STEREOSTRUCTURE OF (-)-MULTIFLORINE N-OXIDE: A NEW LUPIN ALKALOID FROM LUPINUS HIRSUTUS

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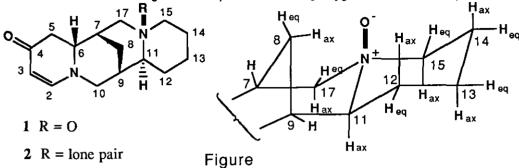
Abstract- A new lupin alkaloid, (-)-multiflorine *N*-oxide (1) was isolated from the seedlings of *Lupinus hirsutus* together with twelve known alkaloids. The structure of 1 was determined by spectroscopic methods and by chemical transformations.

Lupinus hirsutus Linn. (Leguminosae) is a herbaceous annual plant containing lupin alkaloids. We have already reported the isolation of nine lupin alkaloids from the aerial parts of *L. hirsutus.*¹ We also described the alkaloidal components in the earlier stages of the seedlings and the change of alkaloidal pattern with germination.² In the present paper, we describe the structural determination of (-)-multiflorine *N*-oxide (1) isolated from the seedlings of this plant together with twelve known alkaloids, (-)-multiflorine (2), (-)-13 α -tigloyloxymultiflorine (3), (-)-5,6-dehydromultiflorine (4), (+)-epilupinine (5), (+)-epilupinine *N*-oxide (6), (+)-epilupinine acetate *N*-oxide (7), (+)-(*trans*-4'-hydroxy-3'-methoxycinnamoyl)epilupinine (8), (+)-(*trans*-4'-hydroxycinnamoyl)epilupinine (10), (+)-(*trans*-4'-acetoxycinnamoyl)epilupinine (11), (-)-(*trans*-4'- α -L-rhamnosyloxycinnamoyl)epilupinine (12), and (-)-(*cis*-4'- α -L-rhamnosyloxycinnamoyl)epilupinine (13).

Compound 1 was isolated as a colorless oil in a yield of 0.0002 % from the fresh seedlings by repeated chromatography. The molecular formula of 1 was determined as $C_{15}H_{22}N_2O_2$ by the

in-beam hrms. The peak at m/z 246 corresponds to the fragment loosing 16 mass unit from M⁺,

which is the characteristic fragmentation pattern releasing oxygen of N-oxide of lupin



2 R = lone pair

Table 1.¹³C Nmr Data of 1 and 2 Table 2.¹H Nmr Data of 1 and 2

с		1	2	1-2	H	1	2	1-2
2	(d)	155,8	155.6	+0.2	2	6.85	6.84	+0.01
3	(d)	101.9	98.9	+3.0	3	5.60	4.96	+0.64
4	(s)	191.9	192.5	-0.6	5ax	2,45	2.16	+0.29
5	(t)	39.9	39.3	+0.6	5eq	2.27	2.68	-0.41
6	(d)	61.6	60,3	+1.3	6	3.47	3.46	+0.01
7	(d)	30.4	31,1	-0.7	7	2.39	2.03	+0.36
8	(t)	24.5	25.8	-1.3	8ax	3.85	2.20	+1.65
9	(d)	33.5	34,5	-1.0	8eq	1.35	1.28	+0.07
10	(t)	57.5	57.5	0	9	1.88	1.65	+0.23
11	(d)	70.8	63.6	+7.2	10ax	3.07	3.17	-0.07
12	(t)	27.8	31.5	-3.7	10eq	3.11	3.19	-0.08
13	(t)	23.0	24.8	-1.8	11	2.91	2.06	+0.85
14	(t)	20.3	23.7	-3.4	12ax	2.39	1.47	+0.92
15	(t)	69.8	55.2	+14.6	12eq	1.58	1.58	0
17	(t)	65.1	51.1	+14.0	13ax	1.42	1.32	+0.10
					13eq	1.86	1.78	+0.08
					14ax	2.53	1.51	+1.02
					14eq	1.62	1.60	+0.02
					15ax	3.09	2.17	+0.92
					15eq	3.64	2.81	+0.83
					17ax	3.33	2.37	+0.96
					17eq	3.82	2.92	+0.90

alkaloids.³⁻⁸ Other peakes at m/z 134, 110, 97, 83, 55 and 41 were similar to those of 2. The ir absorption at 980 cm⁻¹ of 1 showed the presence of an N-oxide bond. In the ¹³C nmr spectrum of 1 (Table 1), the signals of C-11, C-15 and C-17 were shifted downfield in the range of 7-15 ppm compared to those of 2. In the ¹H nmr spectrum of 1 (Table 2), the protons at C-11, C-15

and C-17 were also appeared in downfield region between δ 2.9 and 3.8 compared to those of 2. The shifts of these signals of the carbons and protons adjacent to the tertiary nitrogen atom were in good agreement with the substituent effects of N-oxide reported in other lupin alkaloids.³⁻⁸ The signals of axial protons at C-8, C-12 and C-14 were observed in downfield range compared to those of 2, due to the anisotropic effects of the axial N-oxide bond at N-16. Consequently, rings C and D in the structure of 1 were assumed to have boat and chair conformations, respectively, from these ¹H nmr data (Table 2, Figure). The final confirmation of the structure of 1 including the absolute configration was performed by chemical interconversions between 1 and 2. The compound (1) was reduced by sulphur dioxide to give 2. Furthermore, 1 was synthesized from 2 by oxidation with *m*-chloroperoxybenzoic acid. In the cd spectrum, 1 showed negative Cotton effects at 326 nm ($[\theta]_{326}$ -13300) and at 227 nm ($[\theta]_{227}$ -1100) and a positive effect at 296 nm ($[\theta]_{296}$ +2700). These are similar Cotton effects to those of 2 ([0]328 -6400, [0]297 +780, [0]224 -2700). The synthetic 1 from 2 showed the same Cotton effects as those of natural 1. Therefore, the absolute configration of 1 was confirmed as 6R, 7S, 9S, 11S, identical to that of (-)-multiflorine (2).⁹ So far, we have isolated a few N-oxides of lupin alkaloids from leguminous plants.¹⁻⁷ Some N-oxides of lupin alkaloids were also reported in the literature, 10, 11 but the N-oxides were not common in the nature. The N-oxidation of alkaloids may occur with specific enzymes in plants considering the role of alkaloidal Noxide.10,11

EXPERIMENTAL

¹H Nmr and ¹³C nmr spectra were recorded at 500 and 125.65 MHz, respectively. TMS was used as an internal standard in CDCl₃. The complete assignments of protons and carbons were made by use of 2D-nmr experiments. TIc was performed on silica gel plates ($60F_{254}$, 0.25 mm, Merck) in CH₂Cl₂-MeOH-28% NH₄OH (90:9:1). Analytical hplc was carried out as described previously.¹² Preparative hplc was performed on Licrosorb Si-60 5µm, (ø 4.6 x 150 mm) column with solvent system of 50% MeOH in ether-5% NH₄OH (500:20) at 220 nm.

Plant materials

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The seeds of *L. hirsutus* were purchased from Sakata Seeds Co. Ltd., Yokohama, Japan. The seedlings were grown in moistened vermiculite under daylight for 7-10 days at 25°C.

Extraction and isolation

The total alkaloidal fraction from the 75% EtOH (121) extracts of the fresh seedlings (5 kg) was obtained in a vield of 0.24% of the fresh weight as described previously.¹ The total base (12.0 g) was subjected to a silica gel column usinig a solvent system of CH2Cl2-MeOH-28% NH4OH as reported in the previous paper.¹ The 1-rich fractions (20 mg) were eluted with the solvent system of CH2Cl2-MeOH-28% NH4OH (100:20:3). The purification of these rich fractions by use of preparative hplc gave pure 1 (11.1 mg), as a colorless oil, $\left[\alpha\right]_{D}^{23}$ -145.8° (c=0.069, EtOH): uv λ_{max} (MeOH) nm (log ϵ) 316 (4.10), 229 (3.25 sh.); cd (c=2.8 x 10⁻⁴, MeOH) [0]₃₂₆ -13300, [0]296 +2700, [0]227 -1100; hrms (in-beam) m/z (%) 262.1680 (M⁺, calcd for C15H22N2O2 262.1680, 30), 246 (39), 134 (100), 110 (24), 97 (12), 83 (28), 55 (14), 41 (28); ir v_{max} (CHCl₃) 2930, 2850, 2770 (C-H), 1640 (conjugated C=O), 1590 (conjugated C=C), 980 (N+-O⁻) cm⁻¹; ¹³C and ¹H nmr chemical shifts were shown in Tables 1 and 2, respectively, ¹H nmr δ 6.85 (1H, d, J=7.7 Hz, 2-H), 5.60 (1H, d, J=7.7 Hz, 3-H), 3.85 (1H, m, 8-Hax.), 3.82 (1H, d, J=12.9 Hz, 17-Heq.), 3.64 (1H, br d, J=11.0 Hz, 15-Heq.), 3.47 (1H, ddd, J=14.9, 5.0, and 2.5 Hz, 6-H), 3.33 (1H, dd, J=12.9 and 2.5 Hz, 17-Hax.), 3.11 (1H, dd, J=12.1 and 3.0 Hz, 10-Heo.), 3.09 (1H, 15-Hax., overlapped with the signal of 10-Hax, and 10-Heq.), 3.07 (1H, dd, J=12.1 and 2.5 Hz, 10-Hax.), 2.91 (1H, ddd, J=12.9, 3.6, and 3.6 Hz, 11-H), 2.53 (1H, ddd, J=14.3, 13.5, and 4.1 Hz, 14-Hax), 2.45 (1H, t, J=16.3 Hz, 5-Hax.), 2.39 (2H, ddd, J=14.0, 13.2, and 4.1 Hz, 12-Hax. and 7-H), 2.27 (1H, ddd, J=16.3, 5.5, and 0.7 Hz, 5-H_{e0.}), 1.88 (1H, d, J=2.2 Hz, 9-H), 1.86 (1H, dd, J=15.7 and 1.9 Hz, 13-Hea.), 1.62 (1H, br d, J=14.3 Hz, 14-Hea.), 1.58 (1H, m, 12-Hea.), 1.42 (1H, dddd, J=15.7, 13.2, 4.2, and 4.2 Hz, 13-Hax.), 1.35 (1H, d, J=12.6 Hz, 8-Heq.).

Reduction of 1 to 2.

Compound (1) (2 mg) was dissolved in 2 ml of MeOH and reduced with SO₂ gas for 10 min at 0 °C. The reaction mixture was analyzed on hplc.¹¹ Compound (2) was identified by direct comparison with an authentic sample on hplc.

Synthesis of 1 from 2.

The compound (1) was synthesized according to the method reported previously.³ Compound(2) (20 mg) was oxidized with *m*-chloroperoxybenzoic acid (21 mg) in 5 ml of CH₂Cl₂. The reacting species was purified by preparative hplc. The pure 1, $[\alpha]D^{24}$ -144.4° (c=0.16, EtOH), was obtained in a yield of 75% (16 mg). The structure of synthetic products was identified by ir spectrum and by co-hplc as compound (1).

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