{A} by conversion to the corresponding chlorides as described. 11 followed by treatment with NaI in acetone. They were obtained as thick oils which were used without purification in the next step.

13S.lRl and **~3S.7S~-7-Fommido-3.8di-~334-dimethox~~hen~l)-2-neth~l-3-octanenitrile** (1-)-9 and $(-)-10$)

To a solution of 21 ml (0.15 mol) of diisopropylamine in 480 ml of THF at -50°C was added 91.8 m1 (0.147 mol) of 1.6W n-butyllithium in hexane. HWPA (56 ml) was added followed by a solution of 15.1 g (0.072 mol) of **3-(3.4dimethoxyphenyl)propionic** add **(3)** in 75 ml of THF. The

resulting solution was allowed to warm to $10^{\circ}C$ over 1 h and was then cooled to $-50^{\circ}C$. A solution of 28.0 g (0.072 mol) of (S) -iodide $\underline{4}$ in 75 ml of THF was added and the mixture was stirred and allowed to warm to room temperature. After 12 h the mixture was poured into water and washed with Et₂0. The aqueous layer was acidified with HCl, extracted with Et₄0, and the Et₂0 extract was washed with aqueous NaHSO₃ and brine, dried (Na₂SO₄), and evaporated to 31.0 g (91%) of crude acid. This material (30.5 g. 0.066 mol) was dissolved in a mixture of 200 ml of toluene and 11.7 ml (0.084 mol) of triethylamine and 15 ml (19.3 g, 0.07 mol) of diphenylphosphoryl azide was added. The mixture was gradually heated to 100°C over 1 h at which time IR analysis showed conversion to the isocyanate. The mixture was concentrated under reduced pressure and the residue was dissolved in 300 ml of OME. The solution was cooled in an ice bath and 4.0 g (0.11 mol) of NaBH_A was added in small portions over a 2 h period. The mixture was added to water and extracted with EtOAc. The extract was washed with brine. dried (Na₂SO_A), and evaporated to afford a mixture of formamides (-)-9 and (-)-10 which were separated by medium pressure chromatography (10% hexane-EtOAc). The first component eluted was $(-)-9:(011)$ 5.8 g (19% overall yield); ¹H NMR (CDC1₃) & 0.76 (d, 3H), 1.00 (m, IH), 1.16 (d. 3H). 1.35-1.60(m, 3H). 1.90-2.10 (m, 3H). 2.68 (dd. 2H. H-7ab). 3.85 (s, 6H). 3.87 (s, 3H), 3.90 (s, 3H), 4.20 (m, 1H, H-6), $6.58-6.90$ (m, 6H), 8.03 (s, 1H, C<u>H</u>O); [ali5 -6.84' (c 0.2. MeOH). **Anal.** Calcd for C2,H36N205: C.69.20; H, 7.74; N. 5.98. Found: C, 68.89; H. 7.80; N. 5.81.

The second component eluted was $(-)-10:(01)$ 8.7 g (28% overall yield); ¹H NMR (CDCl₃) & 0.78 (d, 3H), 1.10 (m, 1H), 1.16 (d, 3H), 1.30-1.55 (m, 3H), 1.75 (m, 1H), 2.00-2.20 (m, 2H), 2.65 (d, 2H), 3.85 (s, 3H), 3.86 (s, 3H), 3.89 (s, 6H), 6.60-6.92 (m, 6H), 8.07 (s, 1H, CHO); $[6]_0^{25}$ -2.98°(c 0.2, MeOH). <u>Anal</u>. Calcd for C₂₇H₃₆N₂O₅: C, 69.20; H, 7.74; N, 5.98. Found: C. 68.81; **H.** 7.85; N. 5.84.

~S)-3-1(S~-4-C~ano-4-(3.4-dimethox~~henyl)-5~thvlhex-l-~l1-1.2.3.4-tetrahydro-6.7-dimethoxy-2 methylisoquinoline hydrochloride ((35.45)-12)

To a solution of 5.67 g (0.012 mol) of $(-)$ -10 in 100 ml of MeCN was added 2.8 ml (0.03 mol) of POC1₃. The resulting solution was stirred 3 h at room temperature, concentrated at reduced pressure, and partitioned between Et₂0 and aqueous NH_AOH. The Et₂O was evaporated and the residue was dissolved in 100 ml of EtOH, cooled in an ice bath, and treated with 1.0 g of NaBH₄. The mixture was stirred for 30 min and then added to water and acidified with HCl. The solution was washed with Et₂0 and the aqueous layer was basified with NH₄OH and extracted with Et₂0. The Et₂0 extract was evaporated and the residue was dissolved in 30 ml of formic acid and 30 ml of 37% aqueous formaldehyde. This solution was stirred at 100°C for 1 h and then added to ice-water which was subsequently basified with NH_AOH . Et₂0 extraction followed by drying ($Na₂SO_A$) and evaporation afforded the crude free base as an oil. This was dissolved in a small amount of EtOH-HCl and crystallization was induced by adding Et₂0. The yield of $(3\underline{S},4\underline{S})-12$ was 4.5 g (74%) : ¹H NMR $(\text{Me}_2S0-d_{\text{K}})$ 0.71 (d, 3H), 1.13 (d. 3H). 2.65 and 2.70 (broad s. 3H. pseudo axial, equitorial NCtj3). 3.73 (s. 6H). 3.78 adding et. The yield or (3<u>5,45)-12</u> was 4.5 g (74%): H NMR (Me₂50-d₆) 0.71 (d, 3H),
1.13 (d, 3H), 2.65 and 2.70 (broad s, 3H, pseudo axial, equitorial NC<u>H</u>3), 3.73 (s, 6H), 3.78
(s, 6H), 6.74 (s, 2H), 6.91 (s, 1H), C. 66.85; H. 7.81; N. 5.51. Found: C. 66.79; H, 7.85; N. 5.57.

$(R)-3-[S-A-Cyano-4-(3.4-dimethoxypheny])-5-methylhex-1-y1]-1,2,3.4-tetrahydro-6.7-dimethoxy-2$ methylisoquinoline hydrochloride ((3R.4S)-11)

This compound was prepared from $(-)-9$ in 67% yield using procedures analogous to those used for

 $(35.45) - 12$: ¹H NMR (Me₂SO-d₆) 6 0.72 (d, 3H), 1.14 (d, 3H), 2.64 and 2.78 (broad s, 3H, NCH3). 3.73 (5. 3H). 3.74 (s. 3H). 3.78 (s. 6W. 6.74 (s. 2H). 6.93 (s. 1H). 6.96 (s, 2H). Anal. Calcd for C₂₈H₃₈N₂O₄-HC1-H₂O: C, 65.54; H, 7.93; N, 5.38. Found: C, 64.10; H. 7.98; N. 5.30.

J3R.75) and **(3R.7Rl-7-Formamido-3.84i-(3.4dimethox~~hen~ll-2+th~l-3-octanenitrile** ((+l-9 and $(+)$ -10)).

A mixture of (+)-9 and (+)-10 was obtained in 78% overall yield from acid 3. The pure compounds were obtained by medium pressure chromatography (10% hexane-EtOAc). The first component eluted was (+)-<u>9</u>:(oil) [']H NMR identical to (-)-<u>9</u>: [۵]p ²³ +8.07° (c 0.33,
MeOH). <u>Anal</u>. Calcd for C₂₇H₃₆N₂O₅: C,69.2O; H, 7.74; N, 5.98. Found: C, 69.09; H, 7.76; N. 5.93. The second component eluted was (+)-10:(oil) ¹H NMR identical to (-)-10;
[a]₀²⁵ +3.84° (c 0.37. MeOH). <u>Anal</u>. Calcd for C₂₇H₃₆N₂O₅: C, 69.20; H, 7.74; N, 5.98. Found : C. 69.00; H, 7.95; N. 5.74.

jRl-3-I(R)-4-C~ano-4-(3.4-dimethox~~hen~ll-5-meth~lhex-l-~ll-1.2.3.4-tetrah~dr0-6,7dimethoxy-2 methylisoquinoline hydrochloride ((3R.4R)-12)

This compound was prepared from **(+)-10** in 76% yield and was spectrally identical to $(35.45) - 12$. Anal. Calcd for C₂₈H₃₈N₂O₄.HCl: C, 66.85; H, 7.81; N, 5.57. Found: C, 66.77; H. 7.86; N. 5.54.

$(S)-3-[R)-4-Cyano-4-(3.4-dimethoxyphenyl)-5-methylher-1-y1]-1,2,3,4-tetrahydro-6.7-dimethoxy-2$ methylisoquinoline hydrochloride ((3S,4R)-11)

This compound was prepared from $(+)$ -9 in 66% yield and was spectrally identical to $(3R,4S)-11$. $\frac{\texttt{Anal.}}{\texttt{A}}$ Calcd for $\frac{\texttt{c}}{28}\text{H}_{38}\text{N}_2\text{O}_4$ *HCl \cdot 0.5 H₂0: C, 65.57; H, 7.87; N, 5.47. Found: C, 65.59; H. 7.89; N. 5.46.

~RSl-3-1(RSl-4-Cvano4-(3.4-dimethoxy~henyll-5+thylhex-l-Y11-1.2.3.4-tetrahydro-6.7dimethoxy-2 methylisoquinoline hydrochloride (2)

This was prepared according to the procedures described for the synthesis of the isomers except that (t)-4 was used and the diastereomeric formamides were not separated. The final product was analyzed by HPLC using a Spherisorb 5 C8 column with a mobile phase of 0.1 M triethylamine adjusted to pH 5.0:keOH:AcN. 60:35:10. The two diastereomers were cleanly separated with the (R,S)/(S,R) pair eluting as the first peak followed by the (R.R)/(S,S) pair as the second. The diastereomer ratio was determined to be ~50:50. Anal. Calcd for C₂₈H₃₈N₂0₄·HCl: C. 66.85; H. 7.81; N. 5.57. Found: C. 66.92; H. 7.58; N. 5.60.

(S)-2-(tert-Butoxycarbonylamino)-3-(3.4-dimethoxyphenyl) propanol (14)

To a solution of 23.3 g (0.11 mol) of 13^{13} in 400 ml of THF was added 24.0 g (0.11 mol) of di-tert-butyldicarbonate and the resulting solution was heated at reflux for 1 h. The mixture was concentrated at reduced pressure and the residue was filtered through silica gel (EtOAc). Evaporation afforded 32.4 g (95%) of (S)-<u>14</u>: mp 92-93°C; [a] $_{\sf D}^{\leq 2}$ -19.6° (c 0.3, MeOH). di-<u>tert</u>-butyldicarbonate and the resulting solution was heated at reflux for 1 h. The mixture
was concentrated at reduced pressure and the residue was filtered through silica gel (EtOAc).
Evaporation afforded 32.4 g (95 4.57.

$(S)-2-(text-Butoxycarbonylamino)-3-(3.4-d1-methoxyphenyl)-1-(4-methylbenzenesulfonyloxy)-1$ propane (15)

A solution of 32.4 g (0.10 mol) of 14, 29.8 g (0.16 mol) of p-toluenesulfonyl chloride, and 0.63 g of OWAP in 100 ml of pyridine was stirred at room temperature for 48 h. The mixture was diluted with ether and washed uith aqueous cupric sulfate and water. The ether was dried (MgSO₄) and evaporated to a residue that was purified by medium pressure chromatography (30%) (mysu₄) and evaporated to a reside that was parried by medium pressure thromatography (30%)
EtOAc-hexane) to afford 43.5 g (90%) of <u>15</u>: mp 146-148°C; [a]₁²⁵ -15.4° (c 0.36,
CHC1₃). <u>Anal</u>. Calcd for C₂₃H₃₁ N0 6.73; N. 2.98.

~RI-2-Itert-Butox~carbon~lamino~-1-I3.44imethox~hen~114-~entene I161

Vinylmagnesium bromide (37.3 ml, 0.06 mol, 1.6M in THF) was added to a -5°C suspension of 5.8 g (0.03 mol) of cuprous iodide in 75 ml of THF. The mixture was cooled to -70°C and a solution of 4.6 g (0.01 moll of **15** in 80 ml of THF was added slowly. The resulting mixture was stirred at -70°C for 0.5 h and then allowed to warm to room temperature. The mixture was poured into aqueous NH_ACl and extracted with Et₂0. The Et₂0 was dried (MgSO₄) and evaporated. Purification of the residue by medium pressure chromatography (30% EtOAc-hexane) gave 2.0 g (63%) of <u>16</u>: mp 91-92°C; ¹H NMR (CDC1₃) 6 1.41 (s, 9H), 2.05-2.30 (m, 2H), 2.70 (m, 2H),
3.86 (s, 3H), 3.86 (m, 1H, C<u>H</u>N), 3.87 (s, 3H), 5.10 (m, 2H), 5.80 (m, 1H), 6.70-6.82 (m, 3H); [6]²⁵ -20.9° (c 0.26, MeOH). <u>Anal</u>. Calcd for C₁₈H₂₇NO₄: C, 67.26; H, 8.47; N, 4.36. Found: C, 67.33; H. 8.49; N. 4.36.

~S~4-(tert-8utox~carbonylamino)-5-(3.44imethoxhenvll-l-~entanol (172

A solution of 1.6 g (0.005 mol) of **16** was added to a 0°C solution of disiamylborane (0.0075 mol, from 1.6 ml of 2-methyl-2-butene and 7.5 ml of 1.0M BH₃) in 25 ml of THF. The mixture was stirred at room temperature for 1 h, cooled to 0°C, and treated with 1.7 ml of 3N NaOH and 1.7 ml of 30% H_2O_2 . The resulting solution was stirred 0.5 h at room temperature, diluted uith ether, and washed with water and brine. Evaporation of the ether and crystallization of the residue from Et₂0-hexane gave 1.2 g (70%) of <u>17</u>: mp 73-75°C; $[a]_D^{25}$ +0.89° (c 0.2, MeOH). Anal. Calcd for C₁₈H₂₉NO₅: C, 63.69; H, 8.61; N, 4.13. Found: C, 63.58; H, 8.64; **1.** 4.06.

1-1-9 and I+)-10 frm 17. Scheme IV

A solution of 1.0 g (0.003 mol) of 17 in 15 ml of CH₂C1₂ and 1.3 ml of triethylamine was cooled to 0°C and 0.26 ml (0.003 mol) of methanesulfonyl chloride was added. After 10 min the mixture was added to water and the CH₂Cl₂ was separated, dried (MgSO₄), and evaporated. The crude mesylate (18) was dissolved in 25 ml of acetone and 3.0 g of Nal were added. The $mixture$ was heated under reflux of 0.5 h. poured into water, and extracted with $Et₂0$. The Et₂0 extract was washed with aqueous NaHSO₃, water, and brine, dried (MgSO₄), and evaporated to crude **20** which was used at once in the next reaction.

A solution of 1.3 g (0.006 mol) of nitrile **19** in 5 ml of THF was added to a -70°C solution of LDA (0.006 mol) in 20 ml of THF. After 20 min at -70°C, the mixture was treated with a solution of crude 20 (from above) in 5 ml of THF and the resulting solution was allowed to warm to 0° C. Aqueous HCl was added and the mixture was extracted with Et₂0. The Et₂0 extract was washed with water and brine. dried (MgSO_A), and evaporated to a residue which was

purified by medium pressure chromatography (30% EtOAc-hexane) to afford 0.6 g (38%) of the diastereomeric mixture 21. This material was dissolved in 5 ml of formic acid and the resulting solution was heated at reflux for 15 min. The mixture was concentrated under reduced pressure and the residue was treated with 2 ml of formic-acetic anhydride. Et_a0 was added and the mixture was washed with water, aqueous NaHCO₃, and brine. After drying (Na₂SO₄) the ether was evaporated and the residue was purified by medium pressure chromatography (EtOAc). The first component eluted was (-)-9. (0.18 g), identical by tlc and nmr analysis with material from the other route: $\begin{bmatrix} a \end{bmatrix}^{25}_{0}$ -4.3° (c 0.28, MeOH). The second component eluted was **(t)-10** (0.04 g) identical by tlc and nmr analysis with material from the other route: $[a]_0^{25}$ +2.3* (c 0.3, MeOH).

1-)-(3R.45)-11 from Scheme **IV**

This was prepared from fommide **(-)-9,** derived from **13** in Scheme IV, in 68% yield according to the previously described procedure. Anal. Calcd for $C_{2R}H_{38}N_2O_4$ -HCl-0.5 H₂0: C, 65.57; H, 1.81; N, 5.47. Found: C, 65.59; H, 7.89; N, 5.46.

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