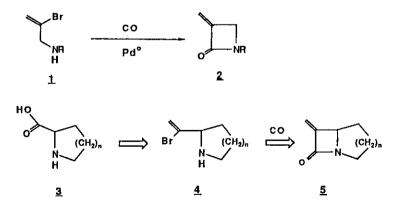
## RING CONSTRUCTION OF BICYCLIC-& LACTAM BY USE OF PALLADIUM CATALYZED CARBONYLATION

Miwako Mori,\* Yukako Higuchi, Katsuji Kagechika, and Masakatsu Shibasaki\*

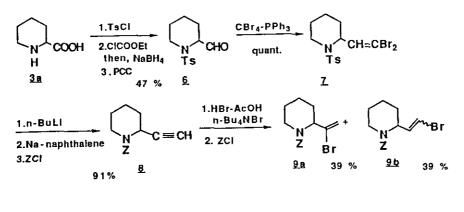
Faculty of Pharmaceutical Sciences, Hokkaido University, Sapporo, 060 Japan

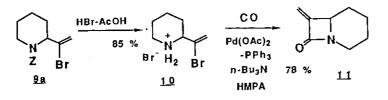
<u>Abstract</u>----Palladium catalyzed carbonylation into vinyl halide <u>10</u> afforded carbacepham <u>11</u> in good yield. The introduction of methoxycarbonyl group at C-4 position of carbacepham was achieved by conversion of methoxy group introduced by anodic oxidation in MeCN-MeOH to carboxyl group.

The search for  $\beta$ -lactam antibiotics possessing enhanced activity and resistance to  $\beta$ -lactamase has generated strong interest in methods of preparing the carbacephem and carbapenem skeletons. We have already reported the new synthetic method of  $\alpha$ -methylene- $\beta$ -lactams by use of palladium catalyzed carbonylation into 2-bromoallylamine derivatives.<sup>1</sup> This procedure prompted us to develope a new synthetic method of bicyclic  $\beta$ -lactam 5 from vinyl halide 4.



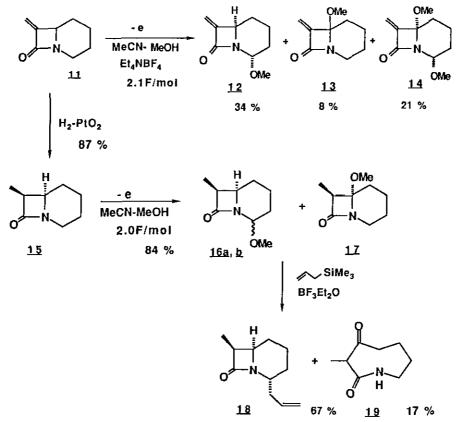
In order to prepare vinyl halide 4, the attempt to convert the carboxyl group of cyclic amino acid 3 such as proline(n=1) or pipecolinic acid(n=2), into vinyl halide was made. Pipecolinic acid 3a was converted to aldehyde 6 by usual method, which was treated with CBr<sub>4</sub>-PPh<sub>3</sub> to afford vinyl dibromide  $\underline{Z}$ . Treatment of  $\underline{Z}$  with excess n-BuLi<sup>2</sup> was followed by conversion of protecting group from tosyl group to benzyloxycarbonyl group<sup>3</sup>. Addition of HBr to compound  $\underline{8}^4$  followed by protection of amino group with ZCl provided vinyl bromides. <u>9a</u> and <u>9b</u> in a ratio of 1 to 1. However, the latter vinyl halide <u>9b</u> could easily give back to acetylene <u>8</u>. Removal of the protecting group of <u>9a</u> with HBr-AcOH afforded the desired vinyl bromide hydrogen bromide <u>10</u>, which was successfully converted to bicyclic  $\beta$ -lactam <u>11</u> by palladium catalyzed carbonylation. Namely, a solution of vinyl halide <u>10</u>, Pd(OAc)<sub>2</sub>(2 mol %), PPh<sub>3</sub>(4 mol %) and n-Bu<sub>3</sub>N(2.5 eq) in hexamethylphosphoric triamide(HMPA) was heated at 100°C for 4 h under carbon monoxide(1 atm) to give  $\beta$ -lactam <u>11</u> in 78 % yield.



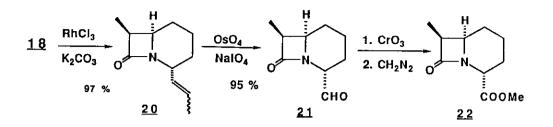


In order to introduce the carboxyl group at C-4 position of carbacepham <u>11</u>, the anodic oxidation should be a suitable method because the methoxy group at the  $\alpha$ -position of lactam<sup>5</sup> introduced by the anodic oxidation could be replaced by carbon nucleophile.<sup>6</sup> Thus, the electrochemical oxidation to  $\beta$ -lactam <u>11</u> was carried out in an undivided cell using platinum plates as electrode in MeCN-MeOH (9:1) containing Et<sub>4</sub>NBF<sub>4</sub> as supporting electrolyte . After 2.1 F/mol of electricity was passed

through the solution, methoxylated compounds 12, 13 and 14 were obtained in 34 %, 8 %, and 21 % yields, respectively.<sup>7</sup> Since the aliylic position should be easy to oxidize for the electrolysis, <sup>5b</sup>  $\alpha$ -methylene- $\beta$ -lactam 11 was hydrogenated with PtO<sub>2</sub> to give compound 15 as a single product. The methyl group of compound 15 should be oriented to the  $\beta$ -position because the catalyst might approach from the less hindered site. When 2.0 F/mol of electricity was passed through the MeCN-MeOH(9:1) solution of compound 15, inseparable mixture of methoxylated compounds 16 and 17 was obtained in 84 % yield. The nmr spectrum indicated that the ratio of 16 to 17 was 7 to 1. Treatment of the mixture of 16 and 17 with allylsilane in the presence of BF3Et2O<sup>8</sup> gave compound 18 in 67 % yield along with compound 19 (17 % yield).<sup>9</sup> The latter compound 19 should be obtained from compound 17 by treatment with BF3Et2O in the presence of a small amount of water.



Compound <u>18</u> was treated with RhCl<sub>3</sub> in the presence of K<sub>2</sub>CO<sub>3</sub> in EtOH followed by treatment with  $OsO_4$  and NaIO<sub>4</sub> to give aldehyde <u>21</u> in good yield. Oxidation of compound <u>21</u> with CrO<sub>3</sub> provided carboxylic acid, which was converted into methyl ester <u>22</u><sup>10</sup> by treatment with CH<sub>2</sub>N<sub>2</sub>.



These results suggested that palladium catalyzed carbonylation into vinyl halide <u>10</u> afforded bicyclic  $\beta$ -lactam <u>11</u> in good yield. In order to introduce the carboxyl group at C-4 position of carbacepham skeletone, introduction of the methoxy group to the  $\alpha$ -position of lactam by anodic oxidation was a good procedure because carbon nucleophile could be introduced to the methoxylated position. If proline was used for this reaction, carbapenam skeleton would be formed.

Further studies are in progress.

## REFERENCES AND NOTES

- M. Mori, K. Chiba, M. Okita, and Y. Ban, <u>Chem. Comm.</u>, 1979, 698. M. Mori, K. Chiba, M. Okita, I. Kayo, and Y. Ban, <u>Tetrahedron</u>, 41, 1985, 375.
- 2. E. J. Corey and P. L. Fuchs, Tetrahedron Lett., 1972, 3769.
- 3. The deprotection of the tosyl group with Na-naphthalene would accompany the debromination of vinyl halide.
- 4. J. Coussean, Sunthesis, 1980, 805.
- a) M. Okita, T. Wakamatsu, and Y. Ban, <u>Chem. Comm.</u> 1979, 749. b) M. Okita, M. Mori, T. Wakamatsu, and Y. Ban, <u>Heterocycles</u>, 23, 1985, 247.
- 6. T. Shono, Y. Matsumura, and K. Tsubata, J. Am. Chem. Soc., 103, 1981, 1172.
- Compounds <u>12</u> and <u>13</u> were inseparable mixture, but the nmr spectrum indicated that compound <u>12</u> was a single isomer. Presumably, methoxy group should attack from the less hindered site of the acyl iminium cation generated by electrolysis.
- 8. G. A. Kraus and K. Neuenschwander, Chem. Comm., 1982, 134.
- 9. From the nmr spectrum of compounds <u>16</u> and <u>17</u>, methoxylated compound <u>16</u> was a mixture of two isomers and the ratio of  $\alpha$ -(<u>16a</u>) to  $\beta$ -methoxylated compound(<u>16b</u>) was 4 to 1. However, compound <u>18</u> was obtained as a single isomer.
- Compound <u>22</u>; ir v max(CHCl<sub>3</sub>) 1730 cm<sup>-1</sup>; ms m/e 197(M<sup>+</sup>), 169(M<sup>+</sup>-CO), 138(M<sup>+</sup>-COOMe), 110, 82, 68, 55, high resolution mass spectrum Calcd for C<sub>10</sub>H<sub>15</sub>NO<sub>3</sub> 196.1069, found 196.1061; nmr δ(CDCl<sub>3</sub>) 1.19(d, J=6Hz, 3 H), 1.4-2.2(m, 6 H), 3.4(m, 1 H), 3.74(s, 3 H, OMe), 3.8(m, 1 H), 4.56(bd, J=7 Hz, 1 H).

Recieved, 23rd January, 1989