

ENANTIOSELECTIVE SYNTHESIS OF INDOLIZIDINE ALKALOID: A FORMAL TOTAL SYNTHESIS OF (-)-PUMILIOTOXIN 251D

Toshio Honda,* Michiyasu Hoshi, and Masayoshi Tsubuki

Institute of Medicinal Chemistry, Hoshi University, Ebara 2-4-41, Shinagawa-ku,
Tokyo 142, Japan

Abstract---An enantioselective formal total synthesis of (-)-pumiliotoxin 251D was achieved by employing the Sharpless kinetic resolution of the 2-furylmethanol derivative (3) and a stereoselective radical cyclisation of the thiocarbonyl-imidazolide (10) as key steps .

Indolizidine alkaloids possessing the 1-azabicyclo[4.3.0]nonane skeleton are widely distributed in nature with their wide range of structural and stereochemical features. Due to their interesting biological properties, this class of alkaloids have provoked an extraordinary amount of activity by synthetic organic chemists.¹ Recently, Gallagher and his co-workers have published² the elegant total synthesis of (-)-pumiliotoxin 251D (1) *via* the lactam (24) and this report has prompted us to disclose our efforts directed toward the synthesis of 1 (Figure 1).

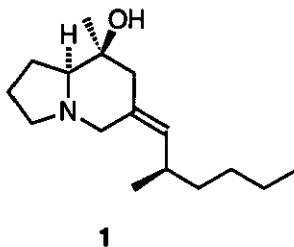
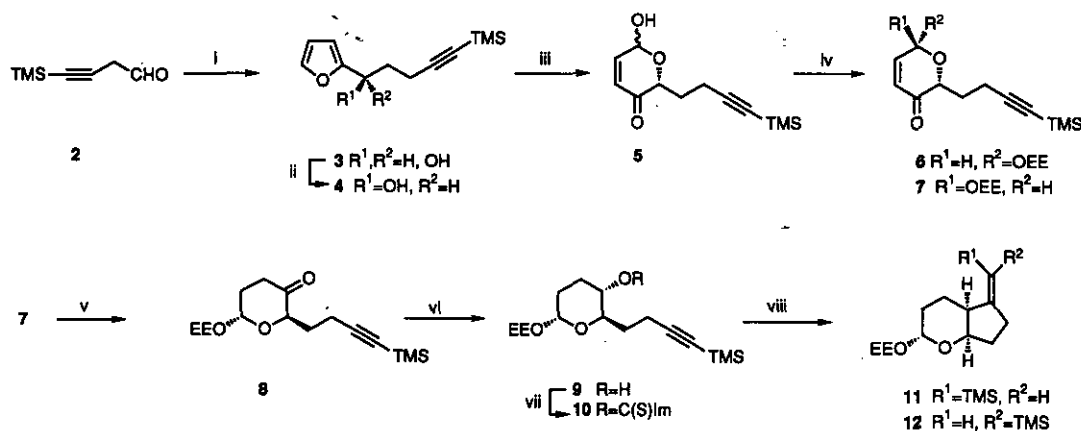


Figure 1

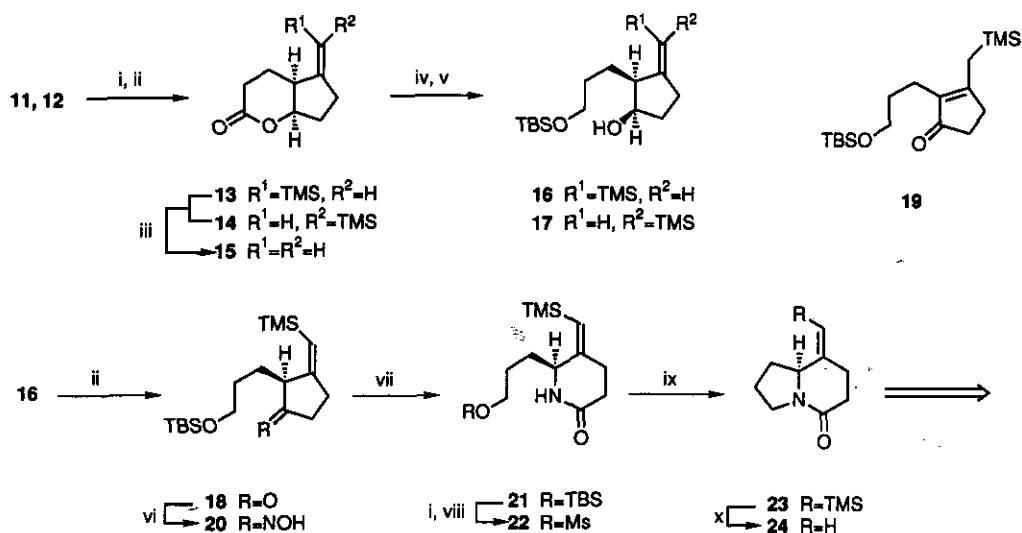
The synthesis that we have developed involves the Sharpless kinetic resolution using L-diisopropyl tartrate, titanium tetrakisopropoxide, and *tert*-butyl hydroperoxide, of the racemic 2-furylmethanol derivative (3),³ readily derived from 2-lithiofuran and 5-trimethylsilyl-4-pentyn-1-al (2), providing the *R*-alcohol (4) in 46.5% yield with >95% enantiomeric excess,⁴ as a starting material. Oxidation of 4 with *N*-bromosuccinimide in aqueous tetrahydrofuran⁵ afforded the lactol (5) in 98% yield, which on treatment with ethyl vinyl ether and pyridinium *p*-toluenesulphonate was converted into the ethoxyethyl ethers (6) and (7) in 93% yield in a ratio of 1:2.4. Reduction of the major isomer (7) with lithium aluminium hydride in the presence of copper(I) iodide⁶ in tetrahydrofuran-hexamethylphosphoric triamide

(4:1,v/v) gave the 1,4-reduction product (**8**) in 70% yield. Further reduction of **8** with sodium borohydride provided the alcohol (**9**) as a major isomer, which was then transformed into the thiocarbonylimidazolide (**10**) in a usual manner⁷ in 71% overall yield (Scheme 1).



Scheme 1 Reagents and conditions: i, 2-lithiofuran, THF, -78°C ; ii, L-DIPT, $\text{Ti}(\text{O}^i\text{Pr})_4$, TBHP, MS 3A, CH_2Cl_2 , -25°C ; iii, NBS, THF- H_2O (4:1), 0°C ; iv, $\text{CH}_2=\text{CHOEt}$, PPTS, CH_2Cl_2 ; v, LiAlH_4 , CuI, THF-HMPA (4:1), -78°C ; vi, NaBH_4 , THF, 0°C ; vii, $\text{Im}_2\text{C}=\text{S}$, $\text{ClCH}_2\text{CH}_2\text{Cl}$, reflux; viii, Bu_3SnH , AIBN, benzene, reflux

We envisioned construction of the chiral indolizidine skeleton from the corresponding chiral 2-substituted cyclopentan-1-one, *via* the piperidone derivative, by using the Beckmann rearrangement with retention of the chiral center, followed by an intramolecular alkylation. Thus, treatment of the imidazolide (**10**) with tri-*n*-butyltin hydride in the presence of azobisisobutyronitrile in refluxing benzene afforded the cyclisation products (**11**) and (**12**) in 84% yield in a ratio of 4:3 as an inseparable mixture, which were further converted to the lactones (**13**) and (**14**) by deprotection of the ethoxyethyl group, followed by oxidation of the resulting lactols with pyridinium chlorochromate in 54% and 41% yields, respectively. The stereochemistry of **13**⁸ was assigned as depicted in Scheme based on NOEs between the olefinic proton and one of the angular protons and also between two angular protons. Since desilylation of both compounds (**13**) and (**14**) with *p*-toluenesulphonic acid⁹ in aqueous acetonitrile yielded the same olefin (**15**), the lactone (**14**) was unambiguously determined to the olefinic isomer of **13**. These results indicated that the radical cyclisation of **10** proceeded in a stereoselective manner to form the *cis*-ring juncture predominantly and none of the *trans* isomer could be isolated in this cyclisation. Lithium aluminium hydride reduction of **13**, followed by selective protection of the primary alcohol with *tert*-butyldimethylsilyl chloride gave the silyl ether (**16**), in 86% yield, which on oxidation with pyridinium chlorochromate in the presence of sodium acetate in dichloromethane afforded the desired chiral cyclopentanone (**18**) in 60% yield. Similar oxidation of **17**, prepared from **14** by two-steps as above, furnished the cyclopentenone (**19**) unfortunately (Scheme 2).



Scheme 2 Reagents and conditions : i, 2N HCl, THF, 0°C; ii, PCC, AcONa, CH₂Cl₂; iii, *p*-MeC₆H₄SO₂H, aq. MeCN; iv, LiAlH₄, Et₂O; v, TBSCl, Et₃N, CH₂Cl₂, 0°C; vi, NH₂OH·HCl, pyridine, MeOH; vii, SOCl₂, THF, 0°C; viii, MeSO₂Cl, Et₃N, CH₂Cl₂, 0°C; ix, K₂CO₃, dioxane-H₂O (4:1), 90°C; x, *p*-MeC₆H₄SO₂H, aq. MeCN, reflux.

With the requisite chiral cyclopentanone available, a study was made of the best condition for the Beckmann rearrangement of the corresponding oxime (20) to the piperidone (21). After the number of attempts were unsuccessful or not efficient in terms of the conversion yield and reproducibility, due to the feasible migration of the exo-methylene to an internal olefin, it was found that the Beckmann rearrangement of the oxime (20),¹⁰ prepared from the ketone (18) and hydroxylamine hydrochloride and pyridine, with thionyl chloride in tetrahydrofuran¹¹ at 0°C resulted in the formation of the lactam (21) as a single compound in 41% yield. Removal of the silyl group of 21 with aqueous hydrochloric acid, followed by treatment with methanesulphonyl chloride in dichloromethane gave the methanesulphonate (22) in 75% yield. An intramolecular alkylation of 22 with potassium carbonate in aqueous dioxane and subsequent desilylation of 23 with *p*-toluenesulphonic acid⁹ in aqueous acetonitrile furnished the lactam (24), in 58% yield, whose spectroscopic data including the specific optical rotation $[\alpha]_D -95.0^\circ (c=0.3, \text{CHCl}_3)$ {lit.,² $[\alpha]_D -98.3^\circ (c=1.2, \text{CHCl}_3)$ } were identical with those reported.²

Since the lactam (24) has already been transformed into (-)-pumiliotoxin 251D (1),² this synthesis constitutes its formal synthesis and this synthetic route would be applicable to the chiral synthesis of other indolizidine alkaloids.

REFERENCES AND NOTES

1. A. S. Howard and J. P. Michael, "The Alkaloids", Vol. 28, p. 183, ed by A. Brossi, Academic Press, Inc., New York, 1986; J. A. Lambertson, *Natural Product Reports*, 1984, **1**, 245; M. F. Grundon, *ibid.*, 1985, **2**, 235; M. F. Grundon, *ibid.*, 1987, **4**, 415; M. F. Grundon, *ibid.*, 1989, **6**, 523; J. P. Michael, *ibid.*, 1990, **7**, 485; J. P. Michael, *ibid.*, 1991, **8**, 553; S. Rajeswari, S. Chandrasekharan, and T. R. Govindachari, *Heterocycles*, 1987, **25**, 659.
2. D. N. A. Fox, D. Lathbury, M. F. Mahon, K. C. Molloy, and T. Gallagher, *J. Am. Chem. Soc.*, 1991, **113**, 2652.
3. T. Kametani, M. Tsubuki, Y. Tatsuzaki, and T. Honda, *J. Chem. Soc., Perkin Trans. 1*, 1990, 639; T. Honda, T. Kametani, K. Kanai, Y. Tatsuzaki, and M. Tsubuki, *ibid.*, 1990, 1733.
4. The enantiomeric excess of **4** was determined on the basis of nmr data of its Mosher's ester and all the compounds described in this synthesis were fully characterised by spectroscopic methods and microanalyses.
5. M. P. Georgiades and E. A. Couladouros, *J. Org. Chem.*, 1986, **51**, 2725.
6. T. Tsuda, T. Fujii, K. Kawasaki, and T. Saegusa, *J. Chem. Soc., Chem. Comm.*, 1980, 1013.
7. J. R. Rasmussen, C. J. Slinger, R. J. Kordish, and D. D. Newman-Evans, *J. Org. Chem.*, 1981, **46**, 4843.
8. Selected spectroscopic data for **13**: $\nu_{\max}(\text{CHCl}_3)$ 1730 and 1620 cm^{-1} ; $\delta(\text{CDCl}_3)$ 0.12(9H, s, SiMe₃), 2.85-3.00(1H, m, CHCO), 4.89(1H, dt, *J* 4.9 and 4.9 Hz, CHOCO), 5.50(1H, dd, *J* 3.7 and 1.8 Hz, CH=C).
For **14**: $\nu_{\max}(\text{CHCl}_3)$ 1730 and 1630 cm^{-1} ; $\delta(\text{CDCl}_3)$ 0.10(9H, s, SiMe₃), 2.75-2.85(1H, m, CHCO), 4.85(1H, dt, *J* 4.9 and 1.8 Hz, CHOCO), 5.43(1H, dd, *J* 4.9 and 2.4 Hz, CH=C).
For **18**: $\nu_{\max}(\text{CHCl}_3)$ 1750 and 1645 cm^{-1} ; $\delta(\text{CDCl}_3)$ 0.03(6H, s, SiMe₂Bu), 0.11(9H, s, SiMe₃), 0.87(9H, s, SiMe₂Bu), 3.50-3.70(2H, m, CH₂OSi), 5.53(1H, dd, *J* 2.4 and 1.2 Hz, CH=C).
For **21**: $\nu_{\max}(\text{CHCl}_3)$ 3400 and 1655 cm^{-1} ; $\delta(\text{CDCl}_3)$ 0.05 and 0.06(each 3H, each s, SiMe₂Bu), 0.12(9H, s, SiMe₃), 0.89(9H, s, SiMe₂Bu), 3.55-3.75(2H, m, CH₂OSi), 4.03-4.13(1H, m, CHNH), 5.42(1H, s, CH=C), 6.57(1H, br s, NH).
For **23**: $\nu_{\max}(\text{CHCl}_3)$ 1635 cm^{-1} ; $\delta(\text{CDCl}_3)$ 0.14(9H, s, SiMe₃), 3.26(1H, dt, *J* 12.2 and 3.7 Hz, CHHNCO), 3.86(1H, dt, *J* 12.2 and 8.6 Hz, CHHNCO), 4.11(1H, dd, *J* 11.6 and 5.5 Hz, NCHC=), 5.49(1H, s, CH=C).
9. J. M. Dener, D. J. Hart, and S. Ramesh, *J. Org. Chem.*, 1988, **53**, 6022; E. Wenkert, B. C. Bookser, and T. S. Arrhenius, *J. Am. Chem. Soc.*, 1992, **114**, 644.
10. The oxime (**20**) was obtained as a mixture of *E:Z* (4:1) isomers, and the *E* isomer was subjected to the Beckmann rearrangement.
11. E. J. Moriconi and M. A. Stemniski, *J. Org. Chem.*, 1972, **37**, 2035.