A SIMPLE SYNTHESIS OF AMPHIMEDINE

Rolf H. Prager^{*} and Chris Tsopelas School of Physical Sciences, The Flinders University of South Australia, Bedford Park, South Australia 5042, Australia

<u>Abstract</u> - Amphimedine may be synthesised in six simple steps from the readily available azafluorenone (2).

Amphimedine (1) is one of a number of pentacyclic alkaloids recently isolated from marine sources $^{1-3}$. Because of its general cytotoxicity¹, we were interested in devising a synthesis that would generate intermediates that might possess more specific anti-tumor properties. We have communicated our synthesis at a Conference⁴, and recently reported syntheses of amphimedine by Echavarren and Stille⁵ and Kubo and Nakahara⁵ prompt us to report our quite different approach.

The fluorenone $(2)^{6-7}$ was silvlated and treated with 4-pyridyllithium at -20°C to give the fluorenol (3)⁸ in 87% yield. By analogy with our previous work⁷, we were confident that reaction of (3) with hydrazoic acid would result in migration only of the most electron-rich benzene ring (cf 4), to give the desired diazaphenanthrene (5), mp 327-328°C, which was isolated in 69% yield. Conversion of (5) to the chloropyridine (6), mp 227-228°C was achieved (90%) with phosphorus oxychloride and dimethylformamide in phosphorus trichloride. Of the three nitrogen atoms in (6) that on the pyridine ring was methylated exclusively with one equivalent of methyl fluorosulfonate, and alkaline ferricyanide oxidation then led cleanly to (7), mp 280-282°C (decomp), which was converted to the nitrile (8) by reaction with cuprous cyanide in hot dimethyl sulfoxide. Finally, reaction of (8) with hot PPA gave amphimedine (1), mp ≥ 340°C, with identical spectroscopic properties (ir, uv, nmr) as those kindly provided by Professor F.J. Schmitz. Prior hydrolysis of the nitrile to the acid, followed by cyclisation proved to be more efficient.⁶ The route described herein is short and efficient, using readily available reagents. A number of analogues of (2) and (5) have been subjected to cytotoxicity and mutagenic activity studies⁹. The compounds showed little mutagenicity activity using the hypoxanthine-guanine phosphoribosyl transferase or cytokinesisblock micronucleus assays. Analogues of (2) showed cytotoxicity, as measured by the frequency of clone-forming cells or blocking of ³H-thymidine uptake.



(a) (i) Me₃SiCl, Et₃N, THF 60°C, 60 min (ii) 4-bromopyridine, BuLi, -40°C~20°C, 2 h, 87%.

(b) NaN₃, PPA, 45°C, 20 h, 69%. (c) PC1₅, DMF (cat.) in POC1₃, 180°C, 20 h, 90%.

(d) (i) MeOSO₂F, 1.3 equiv., 20°C, 40 min (ii) KOH, K₂Fe(CN)₆, 2 equiv., 20°C, 10 h, 61%.

(e) CuCN, DMSO, 150°C, 4 h, 70%. (f) PPA, 90°C, 5 h, 35%.

REFERENCES

- F.J. Schmitz, S.K. Agarwal, S.P. Gunasekera, P.G. Schmidt, and J.N. Shoolery, J.Am.Chem.Soc., 1983, 105, 4835.
- 2. S.J. Bloor and F.J. Schmitz, J.Am.Chem.Soc., 1987, 109, 6134.
- 3. G. Cimeno, S. De Rose, S. De Stefano, and G. Sodano, Pure Appl. Chem., 1986, 58, 375.
- C. Tsopelas and R.H. Prager, '8th National Convention', p.42 Royal Australian Chemical Institute, Sydney, August 1987.
- (a) A.M. Echavarren and J.K. Stille, <u>J.Am.Chem.Soc.</u>, 1988, <u>110</u>, 4051. (b) A. Kubo and S. Nakahara, <u>Heterocycles</u>, 1988, <u>27</u>, 2095.
- 6. J.C. Powers and I. Ponticello, J.Am.Chem.Soc., 1968, 90, 7102.
- 7. T. Duong, R.H. Prager, and S.T. Were, Aust.J.Chem., 1983, 36, 1431.
- 8. All new compounds, except the unstable (3), gave satisfactory microanalytical or high resolution mass spectral figures.
- 9. Carried out by Professor A. Morley and D.B. Tran in the Flinders Medical Centre.

Received, 26th December, 1988