

THE SYNTHESIS OF THIENOTRIAZOLOTHIAZEPINES

Hitoshi Nagaoka, Hiromu Hara, and Toshiyasu Mase*

Central Research Laboratories, Yamanouchi Pharmaceutical Co. Ltd.,

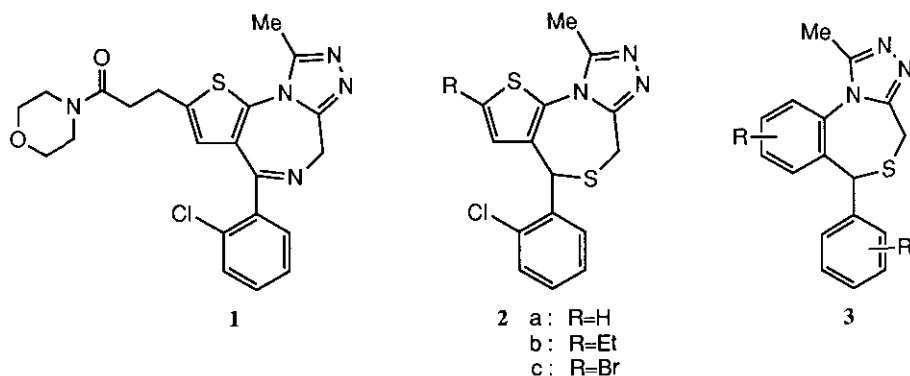
21 Miyukigaoka, Tsukuba, Ibaraki 305, Japan

Abstract—Some derivatives of thienotriazolothiazepine, a novel heteroazepine, were synthesized. These compounds showed anti-PAF activity.

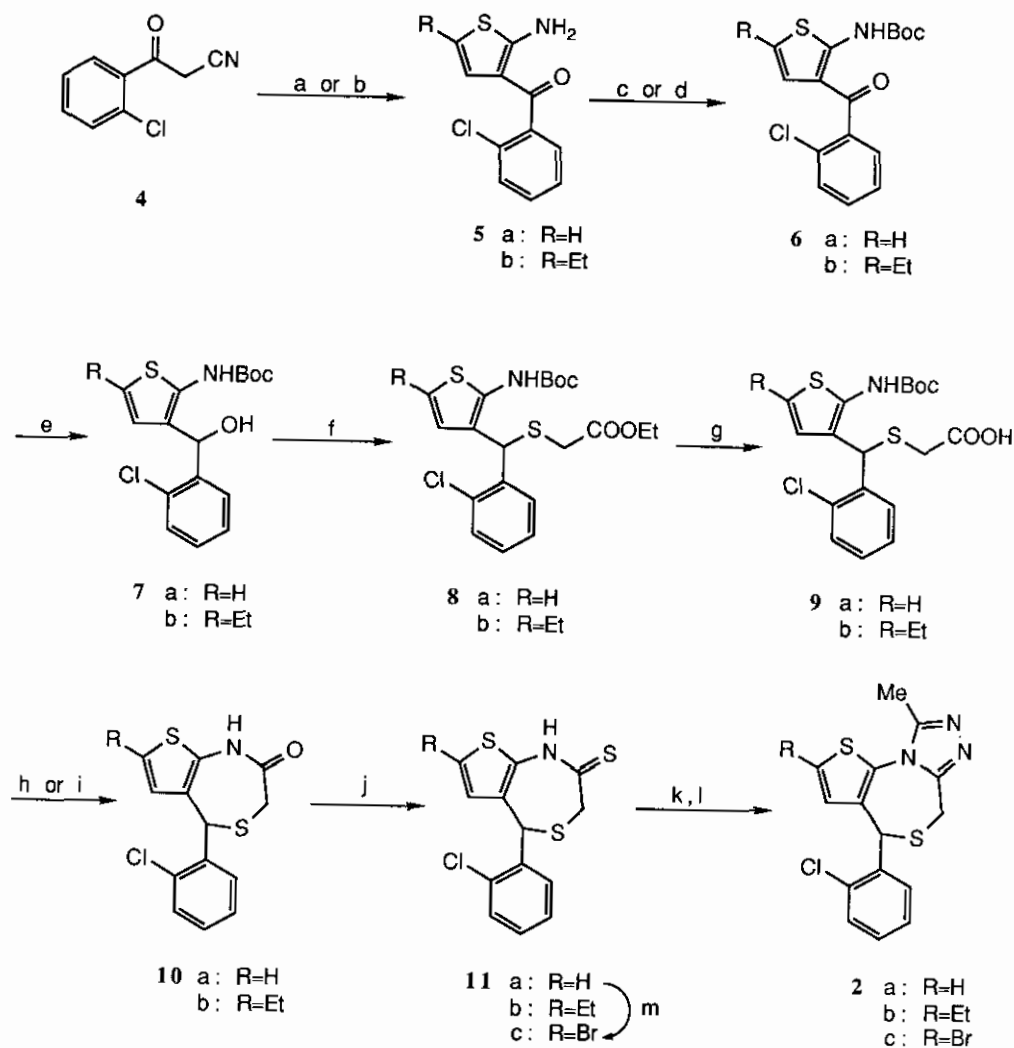
In 1986, WEB-2086 (**1**), which is characterized by its thienotriazolodiazepine skeltone, was reported to exhibit anti-PAF (platelet activating factor) activity.¹ Following this report, a number of thienotriazolodiazepine derivatives have been synthesized and tested for their anti-PAF activity.

In a search for potent and orally active PAF antagonists, we were interested in this structure and planned to synthesize its structurally related analogue, thienotriazolothiazepines (**2**). While some benzotriazolothiazepines (**3**) have been reported,² there have, to the best of our knowledge, been no report on the synthesis of thienotriazolothiazepines. Now, we report the first synthesis of thienotriazolothiazepines.

Synthetic route is shown in Scheme I. First, the α -cyanoketone (**4**)³ was treated with 1,4-dithiane-2,5-diol and Et₃N⁴ or 1-butanal, sulfur and Et₃N⁵ to give the 2-aminothiophene (**5a**, 49%) or its ethyl analogue (**5b**, 57%), respectively. Protection of the amino group of **5** with *t*-butoxycarbonyl (Boc) group (**6a**: 63%, **6b**: 64%) followed by reduction (NaBH₄, DMF) furnished the alcohols (**7a**: 88%, **7b**: 76%).



Scheme I



- a) 1,4-dithiane-2,5-diol, Et₃N, dioxane, b) 1-butanal, sulfur, Et₃N,
 c) Boc₂O, 10%NaOH aq., THF, d) NaH, Boc₂O, DMF, e) NaBH₄, DMF,
 f) PPh₃, DEAD, ethyl thioglycolate, benzene, g) 10%NaOH aq., MeOH,
 h) (COCl)₂, DMF, CH₂Cl₂, i) (COCl)₂, DMF, CH₂Cl₂ then TFA,
 j) Lawesson's reagent, toluene, k) H₂NNH₂·H₂O, THF, l) MeC(OMe)₃,
 m) Br₂, pyridine

The alcohols (7) were treated under Mitsunobu conditions (Ph_3P , DEAD, ethyl thioglycolate, benzene) to provide the sulfides⁶ (**8a**: 35%, **8b**: 68%), which were hydrolyzed to give the carboxylic acids (**9a**: 97%, **9b**: 92%).

Now that we obtained the desired precursor for the thiazepine ring, the stage was set for the cyclization. Cyclization of **9** to the thiazepine (**10**) was effected by treatment with oxalyl chloride (**10a**: 14%) or with oxalyl chloride and trifluoroacetic acid (**10b**: 67%).

The construction of the triazole ring was readily performed under the conventional conditions. Treatment of **10** with Lawesson's reagent⁷ gave the thioamide (**11a**: 97%, **11b**: 77%). At last, **11** was converted into the thienotriazolothiazepine (**2a**: 37%, **2b**: 47%)⁸ by the successive treatment with hydrazine monohydrate and trimethyl orthoacetate. In the same manner, the bromide (**2c**)⁸ was obtained in 57% yield from **11c**, which was prepared from **11a** by bromination (Br_2 , pyridine; 95%).

Thienotriazolothiazepines (**2**) were tested for their activity in inhibiting rabbit platelet aggregation induced by PAF.⁹ All compounds synthesized here exhibited anti-PAF activity (Table 1).

Thus we achieved a synthesis of a novel heteroazepine, thienotriazolothiazepines, which showed anti-PAF activity.

ACKNOWLEDGEMENTS

We are indebted to Drs. N. Inukai and K. Murase for their encouragements. We are also grateful to the members of the Physico-Analytical Center in their laboratories for the measurements of nmr and mass spectra. Finally, we would like to express our thanks to the members of the Department of Pharmacology in their laboratories for the measurements of anti-PAF activity.

Table I. Inhibition of PAF(10^{-8}M)-induced Platelet Aggregation in Rabbit P.R.P.

Compound	$\text{IC}_{50}(10^{-6}\text{M})$
2a	21
2b	(42%)*
2c	26

* Inhibition of platelet aggregation at the cocentration of 10^{-4}M

REFERENCES AND NOTES

- 1) J. Casals-Stenzel, K. H. Weber, G. Walther, and A. Harreus, Pattenblatt DE, 34 35 974 A1 (1986) (Chem. Abstr., 1986, **105**, P 120760f).
- 2) K. Hirai, S. Matsutani, T. Ishida, and I. Makino, Ger. Offen., 2,947,773 (1980) (Chem. Abstr., 1980, **93**, P 204705p).
- 3) F. N. Stepanov and N.S. Vul'fson, Org. Poluprod. i Krasiteli, Nauch.-Isseldovatel. Inst. Org. Poluprod. i Krasitelei im. K. E. Voroshilova, Sbornik Statei, 1959, 222 (Chem. Abstr., 1961, **55**, 18747f).
- 4) O. Hromatka, D. Bintder, and P. Stanetty, Monatsh. Chem., 1973, **104**, 709.
- 5) K. H. Weber, G. Walther, A. Harreus, J. Casals-Stenzel, G. Muacevic, and W. Troeger, Ger. Offen. DE, 3,502,392 (1986) (Chem. Abstr., 1987, **106**, P 156502h).
- 6) Bromination (Br₂, Ph₃P or NBS, Ph₃P) or mesylation (MsCl, Et₃N) of 4a failed, presumably because of the lability of the product.
- 7) S. Raucher and P. Klein, Tetrahedron Lett., 1980, **21**, 4061.
- 8) **2a**, ¹H Nmr (100 MHz, CDCl₃, TMS)δ: 2.43 (3H, s), 3.83 (1H, d, J=14Hz), 3.95 (1H, d, J=14Hz), 5.73 (1H, s), 6.35 (1H, d, J=6Hz), 6.76 (1H, d, J=6Hz), 7.17-7.42 (3H, m), 7.55-7.67 (1H, m); ms m/z: 333 (M⁺).
- 2b**; ¹H Nmr (100MHz, CDCl₃, TMS)δ: 1.30 (3H, t, J=8Hz), 2.39 (3H, s), 2.80 (2H, q, J=8Hz), 3.94 (1H, d, J=14Hz), 3.98 (1H, d, J=14Hz), 5.64 (1H, s), 6.39 (1H, s), 7.19-7.40 (3H, m), 7.48-7.61 (1H, s); ms m/z: 361 (M⁺).
- 2c**; ¹H Nmr (100MHz, CDCl₃, TMS)δ: 2.39 (3H, s), 3.90 (1H, d, J=20Hz), 3.93 (1H, d, J=20Hz), 5.69 (1H, s), 6.74 (1H, s), 7.19-7.36 (3H, m), 7.48-7.64 (1H, m); ms m/z: 413(M⁺).
- 9) G. V. R. Born and M. J. Cross, J. Physiol., 1963, **168**, 178.

Received, 14th May, 1990