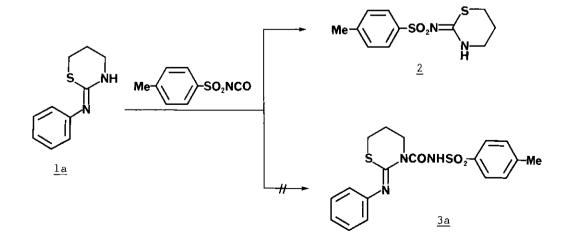
NOVEL EXCHANGE REACTION OF 2-IMINO OR 2-THIONO DERIVATIVES OF 1,3-THIAZINES AND 1,3-OXAZINES WITH p-TOLUENESULFONYL ISOCYANATE

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<u>Abstract</u> — The reaction of 2-arylimino-1,3-thiazine with p-toluenesulfonyl isocyanate led to the formation of 2-(p-toluenesulfonyl)imino-1,3-thiazine. A similar reaction carried out with 1,3-thiazine-2-thione and 1,3-oxazine-2-thione gave 2-(p-toluenesulfonyl)imino-1,3-thiazine or 1,3-oxazine, respectively.

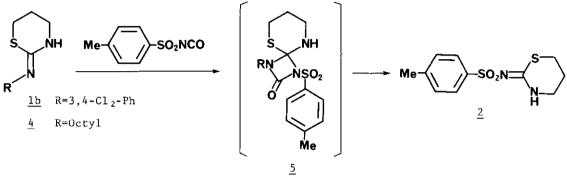
In the course of our study on the synthesis of 1,3-thiazine derivatives $\underline{1}$,^{1,2} Nsulfonyl carbamoyl derivatives $\underline{3}$ had to be prepared. We found that 2-phenylimino-1,3thiazine <u>la</u> readily underwent cycloaddition with <u>p</u>-toluenesulfonyl isocyanate to give <u>p</u>-toluenesulfonylimino-1,3-thiazine $\underline{2}$. We report here on the cycloaddition reaction of 1,3-thiazine and 1,3-oxazine derivatives with <u>p</u>-toluenesulfonyl isocyanate. Reaction of 2-phenylimino-1,3-thiazine <u>la</u>, which was prepared from phenyl isothiocyanate according to the reported procedure,³ with <u>p</u>-toluenesulfonyl isocyanate in benzene or toluene did not give the desired product <u>3a</u>, but the product <u>2</u> (Scheme 1).



Scheme 1

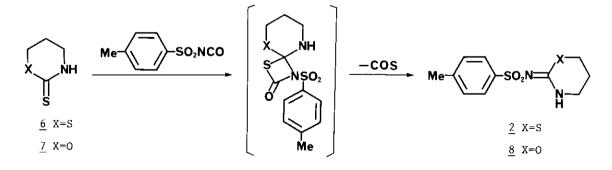
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The structure of $\underline{2}$ was confirmed by spectral data and elemental analysis. In order to examine the generality of our observation, reactions of 1,3-thiazines such as 2-(3,4-dichlorophenyl)imino- $\underline{1b}$ and octylimino-1,3-thiazine $\underline{4}$ were studied under similar conditions. In all the cases, p-toluenesulfonylimino-1,3-thiazine was obtained as the sole product (Scheme 2). This reaction presumably proceeds via the initially





formed [2+2]cycloadducts 5 which undergo subsequent elimination of phenyl or alkyl isocyanate. Although the [2+2]cycloaddition reaction of sulfonyl isocyanate with C=C, C=O or the thiocarbamate C=S bond has been reported, $^{4-6}$ the reaction with the C=N bond of 2-imino-1,3-thiazines has not yet been reported. It is especially interesting that acidic NH in the thiazine ring did not react with isocyanate. Extension of this reaction to 1,3-thiazine-2-thione <u>6</u> or 1,3-oxazine-2-thione <u>7</u> in place of a 2-imino such as <u>la</u>, <u>lb</u> and <u>4</u> gave the desired product <u>2</u> or <u>8</u>, respectively, in high yield as shown in Scheme 3.



Scheme 3

The application and limitations of this reaction are under further investigation.

Compound No.	Solvent	Reaction temp.	Reaction time (h)	Yield (%)
la	toluene	reflux	9	73
lb	benzene- dímethoxyethane	reflux	13	90
4	benzene	r. t.	3	88
<u>6</u>	toluene	reflux	10	69

Table 1. Reaction of 1,3-thiazines with p-toluenesulfonyl isocyanate

EXPERIMENTAL

All melting points are uncorrected. ¹H Nmr spectra were obtained on a JEOL JNM-PMX 60Si spectrometer with TMS as the internal standard. Ir spectra were measured on a Hitachi 260-10 infrared spectrophotometer. Mass spectra were measured on a Hitachi RMU-8GN mass spectrometer.

Preparation of 2-(p-Toluenesulfonyl)imino-5,6-dihydro-4H-1,3-thiazine (2).

The results are summarized in Table 1. A mixture of 2-phenylimino-1,3-thiazine <u>la</u> (0.48 g, 2.5 mmol) and <u>p</u>-toluenesulfonyl isocyanate (0.59 g, 3.0 mmol) in toluene (6 ml) was refluxed for 9 h. The reaction mixture was poured into water and extracted with methylene chloride. The extract was washed with water, dried over anhydrous sodium sulfate and concentrated under reduced pressure. The residue was recrystallized from methylene chloride to afford 0.49 g (73%) of <u>2</u>, mp 193-194°C; ir (KBr) 1582 (C=N), 1377 (SO₂) cm⁻¹; ¹H nmr (CDCl₃) δ 1.92 (2H,m), 2.38 (3H,s), 2.99 (2H,t,J=6.2Hz), 3.29 (2H,m), 7.29 (2H,d,J=8Hz), 7.83 (2H,d,J=8Hz), 8.80 (1H,br); ms m/z: 270 (M⁺); (Found: C, 48.63; H, 5.21; N, 10.18; S, 23.56. Calcd for C₁₁H₁₄N₂O₂S₂: C, 48.87; H, 5.22; N, 10.36; S, 23.72%).

2-(p-Toluenesulfonyl)imino-5,6-dihydro-4H-1,3-oxazine (8).

A mixture of 1,3-oxazine-2-thione (0.47 g, 4.0 mmol) and <u>p</u>-toluenesulfonyl isocyanate (0.95 g, 4.8 mmol) in toluene (8 ml) was treated as described above (refluxed for 4 h) to give 0.90 g (89%) of <u>8</u>, mp 195-196°C (from benzene); ir (CHCl₃) 1630 (C=N), 1310 $(SO_2) \text{ cm}^{-1}$; ¹H nmr (CDCl₃) & 2.02 (2H,m), 2.41 (3H,s), 3.50 (2H,dt,J=6,5Hz), 4.35 (2H,t,J=5.5Hz), 7.30 (2H,d,J=8Hz), 7.85 (2H,d,J=8Hz), 8.70 (1H,br); ms m/z: 254 (M⁺); (Found: C, 51.99; H, 5.55; N, 10.87. Calcd for C₁₁H₁₄N₂O₃S: C, 51.95; H, 5.55; N, 11.02%).

ACKNOWLEDGEMENT

We wish to express our thanks to Dr. Y. Hayashi, Director of Aburahi Laboratories, for his support and encouragement.

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Received, 5th December, 1988