

A NEW SYNTHETIC APPROACH TO (+)-GALANTINIC ACID, DEGRADATION PRODUCT
FROM THE PEPTIDE ANTIBIOTIC GALANTIN I, VIA 4-AMINO-3-HYDROXYPYRANOSE

Shinzo Kano,* Tsutomu Yokomatsu, and Shiroshi Shibuya

Tokyo College of Pharmacy, 1432-1 Horinouchi, Hachioji, Tokyo 192-03, Japan

Abstract — 5-Phenylthioxazolidin-2-ones, derived from L-serine, were subjected to photo-induced radical allylation to give the correspondg 4-substituted 4,5-trans-5-allyloxazolidin-2-ones. 4-Ethoxyethyl derivative (**13c**) was led to N-Boc galantinic acid methyl ester via N-Boc 4-amino-3-hydroxypyranose.

The presence of 2-amino alcohol in biologically active molecules such as amino sugars, antibiotics and peptides has raised the interest in the diastereoselective synthesis of these compounds.¹ Furthermore, 2-amino alcohols have been used as the dipeptide isostere in peptidomimetic chemistry.² In the previous paper,³ we demonstrated a photo-induced radical allylation at the 5-position of oxazolidin-2-ones to get 4-substituted 4,5-trans-5-allyloxazolidin-2-ones, protective form of threo 2-amino alcohols.

The method was extensively applied to a synthesis of (+)-galantinic acid,^{4,5} the degradation product of the peptide antibiotic galantin I.^{5,6} The results of our studies are described in this paper. At first, 4-substituted 5-phenylthioxazolidin-2-ones, used for photo-radical allylation reaction, were prepared by starting from L-serine and L-threonine as outlined in the Scheme. The alcohols (**2a,b**), obtained by reduction of **1a,b** (NaBH₄, MeOH), were treated with diphenyl disulfide in the presence of tri-*n*-butylphosphine⁷ followed by O-deprotection (EtOH, PPTS) of the resulting sulfides (**3a,b**) to give the corresponding sulfur-containing 2-amino alcohols (**4a**),⁸ mp 64-65°C, [α]_D +51.2° (c 0.7, MeOH) and **4b**, mp 83-84°C, [α]_D +22.7° (c 1.0, CHCl₃), respectively. Acetylation of **4a,b** gave **5a,b** chlorination of which with *N*-chlorosuccinimide and subsequent cyclization with SnCl₄ (-78°C, 20 min then rt 20 min in CH₂Cl₂) yielded the corresponding 4-substituted 4,5-trans-5-phenylthioxazolidin-2-ones (**6a**) in 84 % yield, mp 115-118°C, [α]_D -241.8° (c 1.0, CHCl₃) and **6b** in 86 % yield, mp 175-176°C, [α]_D -224.5° (c 1.0, CHCl₃), with high diastereoselectivity. Replacement of acetyl group of **6a** with ethoxyethyl group (i. Cs₂CO₃, MeOH ii. CH₂=CHOEt, PPTS) afforded **7** as an oil. The epimer at 4-(α)- position of **6b** was also easily prepared from **4b** as follows. Methane-sulfonylation of **4b**, followed by treatment with CsOAc in DMF afforded **8** which was led to **9** according

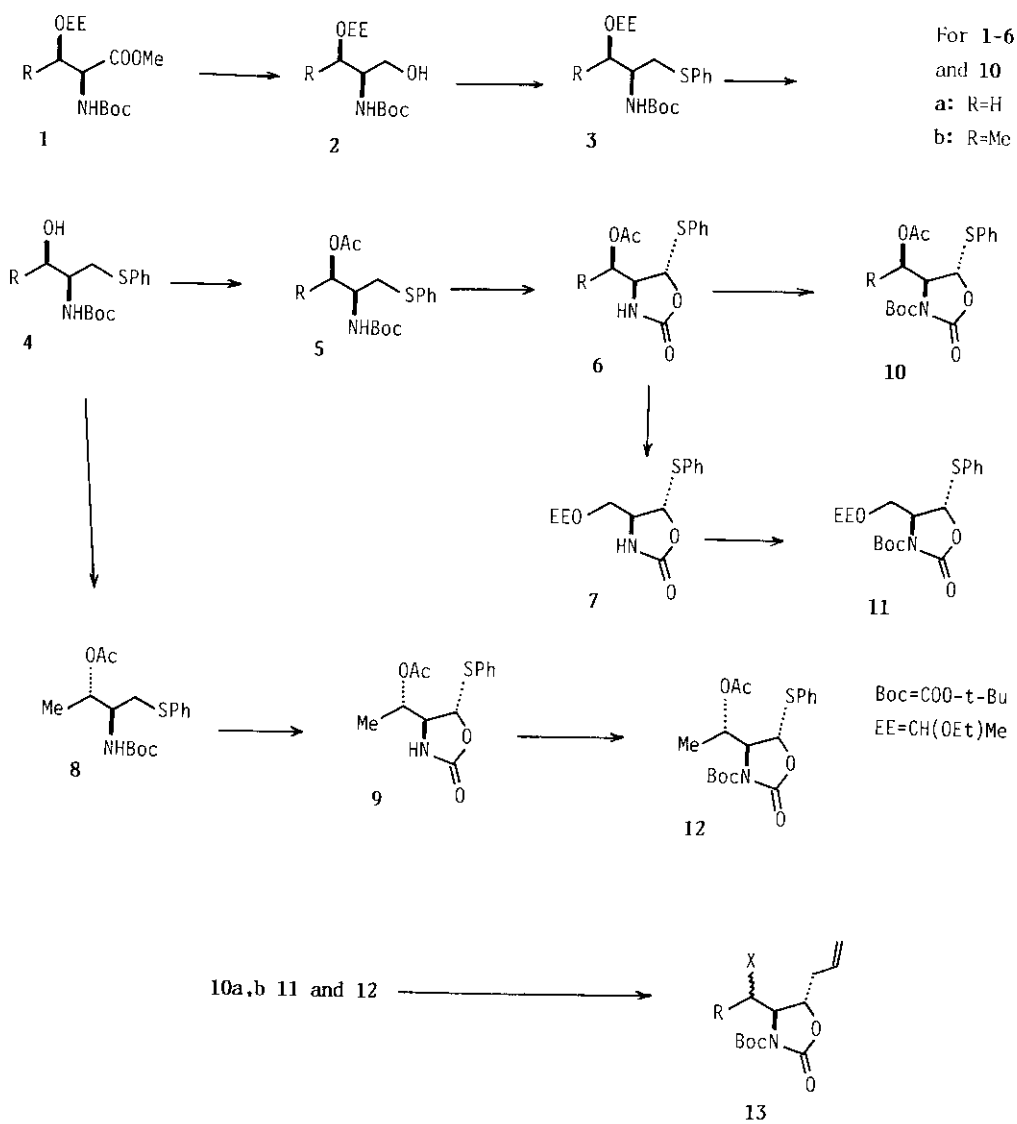
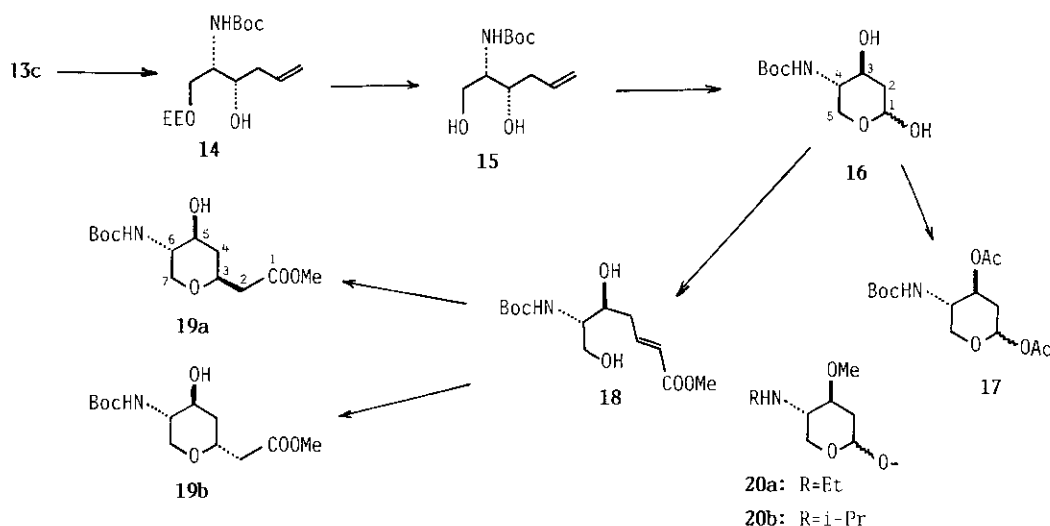


Table 1. Photo-radical Allylation Products (**13a-d**)

Compound	R	X	Yield (%)	mp (°C)	$[\alpha]_D^{25}$ (CHCl ₃)
13a	H	—OAc	80.5	oil	+5.4 (c 1.0)
13b	Me	—OAc	75.0	oil	-9.8 (c 1.0)
13c	H	—OEE	75.8	oil	+3.7 (c 0.7)
13d	MeOAc	81.0	68-71	+8.4 (c 1.0)



to the same conditions as in a synthesis of **6a,b**. *t*-Butoxycarbonylation of **6a,b**, **7** and **9** gave **10a,b**, **11** as an oil and **12**, mp 91-92°C, respectively. These compounds were subjected to photo-induced radical allylation [(*n*-Bu₃Sn)₂ (1 equiv.), *n*-Bu₃SnCH₂CH=CH₂ (4 equiv.) in toluene-MeCN (7:3, 0.5 M solution), *hν* 500W Hg lamp, 6 h] to give the corresponding 5-allyloxazolidin-2-ones (**13a-d**). The results were summarized in the Table 1. In their ¹H-nmr spectra, 4-H resonated at around δ 3.9-4.1 and 5-H at around δ 4.5 ppm.

Next, ring cleavage of **13a** and **13c** was examined to get *N*-Boc 2-amino-1,3-dihydroxy-5-hexene (**15**) which would be the key intermediate for preparation of 4-amino-3-hydroxypyranose. Upon ring cleavage of **13a** (Cs₂CO₃, MeOH), the yield for **15** was quite low. However, ring cleavage of **13c** gave **14** in 84 % yield. Subsequent *O*-deprotection (EtOH, PPTS) of **14** afforded **15**. The desired 4-amino-3-hydroxypyranose (**16**) from **15** was successfully achieved by ozonolysis procedure (O₃, MeOH, -78°C then Me₂S) in 76 % yield, mp 168-171°C (decomp.), [α]_D +7.14° (c 0.3, MeOH). The ratio for anomers was determined as 1:4 by conversion to diacetate (**17**), an oil, [α]_D +14.4° (c 0.7, MeOH), ¹H-nmr (CDCl₃ 400 MHz) δ 6.14 (0.2H, br t, J=3 Hz), 5.87 (0.8H, dd, J=3.59, 5.1 Hz). Wittig reaction of **16** with methyl (triphenylphosphoranylidene)acetate (MeCN, reflux) yielded **18**, cyclization of which (0.1 equiv. K₂CO₃, MeOH) yielded *N*-Boc galantinic acid methyl ester⁵ (**19a**, 43.4 % yield from **16**), mp 104-106°C, [α]_D -4.69° (c 0.9, CHCl₃) [lit.,⁴ mp 104.5-106°C, [α]_D -4.4° (c 1.2, CHCl₃)] and its C₃-epimer (**19b**, 34.2 % yield from **16**), an oil, [α]_D +20.5° (c 0.9, MeOH) [lit.,⁴ [α]_D +20.8° (c 1.5, MeOH)]. These were easily separated by column chromatography on silica gel. The spectral data of **19a,b** were identical with those in the literature.⁴ The method for a preparation of 4-amino-3-hydroxypyranose described here would be potentially useful for a synthesis of this type of amino sugars such as **20a** in calicheamicin⁹ and **20b** in esperamicin.¹⁰

REFERENCES AND NOTES

1. B. Rauge, J. A. Fehrentz, R. Guegan, Y. Chapleur, and B. Castro, Bull. Soc. Chim. France, 1983, 230. M. Fujita and T. Hiyama, J. Am. Chem. Soc., 1984, **106**, 4629; S. Kobayashi, T. Isobe, and M. Ohno, Tetrahedron Lett., 1984, **25**, 5079; M. Hirama, T. Shigemoto, Y. Yamazaki, and S. Ito, J. Am. Chem. Soc., 1985, **107**, 1797; M. Reetz, M. W. Drews, and A. Schmitz, Angew. Chem., Int. Ed. Engl., 1987, **26**, 1141; W. Roush and M. A. Adam, J. Org. Chem., 1985, **50**, 3752; S. Hanessian and J. Kloss, Tetrahedron Lett., 1985, **26**, 1261; T. Mukaiyama, Y. Goto, and S. Shoda, Chemistry Lett., 1985, 671; D. A. Claremon, P. K. Lumma, and B. T. Philips, J. Am. Chem. Soc., 1986, **108**, 8265; A. Bongini, G. Gardillo, M. Orena, G. Porzi, and S. Sandri, Tetrahedron, 1987, **43**, 4377; M. Hirama, T. Shigemoto, and S. Ito, J. Org. Chem., 1988, **53**, 3628; P. Herold, Helv. Chim. Acta, 1988, **71**, 354; R. M. Devant and H.-E. Radunz, Tetrahedron Lett., 1988, **29**, 2307; J. Jurczak, A. Golebiowski, and J. Raczko, J. Org. Chem., 1989, **54**, 2495, and references cited therein.
2. M. W. Hollady and D. H. Rich, Tetrahedron Lett., 1983, **24**, 4401; D. H. Rich, J. Med. Chem., 1985, **28**, 263; D. J. Kempf, E. De Lara, H. H. Stein, J. Cohen, and J. J. Plattner, J. Med. Chem., 1987, **30**, 1978; P. G. M. Wuts, S. R. Putt, and A. R. Ritter, J. Org. Chem., 1988, **53**, 4503; P. Herold, R. Duthaler, G. Rihs, and C. Angst, J. Org. Chem., 1989, **54**, 1178.
3. S. Kano, T. Yokomatsu, and S. Shibuya, J. Org. Chem., 1989, **54**, 513.
4. Y. Ohfuné and N. Kurokawa, Tetrahedron Lett., 1984, **25**, 1587.
5. Recently, the authors received informations concerning the structure revision of galantin I in which the substructure of galantinic acid is not correct. The structure depicted in **19a** was found to be an artifact produced during the chemical degradation studies of galantin I. (Private Communication from Dr. Ohfuné.⁴ Correct structure will be published by his group).
6. J. Shoji, R. Sakazaki, Y. Wakishima, K. Koizumi, M. Mayama, and S. Matsuura, J. Antibiotics, 1975, **28**, 122.
7. T. Nakagawa and T. Hata, Tetrahedron Lett., 1975, 1409.
8. All new compounds gave satisfactory elementary analyses and spectral data.
9. M. D. Lee, T. S. Dunne, M. M. Siegel, C. C. Chang, G. O. Morton, and D. B. Borders, J. Am. Chem. Soc., 1987, **109**, 3464; M. D. Lee, T. S. Dunne, C. C. Chang, G. A. Ellestad, M. M. Siegel, G. O. Morton, W. J. McGahren, and D. B. Borders, J. Am. Chem. Soc., 1987, **109**, 3466.
10. J. Golik, J. Clardy, G. Dubay, G. Groenwold, H. Kawaguchi, M. Konishi, B. Krishnan, H. Ohkuma, K. Saitoh, and T. W. Doyle, J. Am. Chem. Soc., 1987, **109**, 3461; J. Golik, G. Dubay, G. Groenwold, H. Kawaguchi, M. Konishi, B. Krishnan, H. Ohkuma, K. Saitoh, and T. W. Doyle, J. Am. Chem. Soc., 1987, **109**, 3462.

Received, 27th September, 1989