

STRUCTURE AND STEREOCHEMISTRY OF CORYTENSINE, A NEW PHTHALIDEISOQUINOLINE
ALKALOID FROM CORYDALIS OCHOTENSIS

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Abstract— A new phthalideisoquinoline alkaloid, corytensine, has been
isolated from Corydalis ochotensis Turcz., and its structure and stereochem-
istry have been established as (1) from spectral data and a single crystal
X-ray analysis.

In previous paper¹, we reported the isolation and structural elucidation of several new alkaloids
from Corydalis ochotensis Turcz. We now describe the structural elucidation of a further new
phthalideisoquinoline alkaloid, corytensine, isolated from the whole herb of the same plant which
was collected at Nan-Tou, Taiwan.

Corytensine (1) was isolated as colorless prisms², mp 215–215.5 °C (Me₂CO), [α]_D²⁵+168° (c =0.5,
CHCl₃). Elemental analysis established the molecular formula as C₂₀H₁₉NO₆. An absorption band at
239 nm (log ϵ 3.91) and a characteristic tetrahydroisoquinoline band at 290 nm (end absorption) in
the uv spectrum of 1 were very similar to those of (+)-egenine (2)³(with undefined stereochemistry
at C-7') and (-)-narcotinehemiacetal (3)⁴. The ir spectrum of 1 showed a hydroxy absorption at 3400
cm⁻¹. Salient features of its ¹H-nmr (400 MHz, CDCl₃) spectrum were the presence of an N-methyl
singlet at δ 1.96 (3H), four mutually coupled resonance at δ 2.47 (1H, dt, J=15.5 and 3.0 Hz, H-4eq),
2.54 (1H, ddd, J=13.0, 10.5, and 3.0 Hz, H-3ax), 3.00 (1H, dt, J=10.5 and 3.0 Hz, H-3eq), and 3.20
(1H, ddd, J=15.5, 13.0, and 3.0 Hz, H-4ax). Two one-proton benzylic signals for H-1 and H-9 appear-
ed as singlets at δ 3.68 and 5.29 thereby indicating a dihedral angle of ca. 90° between these
atoms. Signals for two methylenedioxy groups occurred at δ 5.90 and 5.94 (each 1H, d, J=1.7 Hz),

and 6.04 and 6.08 (each 1H, d, J=1.5 Hz). An AB quartet at δ 6.83 and 6.85 (J=8 Hz) was assigned to two ortho-related aromatic ring protons (H-2' and H-3'). Other one-proton singlets which appeared at δ 6.25, 6.60 and 6.71 were attributed to H-7', H-5, and H-8, respectively. A singlet at δ 6.25 in the ^1H -nmr spectrum and a carbon resonance at δ 97.7 in the ^{13}C -nmr spectrum indicated the presence of a hemiacetal system. The EI mass spectrum lacked a molecular ion peak but contained instead of several small peaks at m/z 370 ($M+1$)⁺, 368 ($M-1$)⁺, and 352 ($M-17$)⁺; the base peak, which occurred at m/z 190, was due to the familiar benzylic cleavage of phthalideisoquinolines. The foregoing spectral data led to the assignment of structure **1** to corytensine.

The complete structure and relative stereochemistry of corytensine were defined unequivocally by nOe studies (Table 1) and a single-crystal X-ray analysis. Crystal data: $\text{C}_{20}\text{H}_{19}\text{NO}_6$ (**1**), $M = 369.38$, orthorhombic, space group $P2_12_12_1$, $a = 12.936(2)$ Å, $b = 7.649(1)$ Å, $c = 1684.1$ Å, $D_{\text{calcd.}} = 1.457$ g cm^{-3} , μ (Cu-K α radiation, $\lambda = 1.5418$ Å) = 8.6 cm^{-1} . One octant of intensity data (1737 reflections) was recorded from a crystal of dimensions ca. 0.40 x 0.40 x 0.60 mm mounted on an Enraf-Nonius CAD-4 diffractometer (Cu-K α radiation, incident-beam graphite monochromator; θ -2 θ scans, $\theta_{\text{max.}} = 67^\circ$). The data were corrected for the usual Lorentz and polarization effects. The crystal structure was solved by direct methods.⁵ Initial non-hydrogen atom positions were obtained from an

Table 1. ^1H -Nmr Chemical Shift (δ) and NOE Data for Corytensine (**1**)

Proton irradiated	Proton observed	% Area increase
N-Me (1.96)	H-1 (3.68)	8.8
	H-3ax (2.54)	4.5
	H-3eq (3.00)	4.5
H-1 (3.68)	H-9 (5.29)	4.2
	H-8 (6.71)	5.8
	N-Me (1.96)	5.6
	H-2' (6.83)	2.8
H-9 (5.29)	H-1 (3.68)	3.4
	H-8 (6.71)	7.1
	H-2' (6.83)	2.9
H-7' (6.25)	no nOe	

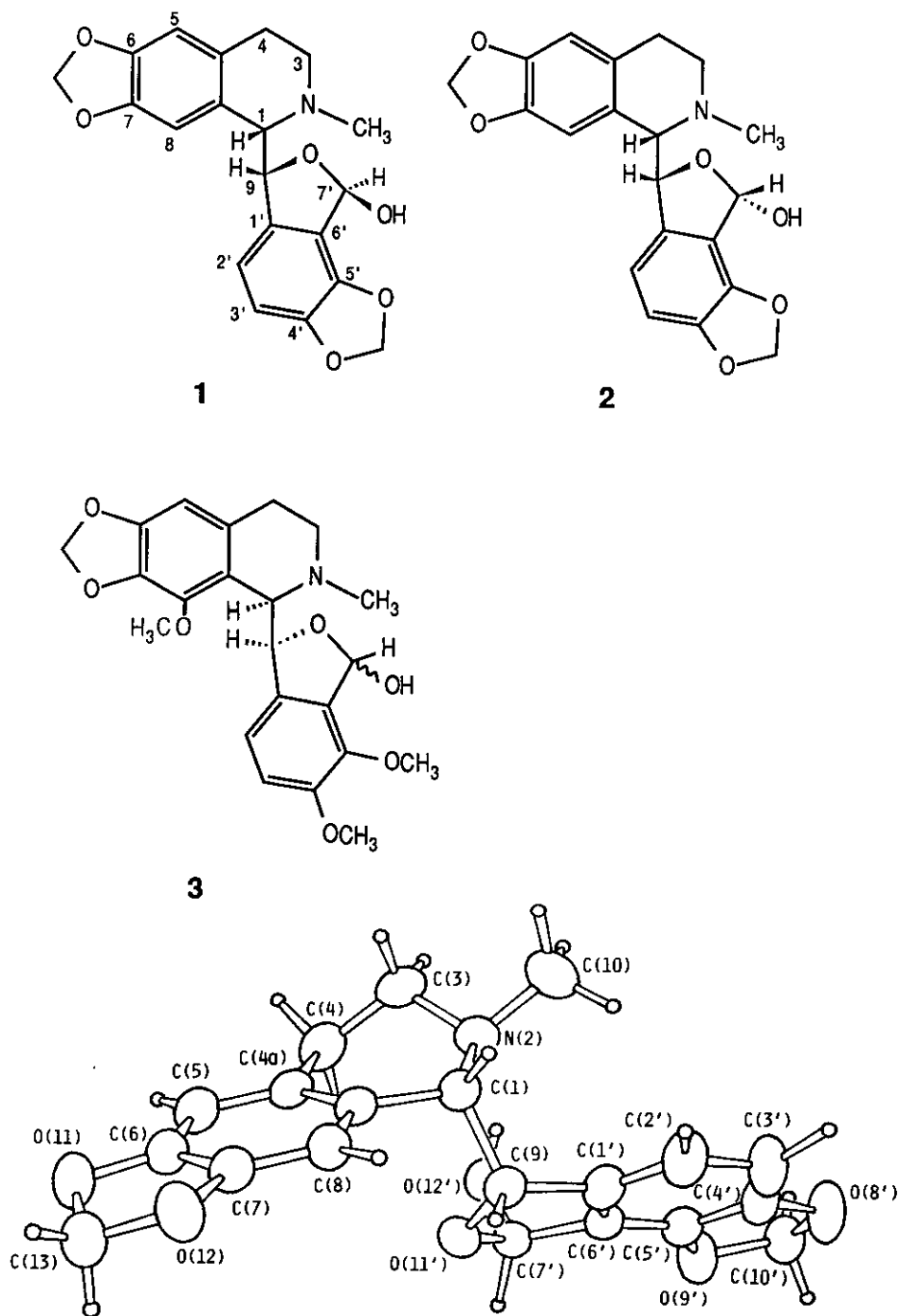


Figure 1. Structure and solid-state conformation of corytensine (1); small circles denote hydrogen atoms.

E-map. Hydrogen atoms were all located in a difference Fourier synthesis evaluated following several rounds of full-matrix least-squares adjustment of nonhydrogen atom positional and anisotropic temperature factor parameters. With the inclusion of hydrogen atom positional and isotropic thermal parameters as variables in the final least-squares iterations, the refinement converged at $R = 0.035$, $R_w = 0.053$ ⁶ over 1684 reflections with $I > 3.0 \sigma(I)$. A view of the solid-state conformation is presented in Figure 1.

Corytensine (1) is the C-7' epimer of (+)-egenine (2). Accordingly, the results of the present study now allow the complete relative stereochemistry of the latter to be defined as shown. Moreover, assuming that their absolute configuration at C-1 is S as in all classical phthalideisoquinolines exhibiting a positive specific rotation,⁷ the absolute stereochemistries of both 1 and 2 must also be as represented.

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REFERENCES AND NOTES

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- The yield of 1 is 0.001% from the whole herb. $\nu_{\text{max}}^{\text{KBr}}$ cm^{-1} : 3400, 1505, 1490, 1475; HRMS: Calcd for $\text{C}_{20}\text{H}_{19}\text{NO}_6$ 368.1132 (M-1)⁺, Found, 368.1124; Calcd for $\text{C}_{20}\text{H}_{20}\text{NO}_6$ 370.1290 (M+1)⁺, Found, 370.1308; Anal. Calcd for $\text{C}_{20}\text{H}_{19}\text{NO}_6$: C, 65.03; H, 5.19; N, 3.79. Found: C, 64.74; H, 5.17; N, 3.70; EIMS m/z: 370 (M+1)⁺, 368, 352, 190(100%), 188, 178, 175, 162, 149, 132; CIMS m/z: 370(M+1)⁺, 192, 190(100%), 179, 163; ¹³C-nmr (CDCl₃, 25 MHz): δ 148.2(s), 146.3(s), 146.1(s), 141.6(s), 135.2(s), 130.5(s), 128.6(s), 124.1(s), 113.8(s), 108.9(d), 108.1(d), 106.8(d), 101.8(t), 100.8(t), 97.7(d), 89.7(d), 68.5(d), 53.8(t), 46.7(q), 29.2(t); CD $\Delta\epsilon$ (nm)(MeOH): +0.254(302), +0.169(297), +0.505(283), +0.169(257), +3.56(231.5), +0(221), -5.17(213).
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- Crystallographic calculations were performed on PDP11/44 and MicroVAX II computers by use of the Enraf-Nonius Structure Determination Package incorporating the direct methods program MULTAN11/82.
- $R = \frac{\sum ||E_0| - |E_C||}{\sum |E_0|}$; $R_w = \left[\frac{\sum w(|E_C| - |E_0|)^2}{\sum w|E_0|^2} \right]^{1/2}$
- For a listing of spectral data for phthalideisoquinoline alkaloids, see G. Blaskó, D. J. Gula, and M. Shamma, *J. Nat. Prod.*, **45**, 105 (1982).

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