

SHORT SYNTHESIS OF (±)-CORYNANTHEIDOL AND
(±)-3-EPICORYNANTHEIDOL

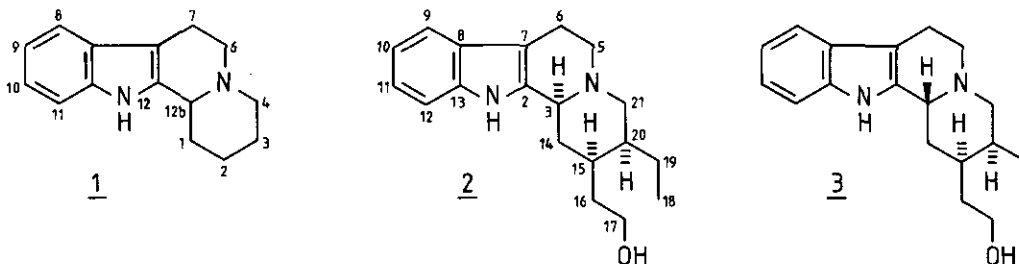
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Abstract - A short synthesis for (±)-corynantheidol and (±)-3-epicorynantheidol is described.

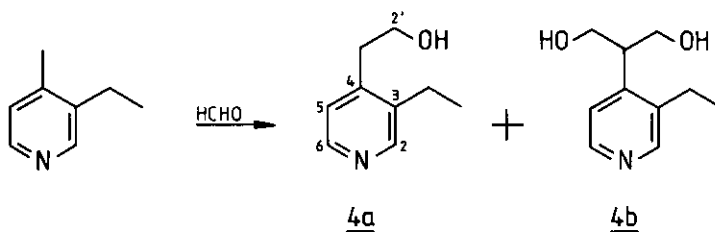
Recent reports from our laboratory have described a new general method,¹⁻⁵ which permits the preparation of 1-, 2- and 3-substituted 1,2,3,4,6,7,12,12b-octahydroindolo[2,3-a]quinolizine (1) derivatives possessing the C(12b)H-C(1)H, C(12b)H-C(2)H and C(12b)H-C(3)H relationship [corresponding to the C(3)H-C(14)H, C(3)H-C(15)H and C(3)H-C(20)H relationship, respectively, when the biogenetic numbering of indole alkaloids is used⁶] cis or trans at will.

Our method appeared to be ideally suited for a short synthesis of both (±)-corynantheidol (2) and (±)-3-epicorynantheidol (3) (biogenetic numbering⁶) of which the former is the racemic form of the known indole alkaloid (-)-corynantheidol found in *Mitragyna parvifolia* (Roxb.) Korth. (Rubiaceae).⁷ In the present paper we describe the results obtained.



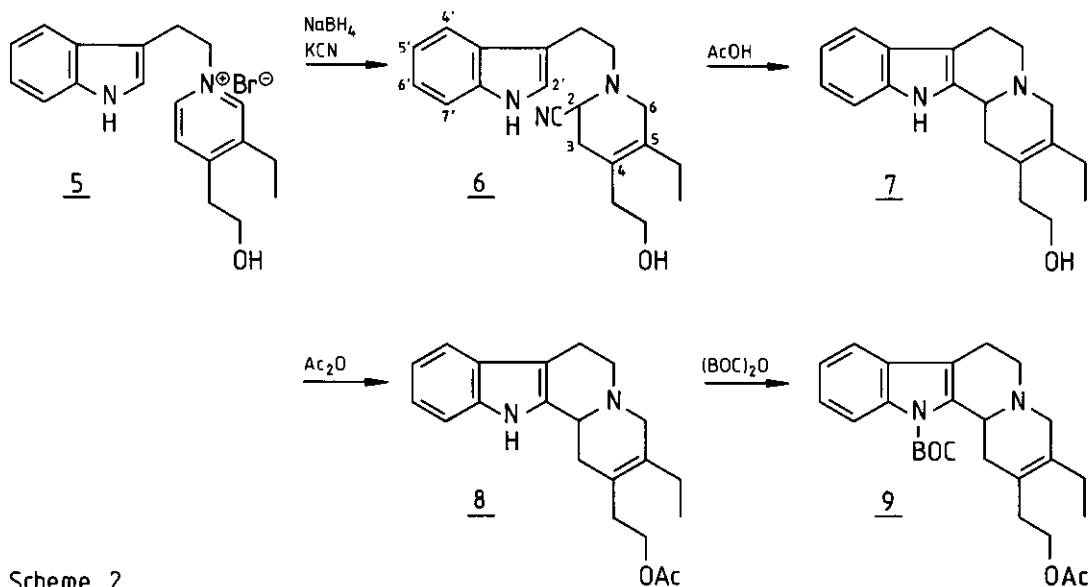
RESULTS AND DISCUSSION

Alkylation of 3-ethyl-4-(2'-hydroxyethyl)pyridine (4a) [prepared together with compound (4b) from 3-ethyl-4-methylpyridine⁸ and formaldehyde; Scheme 1] with tryptophyl bromide⁹ yielded the pyridinium salt (5).



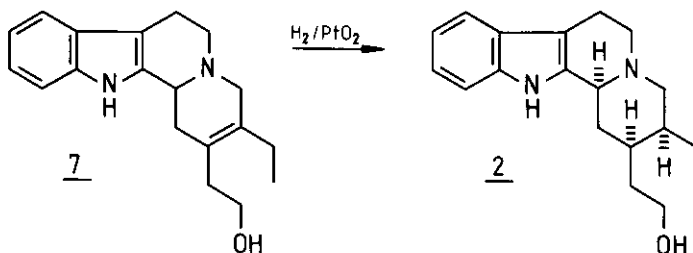
Scheme 1

The pyridinium salt (5) was transformed by NaBH_4 reduction and cyanide trapping¹⁰⁻¹³ to α -aminonitrile (6), which by AcOH treatment¹⁴⁻¹⁶ afforded compound (7). A part of compound (7) was acetylated to compound (8) which was then transformed with di-*t*-butyl dicarbonate [$(\text{BOC})_2\text{O}$] to the corresponding BOC-protected compound (9) (Scheme 2).^{13,17,18}



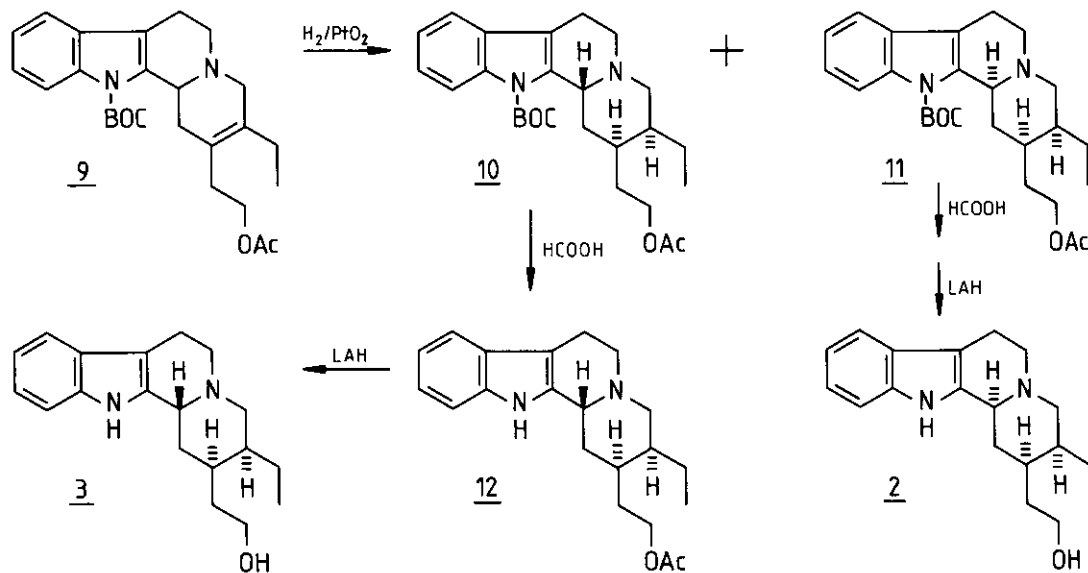
Scheme 2

Catalytic hydrogenation (PtO_2) of compound (7) led directly to (+)-corynantheidol (2) [C(3)H-C(15)H *cis*; C(3)H-C(20)H *cis*]. Thus a very short stereoselective synthesis of (+)-corynantheidol (2) was in hand (Scheme 3).



Scheme 3

Catalytic hydrogenation (PtO_2) of the BOC-protected compound (9) instead of compound (7) led to a 60/40 mixture of compound (10) [C(3)H-C(15)H trans; C(3)H-C(20)H trans] and compound (11) [C(3)H-C(15)H cis; C(3)H-C(20)H cis], which were separated by tlc. Acid-induced cleavage (HCOOH) of compound (10) afforded compound (12), which by LAH treatment yielded (\pm)-3-epicorynantheidol (3). Compound (11) was transformed by consecutive HCOOH and LAH treatments to (\pm)-corynantheidol (2) without the isolation of the acetate intermediate [cf. compound (12)] (Scheme 4¹⁹).



Scheme 4

^{13}C Nmr data of all the compounds formed are given in Figure 1. The proper shift assignment was confirmed by recording single frequency, off-resonance decoupled (sford) spectra.

Comparison of the chemical shifts found for compounds (2), (3), (7), (8), (9), (10), (11) and (12) with those given earlier,^{1,3,20} taking into account the conformational considerations relevant for indolo[2,3-*a*]quinolizidines, provides clear evidence of the stereostructures depicted in the formulae.

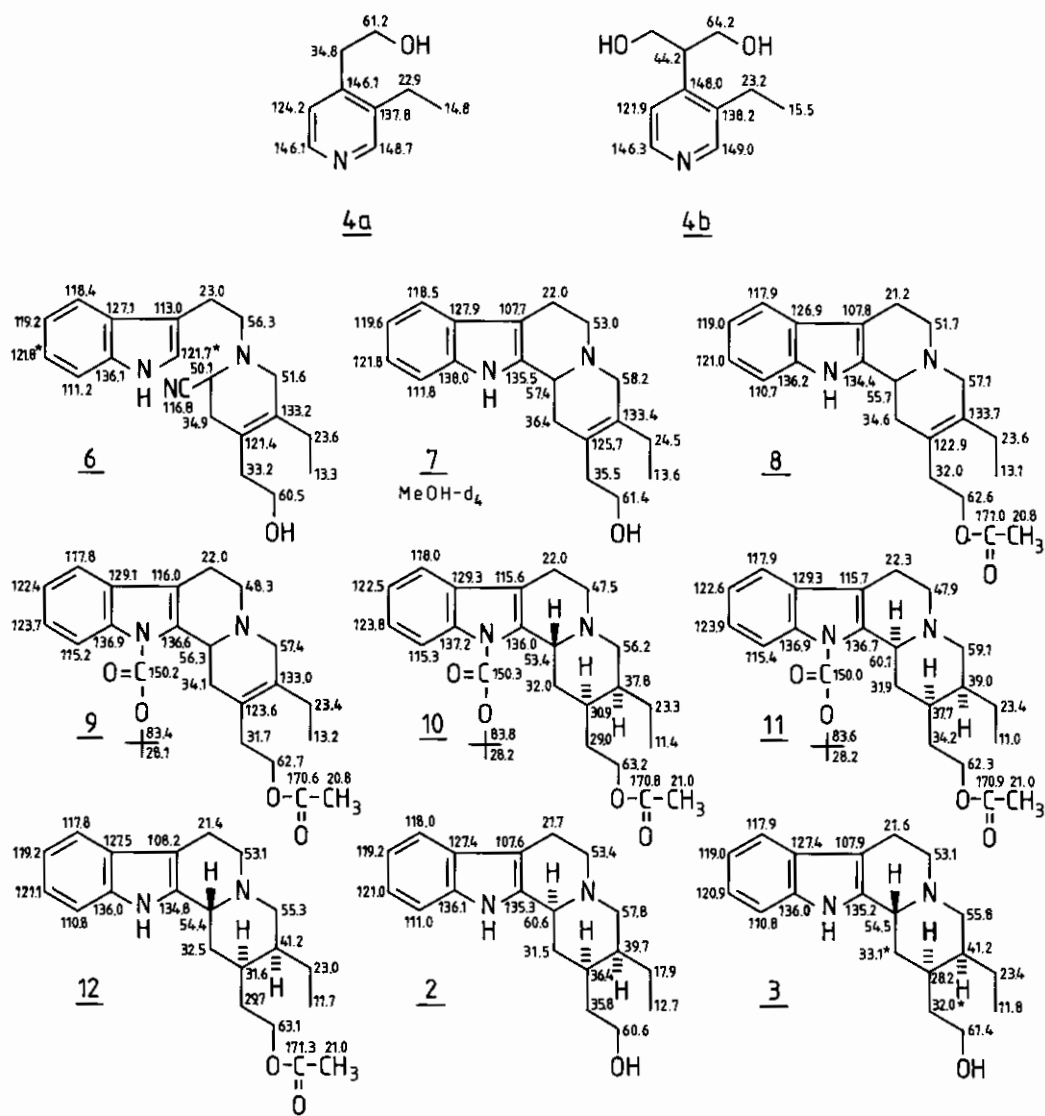


Figure 1

CONCLUSIONS

The results clearly demonstrate that our recently developed method¹⁻⁵ can successfully be applied to a short synthesis of both (\pm)-corynantheidol (2) and (\pm)-3-epicorynantheidol (3). Compared with some earlier syntheses²¹⁻²⁷ our method affords an easier access to these compounds.

EXPERIMENTAL

Ir spectra were recorded with a Perkin-Elmer 700 spectrophotometer. Absorption bands are expressed in reciprocal centimetres (cm^{-1}) using polystyrene calibration. ^1H and ^{13}C nmr spectra were recorded in CDCl_3 (if not otherwise stated) with a JEOL JNM-FX 60 spectrometer working at 59.80 MHz (^1H nmr) and 15.04 MHz (^{13}C nmr). Chemical shift data are given in ppm downfield from TMS. Abbreviations s, d, t, m and br are used to designate singlet, doublet, triplet, multiplet and broad, respectively. For ^{13}C nmr data see Figure 1. Mass spectrometry was done on a JEOL DX 303/DA 5000 instrument.

Compounds (4a) and (4b)

Commercial 3-ethyl-4-methylpyridine⁸ (10.00 g, 82.64 mmol) and formaldehyde (35%, 15 ml) were refluxed for 60 h (Ar-atm). The crude product was purified by column chromatography (alumina, first CH_2Cl_2 , then increasing the polarity of the eluent with gradual MeOH addition). Three compounds were isolated: unreacted starting compound, monoalcohol (4a) and dialcohol (4b).

Compound (4a). Yield: 2.50 g (20%). Oil. Ir 3300 (OH), pmr 1.20 (3H, t, $J=7.0$ Hz, $-\text{CH}_3$), 2.67 (2H, q, $J=7.0$ Hz, $-\text{CH}_2-\text{CH}_3$), 2.89 (2H, t, $J=7.0$ Hz, $-\text{CH}_2-\text{CH}_2\text{OH}$), 3.87 (2H, t, $J=7.0$ Hz, $-\text{CH}_2\text{OH}$), 4.83 (1H, s, -OH), 7.14 (1H, d, $J=6.0$ Hz, H-5), 8.20 (1H, d, $J=6.0$ Hz, H-6), 8.23 (1H, s, H-2), m/z 151 (M^+), 132, 120, 118 (100%), 106; exact mass: 151.1010 (calcd for $\text{C}_9\text{H}_{13}\text{NO}$: 151.0997).

Compound (4b). Yield: 3.74 g (25%). mp 102-103°C (MeOH). Ir 3200 (OH), pmr 1.18 (3H, t, $J=7.0$ Hz, $-\text{CH}_3$), 2.67 (2H, q, $J=7.0$ Hz, $-\text{CH}_2-\text{CH}_3$), 3.88 (4H, 2xd, $J=6.0$ Hz, $-\text{CH}_2\text{OH}$), 4.71 (1H, s, -OH), 7.12 (1H, d, $J=5.0$ Hz, H-5), 8.12 (1H, d, $J=5.0$ Hz, H-6), 8.17 (1H, s, H-2), m/z 182 ($\text{M} + 1$ peak²⁸), 181 (M^+), 145, 133, 118 (100%), 106; exact mass: 181.1110 (calcd for $\text{C}_{10}\text{H}_{15}\text{NO}_2$: 181.1103).

Compound (5)

Alkylation of compound (4a) (1.21 g, 8.01 mmol) with tryptophyl bromide (1.80 g, 8.04 mmol) afforded salt (5). Yield: 2.79 g (93%).

Compound (6)

Hydrochloric acid (6N, 4 ml) was added dropwise to a cooled (0°C) stirred solution of KCN (3.02 g, 46.46 mmol) in H_2O (4 ml) and layered with Et_2O (20 ml). MeOH (7 ml) and the salt (5) (2.79 g, 7.44 mmol) were added, after which NaBH_4 (0.37 g, 9.78 mmol) was added during 0.5 h keeping the solution at 0°C. Stirring was continued for 4 h at room temperature. The ethereal layer was separated and the aqueous layer was extracted several times with ether. The combined organic layers were dried over Na_2SO_4 and evaporated to yield nitrile (6), which was used without purification in the next step. Yield: 2.35 g (98%). Amorphous material. Ir 3420 (NH), 3300 (OH), 2280 (CN), pmr 0.99 (3H, t, $J=7.5$ Hz, $-\text{CH}_3$), 6.94 (1H, d, $J=2.4$ Hz, H-2'), 7.19-7.66 (4H, m, H-4', 5', 6', 7'), 8.26 (1H, br s, NH), m/z 323 (M^+), 308, 296, 168, 144 (100%), 130; exact mass: 323.1984 (calcd for $\text{C}_{20}\text{H}_{25}\text{N}_3\text{O}$: 323.1998).

Compound (7)

Compound (6) (2.17 g, 6.72 mmol) in 50% AcOH (200 ml) was stirred (at room temperature) for 68 h. After evaporation and neutralization (2N Na₂CO₃) the solution was extracted with CH₂Cl₂. The combined organic extracts were dried over Na₂SO₄. The product was purified by column chromatography (alumina, first CH₂Cl₂, then increasing the polarity of the eluent with gradual MeOH addition) to yield compound (7). Yield: 1.18 g (59%). mp 223-224°C (MeOH). Ir 3410 (NH), 3210 (OH), pmr 0.97 (3H, t, J=7.5 Hz, -CH₃), 7.04-7.53 (4H, m, H-9, 10, 11, 12), 8.64 (1H, m, NH), m/z 296 (M⁺), 295, 170 (100%), 169; exact mass: 296.1906 (calcd for C₁₉H₂₄N₂O: 296.1889).

(±)-Corynantheidol (2)

Catalytic hydrogenation (PtO₂, 24 h) of compound (7) (180 mg, 0.61 mmol) in MeOH afforded crude compound (2), which was purified by tlc (silica gel, CH₂Cl₂/MeOH; 95:5). Yield: 100 mg (55%) (after recycling the unreacted starting material). mp 163-165°C (CH₂Cl₂) (lit. 158-162°C;²¹ 158-160°C;²² 157-159°C;²³ 160-161°C;²⁵ 162-164°C²⁶). Ir 3430 (NH), 3280 (OH), 2830 and 2780 (Bohlmann bands), pmr 0.88 (3H, t, J=7.0 Hz, -CH₃), 6.96-7.56 (4H, m, H-9, 10, 11, 12), 8.54 (1H, br s, NH), m/z 298 (M⁺), 297 (100%), 170, 169; exact mass: 298.2061 (calcd for C₁₉H₂₆N₂O: 298.2041).

Compound (8)

Compound (7) (0.60 g, 2.03 mmol), Ac₂O (6.2 ml) and two drops of pyridine were stirred (room temperature, Ar-atm) for 20 h. The solution was poured into ice water, neutralized with aq. NH₄OH and extracted with CH₂Cl₂. The combined extracts were dried over Na₂SO₄ to afford pure compound (8). Yield: 0.64 g (93%). Amorphous material. Ir 3400 (NH), 1740 (C=O), pmr 1.00 (3H, t, J=7.0 Hz, -CH₃), 2.00 (3H, s, AcO-), 7.02-7.55 (4H, m, H-9, 10, 11, 12), 8.40 (1H, br s, NH), m/z 338 (M⁺), 170 (100%), 169; exact mass: 338.1981 (calcd for C₂₁H₂₆N₂O₂: 338.1990).

Compound (9)

To compound (8) (310 mg, 0.92 mmol) in dry CH₂Cl₂ (2 ml) were added 4-dimethylaminopyridine (DMAP) (12 mg, 0.1 equiv.) and di-*t*-butyl dicarbonate [(BOC)₂O] (242 mg, 1.2 equiv.) with stirring (room temperature, Ar-atm). After 2 h the mixture was evaporated and purified by column chromatography (silica gel, CH₂Cl₂/MeOH/Et₃N; 99:0.75:0.25) to afford pure compound (9). Yield: 361 mg (90%). Viscous oil. Ir 1740 (2 x C=O), pmr 1.03 (3H, t, J=7.0 Hz, -CH₃), 1.66 (9H, s, -C(CH₃)₃), 2.02 (3H, s, AcO-), 4.10 (1H, t, J=6.0 Hz, H-3), 7.14-7.42 (3H, m, H-9, 10, 11), 8.06 (1H, m, H-12), m/z 438 (M⁺), 383, 382, 214 (100%), 170, 169; exact mass: 438.2516 (calcd for C₂₆H₃₄N₂O₄: 438.2519).

Compounds (10) and (11)

Catalytic hydrogenation (PtO₂, 20 h) of compound (9) (344 mg, 0.79 mmol) in MeOH afforded a mixture of compounds [(10) and (11) (60:40)] and unreacted compound (9), which was recycled. Compounds (10) and (11) were separated by tlc (silica gel,

CH₂Cl₂/MeOH; 95:5).

Compound (10). Yield: 198 mg (57%). Amorphous material. Ir 1730 (2 x C=O), pmr 0.93 (3H, t, J=6.0 Hz, -CH₃), 1.68 (9H, s, -C(CH₃)₃), 2.05 (3H, s, AcO-), 4.35 (1H, m, H-3), 7.14-7.50 (3H, m, H-9, 10, 11), 7.98 (1H, m, H-12), m/z 440 (M⁺), 384, 383 (100%), 339; exact mass: 440.2663 (calcd for C₂₆H₃₆N₂O₄: 440.2675).

Compound (11). Yield: 132 mg (38%). Amorphous material. Ir 1730 (2 x C=O), pmr 0.98 (3H, t, J=6.0 Hz, -CH₃), 1.67 (9H, s, -C(CH₃)₃), 2.04 (3H, s, AcO-), 4.09 (1H, m, H-3), 7.14-7.43 (3H, m, H-9, 10, 11), 8.07 (1H, m, H-12), m/z 440 (M⁺), 384, 383 (100%), 339; exact mass: 440.2666 (calcd for C₂₆H₃₆N₂O₄: 440.2675).

Compound (12)

Compound (10) (92 mg, 0.21 mmol) was stirred in HCOOH (2 ml) for 70 h (room temperature, Ar-atm). After evaporation and neutralization (10% Na₂CO₃) the solution was extracted with CH₂Cl₂. The combined extracts were dried over Na₂SO₄ and evaporated to yield compound (12). Yield: 57 mg (80%). Amorphous material. Ir 3400 (NH), 1730 (C=O), pmr 0.93 (3H, t, J=5.0 Hz, -CH₃), 2.06 (3H, s, AcO-), 4.20 (1H, m, H-3), 7.05-7.53 (4H, m, H-9, 10, 11, 12), 8.28 (1H, m, NH), m/z 340 (M⁺, 100%), 339, 170, 169; exact mass: 340.2153 (calcd for C₂₁H₂₈N₂O₂: 340.2151).

(±)-3-Epicorynantheidol (3)

LAH treatment of compound (12) (57 mg, 0.17 mmol) in dry THF for 2.5 h (room temperature, Ar-atm) afforded crude product (3), which was purified by tlc (silica gel, CH₂Cl₂/MeOH; 90:10). Yield: 25 mg (50%). mp 189-191°C (CH₂Cl₂) (lit. 191-192°C;^{21,22} 192-194°C²³). Ir 3350-3200 (NH and OH), pmr 0.90 (3H, t, J=5.0 Hz, -CH₃), 3.74 (1H, m, H-3), 7.01-7.42 (4H, m, H-9, 10, 11, 12), 8.49 (1H, m, NH), m/z 298 (M⁺), 297 (100%), 170, 169; exact mass: 298.2019 (calcd for C₁₉H₂₆N₂O: 298.2041).

(±)-Corynantheidol (2)

Consecutive HCOOH and LAH treatments (vide supra) of compound (11) (60 mg, 0.14 mmol) led to crude compound (2), which was purified by tlc (silica gel, CH₂Cl₂/MeOH; 90:10). Yield: 18 mg (45%). mp 163-165°C (CH₂Cl₂) (lit. 158-162°C;²¹ 158-160°C;²² 157-159°C;²³ 160-161°C;²⁵ 162-164°C²⁶).

Spectral data were identical with those described above.

REFERENCES AND NOTES

1. M. Lounasmaa and R. Jokela, Tetrahedron, 1989, **45**, 3975.
2. M. Lounasmaa and R. Jokela, Tetrahedron, 1989, **45**, 7449.
3. M. Lounasmaa, R. Jokela, B. Tirkkonen, and T. Tamminen, Tetrahedron, 1989, **45**, 7615.
4. M. Lounasmaa and R. Jokela, Tetrahedron, 1990, **46**, 615.
5. M. Lounasmaa, R. Jokela, P. Mäkimattila, and B. Tirkkonen, Tetrahedron, 1990, **46**, 2633.

6. J. Le Men and W. Taylor, Experientia, 1965, 21, 508.²⁹
7. E. J. Shellard and P. J. Houghton, Planta Medica, 1973, 24, 13. See also, S. R. Hemingway and J. D. Phillipson, "Indole and Biogenetically Related Alkaloids", eds. J. D. Phillipson, and M. H. Zenk, Academic Press, London, 1980, p. 74.
8. Fluka, Compound 04451.
9. T. Hoshino and K. Shimodaira, Liebigs Ann. Chem., 1935, 520, 19.
10. E. M. Fry, J. Org. Chem., 1964, 29, 1647.
11. E. M. Fry and J. A. Beisler, J. Org. Chem., 1970, 35, 2809.
12. M. Lounasmaa, "Studies in Natural Products Chemistry", ed. Atta-ur-Rahman, Vol. 1, Stereoselective Synthesis (Part A), Elsevier, Amsterdam, 1988, pp. 89-122.
13. T. Tamminen, R. Jokela, B. Tirkkonen, and M. Lounasmaa, Tetrahedron, 1989, 45, 2683.
14. R. Jokela, S. Schüller, and M. Lounasmaa, Heterocycles, 1985, 23, 1751.
15. M. Lounasmaa and R. Jokela, Heterocycles, 1986, 24, 1663.
16. R. Jokela, E. Karvinen, A. Tolvanen, and M. Lounasmaa, Tetrahedron, 1988, 44, 2367.
17. D. S. Grierson, M. Harris, and H.-P. Husson, Tetrahedron, 1983, 39, 3683.
18. M. Lounasmaa, E. Karvinen, A. Koskinen, and R. Jokela, Tetrahedron, 1987, 43, 2135.
19. Note that Scheme 4 is drawn in such a way that optical antipodes of compound (9) are involved in the formation of compounds (10) and (11).
20. M. Lounasmaa, R. Jokela, and T. Tamminen, Heterocycles, 1985, 23, 1367. See also, Ref. 13, Note 15 and Ref. 16, Note 15.
21. E. Wenkert, K. G. Dave, and F. Haglid, J. Am. Chem. Soc., 1965, 87, 5461.
22. E. Wenkert, K. G. Dave, R. G. Lewis, and P. W. Sprague, J. Am. Chem. Soc., 1967, 89, 6741.
23. S. Takano, K. Masuda, and K. Ogasawara, J. Chem. Soc., Chem. Comm., 1980, 887.
24. S. Takano, K. Shibuya, M. Takahashi, S. Hatakeyama, and K. Ogasawara, Heterocycles, 1981, 16, 1125.
25. T. Kametani, N. Kanaya, and T. Honda, Heterocycles, 1981, 16, 1937.
26. T. Imanishi, M. Inoue, Y. Wada, and M. Hanaoka, Chem. Pharm. Bull., 1983, 31, 1551.
27. In references 21 and 22 (±)-3-epicorynantheidol (3) is called dl-3-isocorynantheidol.
28. G. Spiteller, "Massenspektrometrische Strukturanalyse organischer Verbindungen", Verlag Chemie, Weinheim, 1966, pp. 54-55.
29. Biogenetic numbering⁶ is used in the present paper for compounds (2), (3), (7), (8), (9), (10), (11) and (12).

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