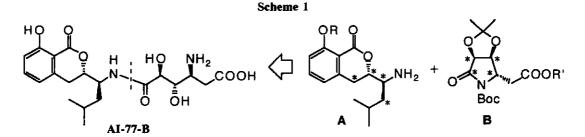
## A NEW CONCISE APPROACH TO THE ENANTIOSELECTIVE SYNTHESIS OF THE HYDROXYAMINO ACID MOIETY OF AI-77-B

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Abstarct---The hydroxyamino acid moiety of AI-77-B has been prepared from Dribose in an optically pure form via stereoselective alkylation of *N*-acylpyrrolidinium ion intermediates as the key step.

In the course of our synthetic efforts toward AI-77-B, a gastroprotective substance isolated from a culture broth of *Bacillus pumilus* AI-77,<sup>1</sup> we have recently disclosed an efficient entry to the amino-dihydroisocoumarin motety **A** by using chiral triflate technology<sup>2</sup> and also a new powerful method for the final stage to condense **A** with *N*-Boc-lactams such as **B** by applying high pressure chemistry (Scheme 1).<sup>3</sup>



Although there are several reports on the synthesis of AI-77-B including its fragments, 4-8 most of the works utilize naturally occurring amino acids as the chiral sources. Our strategy to the construction of **A** and **B** is based on an advantageous structure of D-ribose as an abundant starting material asterisks indicate that the carbon atoms should be derived from D-ribose. In this paper we wish to describe a simple and enantioselective approach to the *N*-Boc-lactam **B**.

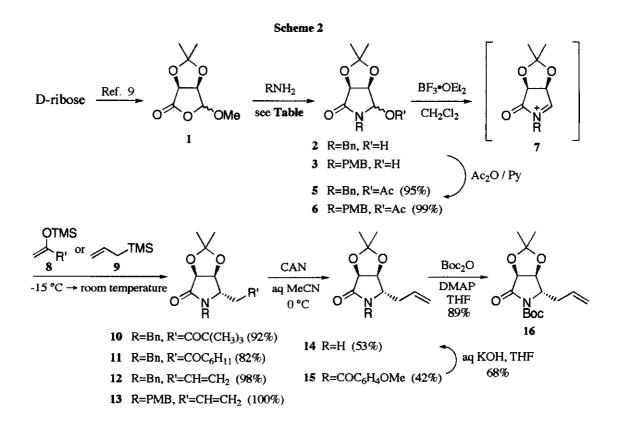


Table. Solvent effect on the formation of lactams from 1

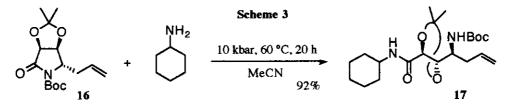
0″		PhCH <sub>2</sub> NH <sub>2</sub> 0 °C, 1 h	P N Bn 2 +	
	Solvent	Benzylamine (equiv.) <sup>a)</sup>	2.4	Total yield, %
	MeCN	18	4 · 6	92
	AcOEt	1.7	4:3	75
	Et <sub>2</sub> O	1.5	7 · 3	99
	EtOH	1.0	>10 : 0	95

a) Benzylamine was added until all of the starting material was consumed.

Scheme 2 illustrates the synthesis of 16 as a synthen of the fragment B. The lactol (1), readily available from Dribose according to the literature procedure,<sup>9</sup> was converted into lactams (2) and (3) by treatment with benzylamine or *p*-methoxybenzylamine. Interestingly, in this case, the product distribution was affected by the solvent used, as summarized in the Table. Thus, in ethanol the desired *N*-benzyl-lactam (2) was obtained exclusively. On the contrary, the use of the other solvent caused a significant side reaction giving aminal (4). The lactams (2) and (3) were converted into the corresponding acetates (5) and (6) to carry out the following alkylation.

The next stage of our plan for introducing an acetic acid unit onto the lactam ring was designed to exploit the intrinsic nature of 5 and 6 to form the N-acylpyrrolidinium ion intermediate (7) during Lewis acid-promoted alkylation.<sup>10</sup> In analogy to the literature method,<sup>11</sup> 10 and 11 were prepared from 5 by the action of 8 with  $BF_3$ -etherate<sup>12</sup> in a complete stereoselectivity. Unfortunately, however, all attempts to perform subsequent Baeyer-Villiger oxidation under a variety of conditions were unsuccessful. As a more concise route, we also examined the alkylation using ketene acetals such as  $CH_2=COTMS(OBu^t)$ , but no detectable amounts of the desired adducts were obtained. Then we turned our attention to the use of allylsilane (9) under similar conditions, providing 12 and 13 in almost quantitative yields. The stereochemical course of this reaction was clearly confirmed by the nmr data,  $J_{4,5} = 0$  Hz, indicating the reaction took place from the convex side.

Although deprotection of the benzyl group of 12 was not so easy, oxidative cleavage of the PMB group in 13 using ceric ammonium nitrate  $(CAN)^{13}$  gave smoothly 14 together with the N-benzoyl derivative (15) which was further hydrolyzed to 13. Finally, transformation of 14 to N-Boc-lactam (16) was proceeded cleanly under normal conditions. In order to realize the feasibility of our general procedure to convert N-Boc-lactams into  $\omega$ -amino-carboxamides at high pressure,<sup>3</sup> 16 was subjected to aminolysis using cyclohexylamine (1 equiv.) at 10 kbar and 60 °C for 20 h to furnish 17 in 92% yield (Scheme 3).



In conclusion, the hydroxyamino acid fragment (16) was assembled from 1 in 72% overall yield via 5-step sequence, in which D-ribose was employed as a convenient chiral source. Further synthetic studies on AI-77-B were now in progress in our laboratory.

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## REFERENCES AND NOTES

- Y. Shimojima, H. Hayashi, T. Ooka, M. Shibukawa, and Y. Iitaka, *Tetrahedron Lett.*, 1982, 23, 5435;
  Y. Shimojima, H. Hayashi, T. Ooka, and M. Shibukawa, *Agric. Biol. Chem.*, 1982, 46, 1823;
  Y. Shimojima, H. Hayashi, T. Ooka, M. Shibukawa, and Y. Iitaka, *Tetrahedron*, 1984, 40, 2519.
- 2. H. Kotsuki, A. Miyazaki, and M. Ochi, Chem. Lett., 1992, 1255.
- 3. H. Kotsuki, M. Iwasaki, and H. Nishizawa, Tetrahedron Lett., 1992, 33, 4945.
- A. Kawai, O. Hara, Y. Hamada, and T. Shioiri, *Tetrahedron Lett.*, 1988, 29, 6331; Y. Hamada, A. Kawai, Y. Kohno, O. Hara, and T. Shioiri, *J. Am. Chem. Soc.*, 1989, 111, 1524; Y. Hamada, A. Kawai, T. Matsui, O. Hara, and T. Shioiri, *Tetrahedron*, 1990, 46, 4823; Y. Hamada, O. Hara, A. Kawai, Y. Kohno, and T. Shioiri, *ibid.*, 1991, 47, 8635.
- 5. J. P. Gesson, J. C. Jacquesy, and M. Mondon, Tetrahedron Lett., 1989, 30, 6503.
- 6. S. D. Broady, J. E. Rexhausen, and E. J. Thomas, J. Chem. Soc., Chem. Commun., 1991, 708.
- 7. R. A. Ward and G. Procter, Tetrahedron Lett., 1992, 33, 3359.
- 8. J.-M. Durgnat and P. Vogel, Helv. Chim. Acta, 1993, 76, 222.
- 9. S. M. Ali, K. Ramesh, and R. T. Borchardt, Tetrahedron Lett., 1990, 31, 1509.
- H. E. Zaugg, Synthesis, 1984, 85 and 181; W. N. Speckamp and H. Hiemstra, Tetrahedron, 1985, 41, 4367.
- 11. A. B. Smith, III, B. A. Salvatore, K. G. Hull, and J. J.-W. Duan, Tetrahedron Lett., 1991, 32, 4859.
- 12. The use of other Lewis acid such as TiCl<sub>4</sub> gave rise to only deacetylation of the substrates.
- 13. J. Yoshimura, M. Yamaura, T. Suzuki, and H. Hashimoto, Chem. Lett., 1983, 1001.

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