## SYNTHESIS OF REGIOSPECIFICALLY SUBSTITUTED PYRIMIDYL DERIVATIVES AND THEIR INCORPORATION INTO PENEMS

Hans-Georg Capraro<sup>\*</sup>, Marc Lang, and Peter Schneider

Research Department, Pharmaceuticals Division, CIBA-GEIGY Limited, CH-4002 Basel, Switzerland

<u>Abstract</u> - Syntheses of regiospecifically substituted pyrimidines are described. Depending on the reaction conditions,  $N_1$ - or  $N_3$ -substituted pyrimidines are obtained. It has been shown that substitution on uracil under Mitsunobu conditions yields  $N_1$ -substituted products. Incorporation of these derivatives into the penem nucleus gives penem antibiotics with extremely long half-lives.

Penems <u>1</u> are highly potent, broad-spectrum  $\beta$ -lactam antibiotics, closely related to the penicillins, the cephalosporins and the carbapenems. Since the publication of the first penem synthesis by the Woodward-CIBA group in the 1970's<sup>1</sup>, chemists have been endeavouring to discover an economical route for the total synthesis of penems possessing a high degree of antibiotic activity<sup>2</sup>. These efforts have led to a series of development compounds, which are or have been in the past under investigation by various companies (Table 1).



The outstanding compound in this list is CGP 31 608; of all the penems prepared and known from published data, CGP 31 608 is the only one displaying not only excellent activity against anaerobes, Gram-positive and the "normal" Gram-negative bacteria, but also against <u>Pseudomonas aeruginosa</u>. For this reason, CGP 31 608 can justly be described as a genuine broad-spectrum penem antibiotic, only comparable to the carbapenems<sup>3</sup>.

Besides good antibacterial activity, an important criterion of the clinical efficacy of  $\beta$ -lactam antibiotics is the duration of their sojourn in the blood plasma. The half-lives of these substances in human plasma is usually limited, ranging from 1 to 2 hours. One of our principal objectives has consequently been to synthesize a new

penem with a longer half-life than the existing representatives of this class, the other semisynthetic penicillins and the cephalosporins.



As rational thinking does not hold the key to success in the design of penem antibiotics with long plasma half-lives, we were pleased, during our systematic derivatization work on CGP 31 608, to discover the pyrimidyl penems  $\underline{3}$  and  $\underline{5}$ , (from  $\underline{2}$  and  $\underline{4}$  respectively) both possessing the desired properties<sup>4,5</sup>.

CGP 31 608, bearing an alkylamino group, proved to be an interesting lead compound that reacts easily with a wide variety of amino reagents, giving acyl derivatives, carbamates, urethanes, etc. The maleic acid anhydride derivative  $\underline{6}^{6}$  reacts, in a rather peculiar fashion, with primary amino groups to yield  $\underline{7}$  (Scheme 2). The reaction (Scheme 2) is thought to proceed by way of the Lossen rearrangement<sup>7</sup> of the cis-oriented hydroxamic acid intermediate  $\underline{a}$  to the isocyanate  $\underline{c}$ . The subsequent cyclization affords an elegant synthesis of regiospecifically substituted pyrimidyl derivatives. By analogy with the reaction  $\underline{6} \rightarrow \underline{7}$ , CGP 31 608 and its homologue  $\underline{8}$  reacted with  $\underline{6}$  to give the penem derivatives  $\underline{9}$  and  $\underline{10}$  (Scheme 3). For n = 1,  $\underline{11}$  was a by-product. Albeit isolated as an isomeric mixture only in a modest yield, its existence supports the proposed reaction mechanism of Scheme 2.

The pyrimidyl penem 9, the structure of which was further confirmed by an independent synthesis (see below), could be prepared in gram quantities in an average, non-optimalized yield of 50-60 %, starting from CGP 31 608. The nmr spectrum of 9 in D<sub>2</sub>O (Table 3), FAB/MS (M-H=366) and <sup>13</sup>C-nmr were in agreement with the proposed structure. Biological data on 9, e.g. elimination half-life in mouse plasma (41 min), protein binding (90 % in human serum) and plasma concentration (19.8  $\mu$ g/ml 50 min after administration) justified the further exploration of the regiospecifically substituted pyrimidyl penems in order to elucidate the relations between the biological properties and the substitution pattern of the pyrimidyl ring.

Scheme 2



a) H<sub>2</sub>O/pH=8-8 5/4-5 h (tic control)

b) chromatography on reversed- phase silica gel in  $\mathrm{H_2O}$ , ca. 50%

Scheme 4 summarizes the reaction sequence carried out with uracil <u>12</u>. Our primary target compounds were the "amide"-acid <u>15</u> and the "imide"-acid <u>24</u>. Alkylations of uracil <u>12</u> are known to yield predominantly  $N_1$ -substituted products. Indeed, the synthesis of <u>15</u> was accomplished by the published procedure<sup>8</sup>.

Alternatively, alkylation of <u>12</u> with ethyl iodoacetate gave the ester <u>13</u>, which was converted to <u>15</u> by hydrolysis with conc. HCl.



The synthesis of penem 5 (n=1; R<sup>1</sup>=R<sup>2</sup>=H), starting from silver salt  $\underline{26}^9$  and acid chloride  $\underline{27}$  via phosphorane  $\underline{28}$  and penem ester  $\underline{29}$ , is described in Scheme 5<sup>10</sup>. To our surprise, the nmr spectroscopic data of 5 were fully identical with the published data of the penem resulting from deprotection of the ester  $\underline{31}$ , postulated (Scheme 6) as the product of a Mitsunobu reaction between uracil  $\underline{12}$  and  $\underline{30}^{11,12}$ .

We therefore began to investigate the reactivity of uracil <u>12</u>, a multifunctional substrate, under Mitsunobu conditions (glycolic acid n-butyl ester, triphenylphosphine, DEAD). Careful chromatography of the resultant complex reaction mixture led to the isolation of only one monosubstituted uracil ester <u>14</u> (26 %; mp: 111-112°C), which after hydrolysis gave the acid <u>15</u>, identical in every respect with the product of the reaction

sequence  $\underline{12} \rightarrow \underline{15}$ , or  $\underline{12} \rightarrow \underline{13} \rightarrow \underline{15}$ . The N<sub>3</sub>-substituted product  $\underline{23}$  (R<sub>2</sub>=n-C<sub>4</sub>H<sub>9</sub>) was not detectable.

O,O-Bisalkylated by-products resulting from the reaction with the respective enolates and N,N-bisalkylated products were identified by nmr spectroscopy, but could not be separated<sup>14</sup>.

In the literature, the synthesis of the isomeric uracil acid <u>24</u> starting from cytosine is reported<sup>15</sup>, but we were not able to reproduce the described procedure. We therefore prepared the N<sub>1</sub>-acetyluracil <u>17</u><sup>16</sup>, carrying an easily removable blocking group. Mitsunobu alkylation of <u>17</u> by analogy with the preparation of <u>14</u> gave an inseparable mixture of deacylated esters <u>23</u> (R<sup>2</sup>=n-C<sub>4</sub>H<sub>9</sub>; major) and the isomeric compound <u>14</u> (minor, ratio by nmr approx. 4:1). However, <u>17</u> on reaction with ethyl iodoacetate (K<sub>2</sub>CO<sub>3</sub>/DMSO/RT/6h) gave a 32% yield of pure <u>23</u> (R<sup>2</sup>=C<sub>2</sub>H<sub>5</sub>; mp: 141°C), easily separated from the N,N-bisalkylated derivative (16%; not shown in the



scheme). The acid <u>24</u> was obtained as usual by hydrolysis from <u>23</u> ( $R^2=C_2H_5$ ). The nmr spectroscopic data of the esters <u>14</u> and <u>23</u> ( $R^2=C_2H_5$ ) and of the corresponding acids <u>15</u> and <u>24</u> are presented in Table 2. By analogy with the synthesis of <u>5</u> (Scheme 5), <u>24</u> served as starting material for the synthesis of the





regioisomeric penem, which results from the deprotection of <u>31</u> (Scheme 6; nmr data see table 3). Thus it was demonstrated that <u>31</u> is clearly not the main product of a Mitsunobu reaction of <u>30</u> and <u>12</u> <sup>17,18</sup>.

Further transformations of uracil <u>12</u> are shown in Scheme 4. Methylation with  $CH_3I/K_2CO_3$  in DMSO led to a mixture of <u>18</u> (major; mp 231-234°C) and <u>19</u> (minor; mp 118-121°C)<sup>19,20</sup>, separated by chromatography. Preparation of <u>20-22</u>, <u>16</u><sup>21</sup> and <u>25</u><sup>8</sup> followed standard procedure.

Scheme 7



ĊOO"Na+

In Scheme 7, additional derivatives are given. Starting from commercially available 5,6-dimethyluracil <u>38</u>, the esters <u>42</u> and <u>44</u> were prepared (ICH<sub>2</sub>COOC<sub>2</sub>H<sub>5</sub>/K<sub>2</sub>CO<sub>3</sub>/DMSO/50°C/15 h) and separated by chromatography. In this reaction, N,N-dialkylated uracil <u>46</u> (ir: 1749, 1702, 1655 cm<sup>-1</sup> in CH<sub>2</sub>Cl<sub>2</sub>; mp 92 - 93°C) was the main product. Acids <u>43</u> and <u>45</u>, obtained from <u>42</u> and <u>44</u> by acidic hydrolysis, gave the penetros <u>54</u> and again <u>9</u>.

Thus, the synthesis of <u>9</u>, accomplished in an independent way, enabled us to assign the correct regiochemistry of the substitution in <u>42</u> and <u>44</u> and the corresponding acids <u>43</u> and <u>45</u>, whose spectroscopic data alone do not permit unambiguous structure assignment (see Tables 2 and 3)<sup>22</sup>.

Compound <u>49</u> was prepared according to the published procedure<sup>8</sup> starting from thymine <u>39</u>; interestingly enough, we were not able to prepare the isomer <u>51</u> (from <u>41</u>) by the same method. Again, Mitsunobu's procedure yielded <u>50</u> and thereafter the desired acid <u>51</u>. Alkylation on 5-fluorouracil (<u>40</u>) is reported to yield mixtures of monosubstituted and disubstituted products on nitrogen, depending on the reaction conditions<sup>24</sup>. By our procedure (ICH<sub>2</sub>COOC<sub>2</sub>H<sub>5</sub>/K<sub>2</sub>CO<sub>3</sub>/DMSO/50<sup>o</sup>), an easily separable, approx. 2:1 mixture of <u>47</u> and N,N-dialkylated fluorouracil (not shown in the scheme) was obtained. <u>48</u> was prepared from <u>47</u> by hydrolysis. Finally, following the reaction sequence depicted in Scheme 2, the acids <u>52</u> (44 %; mp 182 - 184<sup>o</sup>C) and <u>53</u> (57 %; mp 185 - 187<sup>o</sup>C) were prepared.

Table 4 summarizes some of the kinetic parameters of the uracil penems 3 and 5. The synthesis proceeds in the way outlined in Schemes 3 and 5. The results indicate that the long half-life is not primarily dependent on the  $N_1$ - or  $N_3$ -substitution pattern of the pyrimidine ring, but on the spacer -(CH<sub>2</sub>)<sub>n</sub>- and on the substituents on the ring itself.

_	соон	NH	H <sub>3</sub> C-C(6) H-C(6)	H <sub>3</sub> C-C(5) H-C(5)	2H-C(1')
14	-	8.79	7 12	5.77	4.47
<u>23</u>	-	9.05	7 09	5.73	4.43
<u>42</u>	-	8 41	2 18	1 98	4.65
44	-	9 86	2.16	1 93	4.68
<u>15</u>	13.15	11.35	7 61	5.60	4.41
24	12.95	11 28	7.51	5.67	4.41
<u>43</u>	13.18	11 35	2.13	1 83	4 55
<u>45</u>	12 90	11.06	2 08	1.78	4.41

Table 2 <sup>1</sup>H-Nmr spectroscopic data of <u>14,23,42,44</u> (CDCl<sub>3</sub>) and <u>15, 24, 43, 45</u> (DMSO-d<sub>6</sub>; 360 MHz)

H3C-C(1)	1.30	1.29	1.28	1.28
н-С(6)	3.92	3.91	3.90	3.88
H-C(1')	4.24	4.24	4.23	4 23
2H-C(1")	5.30 / 4.98	5 48 / 5.22	5.43 / 5 02	5 43 / 5.03
H-C(5)	5.65	5.63	5.61	5 59
H <sub>3</sub> C-C(5 <sup></sup> ) H-C(5 <sup></sup> )	5.86	1 93	5.89	1.89
H <sub>3</sub> C-C(6''') H-C(6''')	7.70	2.33	754	2.21
'n	н (5)	CH <sub>3</sub> (54)	Н (31)	CH3 (9)
æ	R' 5-14 ML	н Ч Ч Ч Ч Ч Ч Ч Ч Ч Ч Ч Ч Ч Ч Ч Ч	EO a D	× × × × ×
	ß	Hess N N N	COO'Na+	

(D <sub>2</sub> O; 360 MHz)
<u>9</u> and <u>54</u>
(deprotected),
ы
of penems <u>5</u> ,
c data (
<sup>1</sup> H-Nmr spectroscopic
<u>Table 3</u>



5
σ
5
ကျ
2
F
e
ö
ō.
~
б
Ê
È
⋝
Ω.
õ
ŧa
õ
÷.
ß
÷.
о
Ś
Ð
Ħ
Ē.
g.
۳.
ä
$\overline{\alpha}$
¥
ē
5
Ť.
ŏ
ğ
F.
đ
Ē
а_
-
4
e l
٩
CG

Pyrimidyl $H_3^{C} - H_{MH}$ $H_3^{C} - H_3^{C} - H_3^{C}$ $H_3^{C} - H_3^{C} -$						
Pyrimidyl $H_{3C} - H_{NH}$ $H_{3C} - H_{3C}$	- N N N N N N N N N N N N N N N N N N N	-	9	29	0.6	
Pyrimidyl $H_3^{C} - H_3^{H}$ $H_3^{G} - H_3^{HH}$ $H_3^{H} - H_3^{H}$ $H_3^{H} - H_3^{H} - H_3^{H}$ $H_3^{H} - H_3^{H} - H_3^{H}$ $H_3^{H} - H_3^{H} - H_3^{$	H <sup>2</sup> C H <sup>2</sup> C	F	თ	33	3.2	
Pyrimidyl $H_3^{C} - H_3^{MH}$ $H_3^{MH} - H_3^{MH} - H_3^{MH}$ </td <td>°≠_₹_~ ~</td> <td>-</td> <td>13</td> <td>37</td> <td>&lt;0.8</td> <td></td>	°≠_₹_~ ~	-	13	37	<0.8	
Pyrimidyl $H_3^{C} - H_3^{HH}$ $H_3^{G} - H_3^{HH}$ $H_3^{H} - H_3^{H}$ $H_3^{H} - H_3^{H} - H_3^{H}$ $H_3^{H} - H$		-	20	84	21.8	
Pyrimidyl $H_3^{C} \leftarrow H_3^{H}$ $H_3^{L} \leftarrow H_3^{H}$ $H_3^{L} \leftarrow H_3^{H}$ $H_3^{L} \leftarrow H_3^{H}$ $H_3^{H} \leftarrow H_3^{H} \leftarrow H_3$	o=	-	2	44	<0.4	
Pyrimidyl $H_3^{C} - H_3^{HH}$ $H_3^{L} - H_3^{HH}$ $H_3^{L} - H_3^{HH}$ $H_3^{HH}$ $H_3^$	CH <sup>N</sup> C	-	16	25	13.1	
Pyrimidyl $H_{3C} \sim H_{NH}$ $H_{3C} \sim H_{3C}$	HN N -	-	7	27	1.0	
Pyrimidyl $H_3^{G}C_{NH}^{H_3}$ $H_3^{G}C_{N}^{H_3}$ n     1     1       n     1     2       n     1     2 $t_{1/2}$ 41     5     5       PB     90     24     53     90       C(50)     19:8     -     -     29	₹_%_	~	4	37	<0.8	
Pyrimidyl $H_{3}C$		-	7	4		
Pyrimidyl $H_3^{-1}C_{NH}^{-1}$ n         1         3         5           n         1         2         3         5 $t_{1/2}$ 41         5         5         5           PB         90         24         53         90           C(50)         19.8         -         <0.2<	H <sup>3</sup> C	-	25	50	59	
Pyrimidyl         H <sub>3</sub> C         CH3 NH           side-chain         0         NH           n         1         1           n         1         2           t <sub>1</sub> /2         41         5         5           PB         90         24         53           C(50)         19.8         -<<0.2		ъ С	S	06	<0.2	
Pyrimidyl         H <sub>3</sub> C         C           side-chain         n         1         2           n         1         2         1           n         1         2         1           r <sub>1/2</sub> 41         5         2           PB         90         24           C(50)         19.8         -		e	S	ន	<0.2	
Pyrimidyl <sub>Ha</sub> c side- chain o n 1 t <sub>1/2</sub> 41 PB 90 C(50) 19.8	5-(x -	3	5	24		
Pyrimidyl side- chain n t <sub>1/2</sub> PB C(50)	ວິ H	-	41	06	19.8	
	Pyrimidyl side- chain	C	t1/2	84	C(50)	

protein binding in human serum in %, C(SO) concentration of antibiotic in . µg/ mi in plasma of mice after 50 min. 22.22 2 u<sub>rz</sub> : nair-ire in pias

## ACKNOWLEDGEMENTS

The authors thank Mrs. I. Manso and Mrs. G. Winteler for their skilful experimental work, Mr. E. Batt for the kinetic data and Mr. A.H. Kirkwood for checking the manuscript.

## **REFERENCES AND NOTES**

- R.B. Woodward, "Recent Advances in the Chemistry of β-Lactam Antibiotics". Spec.Publ.No.28, ed. by J. Elks, The Chemical Society, Burlington House, London, 1977, pp. 167-180.
- a) M. Alpegiani, C. Battıstini, A. Bedeschi, G. Franceschi, F. Giudici, E. Perrone, C. Scarafile, and F. Zarini, <u>Chim. Ind. (Milan)</u>, 1986, <u>68</u>, 70.
   b) G. Franceschi, M. Alpegiani, C. Battistini, A. Bedeschi, E. Perrone, and F. Zarini, <u>Pure and Appl. Chem.</u>, 1987, <u>59</u>, 467.
- a) O. Zak, M. Lang, R.M. Cozens, E.A. Konopka, H. Mett, P. Schneider, W. Tosch, and R. Scartazzini, J. <u>Clin. Pharmacol.</u>, 1988, <u>28</u>, 128.
   b) R.M. Cozens, and M. Lang, <u>Drugs of the Future</u>, 1988, <u>13</u>, 19.
- 4. Full biological data on all penems mentioned in this paper have been presented at the 28<sup>th</sup> ICAAC<sup>5</sup>.
- H.-G. Capraro, M. Lang, E. Hungerbühler, P. Schneider, W. Tosch, and O. Zak, Program and Abstract of the 28<sup>th</sup> Intersc. Conf. on Antimicrob. Agents Chemother., No. 224, Los Angeles, Oct. 1988.
- Compound <u>6</u> was first prepared in our company by Dr. B. Muller of the Additives Division, who, together with Dr. M. Baumann of our Central Research Laboratories, explored the synthetic usefulness of this reagent (to be published). The authors thank Dr. M. Baumann for supplying our laboratory with <u>6</u>.
- 7. L. Bauer and O. Exner, Angew. Chem., 1974, 86, 419.
- a) H.L. Wheeler and L.M. Liddle, J. Am. Chem. Soc., 1908, <u>30</u>, 1152.
   b) A.S. Jones, P. Lewis, and S.F. Withers, <u>Tetrahedron</u>, 1973, <u>29</u>, 2293.
- 9. M. Lang, Eur. Pat. Appl. 171362, Febr. 12, 1986, to Ciba-Geigy, CA 105(25):226178 n.
- With the exception of penems 9 11, all penems mentioned in this paper were synthesized following the reaction pathway depicted in Scheme 5. This reaction sequence corresponds to the original Woodward procedure<sup>1</sup>.
- 11. Deutsche Offenlegungsschrift DE 3510272 A1, to Farmitalia Carlo-Erba S.p.A., Milano, Italy (1985). In this paper, the reaction under Mitsunobu conditions of the hydroxymethylpenem <u>30</u> with preparations such as <u>12</u> was postulated to yield compounds of type <u>31</u>, with the N<sub>3</sub>-substitution pattern. As will be shown, at least part of these claims must be reconsidered.
- a) O. Mitsunobu, <u>Synthesis</u>, 1981, 1.
  b) R.P. Volante, Tetrahedron Lett., 1981, 22, 3119.

c) M. Varasi, K.A M. Walker, and M.L. Maddox, J. Org. Chem., 1987, 52, 4235.

- H.-G. Capraro, E. Francotte, B. Kohler, G. Rihs, P. Schneider, R. Scartazzini, O. Zak, and W. Tosch, <u>J.</u> Ant<u>ibiotics</u>, 1988, 41, 759.
- 14. In all alkylation reactions of pyrimidines, following Mitsunobu's procedure or by standard methods, mixtures of products were obtained: e.g. the Mitsunobu reaction of 4-chlormethyluracil (32), a compound unsuitable for normal alkylation reactions because of its labile chlorine atom, gave the four products 33 36 depicted below. In this case, it was possible to separate and characterize the products by spectroscopic methods (<sup>1</sup>H-,<sup>13</sup>C-nmr, NOE; structure proposed for 36. 37 by hydrolysis from 33) whereas with the other pyrimidines only the N<sub>1</sub>-monoalkylated main product was isolated.



- 15. R.S. Goody and R.T. Walker, J. Org. Chem., 1971, 36, 727.
- 16. L.B. Spector and E.B. Keller, J. Biol. Chem., 1958, 232, 185.
- 17. Although our conditions in the Mitsunobu reaction with uracils were not identical with the procedure described<sup>11</sup>, the exact reproduction of the published procedure with penem ester <u>30</u> and uracil <u>12</u> likewise gave penem <u>5</u> after deprotection.
- 18. It must be noted that structure assignment of the penems 5 and 31 (deprotected) based on <sup>1</sup>H-nmr data alone is difficult owing to the small difference in the chemical shifts of the two isomers (Table 3).
- 19. K. Yamauchi and M. Kinoshita, J. Chem. Soc., Perkin Trans. I, 1973, 391.
- 20. K. Yamauchi, T. Tanabe, and M. Kinoshita, J. Org. Chem., 1976, 41, 3691.
- 21. T. Ueda and J.J. Fox, J. Org. Chem., 1964, 29, 1762.
- 22. All the pyrimidines prepared follow the bathochromic shift rule for uv spectra when measured in basic solutions<sup>23</sup>.
- 23. a) D. Shugar and J.J. Fox, Biochim. Biophys. Acta, 1952, 9, 199.
  - b) I. Wempen and J.J. Fox, J. Am. Chem. Soc., 1964, 86, 2474.
- a) H. Pischel, A. Holy, J. Vesely, G. Wagner, and D. Cech, <u>Coll. Czech. Chem. Commun.</u>, 1982, <u>47</u>, 2806.
  b) U. Sanyal and S.K. Chakraborti, <u>Synth. Commun.</u>, 1982, <u>12</u>, 1047.

Received, 5th September, 1988