

TWO NEW ACRIDONE ALKALOIDS FROM GENUS CITRUS PLANTS<sup>1</sup>Motoharu Ju-ichi,<sup>\*,a</sup> Mami Inoue,<sup>a</sup> Chihiro Ito,<sup>b</sup>Miyuki Matsuoka,<sup>b</sup> Hiroshi Furukawa,<sup>b</sup> and Ichiro Kajiura<sup>c</sup>Faculty of Pharmaceutical Sciences, Mukogawa Women's University,<sup>a</sup>Nishinomiya, Hyogo 663, Japan; Faculty of Pharmacy, Meijo University,<sup>b</sup>

Tempaku, Nagoya 468, Japan; Okitsu Branch Fruit Tree Research

Station,<sup>c</sup> Minist. Agric. Forest. & Fisheries, Okitsu, Shizuoka

424-02, Japan

**Abstract** — Two new acridone alkaloids named citrusamine (1) and 1,3-dihydroxy-10-methylacridone (2) were obtained from the root of the hybrid seedlings of Okitsu No.17 x Hyuganatsu and also from Kawano-Natsudaidai.

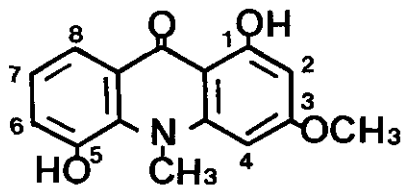
Extensive studies in our laboratories of the constituents of the genus Citrus have resulted in the isolation of a number of acridone alkaloids and coumarins.<sup>1,2-8</sup>

In this paper we wish to report the isolation and structure elucidation of two new acridone alkaloids, citrusamine (1) and 1,3-dihydroxy-10-methylacridone (2), from the roots of several hybrid seedlings<sup>9</sup> resulting from a cross of Okitsu No.17<sup>10</sup> x Hyuganatsu (Citrus tamurana Tan.) and also from Kawano-Natsudaidai<sup>9</sup> (C. natsudaidai Hayata).

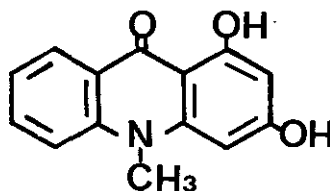
Citrusamine (1), yellow prisms, mp 243-246°C, C<sub>15</sub>H<sub>13</sub>NO<sub>4</sub>. The uv [ $\lambda_{\max}^{\text{EtOH}}$  nm (log  $\epsilon$ ) 264 (4.71), 285 (sh 4.35), 310 (4.16), 406 (3.87)] and ir spectra (3580, 1625, 1595 cm<sup>-1</sup>) showed the characteristic absorptions of the 1-hydroxy-9-acridone system.<sup>11</sup> The presence of an N-methyl and a methoxy group was clear from the <sup>13</sup>C-nmr signals ( $\delta$  40.70 and 55.42) and <sup>1</sup>H-nmr signals ( $\delta$  4.11 and 3.95). The existence of two phenolic hydroxy groups was apparent from the ir (3580 cm<sup>-1</sup>) and <sup>1</sup>H-nmr spectra ( $\delta$  14.67 and 9.34). In the aromatic proton region of the <sup>1</sup>H-nmr spectrum, meta-coupled two-proton signals at  $\delta$  6.22 and 6.49 (each 1H, d, J = 2.44 Hz) and ABC type three-proton signals at  $\delta$  7.91 (1H, dd, J = 7.81, 1.46 Hz), 7.31 (1H, dd, J = 7.81, 1.46 Hz) and 7.16 (1H, t, J = 7.81 Hz) were observed. The lower field signal at

$\delta$  7.91 was characteristic of H-8 in the 9-acridone skeleton, thus suggesting the location of these five aromatic protons at C<sub>2</sub>, C<sub>4</sub>, C<sub>6</sub>, C<sub>7</sub> and C<sub>8</sub>. In NOE experiments, irradiation at the frequency corresponding to the methoxy proton at  $\delta$  3.95 gave enhancements of 15.8% and 8.9% of the signals at  $\delta$  6.22 (H-2) and 6.49 (H-4), respectively. On the other hand, irradiation of the N-methyl proton signal at  $\delta$  4.11 showed 11.3% enhancement merely on the signal at  $\delta$  6.49 (H-4). The above results indicated the location of the methoxy and the hydroxy groups at C-3 and C-5, respectively. Consequently, the structure of citrusamine was established as formula 1.

1,3-Dihydroxy-10-methylacridone (2), mp >295°C, C<sub>14</sub>H<sub>11</sub>NO<sub>3</sub> was obtained as light yellow cubes. The presence of 1-hydroxy-9-acridone nucleus was suggested by the characteristic uv spectrum<sup>11</sup> [ $\lambda_{\text{max}}^{\text{EtOH}}$  nm (log  $\epsilon$ ) 249 (4.44), 262 (4.54), 271 (4.59), 328 (3.84), 394 (3.75)]. The existence of one N-methyl group was apparent from the <sup>13</sup>C-nmr signal at  $\delta$  32.59 and <sup>1</sup>H-nmr signal at  $\delta$  3.88. The <sup>1</sup>H-nmr showed signals at  $\delta$  14.89 and 9.5 (each 1H, s) indicating the presence of two hydroxy groups and in the aromatic proton region, two meta-coupled proton signals at  $\delta$  6.49 and 6.19 (each 1H, d, J= 1.95 Hz) and four proton signals at  $\delta$  8.38 (1H, dd, J= 8.79, 0.98 Hz),  $\sim$ 7.8 (2H, m) and 7.33 (1H, ddd, J= 8.3, 1.95 Hz) were observed. These signal patterns suggested that C<sub>5</sub>, C<sub>6</sub>, C<sub>7</sub> and C<sub>8</sub> were unsubstituted. On the basis of these spectral results, the structure of this alkaloid could be proposed for 2 which had been known as a synthetic compound. To confirm this speculation, authentic 1,3-dihydroxy-10-methylacridone (2) was prepared by the procedure of Hlubucek et al.<sup>12</sup> and was identified with the natural compound by comparisons of nmr, ir spectra and TLC comparisons. This is the first isolation of this compound from natural sources.



(1)



(2)

## EXPERIMENTAL

Extraction and Isolation : The dried root of plant material (hybrid seedlings resulting from a cross of Okitsu No.17 x Hyuganatsu, 500 g)<sup>13</sup> was extracted with acetone. Evaporation of the solvent yielded 54.4g of residue. The residue was chromatographed over silica gel and eluted successively with hexane, benzene, 50% CH<sub>2</sub>Cl<sub>2</sub>-benzene, CH<sub>2</sub>Cl<sub>2</sub>, AcOEt, acetone and MeOH. The acetone eluate was further separated with repeated PTLC using the solvent systems of iso-propyl ether, acetone-CHCl<sub>3</sub> (1:19) and acetone-hexane (1:1) to afford citrusamine (1) (19.8 mg) and 1,3-dihydroxy-10-methylacridone (2) (23.3 mg).

Citrusamine (1): yellow prisms, mp 243-246°C (from acetone); ms m/z: 271 (M<sup>+</sup>, base peak), 256, 242, 228, 213; uv  $\lambda_{\max}^{\text{EtOH}}$  nm (log  $\epsilon$ ): 264 (4.71), 285 (sh, 4.35), 310 (4.16), 406 (3.87); ir (cm<sup>-1</sup>, CHCl<sub>3</sub>): 3580, 1625, 1595; <sup>1</sup>H-nmr ( $\delta$ , acetone-d<sub>6</sub>): 14.67 (1H, s), 9.34 (1H, s), 7.91 (1H, dd, J= 7.81, 1.46 Hz), 7.31 (1H, dd, J= 7.81, 1.46 Hz), 7.16 (1H, t, J= 7.81 Hz), 6.49 (1H, d, J= 2.44 Hz), 6.22 (1H, d, J= 2.44 Hz), 4.11, 3.95 (each 3H, s); <sup>13</sup>C-nmr ( $\delta$ , CDCl<sub>3</sub>+ DMSO-d<sub>6</sub>): 180.48 (s), 165.88 (s), 164.83 (s), 147.34 (s), 147.28 (s), 133.85 (s), 123.92 (s), 123.60 (s), 120.13 (d), 116.36 (d), 105.03 (s), 94.23 (d), 89.79 (d), 55.42 (q), 40.70 (q).

1,3-Dihydroxy-10-methylacridone (2): light yellow cubes, mp >295°C (from CH<sub>2</sub>Cl<sub>2</sub>). ms m/z: 241 (M<sup>+</sup>, base peak), 226, 213, 212, 198, 184; uv  $\lambda_{\max}^{\text{EtOH}}$  nm (log  $\epsilon$ ): 249 (4.44), 262 (4.54), 271 (4.59), 328 (3.84), 394 (3.75); ir (cm<sup>-1</sup>, KBr): 3400, 1630, 1600; <sup>1</sup>H-nmr ( $\delta$ , acetone-d<sub>6</sub>): 14.89 (1H, s), 9.5 (1H, s), 8.38 (1H, dd, J= 8.79, 0.98 Hz), ~7.8 (2H, m), 7.33 (1H, ddd, J= 8.3, 1.95 Hz), 6.49 (1H, d, J= 1.95 Hz), 6.19 (1H, d, J= 1.95 Hz), 3.88 (3H, s); <sup>13</sup>C-nmr ( $\delta$ , acetone-d<sub>6</sub>)<sup>14</sup>: 165.71 (s), 134.90 (d), 126.99 (d), 122.05 (d), 116.21 (d), 96.82 (d), 91.77 (d), 32.59 (q).

## ACKNOWLEDGEMENT

The authors thank Dr. H. Tanino of Meijo University for differential NOE (400 MHz) experiments.

## REFERENCES

- 1) Part V on "Constituents of Domestic Citrus Plant" and Part XVII in the series "Acridone Alkaloids". Part IV: M. Ju-ichi, M. Inoue, R. Tsuda, N. Shibukawa, and H. Furukawa, Heterocycles, 1986, 24, 2777; Part XVI: M. Ju-ichi, M. Inoue, K. Aoki,

- and H. Furukawa, Heterocycles, 1986, 24, 1595.
- 2) T.-S. Wu, H. Furukawa, and C.-S. Kuoh, Heterocycles, 1982, 19, 273.
  - 3) T.-S. Wu, C.-S. Kuoh, and H. Furukawa, Phytochemistry, 1983, 22, 1493.
  - 4) T.-S. Wu, C.-S. Kuoh, and H. Furukawa, Chem. Pharm. Bull., 1983, 31, 895.
  - 5) T.-S. Wu and H. Furukawa, Chem. Pharm. Bull., 1983, 31, 901.
  - 6) M. Ju-ichi, M. Inoue, Y. Fujitani, and H. Furukawa, Heterocycles, 1985, 23, 1131.
  - 7) A. T. McPhail, M. Ju-ichi, Y. Fujitani, M. Inoue, T.-S. Wu, and H. Furukawa, Tetrahedron Let., 1985, 26, 3271.
  - 8) T.-S. Wu, S.-C. Huang, T.-T. Jong, and H. Furukawa, Heterocycles, 1986, 24, 41.
  - 9) The plant was cultivated and collected in Fruit Tree Research Station, Okitsu Branch, Shizuoka.
  - 10) Okitsu No.17 is a hybrid unnamed selection resulting from a cross of Hyuganatsu (C. tamurana Tan.) x Sweet orange (C. sinensis L. Osbeck cv. Fukuhara).
  - 11) J. Reisch, K. Szendrei, E. Minker, and I. Novak, Pharmazie, 1972, 27, 208.
  - 12) J. Hlubucek, E. Ritchie, and W. C. Taylor, Aust. J. Chem., 1970, 23, 1881.
  - 13) Isolation procedures from the root of Kawano-Natsudaidai (C. natsudaidai Hayata) will be reported elsewhere.
  - 14) Because of its low solubility in acetone, other quaternary carbon signals could not be observed.

Received, 5th March, 1987