STUDIES IN SPIROHETEROCYCLES: PART X:REACTIONS OF DIAZOMETHANE WITH 3-AROYLMETHYLENE-INDOL-2-ONES AND SYNTHESIS OF SOME NOVEL FLUORINE CONTAINING SPIRO[CYCLOPROPANE-1,3'(3H)INDOL]-2'(1'H)-ONES

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<u>Abstract</u> - Reaction of ethereal diazomethane with fluorinated 3-aroylmethylene-indol-2-ones has been investigated for the first time yielding spiro [cyclopropane-1,3'(3H)indol]-2' (1'H)-ones. The structures of all compounds have been confirmed on the basis of elemental analysis, ir, ¹H-nmr and mass fragmentation pattern.

The wide spectrum of biological activities associated with indole¹ and the unique structure of isatin makes it an attractive target for synthesizing new spiro systems possessing biological activity. Keeping this in view, we have carried out the reaction of 3-aroylmethylene-indol-2-ones with ethereal diazomethane. A literature survey reports that on addition of diazomethane to 3-alkenyl substituted oxindoles having an acid² or ester³ grouping at the unsaturated carbon atom at C-3. spiropyrrolizines^{2,3} are formed. However, when diphenyl diazomethane is added, spirocyclopropanes³ are obtained. This particular reaction has now been reinvestigated with an acyl group on unsaturated carbon atom at C-3 since this reaction has not been investigated so far and besides, this has an extra reactive site viz C=0 of acyl group. Spectral studies indicate that the cycloaddition occurs at C=C bond and since the presence of electron withdrawing carbonyl group at the unsaturated carbon atom makes the α -carbon a strong nucleophilic centre, it results in the loss of nitrogen and leads to the formation of spiro [cyclopropane-1, 3'(3H)indol]-2'(1'H)-ones instead of spiro [3H-indole-3,3'-pyrazolin]-2-ones as reported earlier. In this investigation,

ethereal diazomethane was treated with 3-aroylmethylene-indol-2-ones(II) at room temperature when spiro[cyclopropane-1,3'(3H)indol]-2'(1'H)-ones(III) were obtained in 60-70% yields. The high yields of the products indicated that cycloaddition reaction occurred only at one unsaturated centre. 3-Aroylmethylene-indol-2-ones(II) were earlier obtained by the dehydration of 3-hydroxy-3phenacylindol-2-one(I) with HCl-CH₃COOH; the latter being prepared by the reaction of indole-2,3-dione with substituted acetophenone in the presence of diethylamine as catalyst (Scheme)

Scheme



X = H, 5-F, 6-F R = H, $COCH_3$, CH_2N R'= 2-F, 4-F; R"=H, 2-CH₃, 3-CH₃, 5-CH₃, 3-Cl, 3-F

The spiro compounds were characterized by correct elemental analysis, ir, ¹H-nmr, ¹⁹F-nmr and mass spectra. The ir spectrum showed characteristic absorption bands at 3030-3280 cm⁻¹ due to N-H stretching and 1710 cm⁻¹ and 1670 cm⁻¹ due to two carbonyl groups. This indicates that none of the $\C=0$ groups of 3-aroylmethylene-indol-2-one are involved in this reaction and the cycloaddition occurs at $\[c]C=C\]$ bond. The ¹H-nmr signals due to $\[c]CH_2\]$ group appear as two double doublets at § 2.11 and § 2.52 ppm indicating a difference in the chemical environment of the two protons. Apart from this, a triplet centred at § 3.6 ppm due to $\[c]CH\]$ proton, a singlet at § 7.7 ppm due to NH proton and a multiplet from § 6.96 to 7.85 ppm for aromatic protons were also observed. The mass spectrum further supported the formation of 2-(4-fluorobenzoyl)-spiro [cyclopropane-1,3'(3H)indol]-2'(1'H)-one (IIIa) as the parent peak at 281(M⁺) corresponded to the molecular weight. The presence and position of fluorine was confirmed by ¹⁹F-nmr, when fluorine attached to indole ring was observed at -113 to -117 ppm; fluorine attached at 4-position of benzoyl ring at -119 to -120 ppm (IIIa-d,g,h) and at 2-position at -105 to -106 ppm (III e,f).

EXPERIMENTAL

All melting points are uncorrected. Ir spectra were recorded on Perkin-Elmer (Model-577) in KBr pellets. ¹H- and ¹⁹F-nmr were recorded on Jeol (Model-FX 90 Q) at 89.5 MHz using TMS as external reference for ¹H-nmr. ¹⁹F- spectra were taken in CDCl₃ at 84.25 MHz using hexafluorobenzene as external reference. Purity of all compounds were checked by tlc on silica gel plates. 5-/6-Fluoroindole-2,3-dione was prepared by literature method⁴⁻⁶.

<u>3-Hydroxy-3-(4-fluorophenacyl)-indol-2-one (Ia)</u>- Equimolar amount of indole-2,3-dione (0.01 M) and 4-fluoroacetophenone (0.01 M) were refluxed in absolute ethanol (30 ml) in presence of 2-3 drops of diethylamine as catalyst for 30-60 min and then kept at room temperature for 4 days. After filtration and recrystallization with ethanol, the obtained compound (Ia) was checked on tlc as a single spot, yield 1.3 g (65%), mp 165°C.) $\frac{\text{cm}^{-1}}{\text{max}}$: 3200-3380 (N-H and 0-H), 1700, 1670 cm⁻¹ (both $\subset =0$); ¹H-nmr (CDCl₃): δ 4.2 (broad OH), 5.1 (s, CH₂), 6.8 to 8.1 (m, aromatic protons) and 7.8 ppm (s, NH).

<u>3-(4-Fluorobenzoyl)methylene-indol-2-one (IIa)</u> - A mixture of 3-hydroxy-3-(4-fluorophenacyl)-indol-2-one (Ia) (0.01M), conc. HCl (0.5 ml) and glacial acetic acid (10 ml) were heated at 95°C for 15-30 min. After the addition of ethanol (10 ml), the solid compound was filtered and recrystallized from ethanol, yield 2.2 g (80%), mp 187°C.) $\frac{\text{cm}^{-1}}{\text{max}}$: 3250 (broad, N-H), 1720, 1690 cm⁻¹

S.No.	х	R	R'	R"	Mp (°C)	Yield %	Molecular formula	Nitrogen Calc.	% Found.
8.	Н	Н	4 - F	Н	19 7	65	^C 17 ^H 12 ^{FNO} 2	4.98	4.79
b.	6 - F	н	4 - F	3-Cl	148	63	^C 17 ^H 10 ^F 2 ^{ClNO} 2	4.19	4.17
с.	5-F	Н	4-F	^{3-СН} 3	225	68	^C 18 ^H 13 ^F 2 ^{NO} 2	4.47.	4.45
d.	5-F	Н	4-F	2-CH3	196	68	^C 18 ^H 13 ^F 2 ^{NO} 2	4.47	4.42
e.	5-F	Н	2 - F	5-CH3	213	60	^C 18 ^H 13 ^F 2 ^{NO} 2	4.47	4.44
f.	5-F	Н	2 - F	3 - F	197	64	^C 17 ^H 10 ^F 3 ^{NO} 2	4.41	4.40
g.	5 - F	соснз	4-F	Н	158	63	^C 19 ^H 13 ^F 2 ^{NO} 3	4.10	4.08
h	5-F	CH ₂ N	4-F	2-CH3	181	68	C ₂₄ H ₂₄ F ₂ N ₂ O ₂	6.82	6.80

Physical and Analytical Properties of Spiro [cyclopropane-1,3'(3H)indol]-2'(1'H)-ones(III)*

* The substituents and S.No. are the same for compounds I and II also.

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Table - I

(both $\geq C=0$), ¹H-nmr (CDCl₃): § 5.3 (s,C<u>H</u>), 6.7 to 8.3 (m, aromatic protons), 7.8 ppm (s,N<u>H</u>).

<u>2-(4-Fluorobenzoyl)-spiro</u> [cyclopropane-1,3' (3H) indol]-2'(1'H)-one (IIIa) -Ethereal solution of diazomethane was added in two portions to a suspension of 3-(4-fluorobenzoyl)methylene-indol-2-one(IIa) (2 g) in ether. The reaction mixture was then stirred at room temperature for 3 h. The excess ether was distilled off giving a sticky compound, which was purified by column chromatography on silica gel. Elution was carried out using solvents of increasing polarity, the white compound from benzene - ethyl acetate (4:1) fraction was collected and recrystallized from benzene - ethyl acetate mixture, yield 1.3 g (65%), mp 197°C (Found: C, 72.38; H,4.22; N,4.78. $C_{17}H_{12}FNO_2$ requires C, 72.59; H, 4.27; N, 4.98%). \mathcal{V}_{max}^{cm-1} : 3280-3030 (broad, N-H), 1700, 1670 cm⁻¹ (both $\bigcirc C=0$); ¹H-nmr (CDCl₃); & 2.11, 2.52 (dd, CH₂), 3.6 (t,CH), 6.96 to 7.85 (m, aromatic protons) and 7.7 ppm (s,NH). MS m/z 281 (M⁺). All other compounds given in table-I (IIIb-h) were prepared in a similar manner.

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