CHEMICAL TRANSFORMATIONS OF QINCHAOSU, A PEROXIDIC ANTIMALARIAL, II*

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<u>Abstract</u>— Qinghaosu ($\underline{1}$) is known to give a plethora of products in alkaline media. This paper is concerned with the amended stereochemistry of products $\underline{3}$, $\underline{4}$ and $\underline{5}$, with attendant mechanistic implications. Lactonization of $\underline{5}$ is also discussed.

This paper reports some new results on the chemistry of the peroxidic antimalarial, qinghaosu (also known as arteannuin, $\underline{1}$)¹. In addition, the stereochemistry of some of the transformation products reported earlier became suspect in the course of time and had to be amended ($\underline{3}$ and $\underline{5}$ in this paper, $\underline{4}$ by other authors²). In alkaline media (K2CO3 or NaOH) qinghaosu (1) suffers lactonic ring opening with simultaneous unlocking of all the latent functionalities to give \underline{A} , as shown in the Scheme. Subsequent condensation of the aldehydic group onto the active methylene gives B which in turn by internal Michael-type reaction gives C, a key intermediate with a dioxetane ring. Rupture of the dioxetane ring (path b) is the major route in aqueous solution (containing 20% methanol for solubility reasons) with an optimized conversion of better than 88%, a fact that has been exploited for the quantitative analysis of qinghaosul. Here the stereochemistry of the product (5) at C-6 has to be amended as shown. The proposed inversion of this center occurs only under slightly more drastic conditions (5% NaOH under reflux), furnishing $\underline{6}$. A synthetic specimen $(\underline{7})^3$ with retention at C-6 was found to be distinctly different from $\underline{6}$ by direct comparison (e.g., Rf values). When the mixed solvent becomes much richer in alcohol (MeOH or EtOH), the yield of 5 drops sharply, giving way to the formation of $\frac{4}{2}$, among others. Evidently $\underline{\underline{\psi}}$ has its genesis from $\underline{\underline{C}}$ via an internal attack of the peroxidic linkage by the enclate anion (path a). This anion is probably better solvated in aqueous media and hence less aggressive. Note the non-inversion at C-5 of compound 4, which was previously wrongly assigned as $\underline{3}$ (with inversion at C-5)¹. As previously proposed 1, the formation of 2 requires an intermolecular reaction between 2 and hydrogen peroxide. Now we did detect a very small amount of $\underline{2}$ in the reaction. mixture. (The configuration at C-5 follows from that of 3). Free hydrogen peroxide was also detected by several independent experiments, including polarography. However, compound 3 was not present in detectable amounts and was to be obtained from an unexpected source (vide infra).

^{*}Dedicated to Professor D.H.R. Barton on the occasion of his 70th birthday.

Scheme 1. Transformations of qinghaosu $(\underline{\underline{1}})$. Structures labeled with capital letters are proposed intermediates. Compounds labeled with asterisks have been analyzed by X-ray diffraction (* previously; ** this work).

Catalytic hydrogenation of qinghaosu (1) gives deoxyqinghaosu (8)4. Quite unexpectedly, the product (2) has suffered an inversion at C-5 (Reaction II). This is attributable to the intervention of $\underline{\mathbf{E}}$, an internal Michael-type adduct, the epoxide of which is vulnerable to a back-side attack by the carboxylate anion (incidently, further treatment of $\underline{2}$ with H_2O_2 gives $\underline{3}^4$, which was fully characterized by X-ray diffraction). We did also isolate another minor isomer 9 with intact C-5 stereochemistry. NOE evidence is only compatible with C-54 and C-134 configurations, confirmed by X-ray diffraction. The inversion at C-13 can be rationalized by the need of releasing steric congestion. The C-13 methyl group of $\underline{4}$ is also subject to severe steric congestion, yet somehow it managed to preserve stereochemical integrity under apparently similar conditions. The hydroxy-acid (5) can be lactonized by storage in acidic media. Since the hydroxy group of $\underline{5}$ is part of a hemi-ketal, reversion to the keto-form sets free the bidentate β -dicarbonyl system, thus offering two choices of re-closure in addition to the two epimeric alternatives at C-5. Altogether there are four isomeric products upon lactonization with the $\underline{10}$ a and $\underline{10}$ b pair predominating. The C-5 epimers can be easily separated, but each epimer still contains a regioisomeric pair (10a and 10b; 11a and 11b; $a:b_{7}$ 4:1), as evidenced by ^{1}H and ^{13}C nmr. C-5 configurational homogeneity in the two isolates, aside from the coincident Rf values within each regio-isomeric pair, is strongly supported by smooth pseudo first-order kinetics. The $\underline{10}$ a and $\underline{10}$ b pair is faster than the $\underline{11}$ pair by a factor of 2 (monitored by the development of the enolate absorption at 292 nm with dil. NaOH in large excess).

The aldehydic form (ca. 20%) of 5 is also detected in solution.

EXPERIMENTAL

M.p.s are uncorrected. Optical rotations were measured with Perkin-Elmer 241 Polarimeter. Uv spectra were recorded with a Shimadzu UV-240 spectrophotometer. Nmr spectra were recorded in CDCl₃ with a JEOL FX-90Q spectrometer. Ms were recorded with a VG ZAB-2F spectrometer in the EI mode. X-Ray single crystal analyses were performed on a Nicolet R3M/E diffractometer, using SHELXTL program for the direct method. The absolute configuration of $\underline{\underline{\mathbf{1}}}$ has been established⁴, from which follow all the other configurational representations. The isolation and spectral data of $\underline{\underline{\mathbf{3}}}$ (from Reaction II), $\underline{\underline{\mathbf{4}}}$, $\underline{\underline{\mathbf{5}}}$, $\underline{\underline{\mathbf{6}}}$ and $\underline{\underline{\mathbf{8}}}$ have been fully described previously^{1,4}.

Identification of 2 in the reaction mixture (Reaction I). With an authentic sample of 2 (from Reaction II) at hand, it became convenient to detect 2 from the K2CO3 treatment of qinghaosu (1) in aqueous methanol. It was found to occur as a "contaminant" of 4. Epoxidation gave 3 which can be easily separated (Estimated yield, under 5%).

Isolation and characterization of $\underline{9}$ The mother liquor of $\underline{2}$ (Reaction II, 28% yield), by repeated preparative tlc (silica, 1:1 BuOAc- pet. ether) afforded $\underline{9}$ in 0.8% yield. Mp. 151.5-152.5°C, ($\underline{4}$) $\underline{1}^{5}$ -179° (c, 0.1, CHCl $_{3}$). Ms m/z 248 (M⁺). Uv (95% EtOH) 229nm (13000). Ir (KBr), 3070, 1767, 1667 cm⁻¹. $\underline{1}_{H}$ Nmr, $\underline{4}_{1}$.05 (3H, \underline{d}_{2} , J=7.2), 1.27 (3H, \underline{d}_{3} ,

J=7.2), 2.25 (3H, \underline{s}), 2.40 (1H, \underline{d} , J=16.8 Hz, 1d-H), 2.54 (1H, \underline{quint} , J=7.2), 2.84 (1H, $\underline{\text{ddd}}$, J=16.8, 5.13, 2.92, 1 β -H), 6.71 (1H, $\underline{\text{s}}$, 4-H). NOE (4-H): 2.25 (2.4%), 2.54 (8%).

Isolation of 10 and 11. Qinghaosu (2.4 g) in 95% EtOH (100 ml) was mixed with 0.2% aqueous NaOH (400 ml) and allowed to react at 50°C for 2.5 h. The usual work-upl gave 5. Silica gel chromatography of the mother liquor gave $\underline{10}$ (290 mg) and $\underline{11}$ (92 mg). $\underline{\text{Compound}}$ $\underline{10}$: Mp. 100.5-102.5°C, $\underline{\text{C}}$ $\underline{\text{C}}$ $\underline{\text{C}}$ +13.5° (c, 0.5, CHCl₃). Ms m/z 264 (M⁺). $\underline{\text{UV}}$ (95% EtOH), 244 nm (10,000). Ir (film), 3090, 1790, 1060, 1627 cm⁻¹. ¹H Nmr, δ 1.03 (3H, \underline{d} , J=7.2), 1.20 (3H, \underline{d} , J=7.2), 2.26 (3H, \underline{s}), 3.38 (1H, quint, J=7.2), 7.44 (0.8H, d, J=0.5), 10.00 (0.2H, s). 13c Nmr, 88.7 (q), 19.8 (\underline{t}) , 24.8 (\underline{q}) , 25.0 (\underline{t}) , 32.2 (\underline{t}) , 32.7 (\underline{d}) , 38.7 (\underline{d}) , 42.5 (\underline{d}) , 43.4 (\underline{d}) , 105.3 (\underline{s}), 152.3 (\underline{d}), 178.0 (\underline{s}), 195.9 (\underline{s}), 119.5 (\underline{s}). Compound 11. Mp. $114-117^{\circ}$ C, Ms, m/z 264 (M⁺). Uv (95% EtoH), 250 nm (13200). Ir (KBr), 3090, 1795, 1658, 1635 cm⁻¹. ¹H Nmr, 61.03 (3H, d, J=7.2), 1.18 $(3H, \underline{d}, J=7.2)$, 2.28 $(3H, \underline{s})$, 3.32 $(1H, \underline{quint}, J=7.2)$, 7.54 $(0.8H, \underline{br} \underline{s})$, 10.00 (0.2H, <u>s</u>). ¹³c Nmr, 58.8 (<u>q</u>), 15.9 (<u>t</u>), 18.1 (<u>q</u>), 24.8 (<u>q</u>), 25.9 (<u>t</u>), 30.2 (\underline{d}), 36.0 (\underline{d}), 38.6 (\underline{d}), 40.4 (\underline{d}), 108.3 (\underline{s}), 152.8 (\underline{d}), 177.8 (\underline{s}), 195.7 (\underline{s}) , 40.6 (d), 116.2 (\underline{s}), along with a small peak at 188.4 (\underline{d}). Compound 10 and 11 were interconvertible upon storage in chloroform solution. X-Ray diffraction analysis showed $\underline{2}$, $\underline{5}$ and $\underline{9}$ to belong to the space group P2 $_1^2$ 1 2_1 , with 4 molecules per unit cell. MoKa $(9 \le 23^\circ)$ was used for 3, cell dimensions 5.496 (4), 12.633 (3) and 20.768 (7) Å, 997 observable reflections out of 1163, R= 0.0585. CuK, $(9 \le 57^{\circ})$ was used for both 5 and 2. For 5, cell dimensions 6.922 (1), 12.122 (1) and 17.543 (3) Å, 1097 observable reflections out of 1174, R= 0.0391. For 9, cell dimensions 8.575 (5), 10.398 (5) and 15. 404 (7) Å, 1003 observable reflections out of 1098, R= 0.0490. Bond lengths and bond angles are all within normal ranges.

REFERENCES AND NOTES

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