GARCINOL, A NOVEL ARYLBENZOFURAN DERIVATIVE FROM GARCINIA KOLA

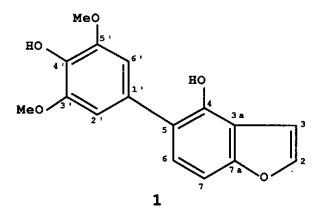
Masatake Niwa,^a* Kenji Terashima,^a and Muhd Aqil^b Faculty of Pharmacy, Meijo University,^a Tempaku, Nagoya 468, Japan, Department of Chemistry, Faculty of Science, University of Maiduguri,^b P. M. B. 1069, Maiduguri, Nigeria

Abstract ----- A novel arylbenzofuran named garcinol was isolated from Garcinia kola Heckel (Guttiferae) collected in Nigeria and the structure was elucidated by spectroscopic methods.

The plants of garcinia genus (Guttiferae) are known as a toxic plant and some of them are used as a traditional herb medicine in the areas of southeastern Asia.¹ But the chemical constituents are little studied. In the course of phytochemical studies of medicinal plants, we isolated a novel arylbenzofuran derivative named garcinol from roots of *Garcinia kola* Heckel (Guttiferae) collected in Nigeria. In this paper, we wish to describe the isolation and structure of garcinol (1).

The ethanol extract was subjected to medium pressure chromatography on silica gel using a mixture of chloroform and methanol (20 : 1) to give garcinol (1) as colorless liquid : $C_{16}H_{14}O_5$ [m/z 286.0848; calcd 286.0840]; uv (MeOH) λ 218 (ϵ 26600), 245 (ϵ 19800), 250 sh. (ϵ 14800), 268 sh. (ϵ 7700) nm; ir (film) v 3400 br. cm⁻¹; δ_H (CD₃OD) 3.86 (6H, s), 6.80 (2H, s, H-2', H-6'), 6.96 (1H, dd, J=2.2, 0.7 Hz, H-3), 7.04 (1H, dd, J=8.4, 0.7 Hz, H-7), 7.17 (1H, d, J=8.4 Hz, H-6), 7.62 (1H, d, J=2.2 Hz, H-2) ppm; δ_C (CD₃OD) 57.6 (OMe x 2), 105.0 (C-7), 105.9 (C-3), 109.2 (C-2', C-6'), 119.6 (C-3a), 124.5 (C-5), 129.0 (C-6), 131.9 (C-1'), 136.4 (C-4'), 145.8 (C-2), 149.0 (C-4), 149.8 (C-3', C-5'), 157.9 (C-7a) ppm.

In the ¹H-nmr spectrum of 1, two AB-type signals were observed. One pair (δ_H 7.04 and 7.17 ppm) having J=8.4 Hz was assigned to ortho-located aromatic protons on a benzene ring and the remaining pair ($\delta_{\rm H}$ 6.96 and 7.62 ppm) having J=2.2 Hz to vicinal-located protons on a furan ring. The small coupling constant (J=0.7 Hz) between a proton at δ_{H} 6.96 ppm and a proton at δ_{H} 7.04 ppm suggested the characteristic long range coupling between the 3- and 7-protons in a benzofuran.² Namely, garcinol was proposed to be 4,5-disubstituted benzofuran. Two substituent groups and their locations were determined as follows. The nmr data (δ_H 3.86 (6H, s) and 6.80 ppm (2H, s), and δ_{C} 57.6, 109.2 and 149.8 ppm) suggested the presence of a symmetrical partial structure; 3,5-dimethoxy-4-hydroxyphenyl group or 2,6dimethoxy-4-hydroxyphenyl group. The adoption of and the location (C-5) of the former group were determined by HMBC experiment.³ These were furthermore supported by NOE experiment. 22.5% enhancement of the signal at δ_H 6.80 (H-2' (6')) was observed by irradiation of the signal at δ_H 3.86 (OMe). On irradiation of the signal at δ_H 6.80, 19.2 and 5.0% enhancements of the signals at δ_H 3.86 and at δ_H 7.17 (H-6), respectively were observed. Consequently, the remaining substituent was a hydroxy group which was located at C-4 of the benzofuran.⁴ This conclusion was also supported by close resemblance of the $\delta_{\rm H}$ values of protons on the benzofuran ring between garcinol and 2'-hydroxy-4-methoxyfurano(3',4'-4",5")chalcone [δ_H 7.02 (dd, J=2.2, 0.8 Hz), 7.08 (dd, J=8.9, 0.8 Hz), 7.58 (d, J=2.2 Hz), and 7.86 (d, J=8.9 Hz)].4



Thus, the structure of garcinol should be 4-hydroxy-5-(3',5'-dimethoxy-4'hydroxyphenyl)benzofuran (1).

Garcinol (1) is the first example of natural 5-arylbenzofuran.

ACKNOWLEDGEMENT

We are grateful to Dr. C. Itoh, Meijo University for the HMBC measurement, and also grateful to Dr. S. Kosemura, Keio University for valuable discussion.

REFERENCES AND NOTE

- 1. "Chugoku Honzou Zukan", ed. by X. Qian, Yukon-Sha, Kyoto, 1986, Vol. 5, p. 80, and Vol. 7, p. 282.
- 2. J. A. Elvidge and R. G. Foster, J. Chem. Soc., 1963, 590.
- 3. In HMBC experiment, cross peaks were observed between H-2 and C-3, H-2 and C-3a, H-2 and C-7a, H-3 and C-2, H-3 and C-3a, H-3 and C-7a, H-6 and C-4, H-6 and C-5, H-6 and C-7a, H-7 and C-3a, H-7 and C-5, H-2' (6') and C-5, H-2' (6') and C-3' (5'), and H-2' (6') and C-4', respectively.
- 4. S. K. Talapatra, A. K. Mallik, and B. Talapatra, Phytochemistry, 1982, 21, 761.

Received, 12th November, 1992