

**THE NICHOLAS REACTION OF INDOLES. PROPARGYLATION OF INDOLES  
WITH (PROPARGYL)DICOBALT HEXACARBONYL CATIONS †**

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**Abstract** - Indole reacted with (propargyl alcohol)Co<sub>2</sub>(CO)<sub>6</sub> complex **1** to give 3-(1,1-dimethylpropargyl)indole **3**, whereas N'-methoxycarbonyltryptamine **7** gave the corresponding N-substituted derivative **11**. The reaction of **7** with (propargyl acetate)Co<sub>2</sub>(CO)<sub>6</sub> complex **23** provided 3a-(1,1-dimethylpropargyl)hexahydropyrroloindole cobalt complex **9**. Oxidative demetalation of **3**, **11**, and **9** with Fe(NO<sub>3</sub>)<sub>3</sub> gave **5**, **13**, and **27**, respectively. Hydrogenation of **5** afforded the corresponding 3-(1,1-dimethylallyl)indole **6**.

There are a number of indole alkaloids such as echinulin,<sup>1)</sup> ilamycins,<sup>2)</sup> brevianamide E,<sup>3)</sup> roquefortine,<sup>4)</sup> LL S490β<sup>5)</sup> and flustramine A and C<sup>6)</sup> which have a 1,1-dimethylallyl group (invert prenyl group) at the 2- or 3-position of the indole ring. Foreseeing selection of an appropriate reagent and conditions for introducing a prenyl group into the indole ring, specially at the 3-position of 3-alkylindoles, is crucial for the successful synthesis of these alkaloids. Three methods<sup>7)</sup> have been reported for introduction of the inverted prenyl group to the indole ring at the 2 or 3 position, which include the rearrangement of 1-prenyl-, 2-prenylthio-, or 3-prenylthioindole derivatives, direct synthesis of 2-(1,1-dimethylallyl)indole from appropriate aniline derivatives, and 1,1-dimethylpropargylation of tryptamine derivatives.<sup>8)</sup> As part of our current interest in the search for new biologically active indole derivatives, we thought it worthwhile to focus our attention on the development of new methods for the synthesis of indoles bearing an invert prenyl group.

The reaction of cobalt-complexed propargylic alcohols with HBF<sub>4</sub> developed by Nicholas<sup>9)</sup> seemed attractive since a cobalt-stabilized carbocation can be reacted with a variety of carbon nucleophiles to provide alkylated products. Recently, Schreiber and coworkers have reported

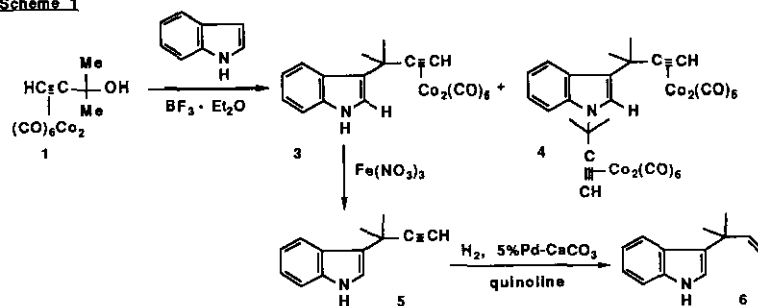
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† We dedicate this paper to the memory of Professor Tetsuji Kametani.

the modified Nicholas reaction which involves the reaction of cobalt-complexed propargylic ethers with Lewis acids.<sup>10)</sup> Here, we report, in full, an investigation of the reaction of indole and tryptamine derivatives with various propargyl dicobalt hexacarbonyl complexes in the presence of Lewis acids.

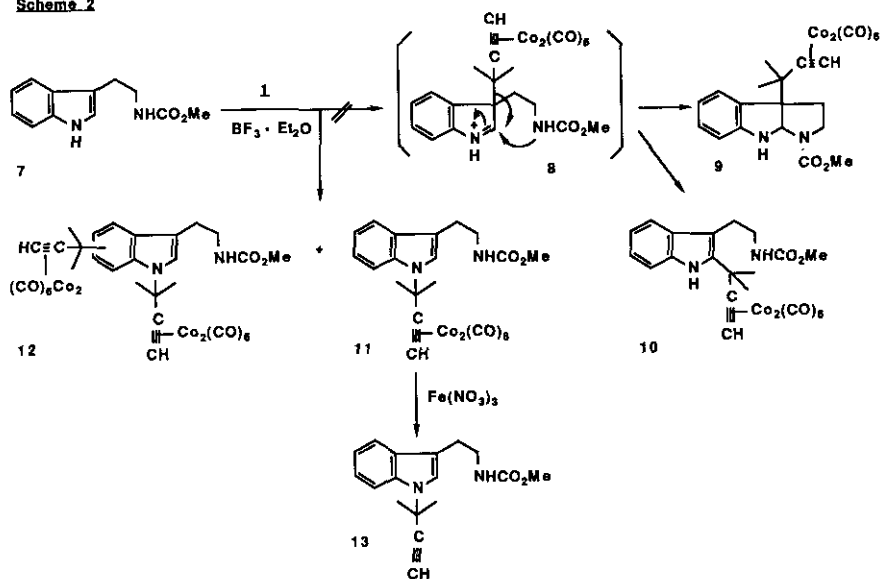
In our first experiment with propargyl dicobalt hexacarbonyl cation, we carried out the reaction of indole with the propargylic alcohol dicobalt hexacarbonyl complex **1** in the presence of  $\text{BF}_3 \cdot \text{etherate}$  instead of  $\text{HBF}_4$  (tetrafluoroboric acid). The reaction proceeded under ice-salt cooling for 1.5 h to give the expected 3-(1,1-dimethylpropargyl)indole complex **3** in 86% yield accompanied by a small amount (7%) of the disubstituted complex **4**. The  $^1\text{H}$  and  $^{13}\text{C}$ -nmr spectra of **3** and **4** confirmed the presence of 1,1-dimethylpropargyl substituent at the 3-position of indole. In the  $^1\text{H}$ -nmr spectrum of **3**, the C-2 proton signal, which was assigned by the selective decoupling method, appeared at  $\delta$  7.05 as a doublet with  $J=1.8\text{Hz}$ . This peak became a singlet upon  $\text{D}_2\text{O}$  addition. The C-4 proton was shifted to down field ( $\delta$  7.93,  $J=7.0\text{Hz}$ ), reflecting the deshielding effect by the presence of the propargyl dicobalt group at the 3-position. Treatment of the complex **3** with an excess of  $\text{Fe}(\text{NO}_3)_3 \cdot 9\text{H}_2\text{O}$  in  $\text{EtOH}$ <sup>11)</sup> afforded the corresponding 3-(1,1-dimethylpropargyl)indole **5** in nearly quantitative yield (97%). Catalytic hydrogenation of **5** using Lindlar catalyst gave 3-(1,1-dimethylallyl)indole **6** quantitatively (Scheme 1).

Scheme 1

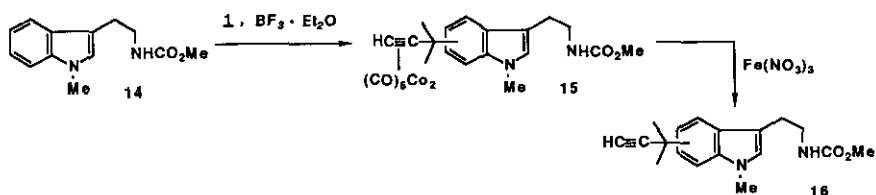


Similar reaction of **1** with *N*'-methoxycarbonyltryptamine **7** under ice-salt cooling for 3 h did not give the expected compound **9** or **10** but resulted in the formation of *N*-substituted complex **11** in 77% yield, whereas the reaction at ambient temperature gave a trace amount of disubstituted complex **12** (0.8%) together with **11** (35%) (Scheme 2). On the other hand, the aromatic substitution reaction on the benzene ring occurred preferentially to give **15** in 32% yield when

Scheme 2

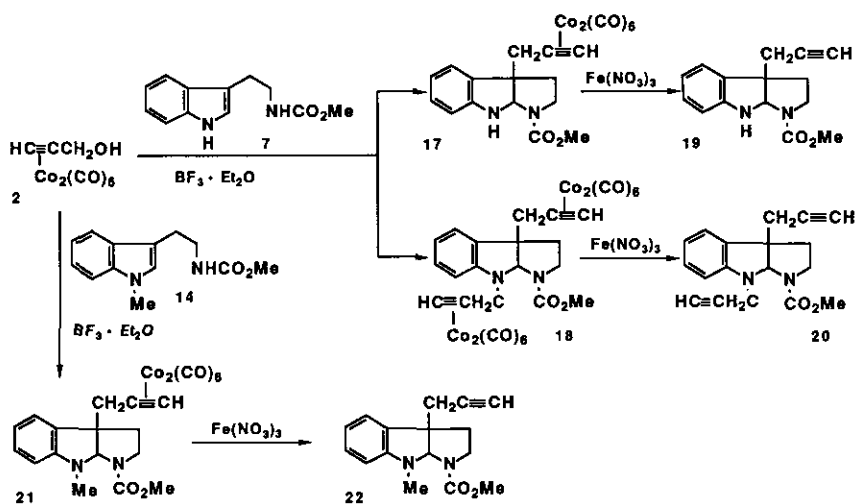


Scheme 3



N-methyl-N'-methoxycarbonyltryptamine 14 was treated with 1 at  $-54^\circ\text{C}$  for 26 h (Scheme 3). The structures of 11 and 15 were confirmed by converting them with  $\text{Fe}(\text{NO}_3)_3$  to the corresponding dimethylpropargyltryptamines 13 and 16, respectively. The *nmr* spectra of 12, 15, and 16 indicated that these compounds consisted of two or more positional isomers which, however, could not be separated. The failure to obtain either 9 or 10 suggested that the formation of 8 was prevented by the bulkyness of 1. Therefore, the reaction of 7 with the less hindered propargyl complex 2 was carried out in the presence of  $\text{BF}_3 \cdot \text{etherate}$  in  $\text{CH}_2\text{Cl}_2$  with ice-salt cooling for 9 h and the expected 3a-propargyl-1-methoxycarbonyl-1,2,3,3a,8,8a-hexahydropyrrolo[2,3-b]indole dicobalt complex 17 and 3a,8-dipropargyl complex 18 were obtained in 40% and 9% yields, respectively. Likewise, when 14 was mixed with 2 in  $\text{CH}_2\text{Cl}_2$  at ambient temperature for 4 h, 21 was formed in 69% yield. Treatment of 17, 18 and 21 with  $\text{Fe}(\text{NO}_3)_3$  provided 19, 20, and 22, respectively (Scheme 4).

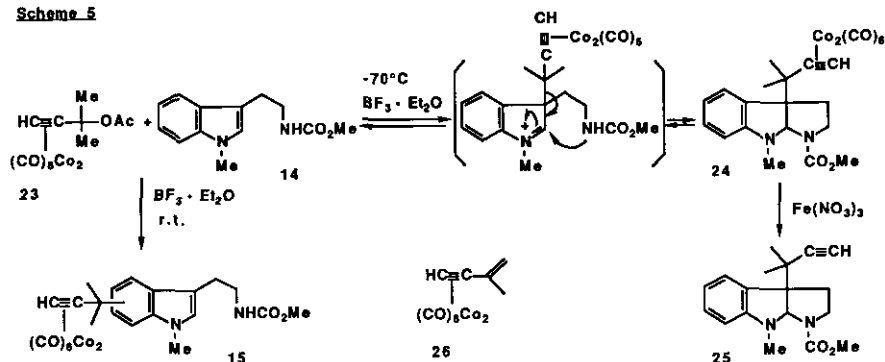
Scheme 4



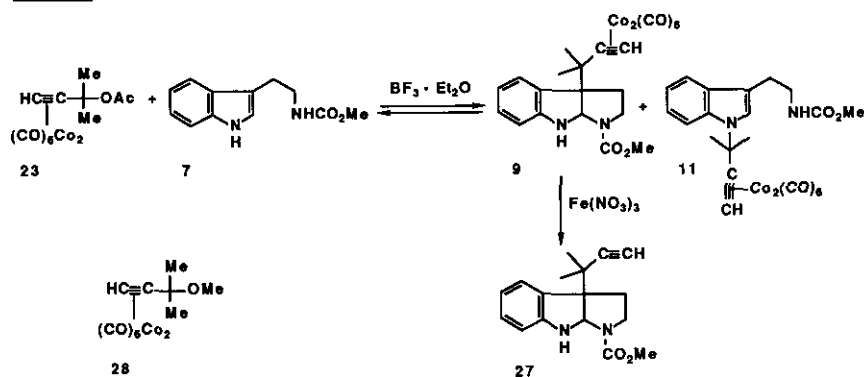
From the above results, steric factors appeared to influence the formation of pyrroloindole cobalt complexes via the reaction at the C-3 position of 3-substituted indole derivatives. As 1,1-dimethylpropargyl alcohol dicobalt complex **1** is too bulky to react at the 3-position, it seemed necessary to increase the reactivity of **1**. Expectedly, more reactive 1,1-dimethylpropargyl acetate dicobalt complex **23** was prepared in an excellent yield (above 90%) from 1,1-dimethylpropargyl acetate<sup>12)</sup> and dicobalt octacarbonyl in hexane at ambient temperature. However, we failed to purify **23** by silica gel column chromatography due to its facile elimination of AcOH to give **26**.<sup>9a)</sup> Therefore freshly prepared crude **23** was used for the reaction with indoles which was found to proceed at lower temperature in the presence of  $\text{BF}_3 \cdot \text{etherate}$ . Thus, when N-methyl-N'-methoxy-carbonyltryptamine **14** was treated with the complex **23** at  $-90^\circ\text{C}$  to  $-80^\circ\text{C}$  for 42 h, 3a-(1,1-dimethylpropargyl)hexahydropyrroloindole dicobalt complex **24** (14%) was formed and a 58% yield of **14** was recovered. Similar treatment of **14** with **23** followed by oxidative demetalation at ambient temperature gave 3a-(1,1-dimethylpropargyl)pyrroloindole **25** in 27% over all yield from **14**. Tlc examination showed that **24** converted to **14** even at low temperature in the presence of Lewis acid like  $\text{BF}_3 \cdot \text{etherate}$ . Accordingly, when the reaction mixture was allowed to warm from  $-85^\circ\text{C}$  to  $20^\circ\text{C}$  for 38 h, **24** was not obtained. Instead, **15** was obtained in 58% yield along with 30% recovery of **14**, indicating that at higher temperature, the initially formed complex **24** at lower temperature reverted to **14** and the 1,1-dimethylpropargyl dicobalt hexacarbonyl cation which underwent substitution reaction at the benzene ring to give the complex **15**. Furthermore, the complex **24** decomposed to **14** (68%), **26**,<sup>9a, 14)</sup> and a trace amount of **15** when dissolved in

$\text{BF}_3$ -etherate- $\text{CH}_2\text{Cl}_2$  solution at ambient temperature for 15 min. Therefore, in order to isolate the complex **24**, the reaction mixture has to be quenched rapidly with saturated  $\text{NaHCO}_3$  at low temperature (Scheme 5)

Scheme 5



Scheme 6



Similar reaction of **7** with the acetate complex **23** at  $-70^\circ\text{C}$  for 28 h produced the N-substituted complex **11** as the main product in 88% yield, but the pyrroloindole complex **9** was obtained in 12% yield which was decomplexed in 93% yield to give **27** (Scheme 6). Treatment of **7** with **23** at low temperature ( $-90^\circ\text{C}$ , 24 h, and  $-70^\circ\text{C}$ , 24 h) followed by immediate demetalation in one pot gave **27** (14%) and **13** (78%). Tlc examination showed that the complex **9** converted to **7** and **11** after 8.5 h when treated with  $\text{BF}_3$ -etherate in  $\text{CH}_2\text{Cl}_2$  at ambient temperature, indicating that **9** reverted to **7** more slowly than **24** to **14**. These reactions were also examined in the presence of  $\text{EtAlCl}_2$ ,  $\text{AlCl}_3$ ,  $\text{Me}_2\text{BBr}$ ,  $\text{TiCl}_4$ ,  $\text{TiCl}_4$ -DABCO,<sup>13</sup>  $\text{SnCl}_4$ , and  $\text{ZnCl}_2$ . However, the reactions did not proceed, with exception of  $\text{TiCl}_4$  (6eq) / DABCO (1eq) ( $70^\circ\text{C}$ , 22.5 h) with which the expected adduct **9** was formed in 4.9% yield. Additionally, attempted reaction of **7** with 1,1-dimethylpropargyl methyl ether

$\nu_{\text{max}}$  3050, 3100, 2980, 2970, 2080, 2050, 2010, 1210, 740 $\text{cm}^{-1}$ ;  $^1\text{H-nmr}$  (270MHz)  $\delta$  1.76 (6H, s,  $\text{CH}_3$ ), 2.07  
 39.0 (s,  $\text{CH}_3\text{C}(\text{CH}_3)_2$ ), 32.0 (q,  $\text{CH}_3$ ), 4: mp 133.5~135.5°C (n-Hexane);  $\nu_{\text{max}}$  2185 $\text{sh}$ , 2605 $\text{sh}$ , 3055 $\text{sh}$ , 3500 $\text{cm}^{-1}$ ; ir  
 122.0 (d,  $\text{C}_5$  or  $\text{C}_6$ ), 121.3 (d,  $\text{C}_4$ ), 119.2 (d,  $\text{C}_6$  or  $\text{C}_5$ ), 120.3 (d,  $\text{C}_2$ ), 111.5 (d,  $\text{C}_7$ ), 110.4 (s,  $\text{C}_3$ ), 74.5 (d,  $\text{C}=\text{CH}$ ),  
 (1H, s, NH, exchangeable);  $^{13}\text{C-nmr}$  (67.8MHz)  $\delta$  200.0 (s, CO), 137.3 (s,  $\text{C}_7\text{a}$ ), 125.5 (s,  $\text{C}_3\text{a}$ ), 125.6 (s,  $\text{C}=\text{CH}$ ),  
 collapses to s with  $\text{D}_2\text{O}$ ), 7.18-7.26 (2H, m,  $\text{C}_5$ , 6-H), 7.34 (1H, d,  $\text{J}=8\text{Hz}$ ,  $\text{C}_7\text{-H}$ ), 7.93 (1H, d,  $\text{J}=\text{THz}$ ,  $\text{C}_4\text{-H}$ ), 7.96  
 1450, 740 $\text{cm}^{-1}$ ;  $^1\text{H-nmr}$  (270MHz)  $\delta$  1.83 (6H, s,  $\text{CH}_3\text{C}(\text{CH}_3)_2$ ), 6.30 (1H, s,  $\text{C}=\text{CH}$ ), 7.05 (1H, d,  $\text{J}=1.8\text{Hz}$ ,  $\text{C}_2\text{-H}$ ,  
 116.5~117.5°C (n-Hexane);  $\nu_{\text{max}}$  218, 2605 $\text{sh}$ , 2925 $\text{sh}$ , 346, 4105 $\text{sh}$  $\text{cm}^{-1}$ ; ir  $\nu_{\text{max}}$  3380, 2080, 2050, 2020, 1950,  
 chromatography (AcOEt/n-Hex=1/5) gave 1.61 g of 3 (86%) and 0.24 g of 4 (7%) as dark red prisms. 3: mp  
 were washed with brine and dried over  $\text{Na}_2\text{SO}_4$  and the solvent was removed by rotary evaporation. Silica gel  
 addition of saturated  $\text{NaHCO}_3$ . The aqueous layer was extracted with  $\text{CH}_2\text{Cl}_2$ . The combined organic extracts  
 $\text{CH}_2\text{Cl}_2$  (5 ml) were added by injection. After stirring for 1.5 h, the reaction mixture was quenched by the  
 (1.64 g, 4.43 mmol) in dry  $\text{CH}_2\text{Cl}_2$  (5 ml) and a solution of  $\text{BF}_3\cdot\text{Et}_2\text{O}$  (distilled over  $\text{CaH}_2$ ) (1.13 g, 7.96 mmol) in  
 salt bath under an argon atmosphere. A solution of (1,1-dimethylpropargyl alcohol) $\text{Co}_2(\text{CO})_8$  complex **14**)  
 A dry flask was charged with a solution of the indole (0.47 g, 4.01 mmol) in dry  $\text{CH}_2\text{Cl}_2$  (30 ml) cooled in a ice-  
 hexagabartboronyl) complex **4**)<sup>14</sup>)

~~Dimethylpropargylindole dicobalt hexacarbonyl complex **3** and 1,3-bis(dimethylpropargyl)indole bis(dicobalt  
 hexacarbonyl) complex **4**)<sup>14</sup>)~~

Representative alkylation of indole derivative with propargyl alcohol dicobalt hexacarbonyl complexes: **3**-  
 to solutions in  $\text{CDCl}_3$ .  
 noted uv spectra ( $\lambda$  in nm) refer to a solution in 95% EtOH, ir spectra ( $\nu$  in  $\text{cm}^{-1}$ ) to KBr disks, and nmr spectra  
 values(ppm). Microanalyses were performed on a Perkin-Elmer 240 C, H, and N analyzer. Unless otherwise  
 spectrometer. All chemical shifts are reported downfield from TMS as an internal standard and expressed as  $\delta$   
 JNM-GX 270 ( $^1\text{H-nmr}$ , 270MHz and  $^{13}\text{C-nmr}$ , 67.8MHz), or a JEOL JNM-GSX 500 ( $^1\text{H-nmr}$ , 500MHz)  
 spectrometer. Nmr spectra were recorded on a Hitachi R-24B ( $^1\text{H-nmr}$ , 60MHz), a JEOL JNM-FX 270, a JEOL  
 Hitachi 260-10 spectrophotometer. Mass spectra were recorded on a Hitachi M-60 or a RMU-7M mass  
 uncorrected. Uv spectra were recorded on a Hitachi 323 spectrophotometer. Ir spectra were obtained with a  
 Melting points were determined with Yamato MP-1 and Yanagimoto micro melting point apparatus, and are

**EXPERIMENTAL**

synthesis of natural products is in progress.  
 the presence of  $\text{TiCl}_4\text{-DABCO}$  **13**) gave **9** in 6.6% yield. Further application of this method to the  
 dicobalt complex **28** with  $\text{BF}_3\cdot\text{etherate}$  **10**) gave starting material, and the reaction of **7** with **28** in

(6H, s, CH<sub>3</sub>), 6.25 (2H, s, C=CH), 7.12 (1H, s, C<sub>2</sub>-H), 7.23-7.14 (2H, m, C<sub>5</sub>, 6-H), 7.69 (1H, d, J=8Hz, C<sub>7</sub>-H), 7.89 (1H, d, J=7Hz, C<sub>4</sub>-H).

N-Dimethylpropargyl-N'-methoxycarbonyltryptamine dicobalt hexacarbonyl complex 11 and

bis(dimethylpropargyl)-N'-methoxycarbonyltryptamine bis(dicobalt hexacarbonyl)complex 12

Similar treatment of N'-methoxycarbonyltryptamine **7** (0.18 g, 0.81 mmol) and **1** (0.40 g, 1.05 mmol) in CH<sub>2</sub>Cl<sub>2</sub> (20 ml) in the presence of BF<sub>3</sub>·Et<sub>2</sub>O (0.23 g, 1.62 mmol) under ice-salt cooling for 3 h, followed by work-up as described above, gave 0.35 g of **11** (77%) as a dark red oil. **11**: uv λ<sub>max</sub> 222, 270<sup>sh</sup>, 300<sup>sh</sup>, 352, 410<sup>sh</sup>nm; <sup>1</sup>H-nmr (270MHz) δ 2.13 (6H, s, CH<sub>3</sub>CCH<sub>3</sub>), 2.92 (2H, t, J=7Hz, CH<sub>2</sub>), 3.40-3.50 (2H, m, CH<sub>2</sub>N), 3.67 (3H, s, OCH<sub>3</sub>), 4.65 (1H, bs, NH), 6.30 (1H, s, C=CH), 7.14 (1H, s, C<sub>2</sub>-H), 7.24-7.09 (2H, m, arom-H), 7.56 (1H, d, J=7Hz, arom-H), 7.73 (1H, d, J=8Hz, arom-H). Similar reaction of **7** (0.20 g, 0.92 mmol) and **1** (0.51 g, 1.38 mmol) in CH<sub>2</sub>Cl<sub>2</sub> (30 ml) carried out in the presence of BF<sub>3</sub>·Et<sub>2</sub>O (0.1 g, 0.74 mmol) at ambient temperature for 24 h, gave 0.19 g of **11** (35%), 0.006 g of **12** (0.8%) as dark red oils, and the recovered 0.12 g of **7** (61%). **12**: uv λ<sub>max</sub> 303, 350nm; <sup>1</sup>H-nmr (270MHz) δ 1.75 (15/2H, s, CH<sub>3</sub>), 1.84 (9/2H, s, CH<sub>3</sub>), 2.85-2.95 and 3.08-3.14 (2H, m, CH<sub>2</sub>), 3.40-3.57 (2H, m, CH<sub>2</sub>N), 3.68 (3H, s, OCH<sub>3</sub>), 4.60 (3/8H, bs, NH), 4.70 (5/8H, bs, NH), 6.15 (3/8H, s, C=CH), 6.21 (5/8H, s, C=CH), 6.25 (1H, s, C=CH), 7.10 (1H, s, C<sub>2</sub>-H), 7.38-7.89 (3H, m, arom-H).

N-Methyl-N'-methoxycarbonyldimethylpropargyltryptamine dicobalt hexacarbonyl complex 15

Similar treatment of N-methyl-N'-methoxycarbonyltryptamine **14** (0.19 g, 0.81 mmol) and **1** (0.40 g, 1.05 mmol) in CH<sub>2</sub>Cl<sub>2</sub> (20 ml) in the presence of BF<sub>3</sub>·Et<sub>2</sub>O (0.42 g, 3.00 mmol) at -54°C for 26 h, followed by work-up as described above, gave 0.15 g of **15** (32%) as a dark red oil and the recovered 0.07 g of **14** (38%). **15**: uv λ<sub>max</sub> 226, 260<sup>sh</sup>, 346, 400<sup>sh</sup>nm; <sup>1</sup>H-nmr (270MHz) δ 1.79 (9/2H, s, CH<sub>3</sub>CCH<sub>3</sub>), 1.80 (3/2H, s, CH<sub>3</sub>CCH<sub>3</sub>), 2.88-2.97 (2H, m, CH<sub>2</sub>), 3.46-3.50 (2H, m, CH<sub>2</sub>N), 3.67 (3H, s, NCH<sub>3</sub>), 3.75 (9/4H, s, OCH<sub>3</sub>), 3.71 (3/4H, s, OCH<sub>3</sub>), 4.71 (1H, bs, NHCO), 6.18 (1/4H, s, C=CH), 6.20 (3/4H, s, C=CH), 6.85 (1H, s, C<sub>2</sub>-H), 7.20-7.62 (3H, m, arom-H). The nmr spectrum of **15** also suggested this consist of two positional isomers which failed to be separated by silica gel chromatography.

1-Methoxycarbonyl-3a-propargyl-1,2,3,3a,8,8a-hexahydropyrrolo[2,3-b]indole dicobalt hexacarbonyl complex

17 and 1-Methoxycarbonyl-3a,8-dipropargyl-1,2,3,3a,8,8a-hexahydropyrrolo[2,3-b]indole bis(dicobalt hexacarbonyl)complex 18

Similar treatment of N'-methoxycarbonyltryptamine **7** (0.80 g, 3.66 mmol) and (propargyl alcohol)Co<sub>2</sub>(CO)<sub>6</sub> complex **2** (1.40 g, 4.09 mmol) in CH<sub>2</sub>Cl<sub>2</sub> (50 ml) in the presence of BF<sub>3</sub>·Et<sub>2</sub>O (0.88 g, 6.20 mmol) under ice-salt cooling for 9 h, followed by work-up as described above, gave 0.80 g of **17** (40%), 0.30 g of **18** (9%) as dark red oils, and the recovered 0.30 g of **7** (38%). **17**: uv λ<sub>max</sub> 240<sup>sh</sup>, 290<sup>sh</sup>, 350nm; <sup>1</sup>H-nmr (270MHz) δ 2.24-2.29 (2H, m, CH<sub>2</sub>), 3.06-3.13 and 3.62-3.65 (2H, m, NCH<sub>2</sub>), 3.42 (1H, d, J=16.5Hz, CH<sub>2</sub>C=CH), 3.53 (1H, d, J=16.5Hz,

CH<sub>2</sub>C=CH), 3.69 (3/2H, s, OCH<sub>3</sub>), 3.77 (3/2H, s, OCH<sub>3</sub>), 4.67 (1/2H, s, NH, exchangeable), 5.10 (1/2H, s, NH, exchangeable), 5.30-5.31 (1H, m, NCHN), 5.72 (1/2H, s, C=CH), 5.75 (1/2H, s, C=CH), 6.60 (1H, d, J=8Hz, arom-H), 6.77 (1H, d, J=7Hz, arom-H), 7.07-7.14 (2H, m, arom-H). 18: uv λ<sub>max</sub> 250<sup>sh</sup>, 315<sup>sh</sup>, 350nm; <sup>1</sup>H-nmr (270MHz) δ 2.02-2.20 (2H, m, CH<sub>2</sub>), 3.12-3.17 and 3.79-3.80 (2H, m, NCH<sub>2</sub>), 3.39 (1H, d, J=16Hz, CH<sub>2</sub>C=CH), 3.50 (1H, d, J=16Hz, CH<sub>2</sub>C=CH), 3.76 (3/2H, s, OCH<sub>3</sub>), 3.78 (3/2H, s, OCH<sub>3</sub>), 4.72-4.90 (2H, m, NCH<sub>2</sub>C=CH), 5.78-5.92 (2H, m, C=CH), 6.07 (1H, s, NCHN), 6.68-6.72 (1H, m, arom-H), 7.03-7.21 (3H, m, arom-H).

1-Methoxycarbonyl-3a-propargyl-8-methyl-1,2,3,3a,8,8a-hexahydropyrrolo[2,3-b]indole dicobalt hexacarbonyl complex 21

Similar treatment of N-methyl-N'-methoxycarbonyltryptamine 14 (0.25 g, 1.08 mmol) and 2 (0.37 g, 1.08 mmol) in CH<sub>2</sub>Cl<sub>2</sub> (30 ml) in the presence of BF<sub>3</sub>·Et<sub>2</sub>O (0.33 g, 2.16 mmol) at ambient temperature for 4 h, followed by work-up as described above, gave 0.41 g of 21 (69%) as a dark red oil and the recovered 0.071 g of 14 (19%). 21: uv λ<sub>max</sub> 248<sup>sh</sup>, 308<sup>sh</sup>, 348, 400<sup>sh</sup>nm; <sup>1</sup>H-nmr (270MHz) δ 2.06-2.18 (2H, m, CH<sub>2</sub>), 2.87 (3/2H, s, NCH<sub>3</sub>), 2.96 (3/2H, s, NCH<sub>3</sub>), 2.96-3.10 and 3.80-3.93 (2H, m, NCH<sub>2</sub>), 3.40 (1H, d, J=17Hz, CH<sub>2</sub>C=CH), 3.58 (1H, d, J=17Hz, CH<sub>2</sub>C=CH), 3.72 (3/2H, s, OCH<sub>3</sub>), 3.77 (3/2H, s, OCH<sub>3</sub>), 5.47 (1/2H, s, NCHN), 5.56 (1/2H, s, NCHN), 5.63 (1/2H, s, C=CH), 5.68 (1/2H, s, C=CH), 6.38 (1H, d, J=8Hz, arom-H), 6.69 (1H, dd, J=7, 7Hz, arom-H), 7.01 (1H, d, J=7Hz, arom-H), 7.14 (1H, dd, J=7, 8Hz, arom-H).

Representative alkylation of N'-methoxycarbonyltryptamine with propargyl acetate dicobalt hexacarbonyl complex: 1-Methoxycarbonyl-3a-dimethylpropargyl-1,2,3,3a,8,8a-hexahydropyrrolo[2,3-b]indole dicobalt hexacarbonyl complex 9

A dry flask was charged with a solution of N'-methoxycarbonyltryptamine 7 (0.50 g, 2.29 mmol) in dry CH<sub>2</sub>Cl<sub>2</sub> (80 ml) and cooled to -70°C under an argon atmosphere. A solution of (1,1-dimethylpropargyl acetate)Co<sub>2</sub>(CO)<sub>6</sub> complex 23<sup>12, 15</sup> (1.89 g, 4.51 mmol) in dry CH<sub>2</sub>Cl<sub>2</sub> (20 ml) and a solution of BF<sub>3</sub>·Et<sub>2</sub>O (distilled over CaH<sub>2</sub>) (1.75 g, 12.3 mmol) in dry CH<sub>2</sub>Cl<sub>2</sub> (5 ml) were added dropwise. After stirring for 28 h, the reaction mixture was quenched by the addition of saturated NaHCO<sub>3</sub>. The aqueous layer was extracted with CH<sub>2</sub>Cl<sub>2</sub>. The combined organic extracts were washed with brine and dried over Na<sub>2</sub>SO<sub>4</sub> and the solvent was removed by rotary evaporation. Silica gel chromatography (AcOEt/n-Hex=1/5) gave 0.15 g of 9 (12%) and 1.07 g of 11 (88%) as dark red oils. 9: uv λ<sub>max</sub> 245<sup>sh</sup>, 310<sup>sh</sup>, 350, 385nm; <sup>1</sup>H-nmr (500MHz) δ 1.16 (2H, s, CH<sub>3</sub>CCH<sub>3</sub>), 1.45 (4H, s, CH<sub>3</sub>CCH<sub>3</sub>), 2.08-2.32 and 2.40-2.54 (2H, m, CH<sub>2</sub>), 2.79-2.03 and 3.54-3.85 (2H, m, NCH<sub>2</sub>), 3.69 (2H, s, OCH<sub>3</sub>), 3.77 (1H, s, OCH<sub>3</sub>), 4.60 (1/3H, s, NH), 5.05 (2/3H, s, NH), 5.42 (1/3H, s, NCHN), 5.46 (2/3H, s, NCHN), 6.01 (1/3H, s, C=CH), 6.06 (2/3H, s, C=CH), 6.58-6.63 (1H, m, arom-H), 6.74-6.77 (1H, m, arom-H), 7.03-7.17 and 7.21-7.24 (2H, m, arom-H).



Similar treatment of **15** (0.23 g, 0.40 mmol) with  $\text{Fe}(\text{NO}_3)_3 \cdot 9\text{H}_2\text{O}$  in ethanol (10 ml) under ice cooling, followed by work-up as described above, gave 0.81 g of **16** (68%) as a colorless oil. IR:  $\nu_{\text{max}}$  230.5, 283<sup>sh</sup>, 291 cm<sup>-1</sup>;  $^1\text{H-NMR}$  (270 MHz)  $\delta$  1.68 (9/2H, s,  $\text{CH}_3\text{CCH}_3$ ), 1.69 (3/2H, s,  $\text{CH}_3\text{CCH}_3$ ), 2.36 (3/4H, s,  $\text{C}=\text{CH}$ ), 2.38 (1/4H, s,  $\text{C}=\text{CH}$ ), 2.95 (2H, t,  $\text{J}=\text{7Hz}$ ,  $\text{CH}_2$ ), 3.45-3.55 (2H, m,  $\text{CH}_2\text{N}$ ), 3.66 (3H, s,  $\text{NCH}_3$ ), 3.74 (3/4H, s,  $\text{OCH}_3$ ), 3.76 (9/4H, s,  $\text{OCH}_3$ ), 4.74 (1H, bs,  $\text{NHCO}$ ), 6.87 (1H, s,  $\text{C}_2\text{-H}$ ).

**N-Methyl-N'-methoxycarbonylpropargyltryptamine 16**

was confirmed by comparison with the sample prepared by the other method.<sup>8</sup> followed by work-up as described above, gave **13** (0.12 g) quantitatively as a yellowish oil. This structure of **13**

Similar treatment of **11** (0.24 g, 0.42 mmol) with  $\text{Fe}(\text{NO}_3)_3 \cdot 9\text{H}_2\text{O}$  in ethanol (10 ml) under ice-salt cooling,

**N-Dimethylpropargyl-N'-methoxycarbonyltryptamine 13:**

$\text{C}_{13}\text{H}_{13}\text{N}$  183.1046. Found 183.1046 ( $\text{M}^+$ ).

exchangeable), 7.94 (1H, d,  $\text{J}=\text{8Hz}$ ,  $\text{C}_4\text{-H}$ );  $m/z$  (%) 183 ( $\text{M}^+$ , 53), 168 (100), 141 (18); HRMS calcd for 2.29 (1H, s,  $\text{C}=\text{CH}$ ), 7.10-7.23 (3H, m,  $\text{C}_2\text{-H}$  and  $\text{arom-H}$ ), 7.34-7.37 (1H, m,  $\text{arom-H}$ ), 7.91 (1H, bs,  $\text{NH}$ ), 291 cm<sup>-1</sup>;  $^1\text{H-NMR}$  (270 MHz)  $\delta$  1.74 (6H, s,  $\text{CH}_3\text{CCH}_3$ ), 2.82, chromatography (AcOEt/n-Hex=1/5) to give **5** (97%) as a pink solid; IR:  $\nu_{\text{max}}$  222.5, 275<sup>sh</sup>, 282, dried over  $\text{Na}_2\text{SO}_4$  and concentrated by rotary evaporation. The residue was subjected to silica gel flash aqueous layer was extracted with  $\text{CH}_2\text{Cl}_2$ . The combined organic extracts were washed with water and brine, of  $\text{CO}_2$  ceased and  $\text{Fe}(\text{NO}_3)_3$  color persisted. This solution was then partitioned between  $\text{CH}_2\text{Cl}_2$  and  $\text{H}_2\text{O}$ . The The 3-(1,1-dimethylpropargyl)indole dicobalt hexacarbonyl complex **3** (0.50 g, 1.07 mmol) was dissolved in ethanol (10 ml). Ferric nitrate $\cdot 9\text{H}_2\text{O}$  was added in portions with stirring at ambient temperature until evolution

**Formation of 3-Dimethylpropargylindole 5**

Representative oxidative demetalation of propargyl indole dicobalt hexacarbonyl complex with ferric nitrate: 6.67 (1H, t-like,  $\text{arom-H}$ ), 7.09-7.18 (2H, m,  $\text{arom-H}$ ), (1/2H, s,  $\text{NCHN}$ ), 5.62 (1/2H, s,  $\text{NCHN}$ ), 5.83 (1/2H, s,  $\text{C}=\text{CH}$ ), 5.93 (1/2H, s,  $\text{C}=\text{CH}$ ), 6.37 (1H, d-like,  $\text{arom-H}$ ), (3/2H, s,  $\text{NCH}_3$ ), 3.74 (3/2H, s,  $\text{OCH}_3$ ), 3.80 (3/2H, s,  $\text{OCH}_3$ ), 2.86-2.94 and 3.78-4.06 (2H, m,  $\text{NCH}_2$ ), 5.48 (3H, s,  $\text{CH}_3\text{CCH}_3$ ), 1.45 (3H, s,  $\text{CH}_3\text{CCH}_3$ ), 2.02-2.08 and 2.22-2.32 (2H, m,  $\text{CH}_2$ ), 2.86 (3/2H, s,  $\text{NCH}_3$ ), 2.94 as a dark red oil and the recovered 0.23 g of **14** (58%). IR:  $\nu_{\text{max}}$  252, 315, 350<sup>sh</sup> cm<sup>-1</sup>;  $^1\text{H-NMR}$  (270 MHz)  $\delta$  1.12 addition of saturated  $\text{NaHCO}_3$  at reaction temperature and work-up as described above, gave 0.20 g of **24** (14%) (mmol) in  $\text{CH}_2\text{Cl}_2$  (20 ml) in the presence of  $\text{BF}_3 \cdot \text{Et}_2\text{O}$  (1.85 g, 13.0 mmol) at  $-90 \sim -80^\circ\text{C}$  for 42 h, followed by Similar treatment of N-methyl-N'-methoxycarbonyltryptamine **14** (0.40 g, 1.72 mmol) and **23** (1.05 g, 2.51

**hexacarbonyl complex 24**

**1-Methoxycarbonyl-3-(1,1-dimethylpropargyl)-8-methyl-1,2,3,3a,8,8a-hexahydroindolo[2,3-b]indole dicobalt**

7.24-7.32 (1H, m, arom-H), 7.44-7.52 (1H, m, arom-H), 7.73 (1H, d, J=2Hz, arom-H); ms  $m/z$  (%) 298 ( $M^+$ , 31), 210 (100), 143 (5).

1-Methoxycarbonyl-3a-propargyl-1,2,3,3a,8,8a-hexahydropyrrolo[2,3-b]indole 19

Similar treatment of 17 (0.89 g, 1.56 mmol) with  $Fe(NO_3)_3 \cdot 9H_2O$  in ethanol (20 ml) at ambient temperature, followed by work-up as described above, gave 0.41 g of 19 (99%) as colorless needles. 19: mp 95~97°C (AcOEt-n-Hexane); uv  $\lambda_{max}$  244, 299nm; ir  $\nu_{max}$  3370, 3270, 2940, 1700, 750, 740 $cm^{-1}$ ;  $^1H$ -nmr (270MHz)  $\delta$  2.05 (1H, t, J=3Hz, C=CH), 2.22-2.46 (2H, m, CH<sub>2</sub>), 2.52 (1H, dd, J=7, 3Hz, CH<sub>2</sub>C=CH), 2.59 (1H, dd, J=7, 3Hz, CH<sub>2</sub>C=CH), 3.02-3.12 and 3.61-3.81 (2H, m, CH<sub>2</sub>N), 3.69 (12/7H, s, OCH<sub>3</sub>), 3.78 (9/7H, s, OCH<sub>3</sub>), 4.73 (3/7H, s, NH), 5.15 (4/7H, s, NH), 5.22 (3/7H, s, NCHN), 5.26(4/7H, s, NCHN), 6.61(1H, d, J=8Hz, arom-H), 6.73-6.80(1H, m, arom-H), 7.07-7.18 (2H, m, arom-H); ms  $m/z$  (%) 256 ( $M^+$ , 94), 217 (100), 130 (59), 77 (16). Anal. Calcd for C<sub>15</sub>H<sub>16</sub>N<sub>2</sub>O<sub>2</sub>: C, 70.10; H, 6.30; N, 10.82. Found: C, 70.10; H, 6.30; N, 10.82.

1-Methoxycarbonyl-3a,8-dipropargyl-1,2,3,3a,8,8a-hexahydropyrrolo[2,3-b]indole 20

Similar treatment of 18 (0.51 g, 0.59 mmol) with  $Fe(NO_3)_3 \cdot 9H_2O$  in ethanol (15 ml) at ambient temperature, followed by work-up as described above, gave 0.16 g of 20 (90%) as colorless oil. 20: uv  $\lambda_{max}$  249, 300nm; ir  $\nu_{max}$  3270, 2940, 1680, 740 $cm^{-1}$ ;  $^1H$ -nmr (270MHz)  $\delta$  2.03 (1H, t, J=3Hz, CH<sub>2</sub>C=CH), 2.08 (1H, s, NCH<sub>2</sub>C=CH), 2.13-2.35 (2H, m, CH<sub>2</sub>), 2.55 (1H, dd, J=17, 3Hz, CH<sub>2</sub>C=CH), 2.65 (1H, dd, J=17, 3Hz, CH<sub>2</sub>C=CH), 3.11-3.19 and 3.82-3.91 (2H, m, NCH<sub>2</sub>), 3.73 (18/11H, s, OCH<sub>3</sub>), 3.82 (15/11H, s, OCH<sub>3</sub>), 4.08-4.37 (2H, m, NCH<sub>2</sub>C=CH), 5.44 (5/11H, s, NCHN), 5.52 (6/11H, s, NCHN), 6.60 (1H, dd, J=7, 1Hz, arom-H), 6.79 (1H, dd, J=8, 1Hz, arom-H), 7.16-7.22 (2H, m, arom-H); ms  $m/z$  (%) 294 ( $M^+$ , 78), 255 (100), 216 (40), 180 (66), 168 (77), 115 (20); HRms calcd for C<sub>18</sub>H<sub>18</sub>N<sub>2</sub>O<sub>2</sub> 294.1369. Found 294.1368 ( $M^+$ ); Anal. Calcd for C<sub>18</sub>H<sub>18</sub>N<sub>2</sub>O<sub>2</sub>: C, 73.45; H, 6.07; N, 9.52. Found: C, 73.14; H, 6.34; N, 9.24.

1-Methoxycarbonyl-(3a-propargyl-8-methyl-1,2,3,3a,8,8a-hexahydropyrrolo[2,3-b]indole 22

Similar treatment of 21 (0.51 g, 0.9 mmol) with  $Fe(NO_3)_3 \cdot 9H_2O$  in ethanol (15 ml) under ice cooling, followed by work-up as described above, gave 0.19 g of 22 (78%) as a colorless oil. 22: uv  $\lambda_{max}$  253, 308nm; ir  $\nu_{max}$  (neat) 3270, 3040, 2940, 1690, 1600, 740 $cm^{-1}$ ;  $^1H$ -nmr (270MHz)  $\delta$  2.00 (1H, t, J=3Hz, C=CH), 2.04-2.33 (2H, m, CH<sub>2</sub>), 2.52 (1H, dd, J=17, 3Hz, CH<sub>2</sub>C=CH), 2.54 (1H, dd, J=17, 3Hz, CH<sub>2</sub>C=CH), 2.91 (6/5H, s, NCH<sub>3</sub>), 3.91 (9/5H, s, NCH<sub>3</sub>), 3.06-3.09 and 3.83-3.94 (2H, m, NCH<sub>2</sub>), 3.72 (9/5H, s, OCH<sub>3</sub>), 3.78 (6/5H, s, OCH<sub>3</sub>), 5.32 (2/5H, s, NCHN), 5.40 (3/5H, s, NCHN), 6.40 (1H, d, J=8Hz, arom-H), 6.66-6.71 (1H, m, arom-H), 7.11-7.17 (2H, m arom-H); ms  $m/z$  (%) 270 ( $M^+$ , 100), 231 (95), 216 (18), 144 (65), 115 (11).

1-Methoxycarbonyl-3a-dimethylpropargyl-8-methyl-1,2,3,3a,8,8a-hexahydropyrrolo[2,3-b]indole 25

Similar treatment of 24 (0.18 g, 0.33 mmol) with  $Fe(NO_3)_3 \cdot 9H_2O$  in ethanol (10 ml) under ice cooling, followed by work-up as described above, gave 0.06 g of 25 (61%) as a colorless oil. 25: uv  $\lambda_{max}$  255, 310nm; ir  $\nu_{max}$

(neat) 3270, 2960, 2100, 1700, 1600, 740 $\text{cm}^{-1}$ ;  $^1\text{H-nmr}$  (270MHz)  $\delta$  1.02 (9/5H, s,  $\text{CH}_3$ ) and 1.05 (6/5H, s,  $\text{CH}_3$ ) (this splitting of the signals of the methyl protons can be ascribed to hindered rotation about the amide group. At 55°C, the methyl protons appear as a sharp singlet at 1.05), 1.32 (9/5H, s,  $\text{CH}_3$ ) and 1.34 (6/5H, s,  $\text{CH}_3$ ) (1.31, s, at 55°C), 2.03-2.09 and 2.43-2.49 (2H, m,  $\text{CH}_2$ ), 2.17 (2/5H, s,  $\text{C}=\text{CH}$ ) and 2.18 (3/5H, s,  $\text{C}=\text{CH}$ )(2.16, s, at 55°C), 2.90 (9/5H, s,  $\text{NCH}_3$ ), 2.99 (6/5H, s,  $\text{NCH}_3$ ), 3.71 (9/5H, s,  $\text{OCH}_3$ ) and 3.78 (6/5H, s,  $\text{OCH}_3$ )(3.73, s, at 55°C), 2.90-2.99 and 3.81-4.02 (2H, m,  $\text{NCH}_2$ ), 5.51 (2/5H, s,  $\text{NCHN}$ ) and 5.60 (3/5H, s,  $\text{NCHN}$ )(5.55, s, at 55°C), 6.35 (1H, d-like, arom-H), 6.64 (1H, t-like, arom-H), 7.09-7.16 (2H, m, arom-H); *ms m/z* (%) 298 ( $\text{M}^+$ , 20), 231 (100), 216 (4), 171 (12), 144 (29); HRms calcd for  $\text{C}_{18}\text{H}_{22}\text{N}_2\text{O}_2$  298.1676. Found 298.1671 ( $\text{M}^+$ ).

**1-Methoxycarbonyl-3a-dimethylpropargyl-1,2,3,3a,8,8a-hexahydropyrrolo[2,3-b]indole 27**

Similar treatment of 9 (0.13 g, 0.25 mmol) with  $\text{Fe}(\text{NO}_3)_3 \cdot 9\text{H}_2\text{O}$  in ethanol (10 ml) under ice cooling, followed by work-up as described above, gave 0.07 g of 27 (93%) as a yellowish oil. 27: *uv*  $\lambda_{\text{max}}$  245, 300nm; *ir*  $\nu_{\text{max}}$  3380, 3280, 2970, 2050, 2000, 1689, 1600, 1200, 740 $\text{cm}^{-1}$ ;  $^1\text{H-nmr}$  (500MHz)  $\delta$  1.08(1H, s,  $\text{CH}_3$ ), 1.10 (2H, s,  $\text{CH}_3$ ), 1.33 (1H, s,  $\text{CH}_3$ ), 1.34 (2H, s,  $\text{CH}_3$ ), 2.19 (1/3H, s,  $\text{C}=\text{CH}$ ), 2.20 (2/3H, s,  $\text{C}=\text{CH}$ ), 2.13-2.18 and 2.56-2.66 (2H, m,  $\text{CH}_2$ ), 2.97-3.03 and 3.65-3.82 (2H, m,  $\text{NCH}_2$ ), 3.69 (2H, s,  $\text{OCH}_3$ ), 3.78 (H, s,  $\text{OCH}_3$ ), 4.63 (1/3H, s, NH, exchangeable), 5.09 (2/3H, s, NH, exchangeable), 5.42 (1/3H, s,  $\text{NCHN}$ ), 5.48 (2/3H, s,  $\text{NCHN}$ ), 6.57-6.58 (1H, m, arom-H), 6.71-6.75 (1H, m, arom-H), 7.08-7.15 (2H, m, arom-H); *ms m/z* (%) 284 ( $\text{M}^+$ , 21), 217 (100), 157 (14), 130 (19). These spectral data were identical with those obtained by our previous method.<sup>8)</sup>

**Reduction of 3-(1,1-dimethylpropargyl)indole 5 to 3-(1,1-dimethylallyl)indole 6**

5% Pd- $\text{CaCO}_3$  (4 mg) and quinoline (0.02 ml) were added to the flask charged with 3-(1,1-dimethylpropargyl)-indole 5 (66.7 mg, 0.36 mmol) and dry benzene (10 ml) under a nitrogen atmosphere. Hydrogen was introduced with vigorously stirring for 75 min, and the reaction mixture was filtered through celite. The filtrate was washed with 5% HCl,  $\text{H}_2\text{O}$ , and brine and dried over  $\text{Na}_2\text{SO}_4$ . After concentration by rotary evaporation, the residue was purified by flash chromatography ( $\text{AcOEt/n-Hex}=1/5$ ) to give 6 (67.8 mg) quantitatively as a pinkish oil. 6:<sup>16)</sup> *uv*  $\lambda_{\text{max}}$  223.5, 278, 283, 292nm; *ir*  $\nu_{\text{max}}$  (neat) 3400, 3060, 3040, 2950, 2910, 2850, 1630, 1610, 1450, 990, 740 $\text{cm}^{-1}$ ;  $^1\text{H-nmr}$  (60MHz)  $\delta$  1.52 (6H, s,  $\text{CH}_3$ ), 5.02 (1H, dd,  $J=1.5, 11\text{Hz}$ ,  $\text{C}=\text{CH}_a$ ), 5.15 (1H, dd,  $J=1.5, 17\text{Hz}$ ,  $\text{C}=\text{CH}_b$ ), 6.14 (1H, dd,  $J=11, 17\text{Hz}$ ,  $\text{CH}=\text{C}$ ), 6.85-7.35 (4H, m,  $\text{C}_2\text{-H}$  and arom-H), 7.64-7.70 (1H, m, arom-H), 7.78 (1H, bs, NH); *ms m/z* (%) 185 ( $\text{M}^+$ , 39), 170 (100), 158 (39), 155 (18), 143 (16), 115 (13); HRms calcd for  $\text{C}_{13}\text{H}_{15}\text{N}$  185.1204; Found 185.1204 ( $\text{M}^+$ ).

**ACKNOWLEDGEMENT**

This work was supported by a Grant-in-Aid for Scientific Research (62470134) from the Ministry of Education, Science and Culture, Japan. We thank Mrs. H. Seki, Miss R. Hara, Mrs. S. Yamada and

Mr. T. Kuramochi in the Analytical Center of our University for measurements of spectral data (nmr and ms) and microanalytical data. We are grateful to Professor P. Magnus, University of Texas at Austin, and Dr. H. Yamazaki, Rikken, for helpful suggestions concerning this work.

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14. The formation of ene-yne complex **26** was observed in all experiments. mp 31.5~33.5°C (sublimed), ir  $\nu_{\max}$  2090, 2040, 2010, 1630, 1450 $\text{cm}^{-1}$ ;  $^1\text{H-nmr}$  (270MHz)  $\delta$  2.07 (3H, s,  $\text{CH}_3$ ), 5.29 (1H, s,  $\text{C}=\text{CH}_2$ ), 5.39 (1H, s,  $\text{C}=\text{CH}_b$ ), 6.19 (1H, s,  $\text{C}=\text{CH}$ ).
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Received, 24th July, 1989