$(R)-(+)$ -NORANICANINE A NEW TYPE OF TRIOXYGENATED BENZYLISOQUINOLINE ISOLATION AND SYNTHESIS

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Abstract - (R)-(+)-Noranicanine (1a), a new type of benzylisoquinoline precursor was isolated from *Aniba canelilla* **(Lauraceae)** and its structure was determined by means of spectroscopic data as well as by synthesis.

1-Benzylisoquinolines and their derivatives represent one of the preeminent group of alkaloids in plants.' The benzylisoquinoline skeleton is known to be produced, in vivo, via cyclization of Schiff bases formed from condensation of dopamine with either 4-hydroxyphenylacetaldehyde or 3,4-dihydroxyphenylacetaldehyde.² Natural known **1-benzyltetrahydroisoquinolines** with a monosubstituted benzyl group therefore always exhibit a para functionality in ring C. We now have isolated from a Bolivian Lauraceae, Aniba canelilla λ^3 (+)noranicanine (1), a trioxygenated benzyltetrahydroisoquinoline with a different substitution pattern in ring C, the oxygenated substituent being in the *meta* position. The structure of this new alkaloid has been firmly established by synthesis.

The EI mass spectrum of 1, C₁₈H₂₁NO₃, exhibited a typical benzyltetrahydroisoquinoline type fragmentation pattern⁴ with a very weak molecular ion at m/z 299 and a base peak at m/z 192 representing the isoquinoline moiety. The nmr spectrum of $(+)$ -noranicanine (CDCl₃, 200 MHz, TMS=0) has been summarized around structure 1a. Noteworthy were the absence of a N-methyl singlet and the presence of two methoxy singlets at δ 3.79 and 3.85 ppm and two aromatic singlets at **6** 6.57 and 6.50 ppm for the isoquinoline moiety. Unlike coclaurine or other compounds monosubstitoted in ring C, H-13 was there in evidence as a deshielded doublet of doublets at δ 7.09 ppm $(I_{12,13}=J_{13,14}=7.5 \text{ Hz})$. However, the unresolved multiplet centered at δ 6.67 ppm corresponding to the three remaining aromatic protons did not allow us to unambiguously establish the substitution pattern of the benzene ring.

Different assays of solvent effects failed in offering a better resolution than CDC13, however the hydroxyl substituent of ring C should be undoubtedly located either in the **meta (la)** or in the **ortho** position **(lb).** We thus undertook the synthesis of both possible compounds via a classical route **5,6** in order to confirm the assigned structure 1 (Scheme 1).

Condensation of 3.4-dimethoxyphenethylamine with m-hydroxyphenylacetic acid or o-hydroxyphenylacetic acid respectively led to amides (2a) (89%,mp 149-150 °C) and 2b (97%, mp 108-109°C) which were benzylated to give 3a (72%. mp 89-90 "C) and 3b (85%, mp 112-113'C). A Bischler-Napieralski cyclization with 3a and 3b then provided 3,4-dihydroisoquinolines (4a) $(71\%$, mp 79-80 °C) and (4b) $(73\%$, mp 184-18S°C) which were further reduced with sodium borohydride into 5a (86%. mp 160-161°C) and 5b (82%. mp 165-166°C), respectively..Debenzylation of 5a and 5b gave the corresponding phenol (la) (61%, mp 155-156 $^{\circ}$ C) and (1b) (76%, mp 147-148 $^{\circ}$ C) as their hydrochlorides.⁷ Compound (1a), in its base form, finally was spectroscopically identical with natural $(+)$ -noranicanine.⁸

The 1-R configuration of the natural (+)-noranicanine is suggested by the positive optical rotation 9 and confirmed by the cd curve which presents a negative Cotton effect at 284 nm and a negative tail at 240 nm $.10$

To our knowledge, this is the fust repon of anatural **benzylteuahydroisoquinoline** monosubstituted in ring C with the hydroxyl group being in the *meta* position. Due to the number of alkaloids derived from this basic skeleton, (+)-noranicanine may be therefore regarded as a new type of benzylisoquinoline precursor.

REFERENCES AND NOTES

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- $\overline{\mathbf{3}}$ Plant collection, extraction and alkaloid isolation- The stem barks of *Aniba canelilla* (1.1 Kg) were collected in 1988 at Fatima de Chimanes, Bolivia. A voucher specimen (ref AF855) was deposited at the herbarium of ORSTOM-IBBA center of La Paz. The crude alkaloids (1.04%) were obtained from the powdered defatted stem barks following a classical procedure. Specific extraction was followed by MPLC (CHCI₃ gradually enriched with MeOH) and preparative TLC (CHCI₃-MeOH 13%) on silica gel. This way, 16 mg of amorphous (+)-noranicanine were isolated.
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- $\overline{7}$ Eims data (70 eV) for compounds $2a \rightarrow 5a$ and $2b \rightarrow 5b$: $2a$ m/z : 315 (M⁺, C₁₈H₂₁NO₄, 8.7%), 165 (10.4%), 164 (100%). 152 (4.8%), 151 (20%). 149 (8.8%). 108 (3.1%), 107 (14.8%). 91 (3.1%). 78 (4.8 %), 77 (8.4%), 44 (3.4%); 3a m/z: 405 ($M⁺$, C₂₅H₂₇NO₄, 7.6%), 165 (12.3%), 164 (100%), 151 (12.5%) , 149 (4.5%) , 134 (3.9%) , 107 (3%) , 92 (3.8%) , 91 (42.3%) , 65 (5%) ; 4a m/z : 387 (M^{+})

C2sH2sN03. 5.7%). 386 (6.9%), 297 (22.3%). 296 (loo%), 280 (4.6%), 268 (11.5%), 252 (3.5%). 192 (2.1%). 152 (1.7%). 92 (5.7%), 91 (53.1%), 77 (3.9%). 65 (7.4%), 38 (7%), 36 (18.9%), 35 (3.2%) ; **5a** m/z : 389 (M+, C25Hz7N03, 0.06%). 280 (1.3%), 193 (12.7%). 192 (100%). 176 (5.8%), 148 (3.5%), 147 (1.6%), 91 (9.5%) ; **2b** m/z : 315 (M+, CI8Hz1N04, 9.6%). 165 (12.6%), 164 (100%), 152 (4.4%). 151 (15.6%). 107 (12.8%), 106 (3.2%), 91 (3.1%), 79 (3.9%). 78 (4.8%). 77 (8.6%) ; 3b m/z : 405 (M+, C25Hz7N04,0.8%), 165 (13.5%), 164 (100%). 151 (13.5%), 149 (5.4%). 107 (12.8%). 106 (4.4%), 92 (5.5%) , 91 (62.0%) , 79 (3.4%) , 78 (8.0%) , 77 (6.9%) , 65 (13.0%) ; 4b m/z : 387 $(M^{+}, C_{25}H_{25}NO_3,$ 0.4%), 377 (3.4%). 296 (6.7%). 281 (13.7%). 280 (71.9%). 264 (5.1%). 128 (5.6%), 126 (15.0%), 92 (8.1%). 91 (100%). 89 (7.5%), 65 (12.5%), 63 (7.1%). 52 (3.1%). 51 (4.7%). 50 (6.2%). 38 (3.2%), 36 (4.5%) ; **5b** mlz : 389 (M+, CzH27N03,0.03%), 280 (3.0%), 193 (13.0%). 192 (loo%), 176 (7.0%), 148 (3.0%), 91 (11.3%).

- 8 Spectral data for compounds **la** and **lb** : $\mathbf{1a}$: C₁₈H₂₁NO₃, uv λ max (MeOH) nm (log ε) : 207 (4.55), 229 sh (4.03). 282 (3.71), *h* max (MeOH-KOH) nm (loge) : 212 (4.66). 235 sh (3.96), 288 (3.61) ; **ir** ^v max (KBr) : 3450, 3260, 1610, 1585 cm⁻¹ ; eims (70 eV) m/z : 299 (M⁺, 0.2%), 193 (12.5%), 192 (100%) , 177 (3.7%) , 148 (4.0%) , cims (NH_3) m/z : 300 $(NH_1^{\dagger}$, 100%), 193 (3.1%) , 192 (24.2%) ; ¹Hnmr (200 MHz, CDCl₃, δ ppm) : see 1a, ¹³C-nmr (68 MHz, CDCl₃, δ ppm) : 56.3 (C-1), 40.3 (C-3), 28.5 (C-4), 126.5 (C-4a), 111.7 (C-5). 147.1 and 147.7 (C-6 and C-7), 109.3 (C-8). 128.9 (C-8a), 41.9 (C-a), 139.4 (C-9), 116.2 (C-10). 157.6 (C-11), 114.7 (C-12), 130.0 (C-13). 120.4 (C-14), 55.8 and 55.9 (2 OCH₃); spectral data for natural (+)-noranicanine, $[\alpha]_D+36^{\circ}$ (c=1, CHCl₃), cd (MeOH) $\Delta \varepsilon$ (nm) 0 (299), -0.15 (284), -0.04 (255), negative tail at 240 nm. **lb**: C₁₈H₂₁NO₃, uv λ max (MeOH) nm (log ε) : 206 (4.63). 230 sh (4.06), 281 (3.79). *h* max (MeOH-KOH) nm (loge) : 210 (4.76). 233 sh (4.11), 286 (3.80) ; irv max KBr) : 3440,3290,1610, 1585 cm-1 ; eims (70 eV) m/z : 299 (M+, 0.2%). 193 (15.7%), 192 (loo%), 178 (8.7%). 177 (4.0%), 176 (10.9%). 148 (5.7%), 131 (4.6%), 77 (7.8%), cims (NH3) m/z : 300 ([MHl+, 100%), 286 (6.7%), 193 (3.2%), 192 (23.9%), 178 (1.7%) ; 'H-nmr (200 MHz, CDCIj, **6** ppm) : 2.83 and 3.35 (2m, 6H, 3 **CH2),** 3.83 and 3.85 (2s. 6H, 2OM3), 4.36 (dd, lH, 1=8.0, 1.0 Hz, H-1), 6.54 (s, lH, H-8), 6.60 (s, lH, H-5). 6.73 (dd, lH, J=7.0, 7.0 Hz, H-13), 6.87 (d, J=7.0 Hz, H-11). 7.03 (d, J=7.0 Hz, H-14), 7.10 (dd, J=7.0, 7.0 Hz, H-12) ; ¹³C-nmr (68 MHz, CDCl₃, δ ppm) : 55.9 (C-1), 38.5 (C-3), 27.1 (C-4), 125.6 and 127.1 and 125.0 (C-4a and C-8a and C-9). 111.0 (C-5), 147.6 and 147.9 (C-6 and C-7), 109.9 (C-8), 40.3 (C-α), 156.7 (C-10), 117.4 (C-11), 128.4 (C-12), 119.1 (C-13), 131.6 (C-14), 55.8 and 55.9 (2 OCH3).
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