SODIUM BOROHYDRIDE DESULPHURIZATION OF THIOLACTAMS TO AMINES

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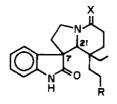
Abstract - Sodium borohydride desulphurizes a variety of thiolactams(1-8) in t-butanol/methanol mixture under reflux to the respective amines (9-16) in good yield.

Thiolactams are usually reduced to amines by Raney nicket which can also desulphurize many sulphur containing groups. Aluminium amalgam in neutral alcoholic solution has also been used in some cases. However, to the best of our knowledge, the use of NaBH, in this conversion is not yet known. Herein we wish to report an efficient method of desulphurization of thiolactams (1-8) to the corresponding amines (9-16) (Scheme 1) in ca. 70-94% yield (Table 1) using NaBH, in t-butanol/methanol mixture, a reagent recently shown to reduce esters and tertiary &-lactams.

In a typical experiment, MeOH (4 ml) was added dropwise over a period of 2-5 h to a refluxing mixture of a thiolactam (0.5 mmol) and NaBH₄ (6-8 mmol) in t-BuOH (6 ml). Usual work-up followed by column chromatography and crystallization afforded the desired amine (70-94%).

In case of compounds 5 and 6, since during purification the products were getting easily transformed to the lactams, further reduction of the enamine double bond was ensured to fully characterise

Scheme I



the product. This was done by cooling the reaction mixture to 0° C followed by successive addition of HOAc (5 ml) and NaBH₄ (\sim 2 mmol) with stirring for 20 min at room temperature. It was basified and after usual work-up the product was purified by chromatography.

Table 1, Desulphurization of Thiolactams (1-8) to Amines (9-16)

Thiolac ~	NaBH ₄ (mol	Time (h)	Product (Yield%)	Observed mp	Reported mp	Ref.
	 [12				
?	16	2	10(90)	126-128	128-129	5
}	12	2	11(85)	88-90	90-93	7
1	16	2	12(90)	107-109	109-110	5
5	12	3	13(70)	98-101	100-102	8
•	12	4	14(78)	Foam	_	9
•	16	3	15(94)	Wax	-	9
1	16	2	16(71)	104-107	108-110	10

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- All the products were characterized by ir, nmr and ms analyses and all new compounds gave satisfactory elemental analyses.
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