ALKALOID SYNTHESIS USING FUROPYRIDONE AS SYNTHON —SYNTHESIS OF KEY INTERMEDIATES FOR THE SYNTHESES OF (±)-QUININE, (±)-AJMALICINE, AND (±)-7-DEMETHYLTECOMANINE—

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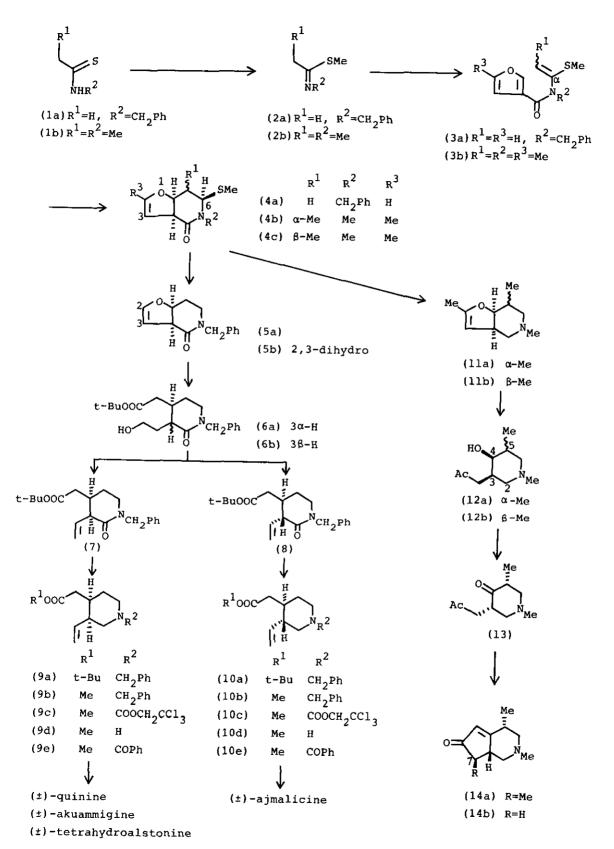
<u>Abstract</u>—Furopyridones (**4a**, **b**, and **c**) have been proved as the potential synthons for alkaloid synthesis from their facile conversion to the key intermediates (**9e**), (**10e**), and (**13**) for the synthesis of quinine, ajmalicine, and 7-demethyltecomanine.

Biogenetically loganin and secologanin stand as key intermediates in the biosynthetic pathway of monoterpenoid alkaloids.¹ Considering this, we have now explored a general and divergent synthetic route for a large number of monoterpenoid alkaloids employing unnatural heterocycle, furopyridones, as common synthons which were prepared <u>via</u> the route involving reductive photocyclization² of enamides substituted with α -alkylthic group³ which can be removed at a later step. N-Benzylthicacetamide (1a) and N-methylthicpropionamide (1b) were alkylated with dimethyl sulfate to give respective thicmidates (2a) and (2b) in quantitative yields which were then acylated with either 3-furoyl or 5-methyl-3-furoyl chloride to afford two enamides (3a)⁴ or (3b)⁴ both quantitatively. The latter enamide (3b) was found to be a 1:2 separable mixture of two geometrical isomers, though their stereochemistries remained unclarified. Reductive photocyclization² of the enamides (3a) and (3b) in the presence of sodium borohydride in acetonitrile-methanol proceeded smoothly to give the hydrogenated lactams (4a)⁵. [55% from (3a)] and (4b)⁵ and (4c)⁵[41% and 48% from (3b)].

[Synthesis of the Key Intermediates (9e) and (10e) for (\pm) -Quinine, (\pm) -Akuammigine, and (\pm) -Ajmalicine] Reduction of 6-methylthiofuropyridone (4a) with tributyltinhydride and 2,2'-azo-bisisobutyronitrile⁷ followed by catalytic hydrogenation of the resulting 6unsubstituted furopyridone (5a) over platinum dioxide under hydrogen atmosphere afforded the tetrahydrofuran (5b)⁵ in 71% yield which was also prepared by catalytic reduction of the 6-methylthiofuropyridone (4a) in the presence of Raney-Ni as catalyst in 51% yield. In order to introduce two carbon unit at the 4-position of the piperidone ring, the furopyridone (5b) was subjected to the elimination-addition reaction 8 which consists of opening reaction of the furan ring in (5b) by lithiation with lithium diisopropylamide followed by addition of the 2lithioacetate to give the desired adduct $(6a)^5$ and $(6b)^5$ as a 1:1 diastereomeric mixture at the 3-position in 69% yield. Phenylselenylation of a mixture of the ethylols (6a) and (6b) with o-nitrophenylselenocyanate-tributylphosphine followed by oxidation with hydrogen peroxide gave a 1:1 mixture of the vinyl esters $(7)^5$ and $(8)^5$ in 64% yield which was separated by column chromatography. The cis-lactam (7) was converted into the known key intermediate (9e) for (\pm) guinine, $9(\pm)$ -akuammigine, 10 and (\pm) -tetrahydroalstonine 10 by the following reaction sequence. Chemoselective reduction of the lactam carbonyl group (AlHa at -50°C), transesterification (MeOH-H₂SO₄), carbamoylation (ClCOOCH₂CCl₃-NaHCO3), reductive decarbamoylation (Zn-AcOH), and finally benzoylation (PhCOCL- Et_3N) afforded the desired <u>cis</u>-N-benzoate (9e)⁶ in almost quantitative yield. Similarly, the trans-lactam (8) was also converted into the known synthetic intermediate $(10e)^6$ of (\pm) -ajmalicine¹¹ quantitatively.

[Synthesis of the Key Intermediate (13) for (\pm) -7-Demethyltecomanine] Tecomanine (14a) is a monoterpenoid alkaloid having hypoglycemic activity and has been recently synthesized by two groups.^{12,13} The furopyridones (4b) and (4c) were converted into the known synthetic intermediate (13)¹⁴ of 7-demethyltecomanine (14b). Reduction of the 6-methylthiolactams (4b) and (4c) with lithium aluminum hydride by refluxing in tetrahydrofuran gave the desulfurizated amines (11a)⁵ and (11b)⁵ in 91 and 66% yields which were hydrolyzed with 10% hydrochloric acid to give the hydroxyketones (12a)⁵ and (12b)⁵ in 82 and 94% yields, respectively. Jones oxidation of the hydroxyketone (12b) gave the diketone (13)⁶ in quantitative yield which was also prepared by the same reaction of the another hydroxyketone (12a) in 55% yield <u>via</u> presumable isomerization at the 3- or 5-position of the resulting thermodynamically unstable 3,5-<u>trans</u>-diketone. The <u>cis</u>-diketone (13) had been utilized as a key intermediate for the synthesis of (\pm)-7-demethyltecomanine.¹⁴

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