

the reaction is worked up before the formation of lactone (IV) starts. By using this technique, the hydroxy ester (III) was afforded in 76-90% yield during 3 to 5 h of stirring (depends upon nature of aldehyde). PCC oxidation of the alcohol (III) was smoothly carried out to furnish the keto ester (V) in 85-90% yield. Conjugate addition of thiophenol in V gave phenylsulfide (IV) in 83-89% yield after chromatography on silica gel. The sulfide (VI) was treated with *N*-chlorosuccinimide and the crude chlorinated product thus obtained, was hydrolysed in aq. acetone⁵ containing CuCl_2 -CuO mixture to generate keto aldehyde (VII) which on stirring with H_2SO_4 formed 2-substituted 4-ethoxycarbonylfuran (I) in 60-72% yield.

EXPERIMENTAL (General Procedure):

Ethyl 4-alkyl/aryl-4-oxo-2-methylenobutanoate (V)

A mixture of aldehyde (II, 0.05 mol), ethyl bromoacrylate (9.65 g, 0.05 mol), Zn dust (3.60 g, 0.055 mol) was stirred in sat. NH_4Cl solution (100 ml) containing THF (20 ml) at 60°C. The reaction was monitored by tlc and was worked up when the starting aldehyde had completely reacted, by extraction with ether. The ether layer was washed with brine, dried over anhyd. Na_2SO_4 and the solvent was removed. The crude hydroxy ester (III) was dissolved in CH_2Cl_2 (25 ml) and added to PCC (0.057 mol) in CH_2Cl_2 (120 ml). It was stirred for 5 h and diluted with ether (100 ml). The organic layer was passed through silica gel column using ether as the eluent. The solvent was removed and the residue was distilled in vacuo (Va): bp 120-124°C/7mm; yield 88%. $\bar{\nu}_{\text{max}}$ (neat): 1720, 1680, 1665 cm^{-1} . Nmr (CDCl_3): δ 1.31 (3H, t, J=7 Hz), 3.88 (2H, s), 4.20 (2H, q, J=7 Hz), 5.81 (1H, m), 6.41 (1H, m), 7.50 (3H, m), 7.98 (2H, m). Anal. Calcd for $\text{C}_{13}\text{H}_{14}\text{O}_3$: C, 71.54; H, 6.47. Found: C, 71.26; H, 6.79. (Vb): bp 110-112°C/8-9mm; yield 90%. $\bar{\nu}_{\text{max}}$ (neat) 3037, 1725-1710, 1622, 1385, 1368 cm^{-1} . Nmr (CDCl_3): δ 0.92 (6H, d, J=6 Hz), 1.25 (3H, t, J=7 Hz), 2.05 (1H, m), 2.25 (2H, d, J=6 Hz), 3.28 (2H, s), 4.08 (2H, q, J=7 Hz), 5.63 (1H, m), 6.26 (1H, m). Anal. Calcd for $\text{C}_{11}\text{H}_{18}\text{O}_3$: C, 66.64, H, 9.15.

Found : C, 66.54, H, 9.18 (Vc) : bp 150-154°C/1mm; yield 88%. $\bar{\nu}_{\max}$ (neat) 1723, 1712 cm^{-1} . Nmr (CDCl_3) : δ 1.31 (6H, m), 2.38 (3H, s), 2.68 (2H, d, J=8 Hz), 3.05-3.68 (3H, m), 4.32 (2H, q, J=7 Hz), 5.52 (1H, m), 6.38 (1H, m), 7.21 (4H, s). Anal. Calcd for $\text{C}_{17}\text{H}_{22}\text{O}_3$: C, 74.42, H, 8.08. Found : C, 74.58, H, 8.25.

Ethyl 4-alkyl/aryl-4-oxo-2-phenylthiomethylbutanoate (VI)

To a solution of NaOC_2H_5 (2.31 g, 0.034 mol) in $\text{C}_2\text{H}_5\text{OH}$ (85 ml) was added $\text{C}_6\text{H}_5\text{SH}$ (13.94 ml, 0.136 mol). After 0.5 h, keto ester (V, 0.034 mol) in $\text{C}_2\text{H}_5\text{OH}$ (20 ml) was added and then stirred for 4 h. It was poured in ice water and extracted with CH_2Cl_2 and dried over anhyd. Na_2SO_4 . The solvent was removed and the residue was chromatographed on silica gel using *n*-hexane as eluent. (VIa) : yield 83%. $\bar{\nu}_{\max}$ (neat) 1745, 1680 cm^{-1} . Nmr (CDCl_3) : δ 1.18 (3H, t, J=7 Hz), 2.48-2.88 (5H, m), 4.18 (2H, q, J=7 Hz), 7.51 (8H, m), 7.98 (2H, m). Anal. Calcd for $\text{C}_{19}\text{H}_{20}\text{O}_3\text{S}$: C, 69.49; H, 6.14. Found : C, 69.38; H, 6.09. (VIb) : yield 88%. $\bar{\nu}_{\max}$ (neat) 3050, 1740, 1715, 1580, 1400, 1380, 742 cm^{-1} . Nmr (CDCl_3) : δ 0.90 (6H, d, J=6 Hz), 1.21 (3H, t, J=7 Hz), 2.05 (1H, m), 2.32 (4H, d, J=6 Hz), 2.68 (3H, m), 4.12 (2H, q, J=7 Hz), 7.32 (5H, m). Anal. Calcd for $\text{C}_{17}\text{H}_{24}\text{O}_3\text{S}$: C, 66.20; H, 7.84. Found : C, 66.18; H, 8.16. (VIc) : yield 89%. $\bar{\nu}_{\max}$ (neat) 1744, 1712 cm^{-1} . Nmr (CDCl_3) : δ 1.33 (6H, m), 2.36 (3H, s), 2.68-3.20 (8H, m), 4.12 (2H, q, J=7 Hz), 7.08-7.30 (9H, m). Anal. Calcd for $\text{C}_{23}\text{H}_{28}\text{O}_3\text{S}$: C, 71.84; H, 7.34. Found : C, 72.11; H, 7.62.

2-Alkyl/aryl-4-ethoxycarbonylfuran (I)

To a solution of thioether (VI, 0.039 mol) in CCl_4 (300 ml) was added NCS (7.48 g, 0.056 mol) at 0-5°C and the mixture stirred at 30°C for 3.5 h under N_2 atmosphere. It was filtered and the residue after removal of solvent from filtrate was dissolved in acetone (225 ml), H_2O (4.5 ml) followed by addition of $\text{CuCl}_2 \cdot 2\text{H}_2\text{O}$ (12.0 g, 0.07 mol) and CuO (12.0g, 0.15 mol). It was refluxed under N_2 atmosphere for 1.5 h and then filtered, the solvent from the filtrate was removed. The residue was taken in ether, washed with H_2O , dried over anhyd. Na_2SO_4 and the solvent was removed. The residue was chromatographed

on silica gel using n-hexane as eluent. To this keto aldehyde (VII) at 0-5°C was added conc. H₂SO₄ (5 ml) dropwise and then the mixture was stirred for 5 min at 50°C. It was extracted with ether and the ether layer was washed with H₂O and dried over anhyd. Na₂SO₄. The solvent was evaporated and the residue was chromatographed on silica gel using n-hexane as the eluent. (Ia) : bp 100-110°C/6-7mm; yield 72%. $\bar{\nu}$ max (neat) 1714 cm⁻¹. Nmr (CDCl₃) : δ 1.32 (3H, t, J=7 Hz), 4.32 (2H, q, J=7 Hz), 6.91 (1H, s), 7.38 (3H, m), 7.61 (2H, m), 7.98 (1H, s). Anal. Calcd for C₁₃H₁₂O₃ : C, 72.21; H, 5.59. Found : C, 77.09; H, 5.52. (Ib) : bp 85-90°C/7-8 mm; yield 65%. $\bar{\nu}$ max (neat) 1730, 1390, 1382 cm⁻¹. Nmr (CDCl₃) : δ 0.92 (6H, d, J=6 Hz), 1.28 (3H, t, J=7 Hz), 1.98 (1H, m), 2.44 (2H, d, J=7 Hz), 4.32(2H, q, J=7 Hz), 6.31 (1H, s), 7.84(1H, s). Anal. Calcd for C₁₁H₁₆O₃ : C, 67.32; H, 8.22. Found : C, 67.65; H, 8.55. (Ic) : yield 58%. $\bar{\nu}$ max (neat) 1725 cm⁻¹. Nmr (CDCl₃) : δ 1.28 (6H, m), 2.28 (3H, s), 2.82 (2H, m), 3.12 (1H, m), 4.26(2H, q, J=7 Hz), 6.20 (1H, s), 7.19 (4H, s), 7.83(1H, s). Anal. Calcd for C₁₇H₂₀O₃ : C, 74.97; H, 7.40. Found : C, 75.23; H, 7.64.

REFERENCES

1. 'Dictionary of Organic Compounds', Vol. I, ed. by J. Buckingham, Chapman and Hall, London, 1982, p. 670; Vol. II, p. 2184; Vol. III, p. 3366; VII Suppl., pp. 211 and references cited therein.
2. (a) M. Elliott, N. F. Janes, and B. C. Peason, J. Chem. Soc. (C), 1971, 2552 (limited to 4-oxo ester as the starting compound which are prepared by laborious routes). (b) H. Escalona and L. A. Maldonado, Syn. Comm., 1980, 10, 857 (not a general route). (c) S. S. Murphree, C.L. Muller, and A. Padwa, Tetrahedron Lett., 1990, 31, 6145 (the preparation is quite cumbersome).
3. P.A. Grieco, Synthesis, 1975, 67.
4. H. Mattes and C. Benezra, Tetrahedron Lett., 1985, 26, 5697.
5. P. Bakuzis, M.L.F. Bakuzis, and T.F. Weingartner, Tetrahedron Lett., 1978, 2371.

Received, 2nd December, 1992