

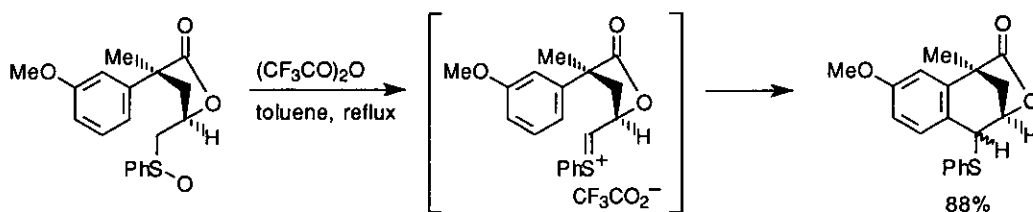
## A FACILE ROUTE TO TETRAHYDROISOQUINOLINE ALKALOIDS VIA SULFOXIDE MEDIATED CYCLIZATION

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**Abstract** — A facile route to 1,2,3,4-tetrahydroisoquinoline framework has been developed by employing the sulfoxide mediated cyclization reaction. Utilizing the reaction developed some naturally occurring isoquinoline alkaloids have been synthesized.

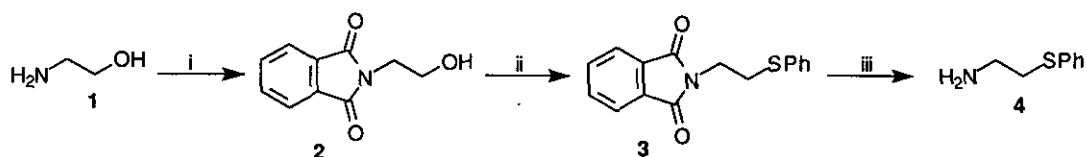
Although it has been well-recognized that activated sulfoxides such as  $\beta$ -keto sulfoxides can participate in cyclization reactions forming carbocycles and heterocycles under acidic conditions *via* generation of highly activated  $\beta$ -keto sulfenium intermediate,<sup>1,2</sup> only a small number of examples where simple sulfoxides involved in the cyclization reaction were so far reported.<sup>3,4</sup> In relation to the simple sulfoxide-mediated hydronaphthalene formation reaction which we developed recently (Scheme 1),<sup>4</sup> we now examined the construction of isoquinoline framework as an extension.



Scheme 1

We first prepared 2-aminoethyl phenyl sulfide<sup>5</sup> (4) from ethanolamine (1) in three steps. Thus, condensation of ethanolamine with phthalic anhydride followed by treating the resulting imide (2) with diphenyl disulfide

in the presence of tri-*n*-butylphosphine<sup>6</sup> yielded the phenyl sulfide (3) which was exposed to hydrazine hydrate to give the amino sulfide (4) in 60% overall yield (Scheme 2).

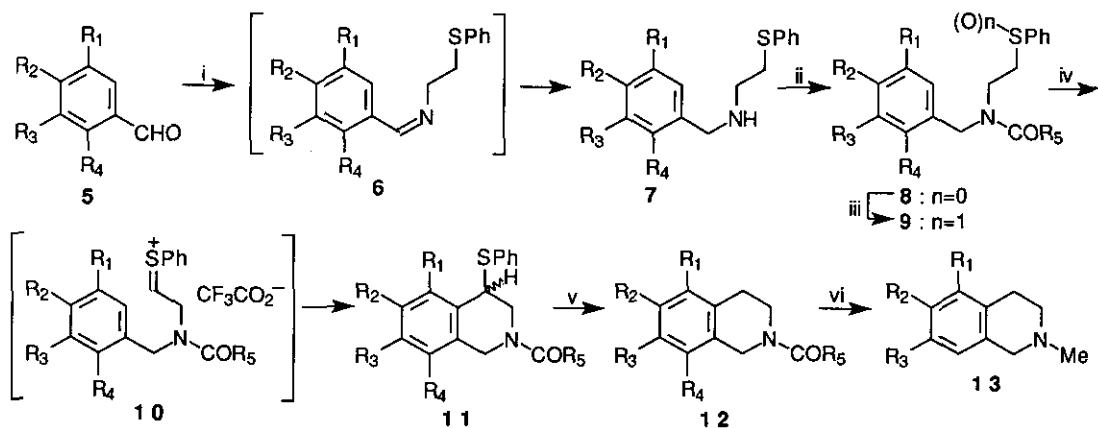


Scheme 2

*Reagents and conditions:* i) phthalic anhydride (1.0 equiv.), toluene, reflux, 3 h; ii) (PhS)<sub>2</sub> (1.5 equiv.), *n*-Bu<sub>3</sub>P (1.5 equiv.), pyridine, 60 °C, 4 h; iii) H<sub>2</sub>NNH<sub>2</sub>·H<sub>2</sub>O (3 equiv.), EtOH, reflux, 1 h.

Condensation of the amine (4) with aromatic aldehydes (5a~f), followed by reduction of the resulting Schiff bases (6a~f) with sodium borohydride afforded the secondary amines (7a~f) without difficulty in satisfactory overall yields. On sequential acylation and oxidation with *m*-chloroperbenzoic acid, the amines (7a~f) afforded the substrate sulfoxides (9a~g) in good overall yields via the sulfides (8a~g), respectively.

Upon exposure to 1,3 equiv. of trifluoroacetic anhydride in refluxing toluene,<sup>4</sup> the sulfoxides (9a~g) reacted within a minute to give the unstable cyclization products (11a~g) which were immediately treated with Raney



Scheme 3

*Reagents and conditions:* i) 4 (1.0 equiv.), *p*-TsOH (cat.), benzene, reflux, overnight, then NaBH<sub>4</sub> (4.0 equiv.), MeOH, room temperature, 2 h; ii) methyl chlorocarbonate (1.2 equiv.) for a~f, or acetyl chloride for g, pyridine (1.2 equiv.), CH<sub>2</sub>Cl<sub>2</sub>, 0 °C, 2 h; iii) *m*-chloroperbenzoic acid (*m*-CPBA) (1.1 equiv.), NaHCO<sub>3</sub> (3.0 equiv.), CH<sub>2</sub>Cl<sub>2</sub>, -30 °C, 2.5 h; iv) trifluoroacetic anhydride (TFAA) (3.0 equiv.), toluene, reflux, 0.5 min; v) Raney Ni, EtOH, reflux, 3 h; vi) LiAlH<sub>4</sub> (9.0 equiv.), THF, reflux, 3 h.

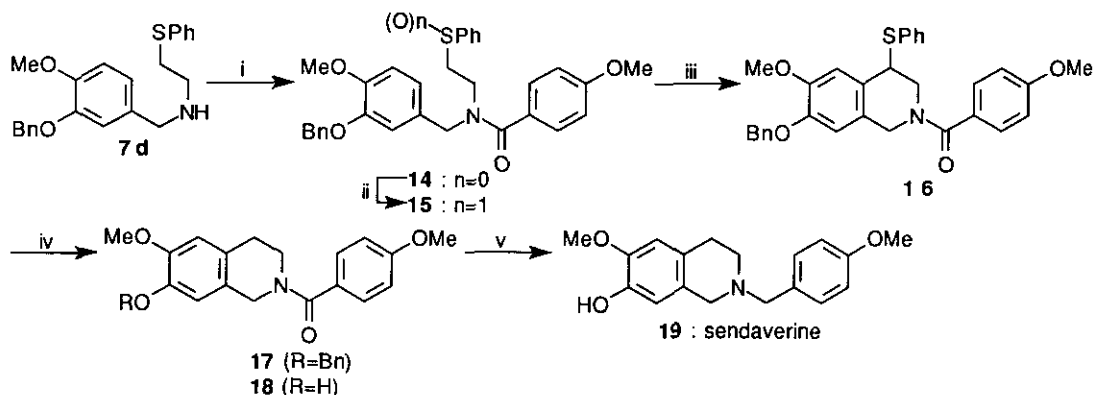
**Table 1:** Formation of the tetrahydroisoquinolines (12) from the Sulfoxides (9) via 11

Sulfoxide (9)	R <sub>1</sub>	R <sub>2</sub>	R <sub>3</sub>	R <sub>4</sub>	R <sub>5</sub>	Isoquinoline (12)	R <sub>1</sub>	R <sub>2</sub>	R <sub>3</sub>	R <sub>4</sub>	R <sub>5</sub>	Yield (%)
(a)	H	-OCH <sub>2</sub> O-	H	OMe		(a)	H	-OCH <sub>2</sub> O-	H	OMe		38
(b)	H	MeO	MeO	H	OMe	(b)	H	MeO	MeO	H	OMe	36
(c)	MeO	MeO	MeO	H	OMe	(c)	MeO	MeO	MeO	H	OMe	44
(d)	H	MeO	BnO	H	OMe	(d)*	H	MeO	HO	H	OMe	62
(e)	H	H	MeO	MeO	OMe	(e)	H	H	MeO	MeO	OMe	30
(f)	H	H	MeO	BnO	OMe	(f)*	H	H	MeO	HO	OMe	46
(g)	H	MeO	MeO	H	Me	(g)	H	MeO	MeO	H	Me	53

\* Concomitant debenzoylation occurred under the desulfurization conditions.

nickel in ethanol to furnish the *N*-acyl-1,2,3,4-tetrahydroisoquinolines (12a-g) in moderate overall yields (Table 1). Among these products, reduction of the carbamates (12a-d) with lithium aluminum hydride in refluxing tetrahydrofuran furnished the naturally occurring alkaloids, hydrohydrastinine<sup>7</sup> (13a), methylcorypalline<sup>8</sup> (13b), tehaunine<sup>9</sup> (13c), and corypalline<sup>10</sup> (13d), respectively (Scheme 3).

On the other hand, the amine (7d) was transformed into the *p*-methoxybenzamide (14) which then was oxidized to the sulfoxide (15). Upon the same treatments above 15 furnished the *N*-*p*-methoxybenzoyltetrahydroisoquinoline (18) via 16 and 17 in 33% overall yield with spontaneous debenzoylation. The resulting

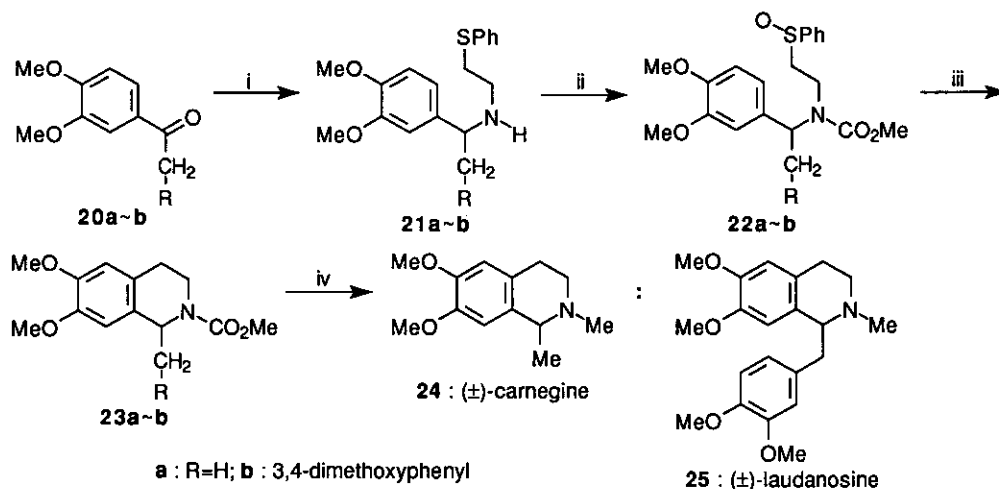
**Scheme 4**

**Reagents and conditions:** i) *p*-methoxybenzoyl chloride (1.2 equiv.), Et<sub>3</sub>N (1.2 equiv.), CH<sub>2</sub>Cl<sub>2</sub>, room temperature, 1.5 h; ii) *m*-CPBA (1.2 equiv.), NaHCO<sub>3</sub> (3.0 equiv.), CH<sub>2</sub>Cl<sub>2</sub>, -30 °C, 2 h; iii) TFAA (3.0 equiv.), toluene, reflux, 0.5 min; iv) Raney Ni, EtOH, reflux, overnight; v) LiAlH<sub>4</sub> (9.0 equiv.), THF, reflux, 3 h.

amide (18) was reduced with lithium aluminum hydride to afford another naturally occurring alkaloid, sendaverine<sup>11</sup> (19), in 67% yield (Scheme 4).

To obtain 1-substituted isoquinoline alkaloids, we next carried out the reaction starting from the aryl ketones (20a~b). On reductive condensation with the sulfide amine (4), these gave the corresponding secondary amines (21a~b) which were sequentially carbamoylated and oxidized to furnish the corresponding sulfoxides (22a~b), respectively, in satisfactory yields *via* the sulfide (21a~b).

Upon exposure to 1.3 equivalent of trifluoroacetic anhydride in refluxing toluene followed by desulfurization with Raney nickel in refluxing ethanol as above, both of the sulfoxides (22a~b) afforded the corresponding 1-substituted tetrahydroisoquinolines (23a~b) in yields of 82 and 81%, respectively. The resulting carbamates (23a~b) were then reduced with lithium aluminum hydride to give (±)-carnegine<sup>12</sup> (24) and (±)-laudanose<sup>13</sup> (25) in 79 and 80% yields, respectively (Scheme 5).



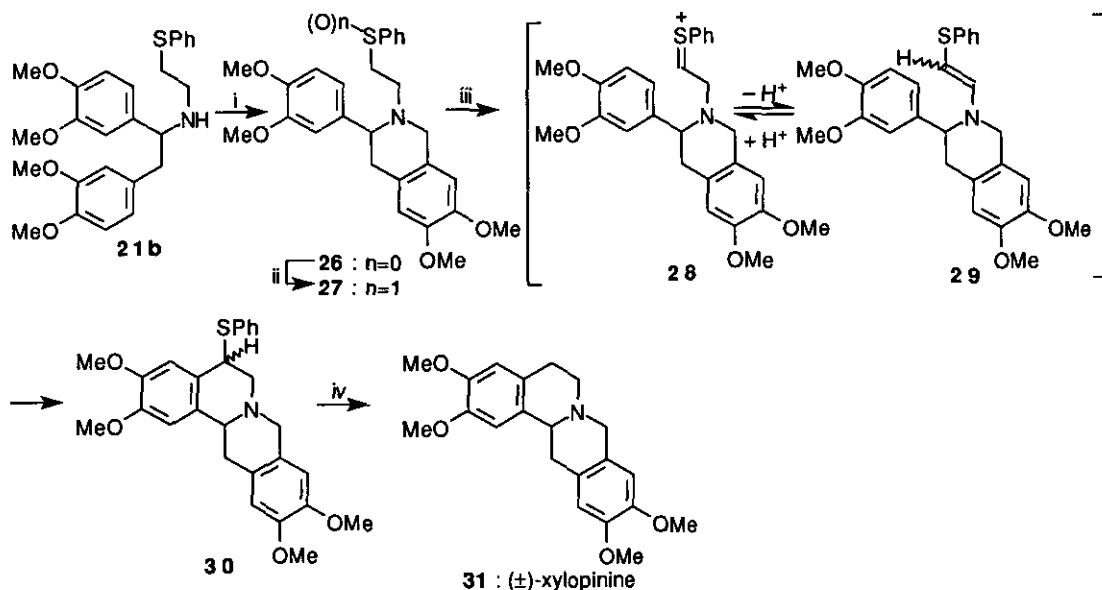
#### Scheme 5

*Reagents and conditions:* i) 4 (1.0 equiv.), *p*-TsOH (cat.), benzene, reflux, overnight, then NaBH<sub>4</sub> (4.0 equiv.), MeOH, room temperature, 5 h; ii) methyl chlorocarbonate (1.2 equiv.), pyridine (1.2 equiv.), CH<sub>2</sub>Cl<sub>2</sub>, 0 °C, 8 h, then *m*-CPBA (1.1 equiv.), NaHCO<sub>3</sub> (3.0 equiv.), CH<sub>2</sub>Cl<sub>2</sub>, -30 °C, 2 h; iii) TFAA (5.0 equiv.), toluene, reflux, 0.5 min then Raney Ni, EtOH, room temperature, overnight; iv) LiAlH<sub>4</sub> (9.0 equiv.), THF, room temperature, 3.5 h.

The secondary amine (21b), on the other hand, was treated with 30% formalin in refluxing methanol containing hydrochloric acid to give the 2,3-disubstituted tetrahydroisoquinoline (26) in 78% yield. Treatment of the resulting tertiary amine (26) with one equivalent of *m*-chloroperbenzoic acid in

dichloromethane at  $-30\text{ }^{\circ}\text{C}$  in the presence of sodium hydrogen carbonate allowed chemoselective oxidation to give the sulfoxide (**27**) in 60% yield.

Upon the same cyclization conditions above, the sulfoxide (**27**) did not furnish the expected tetracyclic amine (**30**), but the unstable vinyl sulfide (**29**) as a stereoisomeric mixture which, presumably, was generated from the sulfenium intermediate (**28**) preceding to the cyclization. We, therefore, added trifluoroacetic acid to the reaction medium to maintain the sulfenium structure (**28**) with expectation to give rise to the cyclization product (**30**) rather than the vinyl sulfide (**29**). As expected when the sulfoxide (**27**) was treated with three equivalents of trifluoroacetic anhydride in refluxing toluene containing three equivalents of trifluoroacetic acid, the cyclization occurred to give the tetracyclic sulfide (**30**) as a mixture of diastereomers which afforded ( $\pm$ )-xylopinine<sup>14</sup> (**31**) in 42% overall yield on desulfurization with Raney nickel in refluxing ethanol (Scheme 6).



**Scheme 6**

*Reagents and conditions:* i) 30% formalin (1.2 equiv.), aq. HCl (cat.), MeOH, reflux, overnight; ii) *m*-CPBA (1.0 equiv.), NaHCO<sub>3</sub> (5 equiv.), CH<sub>2</sub>Cl<sub>2</sub>,  $-30\text{ }^{\circ}\text{C}$ , 3.5 h; iii) trifluoroacetic anhydride (3.0 equiv.), trifluoroacetic acid (3.0 equiv.), toluene, reflux, 0.5 min; iv) Raney Ni, EtOH, reflux, overnight.

In conclusion the present method may be generally applicable for the construction of a variety of the isoquinoline alkaloids having 1,2,3,4-tetrahydroisoquinoline framework as a complement to the conventional Pomeranz-Fritch reaction<sup>15</sup> and the related isoquinoline formation reactions.

## REFERENCES

1. Y. Oikawa and O. Yonemitsu, *Tetrahedron Lett.*, 1972, 3393.
2. Y. Tamura, H. Maeda, S. Akai, K. Ishiyama, and H. Ishibashi, *Heterocycles*, 1982, **19**, 172.
3. C. Exon, T. Gallagher, and P. Magnus, *J. Am. Chem. Soc.*, 1983, **105**, 4739; K. Cardwell, B. Hewitt, M. Ladlow, and P. Magnus, *ibid.*, 1988, **110**, 2242.
4. S. Takano, K. Inomata, T. Sato, and K. Ogasawara, *J. Chem. Soc., Chem. Commun.*, 1989, 1591; S. Takano, K. Inomata, and K. Ogasawara, *ibid.*, 1990, 1544. See also: S. Takano, K. Inomata, T. Sato, M. Takahashi, and K. Ogasawara, *ibid.*, 1990, 290.
5. Alternative synthesis: S. Gabriel and J. Colman, *Ber.*, 1911, **44**, 3628.
6. I. Nakagawa and T. Hata, *Tetrahedron Lett.*, 1975, 1409; I. Nakagawa, K. Aki, and T. Hata, *J. Chem. Soc., Perkin Trans. 1*, 1983, 1315.
7. J. M. Bobbitt, D. N. Roy, A. Marchand, and C. W. Allen, *J. Org. Chem.*, 1967, **32**, 2225.
8. T. -H. Yang and C. -M. Chen, *J. Chinese. Chem. Soc. (Formosa)*, 1970, **17**, 541 (*Chem. Abstr.*, 1970, **73**, 99072s).
9. G. J. Kapadia, M. B. E. Fayeze, M. L. Sethi, and G. S. Rao, *J. Chem. Soc., Chem. Commun.*, 1970, 856.
10. S. M. Kupchan and A. Yoshitake, *J. Org. Chem.*, 1969, **34**, 1062; H. Richard and R. H. F. Manske, *Can. J. Res.*, 1937, **15B**, 159.
11. T. Kametani and K. Ohkubo, *Chem. Pharm. Bull.*, 1967, **15**, 608.
12. J. E. Hodkins, S. D. Brown, and J. L. Massingill, *Tetrahedron Lett.*, 1967, 1321.
13. A. H. Jackson and J. A. Martin, *J. Chem. Soc. (C)*, 1966, 2061.
14. M. Tomita, M. Kozuka, and S. Uyeo, *J. Pharm. Soc. Jpn.*, 1966, **86**, 460.
15. W. G. Gensler, *Org. Reactions*, 1951, **6**, 191.

Received, 12th October, 1992