

SYNTHESIS OF [1,2,3]TRIAZOLO[1,5-*a*][4,1]BENZOXAZEPINES *via* INTRAMOLECULAR AZIDE CYCLOADDITION

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Abstract - Starting from isatoic anhydride and propargyl alcohols, we developed a synthetic approach to the title compounds, where the key step is an intramolecular cycloaddition of the azido group onto the acetylenic bond.

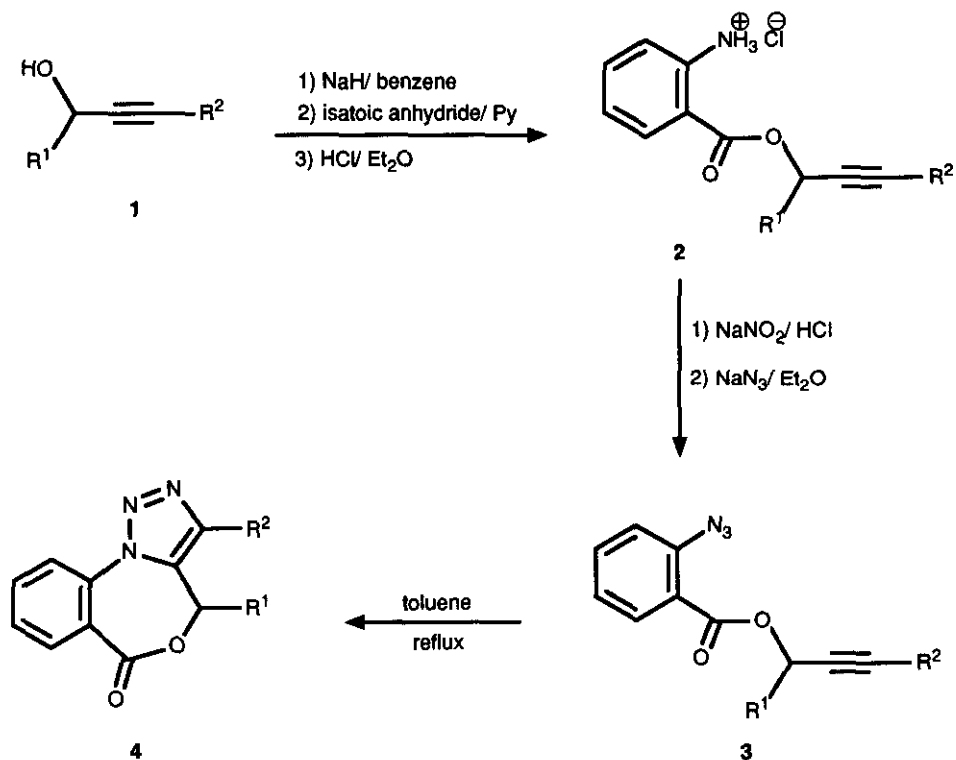
In the last decade, intramolecular 1,3-dipolar cycloadditions have attracted an increasing attention due to their versatile potential in heterocyclic syntheses.¹⁻³ On pursuing our research line in this field,⁴⁻⁷ we report here the synthesis of [1,2,3]triazolo[1,5-*a*][4,1]benzoxazepines (**4**) by intramolecular cycloaddition of the properly *ortho*-substituted aryl azides (**3**). Compounds (**4**) could be of pharmacological interest in view of the known anxiolytic activity of some [1,3,4]triazolo[1,2-*a*][1,4]benzodiazepines such as triazolam, alprazolam and estazolam.⁸

The reaction of isatoic anhydride with the sodium salts of propargyl alcohols (**1**) gave the corresponding alkynyl antranilates, which were characterised as hydrochlorides (**2**) (Table 1).

Diazotisation of the latter and subsequent treatment with sodium azide furnished the appropriate azido derivatives (**3**) (Table 2). The intramolecular cycloaddition step leading to the tricyclic products (**4**) was accomplished upon refluxing (**3**) in toluene. Reaction times and yields are given in Table 3, along with the characteristic data of the products.

It must be noted that compounds (**4**) are formed in a yield ranging from modest to excellent as a function of the substituents R¹ and R². Steric encumbrance by the latter reasonably hinders the intramolecular approach of the reactive π systems in parallel planes, as required for 1,3-dipolar cycloadditions.⁹ Whenever the intramolecular cycloaddition pathway is made impervious, the concurrent thermolysis of the azido group

can occur to generate a transient nitrene which evolves to disappointing side-products.



	a	b	c	d	e
R ¹	H	Me	Ph	H	H
R ²	H	H	H	Me	Ph

EXPERIMENTAL

Melting points were taken on a Büchi apparatus and are not corrected. Ir spectra were recorded on a FT IR Perkin Elmer 1725 X spectrophotometer. ¹H Nmr spectra were obtained on a Bruker 300 MHz apparatus; chemical shifts are given as ppm from tetramethylsilane.

Compound (1a-d) were commercially available. Compound (1e) was prepared according to the literature.¹⁰

Alkynyl Antranilates Hydrochlorides (2a-e); General Procedure. A solution of alcohol (1) (0.20 mol) in dry benzene (100 ml) was treated with 80% NaH (7.50g, 0.25 mol) and refluxed for 1 h. Isatoic anhydride (32.6g, 0.20 mol) in pyridine (100 ml) was added and the solution was refluxed for 4 h. The resulting mixture was poured into water and extracted with ether. The organic layer was washed with water, dried (Na_2SO_4), and evaporated. The oily residue was dissolved in anhydrous ether and a 4M solution of HCl in anhydrous ether (75 ml) was added dropwise under stirring. The precipitate was collected by filtration and washed with anhydrous ether to afford pure hydrochloride (2) (Table 1).

Table 1. Alkynyl Antranilates Hydrochlorides (2)

Compd	Yield (%)	mp (°C)	Microanalyses (%)				¹ H Nmr ^a δ, J (Hz)
			C calcd found	H calcd found	N calcd found	Cl calcd found	
2a	92	169	56.7	4.7	6.6	16.8	2.90 (1H, t, J=3), 5.11 (2H, d, J=3), 7.65-8.25 (4H, m)
			56.6	4.6	6.4	16.9	
2b	71	161	58.5	5.3	6.2	15.7	1.70 (3H, d, J=6), 2.98 (1H, d, J=2.5), 5.71 (1H, m), 7.50-8.50 (4H, m)
			58.5	5.4	6.0	15.6	
2c	51	160	66.8	4.9	4.9	12.3	2.70 (1H, d, J=2.5), 6.65 (1H, d, J=2.5), 7.10-8.10 (9H, m)
			66.9	5.0	4.8	12.1	
2d	91	162	58.5	5.3	6.2	15.7	1.85 (3H, t, J=2.5), 4.95 (2H, q, J=2.5), 7.45-8.85 (4H, m)
			58.4	5.1	6.3	15.9	
2e	88	168	66.8	4.9	4.9	12.3	5.15 (2H, s), 7.30-8.30 (9H, m)
			66.7	4.7	5.0	12.4	

^aSolvent: CD₃COOD for 2a, D₂O for 2b-e.

Alkynyl 2-Azidobenzoates (3a-e); General Procedure. Sodium nitrite (1.52g, 22 mmol) was added portionwise to a solution of **2** (11 mmol) in 1N aqueous HCl (55 ml) under stirring and cooling at 0°C. After addition of ether (100 ml), NaN₃ (3.58g, 55 mmol) was added portionwise under vigorous stirring and ice-cooling. After 30 min, the organic layer was separated, washed with 5% aqueous NaHCO₃, and dried (Na₂SO₄). The solvent was removed under reduced pressure and the residue was recrystallised from a mixture of diisopropyl ether and pentane to give pure **3** (Table 2).

Table 2. Alkynyl 2-Azidobenzoates (**3**)

Compd	Yield (%)	mp (°C)	Microanalyses (%)			Ir (Nujol) ν (cm ⁻¹)	¹ H Nmr (CDCl ₃) δ, J (Hz)
			C calcd found	H calcd found	N calcd found		
3a	86	65	59.7	3.5	20.9	2100	2.51 (1H, t, J=3), 4.90 (2H, d, J=3),
			59.8	3.6	21.0	1725	7.10-7.90 (4H, m)
3b^a	27	44	61.4	4.2	19.5	2110	1.60 (3H, d, J=6), 2.52 (1H, d, J=2.5),
			61.5	4.3	19.6	1730	5.65 (1H, m), 7.20-7.90 (4H, m)
3c^a	40	61	69.3	4.0	15.2	2110	2.71 (1H, d, J=2.5), 6.68 (1H, d, J=2.5),
			69.1	4.1	15.3	1730	7.10-7.90 (9H, m)
3d	77	46	61.4	4.2	19.5	2110	1.85 (3H, t, J=2.5), 4.90 (2H, q, J=2.5),
			61.2	4.1	19.3	1730	7.10-7.90 (4H, m)
3e	82	67	69.3	4.0	15.2	2120	5.15 (2H, s), 7.10-8.00 (9H, m)
			69.3	3.9	15.2	1735	

^aThis compound decomposed in part during the chromatographic separation.

[1,2,3]Triazolo[1,5-a][4,1]benzoxazepines (4a-e); General Procedure. A solution of **3** (4 mmol) in toluene (150 ml) was refluxed for the time indicated in Table 3, then the solvent was removed under reduced pressure. In the case of **3a,b**, the residue was recrystallised from diisopropyl ether to give pure **4a,b**. In the case of **3c-e**, the residue was chromatographed on a silica gel column with CH₂Cl₂-AcOEt 9:1 as eluant to give (**4c-e**) (Table 3).

Table 3. [1,2,3]Triazolo[1,5-a][4,1]benzoxazepines (4)

Compd	Time (h)	Yield (%)	mp ^a (°C)	Microanalyses (%)			Ir (Nujol) ν (cm ⁻¹)	¹ H Nmr (CDCl ₃) δ, J (Hz)
				C calcd found	H calcd found	N calcd found		
4a	12	82	194	59.7 59.8	3.5 3.6	20.9 21.0	1710	5.25 (2H, s), 7.90 (1H, s), 7.65-8.15 (4H, m)
4b	16	75	190	61.4 61.2	4.2 4.2	19.5 19.6	1710	1.90 (3H, d, J=6), 5.35 (1H, q, J=6), 7.82 (1H, s), 7.60-8.10 (4H, m)
4c	22	32	141	69.3 69.5	4.0 4.1	15.2 15.1	1725	6.25 (1H, s), 7.32 (1H, s), 7.40-8.20 (9H, m)
4d	24	50	155	61.4 61.5	4.2 4.1	19.5 19.5	1720	2.50 (3H, s), 5.22 (2H, s), 7.55-8.05 (4H, m)
4e	24	31	165	69.3 69.2	4.0 4.0	15.2 15.4	1730	5.40 (2H, s), 7.40-8.20 (9H, m)

^aRecrystallised from diisopropyl ether.

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