## **BIOSYNTHESIS OF SOME CURVULARIN-TYPE METABOLITES BY A HYBRID STRAIN ME** 0005 **DERIVED FROM** *PENZCILLIUM CITREO-VIRIDE* **B. IF0** 6200 **AND** 4692

Sheng Lai, Yoshikazu Shizuri, Shosuke Yamamura,' Kazuaki Kawai,+ Masatake Niwa,<sup>+</sup> and Hideyuki Furukawa<sup>+</sup>

Department of Chemistry, Faculty of Science and Technology, Keio University, Hiyoshi, Yokohama 223, Japan +Faculty of Pharmacy, Meijo University, Tenpaku-ku, Nagoya 468, Japan

Abstract  $-$  <sup>13</sup>C Nmr analyses were carried out to deduce the labeling patterns of the curvularin-type metabolites of the hybrid strain ME 0005 derived from  $[1,2^{-13}C_2]$ acetate, indicating that eight acetate molecules are incorporated into these metabolites. And citreofuran, 12-oxocurvularin, and 11ß-hydroxy-12-oxocurvularin are presumably formed from curvularin by enzymatic oxygenation. In addition, the structure of **Ilp-hydroxy-12-oxocurvularin** was also supported by the present experiments.

From view points of physiological activity and biogenesis, both curvularin (1) and dehydrocurvularin (2) are quite attractive because of remarkable activities against sea urchin embryo cells: These macrolides induce barrellike spindles resulting in inhibition of cell proliferation, as reported by Kobayashi **et al.** ' Recently, furthemore, biosynthetic study on dehydrocurvularin (2) was carried out by Vederas *et al.* <sup>2</sup> in detail, indicating that 2 is a typical polyketide with a head-to-tail arrangement of eight acetate units. Their further experiments have supported the hypothesis that both **3** and 4 are enzyme-bound intermediates during the assembly of dehydrocurvularin (2)? However, biogenetic relationship between curvularin (1) and dehydrocurvularin **(2)** is still uncertain, because a possibility that the former is derived from dehydrocurvularin (2) by enzymatic reducrion seems not to he always ruled out. We describe herein the biosynthesis of citreofuran **(5),** 12 oxocurvularin (6), and  $11\beta$ -hydroxy-12-oxocurvularin (7) using  $\lceil 1,2^{-13}C_2 \rceil$  acetate.

According to essentially the same procedure as described in the previous papers, $<sup>4</sup>$  the EtOAc extract was</sup> chromatographed on silica gel using a gradient solvent of MeOH-CHCl3. Further separation and purification by recrystallization or repeated preparative tlc led to the isolation of labeled citreofuran (5,0.16% ), I?-~)xocurvolarin (6,0.27% ), and **11P-hydmxy-12-oxocurvularin** (7,0.18% )in addition to labeled cuwularin (I, 3.3% )>.6dehydrocurvularin (2, 16% **),7** cis-dehydrocurvularin (0.12% **),4** and 11-hydroxycurvularins  $(2.3\%$ ).<sup>4</sup>

In the <sup>13</sup>C nmr spectra of citreofuran (5), 12-oxocurvularin (6), and 11 $\beta$ -hydroxy-12-oxocurvularin (7), <sup>13</sup>C-<sup>13</sup>C coupled signals due to doubly enriched carbons from [1,2-<sup>13</sup>C<sub>2</sub>] acetate were observed and summarized as shown in Table 1.



The results show that citreofuran  $(5)$ ,  $12$ -oxocurvularin  $(6)$ , and  $11\beta$ -hydroxy-12-oxocurvularin  $(7)$  must be octnketides with a head-to-tail linkage of the eight acetate units, in accordance with the labeling patterns of curvularin<sup>6</sup> and dehydrocurvularin.<sup>2, 3</sup> In these metabolites, furthermore, the carbon atom ( $C_{12}$ ) bearing the oxygen function is originally derived from the methyl group of acetate molecule, clearly indicating that they are not formed in such a  $\beta$ -oxidation manner as seen in dehydrocurvularin (2)<sup>2,3</sup> but directly formed by enzymatic oxygenation from such a less oxidized precursor as cumularin **(1)** which has been proved to be biosynthesized from **4.1** 

<sup>I</sup>**ID-Hydroxy-12-oxocurvularin** (7) was obtained this time as colorless needles (mp 67-69 OC) by recrystallization from CHCl<sub>3</sub>. The <sup>13</sup>C nmr data [  $\delta$  197.8 ( J= 40.1 Hz ), 43.2 ( J= 40.1 Hz); and  $\delta$  82.7 ( J= 40.8 Hz ), 207.3 (J= 40.8 Hz ) ] indicate that the secondary hydroxyl group is adjacent to the carbonyl group at  $C_{12}$ -position, and also support the structure (7) which was previously proposed based on the <sup>1</sup>H nmr spectral  $data$ , particularly J values, coupled with molecular mechanics calculations.<sup>4</sup>

## **EXPERIMENTAL**

Melting points were determined on a Mitamura Riken apparatus and uncorrected. Optical rotations were measired with a JASCO DIP-360 polarimeter. Infrared spectra were recorded on a JASCO A-202 spectrophotometer. <sup>1</sup>H Nmr and <sup>13</sup>C nmr spectra were recorded on a JEOL JNM-GX 400 NMR spectrometer in methanol-d<sub>4</sub>.

Incubation Polished rice (300 g) in deionized water (800 ml) was cooked using an electric cooker (99 °C, 20 min), then transfered into an Erlenmyer flask (5 1) and pasteurized at 120  $\degree$ C at 2 atm. The pasteurized rice was, after addition of sodium  $[1,2^{-13}C_2]$  acetate (1 g), successively inoculated with a suspension of mycelium of the hybrid strain ME 0005 in sterilized water, incubated stationarily at 25 °C for 23 days, and then extracted with acetone.

308

	$\mathbf{1}$		$2***$		5		6		7	
Carbon	δ	J	δ	J	δ	J	δ	J	δ	J
1	170.8	57.0	172.9	58.6	1733	59.4	172.9	57.3	175.8	60.1
$\overline{c}$	38.5	57.0	42.6	58.6	41.6	59.4	40.2	57.3	38.8	60.1
3	135.2	61.9	137.0	62.2	138.9	62.2	137.6	61.9	136.2	64.0
4	110.2	61.9	112.4	62.2	111.3	62.2	111.9	61.9	113.7	64.0
5	159.1	67.5	162.2	67.1	158.7	68.2	161.7	67.5	166.3	67.1
6	100.7	67.5	102.9	67.1	101.6	68.2	102.9	67.5	97.8	67.1
7	157.4	68.9	162.4	65.3	156.5	70.3	159.1	68.6	169.2	57.7
8	118.8	68.9	117.4	65.3	111.0	70.3	121.2	68.6	113.1	57.7
9	207.8	40.4	199.5	54.9	155.2	71.0	208.2	40.8	197.8	40.1
10	42.6	40.4	133.0	54.9	106.7	71.0	40.9	40.8	43.2	40.1
11	25.7	34.5	153.6	41.2	110.1	73.1	39.3	39.4	82.7	40.8
12	21.8	34.5	34.0	41.2	147.8	73.1	213.0	39.4	207.3	40.8
13	22.8	34.8	25.2	35.4	36.3	34.8	38.9	35.2	42.5	37.6
14	30.9	34.8	35.0	35.4	25.7	34.8	31.0	35.2	26.3	37.6
15	71.8	39.4	74.1	39.7	73.0	39.7	72.3	39.4	73.2	38.0
16	18.5	39.4	20.3	39.7	20.3	39.7	20.2	39.4	20.3	38.0

Table 1 <sup>13</sup>C Nmr spectral data\* of the  $[1,2^{-13}C_2]$  enriched curvularin-type compounds

 $*$  13C Nmr spectra were measured in CD<sub>3</sub>OD at 100.4 MHz on a JEOL JNM-GX 400 NMR spectrometer. Chemical shifts ( 6) **are** given in ppm, and coupling constants ( J ), in Hz.

\*\*The I3C nmr spectral **data are** compatible with those cited in the reference 2.

Isolation and Purification As reported in the previous papers,<sup>4</sup> the acetone extract was concentrated in vncuo to an acetone-free aqueous solution (170 ml) and then extracted with EtOAc (150 ml **x** 7). The EtOAc extract ( dark brown syrup, 4.424 g ) was subjected to a column chromatography on silica gel (300 g, Katayama Chemicals, Type 60) using CHCl<sub>3</sub> (1500 ml) and then a gradient solvent of MeOH-CHCl<sub>3</sub> (1-2 % MeOH-CHCI<sub>3</sub>, 1000 ml for each; 3%, 5%, 10%, 15%, and 20% MeOH-CHCI<sub>3</sub>, 500 ml for each ). Further separation and purification were performed by recrystallization or repeated preparative tlc (Kieselgel 60  $PF_{254}$  and  $F_{254}$ ) with conceivable solvent systems as follows. Elution with 5% MeOH-CHCl<sub>3</sub> afforded a dark brown oil. The oil was separated by repeated preparative tlc using hexane-EtOAc  $(1:2)$ , CHCl<sub>3</sub>-MeOH  $[(15:1) \times 2]$  followed by recrystallization, yielding the labeled citreofuran (5; 7.0 mg,  $0.16\%$ ) as brownish needles (from CHCl<sub>3</sub>-MeOH ) in addition to labeled dehydrocurvularin (2; 710 mg, 16%; yellowish needles from  $C_6H_6$ -MeOH ) and cis-dehydrocurvularin  $<sup>4</sup>$  ( yellowish powder; 5.2 mg, 0.12%). Concentration of the 10% MeOH-CHCl<sub>3</sub> eluant</sup>

gave a yellow residue, which was separated by repeated preparative tlc with hexane-EtOAc  $(1: 2)$ , hexaneacetone (2 : 1; 1 : 1) and purified by recrystallization, affording labeled 11 $\beta$ -hydroxy-12-oxocurvularin (7; 8.02) mg,  $0.18\%$ ) as colorless needles (from CHCl<sub>3</sub>) and curvularin (1; 146 mg, 3.3%) as colorless plates (from  $C_6H_6$ -MeOH). The brown syrup (410 mg) from 15% and 20% MeOH-CHCl<sub>3</sub> elutes was separated by repeated preparative tlc using CHCl<sub>3</sub>-acetone  $(2:1)$ , hexane-acetone  $(2:1)$ , CHCl<sub>3</sub>-MeOH (10 : 1) and CHCl<sub>3</sub>-MeOH [(I5 : 1) x **21** to give rise to labeled 12-oxocurvularin (6; yellowish powder, 12 mg, 0.27% ) and 11 hydroxycurvularins<sup>4</sup> (colorless needles, 101.1 mg, 2.3%).

The physical constants ( mp,  $[\alpha]_D$  ) and spectral data ( ir, <sup>1</sup>H nmr ) of these 1,2-<sup>13</sup>C enriched macrolides are completely identical with those of the authentic non-enriched compounds.<sup>4-6</sup> respectively.

I3c Nmr spectral data of citreofuran ( 5 ), 12-oxocurvularin ( 6 ), and **llp-hydroxy-12-oxocurvularin** ( 7 ), as well as curvularin ( $1$ ) and dehydrocurvularin ( $2$ ) were recorded as shown in Table 1.

## ACKNOWLEDGMENT

We are indebted to the Ministry of Education, Science and Culture as well as the Terumo Foundation for financial assistance to this research. One of us ( S. L. ) also wishes to thank the Fujisawa Foundation for a financial support through this research.

## **REFERENCES**

- I. A. Kobayashi, T. Hino, K. Uneyama, and K. Kawazu, 27th Symposium on the Chemistry of Natural Products, Hiroshima, October 1985, Abstract Papers pp. 343 - 350; T. I. Itoh, H. Sato, and A. Kobayashi, *Zool. Sci.,* 1986, **3,** 255 ; A. Kohayashi, T. Hino, S. Yata, T. I. Itoh, and H. Sato, *Agric.*  Bid. *Chem.,* 1988,52, 3119
- 2. K. Arai, B. **1.** Rawlings, Y. Yoshizawa, and I. C. Vederas, J. *Am. Chem. Soc.,* 1989, 111, 3391.
- 3. Y. Yoshizawa, Z. Li, P. B. Reese, and I. C. Vederas, J. *Am. Chem. Soc.,* 1990, 112, 3212
- 4. S. Lai, Y. Shizuri, Y. Yamamura, K. Kawai, Y. Terada, and H. Fumkawa, *Tetrahedron Len.,* 1989, *30,* 2241; *Chemistry Len.,* 1990, 589.
- 5. 0. C. Musgrave, J. *Chem. Soc.,* 1956,4301; 1957, 1104.
- 6. A. I. Birch, 0. C. Musgrave, R. W. Rickards, and H. Smith, J. *Chem. Sac.,* 1959, 3146.
- 7. H. D. Munro, 0. C. Musgrave, and R. Templeton, J. *Chem. Sac.,* 1967,947.

**Received, 7th December, 1990**