UNEXPECTED RESULTS IN THE REDUCTION OF TETRACYCLIC ENAMIDES. STRUCTURE, STEREOCHEMISTRY AND CONFORMATION OF A 208-HYDROPEROXYIMINE

Denise Dugat,* Gérard Dauphin, and Jean-Claude Gramain

Synthèse et Etude de Systèmes à Intérêt Biologique, URA CNRS 485, Université Blaise Pascal de Clermont-Ferrand, 63177 Aubière, France

Abstract - Reduction of N_b -allyl-20-ethyltetracyclic enamide (7) with lithium aluminium hydride afforded the expected enamine (12) and 4a-oxoethylhexahydrocarbazolones (18a,b). Reduction of N_b -unsubstituted enamide (6), under the same conditions, gave imine (9), enamine (11) and a 20 β -hydroperoxytetracyclic imine (19) resulting from the peroxydation of 9 and 11. The C-20 stereochemistry of 19 was established by means of 2D ¹H nmr and NOE difference spectroscopy.

Alkaloids of the *Aspidosperma* genus are biologically important¹ and their pharmacological activity has stimulated a great deal of work on their synthesis. In our program devoted to studies in this series,^{2,3} tetracyclic enamides (4)² and (5)³ were used as key intermediates in a total synthesis of N_a -benzyl-20-desethylaspidospermidine (1). Reduction of 4 with LiAIH₄ afforded imine (8)² while reduction of 5 afforded enamine (10)³ (Scheme 1). However most of the *Aspidosperma* alkaloids possess an ethyl chain in C-20⁴ [*e.g.* aspidospermidine (2)]. One possible strategy involves the introduction of the 20-ethyl group at the first step of the synthesis *i.e.* into the original *N*-benzylenaminone from which hexahydrocarbazolones (13a,b) were synthesized.⁵ We report herein our results on the reduction of 20-ethyl tetracyclic enamides (6) and (7).

Compounds (6) and (7) were prepared from 13a,b which were isolated in a 30:70 ratio as previously described.⁵ The thermodynamic enolates of 13a,b were alkylated with iodoacetamide or *N*-allyliodoacetamide according to published methods.^{2,3} The reaction afforded a mixture of 4a-carbamoylmethylhexahydrocarbazolones (14a) and tetracyclic hydroxyamides (15a,b) in a 10:90 ratio and 70% yield in the *N*_b-unsubstituted series and a mixture of (16a,b) and (17a,b) in a 60:40 ratio and 69% yield in the *N*_b-allyl series (Scheme 2). In contrast to the results observed in the absence of a C-3 substituent (R¹ = H),^{2,3} the cyclization of 14 and 16 to 15 and 17 is not completely spontaneous. Compounds (16a) and (16b) were separated easily while 15a,b and 17a,b were both isolated as a mixture of C-20 isomers. The stereochemistry of the B/C ring junction has been found to be *cis* for compounds (14-17^{2,5-7} and thermodynamic considerations led us to assign a *cis* C/E ring junction for 15 and 17 as previously mentioned in the series R¹ = H.^{2,3} Moreover 15, 16 and 17 were obtained as a mixture of isomers at C-20. Although the C-3 (future C-20) stereochemistry of the ethyl chain would have to be destroyed in the subsequent dehydration step, we established it for 14a, 16a and 16b. This stereochemistry is supported by spectroscopic data in agreement with a previous article.⁵ Thus, in ¹H nmr ³Jga-1ax and ³Jga-1eq are almost identical for the β-ethyl compound (16b) (≈ 4 Hz) and different for the α-ethyl derivatives (14a) and (16a) (≈ 9 and 6 Hz). Concerning amido alcohols (15a,b) and (17a,b), the C-20

stereochemistry was attributed tentatively by comparison with some other tetracyclic compounds, in particular with tetracyclic nitrones and amines.⁸ In the ¹H and ¹³C nmr, Me-18 of tetracyclic derivatives appears at higher field for β -ethyl isomers than for α -ethyl isomers.

Scheme 1



Acidic treatment of **15a,b** and **17a,b** (CH₂Cl₂, reflux, camphorsulfonic acid, molecular sieves) led to enamides (6) and (7) respectively, in good yield (\approx 90%). Formation of 6 from **14a** and formation of 7 from **16a** or **16b** required stronger conditions (C₆H₆, reflux, camphorsulfonic acid, Dean Stark).





Reduction of **6** and **7** was performed under the usual conditions^{2,3,9} (LiAlH₄, THF, 65°C). In the *N*_b-allyl series, the reaction gave the expected and unstable enamine (**12**) in a yield estimated to be 25-30% and afforded essentially 4a-oxoethylhexahydrocarbazolones (**18a,b**) (25:75) as a mixture of isomers at C-3 (isolated yield 62%) (Scheme 3). Compounds **18a** and **18b** are characterized by a singlet at $\delta \approx 9.6$ ppm (aldehydic H) in the ¹H nmr spectra and by a signal at $\delta \approx 200$ ppm (formyl CO) in the ¹³C nmr spectra. The α or β -ethyl stereochemistry was established as previously indicated for **14** and **16**. In both compounds, the ethyl group is quasi-equatorial and the two isomers correspond to the two chair forms of the C ring.⁵

Scheme 3



In the *N_b*-unsubstituted series, the reaction led to the unstable enamine (11) in equilibrium with its imine tautomer (9) and to a tetracyclic hydroperoxyimine (19) in a range of 30-50% yield depending on the reaction conditions (air or inert atmosphere) (Scheme 3). Compound (19) was isolated as a single isomer. It shows OH and C=N absorptions at 3620 and 1640 cm⁻¹ respectively in the ir spectrum and a C-21 signal (C=N) at δ 177.4 ppm in the ¹³C nmr spectrum. Its C-20 stereochemistry was established by 2D ¹H nmr and NOE difference spectroscopy. The sequence H-2, H-16ax, H-17ax with large coupling constants (*J* = 8.0-9.5 Hz) was deduced from a ¹H-¹H COSY experiment, and it is characteristic of a chair C ring conformation with an axial H-2 proton. Moreover irradiation of the H-17ax hydrogen (1.45 ppm) led to a nuclear Overhauser effect on the CH₃-18 group (0.95 ppm, 8%) and irradiation of OOH (7.08 ppm) produces a nuclear Overhauser effect on H-9 proton (6.96 ppm, 7%). Such effects which would occur exclusively on the isomer with an axial OOH enable us to conclude unambiguously on a β-OOH stereochemistry for carbon C-20 (Scheme 3). Distances of ≈ 2.0-2.5 Å can be approximately measured, on molecular models, between the concerned hydrogens. The distance H₉-H_{OOH} has been estimated in a favored conformation which allowed the formation of an hydrogen bond between the hydrogen H_{OOH} and the nitrogen N_b.

DISCUSSION

* Formation of 4a-oxoethyl hexahydrocarbozolones (18a,b)

The mechanism of amide reduction postulated by Weygand^{10,11} invokes a tetrahedral intermediate type **20** or **21**. This presumed complex can react further by different routes : (i) a nucleophilic attack by a second hydride on the carbonoxygen bond will form imine (**9**) and enamine (**11**) from **20** or enamine (**12**) from **21** (path *a*) ; (ii) a hydrolysis will prevent the second step of the reduction and convert the intermediate (**20**) or (**21**) into the keto aldehydes (**18a,b**) *via* a presumed enamino aldehyde (**22**) (path *b*) (Scheme 4).Thus, aldehydes (**18a,b**) corresponds to partial reduction derivatives. In the N_b -unsubstituted series, the reaction evolves exclusively by path *a* as usually observed on reduction of secondary amides.¹¹ It leads to the mixture of imine (9) and enamine (11). In contrast, both pathways are followed in the N_b -allyl series as mentioned for the reduction of tertiary amides.¹¹ The reaction gives rise to enamide (12) and 4a-oxoethylhexahydrocarbazolones (18a,b).

Scheme 4



22 : $R^2 = allyl$

18a,b

* Formation of the hydroperoxylmine (19)

The literature relates few examples of the introduction of an hydroperoxy linkage α to an imine function. Thus, autoxidation of cyclic imines or enamines by molecular oxygen is known to give the corresponding hydroperoxyimine.^{12 - 14} On another hand, it has been observed that molecular oxygen intervenes in the metal hydride reduction of compounds containing carbon-nitrogen double bonds; thus reduction by LiAlH₄ of an arylketimine is reported to lead to the corresponding α -hydroxyamine *via* an hydroperoxy intermediate¹⁵ (Scheme 5).

In the case concerning us, cyclic imine (9) and enamine (11) undergo reduction with hydride. The resulting metal salt is very reactive to molecular oxygen as was the Stevens and Gasser's analogue.¹⁵ It gives hydroperoxide (19) whose fair stability in a crystalline state is probably due to its tertiary nature in a crowded polycyclic molecule and to the existence of intramolecular hydrogen bonds.

Scheme 5



CONCLUSION

Reduction of tetracyclic enamides (6) ($R^2 = H$) and (7) ($R^2 = allyl$) by LiAlH₄ gives unexpected results. In the N_b-allyl series, formation of 4a-oxoethylhexahydrocarbazolones (**18a,b**) prevents further investigations towards the building of the pentacyclic framework. In contrast isolation of the hydroperoxyimine (**19**), in the N_b-unsubstituted series, represents an interesting result. This compound, with an appropriate side chain on carbon C-20 [$R^1 = (CH_2)_3X$] could be a key intermediate in the synthesis of alkaloids which possess a 20β-hydroxy group such as deoxyaspidodispermine (**3**).

EXPERIMENTAL

Organic layers were dried over MgSO₄. Thin layer chromatography was performed with Merck silica gel 60 F254 and flash column chromatography was carried out with Merck silica gel 0.040-0.063 nm. Melting points were taken on a Reicher hot-stage microscope and are uncorrected. Infrared spectra were run on a Perkin-Elmer 377 spectrophotometer (v values in cm⁻¹). Mass spectra were measured on a Varian CH5 or on a Varian VG 30F apparatus under electronic impact (ei). 13C Nmr spectra were recorded on Jeol FX60 or Bruker MSL 300 spectrometers and 1H nmr spectra on Jeol C60H, Bruker MSL 300 or Bruker AC 400 instruments (δ values in ppm, J values in Hz). The applied pulse sequence was ($\pi/2$), t₁, ($\pi/4$), FID, t₂, for the 1H nmr COSY spectrum. Homonuclear Overhauser effects were generated by presaturating selected proton signals with a low-power 4s decoupler pulse. NOE difference spectra were obtained by substracting alternately right-off resonance-free induction decays (FIDs) from right-on resonance-induced FIDs.

cis-3-Ethyl-4a-(carbamoylmethyl)-9-benzylhexahydrocarbazol-4-one (14a)

21-Hydroxy-20-ethyl-5-oxo-1-benzyloctahydropyrrolocarbazoles (15a,b)

A solution of **13a,b** (30:70) (915 mg, 3.0 mmol) in THF (18 ml) was added dropwise to a suspension of KH (33% in oil, 468 mg, 3.9 mmol) in THF (9 ml) and stirred at room temperature for 15 min under an atmospher of argon. The resulting medium was added to a solution of iodoacetamide (666 mg, 3.6 mmol) in THF (18 ml). The mixture was stirred again for 30 min, then water was added and the majority of THF was distilled. The aqueous phase was extracted with AcOEt. The

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organic layer was dried and concentrated. The crude product was purified by flash chromatography with 60:40 hexane-AcOEt to give **14a** (76 mg, 7%) and **15a,b** (684 mg, 63%, ratio 75:25).

14a : white crystals; mp 155-157°C (hexane-AcOEt); ir (CHCl₃) 3505, 3395 (NH), 1685, 1635 (C=O), 1595;-¹H nmr (CDCl₃, 300 MHz) δ 0.80 (t, 3H, Me-11, *J* = 7.0), 1.35 (ddq, 1H, H-10B, *J* = 14.0, 7.0, 7.0), 1.47 (m, 2H, H-1ax, H-2ax), 1.75-1.90 (m, 3H, H-10A, H-1eq, H-2eq), 2.00 (m, 1H, H-3), 2.76 (AB, 2H, CH₂CO, *J* = 16.0, Δv = 128.0), 4.06 (dd, 1H, H-9a, *J* = 9.5, 5.5), 4.25 (AB, 2H, NCH₂Ph, *J* = 15.5, Δv = 49.0), 5.45 (d, 2H, NH₂, *J* = 34.0), 6.30 (d, 1H, H-8, *J* = 8.0), 6.56 (t, 1H, H-6, *J* = 8.0), 6.81 (d, 1H, H-5, *J* = 8.0), 7.00 (t, 1H, H-7, *J* = 8.0), 7.30 (m, 5H, ArH); ¹³C nmr (CDCl₃, 75 MHz) δ 11.4 (CH₃-11), 22.4 (C-10), 24.1 (C-1), 25.4 (C-2), 41.9 (*CH*₂CO), 48.7 (C-3), 49.2 (NCH₂Ph), 58.6 (C-4a), 70.1 (C-9a), 107.4 (C-8), 117.9 (C-6), 123.3 (C-5), 127.3 (C-7), 127.5-129.2 (5 Ar CH), 129.6 (C-4b), 138.3 (C-1'), 149.8 (C-8a), 173.3 (CONH₂), 209.6 (CO); ms (*m*/z, %) 362 (M^{+*}, 6), 304 (30), 290 (1), 275 (18), 246 (11), 234 (48), 220 (44), 198 (10), 169 (8), 156 (14), 144 (11), 119 (17), 117 (21), 91 (100), 85 (16), 77 (16), 65 (16), 47 (37), 35 (29); exact mass calcd for C₂₃H₂₆N₂O₂ 362.1994, found 362.1991. Anal. Calcd for C₂₃H₂₆N₂O₂: C, 76.21; H, 7.23; N, 7.73. Found: C, 75.97; H, 7.38; N, 7.45.

15a,b: white solid; mp 100-105°C (ether); ir (CHCl₃) 3510 (OH), 3400 (NH), 1690 (CO), 1600; ¹H nmr (CD₃ COCD₃, 400 MHz) δ 0.75 and 0.90* (major isomer peak is indicated by an asterisk) (2t, 3H, Me-18, J = 7.5), 1.05-2.10 (m, 7H, 2H-16, 2H-17, 2H-19, H-20), 2.58* (AB, 2H, 2H-6, J = 15.0, Δv = 96.0) and 2.60 (AB, 2H, 2H-6, J = 17.0, Δv = 66.5), 2.90 (s, 1H, OH), 3.45* (t, 1H, H-2, J = 4.5) and 3.54 (dd, 1H, H-2, J = 7.0, 5.0), 4.28 (AB, 2H, NCH₂Ph, J = 16.5, Δv = 78.3) and 4.30* (AB, 2H, NCH₂Ph, J = 16.5, Δv = 129.0), 6.37 and 6.45* (2d, 1H, H-12, J = 7.8), 6.62 and 6.66* (2t, 1H, H-10, J = 7.8), 6.90 and 7.03* (2t, 1H, H-11, J = 7.8), 7.22 and 7.23* (2d, 1H, H-9, J = 7.8), 7.35 (m, 5H, ArH), 9.10 (br s, 1H, NH); ¹³C nmr (CD₃COCD₃, 15 MHz) δ 11.9 and 12.7* (CH₃-18), 21.2* and 23.3 (C-19), 21.5* and 25.0 (C-16), 23.0* and 26.2 (C-17), 42.6* and 46.3 (C-6), 47.2* and 49.9 (C-20), 50.3 and 50.9* (NCH₂Ph), 54.9* and 58.1 (C-7), 69.1* and 70.1 (C-2), 90.8 and 90.9* (C-21), 107.0 and 108.9* (C-12), 117.6 and 119.1* (C-10), 124.8* and 124.9 (C-9), 126.7 and 127.7* (C-11), 128.2-129.5 (5 Ar CH), 131.0 and 131.7* (C-8), 139.9 (C-1'), 152.1 and 153.5* (C-13), 173.3 and 174.3* (CO); ms (*m/z*, %) 362 (M^{+*}, 19), 344 (10), 304 (23), 290 (5), 275 (10), 253 (7), 246 (6), 234 (12), 220 (10), 130 (6), 91 (100), 65 (8), 41 (6); exact mass calcd for C₂₃H₂₆N₂O₂ 362.1994, found 362.1991.

c/s-3-Ethyl-4a-(N-allylcarbamoylmethyl)-9-benzylhexahydrocarbazol-4-one (16a,b)

21-Hydroxy-20-ethyl-5-oxo-4-allyl-1-benzyloctahydropyrrolocarbazoles (17a,b)

These compounds were prepared from hexahydrocarbazolones (**13a**,**b**) (30:70) (915 mg, 3.0 mmol) and *N*-allyliodoacetamide (810 mg, 3.6 mmol) following the conditions described for the preparation of **14a** and **15a**,**b**. Flash chromatography on silica gel, with 80:20 hexane-AcOEt, gave **16a** (274 mg, 23%), **16b** (225 mg, 19%) and **17a**,**b** (333 mg, 27%, ratio 30:70).

16a: white solid; mp 124-125°C (cyclohexane); ir (CCl₄) 3445 (NH), 1675 (CO), 1590; ¹H nmr (CDCl₃, 300 MHz) δ 0.80 (t, 3H, Me-11, *J* = 7.5), 1.33 (ddq, 1H, H-10B, *J* = 14.0, 7.5, 7.5), 1.46 (m, 2H, H-1ax, H-2ax), 1.70-1.90 (m, 3H, H-1eq, H-2eq, H-10A), 2.0 (m, 1H, H-3), 2.75 (AB, 2H, CH₂CO, *J* = 17.0, Δv = 135.0), 3.70 (t, 2H, N*CH*₂-CH=CH₂, *J* = 5.5), 4.10 (dd, 1H, H-9a, *J* = 9.0, 5.0), 4.27 (AB, 2H, NCH₂Ph, *J* = 16.0, Δv = 47.0), 5.03 (m, 2H, NCH₂-CH=*CH*₂), 5.50 (br s, 1H, NH), 5.70 (m, 1H, NCH₂-*CH*=CH₂), 6.30 (d, 1H, H-8, *J* = 8.0), 6.56 (t, 1H, H-6, *J* = 8.0), 6.80 (d, 1H, H-5, *J* = 8.0), 6.98 (t, 1H, H-7, *J* = 8.0), 7.30 (m, 5H, ArH); ¹³C nmr (CDCl₃, 15 MHz) δ 11.4 (CH₃-11), 22.4 (C-10), 24.0 (C-1), 25.3 (C-2), 41.8 (*CH*₂CO), 42.4 (N*CH*₂-CH=CH₂), 48.7 (C-3), 49.2 (NCH₂Ph), 58.7 (C-4a), 70.0 (C-9a), 107.4 (C-8), 116.2 (NCH₂-CH=*CH*₂), 117.9 (C-6), 123.3 (C-5), 128.6 (C-7), 127.5-129.1 (5 Ar CH), 129.8 (C-4b), 134.2 (NCH₂-*CH*=CH₂), 138.4 (C-40)

1'), 149.8 (C-8a), 170.6 (CONH), 211.1 (CO); ms (m/z, %) 402 (M^{+*}, 9), 304 (90), 275 (8), 246 (6), 234 (12), 220 (8), 130 (7), 91 (100); exact mass calcd for C₂₆H₃₀N₂O₂ 402.2307, found 402.2321. Anal.Calcd for C₂₆H₃₀N₂O₂ : C, 77.58; H, 7.51; N, 6.96. Found: C, 78.01; H, 7.59; N, 6.66.

16b: yellow crystals; mp 114-115°C (cyclohexane); ir (CCl₄) 3440 (NH), 1680 (CO), 1590; ¹H nmr (CDCl₃, 300 MHz) δ 0.80 (t, 3H, Me-11, *J* = 7.5), 1.17 (ddq, 1H, H-10B, *J* = 14.0, 7.0, 7.5), 1.57-1.70 (m, 2H, H-10A, H-2ax), 1.77-1.92 (m, 3H, H-2eq, 2H-1), 2.30 (dddd, 1H, H-3, *J* = 9.0, 7.0, 7.0, 6.0), 2.82 (AB, 2H, CH₂CO, *J* = 16.0, Δv = 196.0), 3.76 (t, 2H, NCH₂-CH=CH₂, *J* = 5.5), 4.16 (dd, 1H, H-9a, *J* = 4.5, 3.5), 4.38 (AB, 2H, NCH₂Ph, *J* = 16.0, Δv = 47.0), 5.10 (m, 2H, NCH₂-CH=CH₂), 5.68 (br s, 1H, NH), 5.75 (m, 1H, NCH₂-*CH*=CH₂), 6.35 (d, 1H, H-8, *J* = 8.0), 6.56 (t, 1H, H-6, *J* = 8.0), 6.88 (d, 1H, H-5, *J* = 8.0), 7.04 (t, 1H, H-7, *J* = 8.0), 7.30 (m, 5H, ArH); ¹³C nmr (CDCl₃, 300 MHz) δ 11.7 (CH₃-11), 22.5 (C-10), 25.0 (C-1), 26.4 (C-2), 41.8 (*CH*₂CO), 45.5 (N*CH*₂-CH=CH₂), 49.8 (NCH₂Ph), 50.2 (C-3), 58.3 (C-4a), 69.1 (C-9a), 106.4 (C-8), 116.2 (NCH₂-CH=*CH*₂), 117.2 (C-6), 124.2 (C-5), 127.3-129.3 (5 Ar CH), 128.1 (C-4b), 134.2 (NCH₂-*CH*=CH₂), 138.6 (C-1'), 151.3 (C-8a), 170.3 (CONH), 211.9 (CO); ms (*m*/*z*, %) 402 (M^{+•}, 12), 304 (100), 290 (3), 275 (8), 246 (6), 234 (12), 220 (7), 198 (2), 130 (7), 91 (70); exact mass calcd for C₂₆H₃₀N₂O₂ 402.2307, found 402.2301. Anal. Calcd for C₂₆H₃₀N₂O₂: C, 77.58; H, 7.51; N, 6.96. Found: C, 77.44; H, 7.55; N, 6.82.

17a,b: white solid; mp 144-149°C (ether); ir (KBr) 3315 (br, OH) 1670 (CO), 1600; ¹H nmr (CD₃COCD₃, 300 MHz) δ 0.84* and 0.88 (major isomer peak is indicated by an asterisk) (2t, 3H, Me-18, J = 7.5), 1.17-2.05 (m, 7H, 2H-16, 2H-17, 2H-19, H-20), 2.60 (AB, 2H, 2H-6, J = 17.5, $\Delta v = 18.0$) and 2.61* (AB, 2H, 2H-6, J = 16.5, $\Delta v = 91.0$), 2.86 (s, 1H, OH), 3.32* (dd, 1H, H-2, J = 7.0, 5.0) and 3.58 (dd, 1H, H-2, J = 4.0, 3.0), 3.91 (m, 2H, NCH₂-CH=CH₂), 4.32 (AB, 2H, NCH₂Ph, J = 16.0, $\Delta v = 90.0$) 5.13 (m, 2H, NCH₂-CH=*CH₂*), 5.90 (m, 1H, NCH₂-*CH*=CH₂), 6.45* and 6.47 (2d, 1H, H-12, J = 8.0), 6.65* and 6.70 (2t, 1H, H-10, J = 8.0), 7.03* and 7.05 (2t, 1H, H-11, J = 8.0), 7.12 and 7.28* (2d, 1H, H-9, J = 8.0), 7.40 (m, 5H, ArH); ¹³C nmr (CD₃COCD₃, 75 MHz) δ 12.5* and 12.8 (CH₃-18), 19.2 and 20.8* (C-19), 20.0 and 21.3* (C-16 and C-17), 42.1* and 42.6 (C-6), 42.9 (NCH₂-CH=CH₂), 43.6 and 44.0* (C-20), 50.1* and 51.3 (NCH₂Ph), 53.2 and 53.9* (C-7), 69.9 and 70.2* (C-2), 93.6 and 94.6* (C-21), 108.4* and 108.9 (C-12), 116.8* and 117.0 (NCH₂-CH=*CH₂*), 118.8* and 119.1 (C-10), 124.9 and 125.5* (C-9), 127.7* and 127.9 (C-11), 128.2-129.7 (5 Ar CH), 131.5* and 131.6 (C-8), 136.0 (NCH₂-*CH*=CH₂), 139.7* and 140.0 (C-1'), 152.5* and 153.8 (C-13), 172.9 and 173.8* (CO); ms (*m*/*z*, %) 402 (M^{+*}, 25), 331 (5), 304 (46), 275 (5), 246 (6), 234 (9), 220 (21), 198 (4), 144 (5), 130 (9), 91 (100), 65 (7), 41 (12). 32 (29), 28 (98); exact mass calcd for C2₆H₃₀N₂O₂ 402.2307, found 402.2313. Anal. Calcd for C2₆H₃₀N₂O₂: C, 77.58; H, 7.51; N, 6.96. Found: C, 77.49; H, 7.54; N, 6.88.

20-Ethyl-5-oxo-1-benzylhexahydropyrrolocarbazole (6)

Method A : A solution of **15a,b** (615 mg, 1.7 mmol) and (\pm)-10-camphorsulfonic acid (10 mg) in CH₂Cl₂ (85 ml) was refluxed in the presence of molecular sieves (4 Å) (17 g) for 8 h. The reaction mixture was then neutralized at room temperature with solid K₂CO₃. After filtration and concentration of the organic phase, compound (**6**) (550 mg, 94%) was purified by crystallization from AcOEt.

Method B : A solution of **14a** (72 mg, 0.2 mmol) and (±)-10-camphorsulfonic acid (2 mg) in benzene (20 ml) was refluxed under a Dean-Stark trap for 10 h. The reaction mixture was then neutralized at room temperature with solid K₂CO₃. After filtration and concentration of the organic phase, compound (6) (63 mg, 92%) was purified by crystallization from AcOEt. White crystals; mp 196-198°C (AcOEt); ir (CHCl₃) 3430 (NH), 1735, 1695 (C=O), 1600; ¹H nmr (CDCl₃, 300 MHz) δ 0.92 (t, 3H, Me-18, *J* = 7.0), 1.50-2.10 (m, 6H, 2H-16, 2H-17, 2H-19), 2.58 (s, 2H, 2H-6), 3.77 (br s, 1H, H-2), 4.34 (AB, 2H, CHCl₃) and the component of the second secon

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NCH₂Ph, J = 16.0, $\Delta v = 49.5$), 6.32 (d, 1H, H-12, J = 7.5), 6.57 (t, 1H, H-10, J = 7.5), 6.99 (t, 1H, H-11, J = 7.5), 7.13 (d, 1H, H-9, J = 7.5), 7.30 (m, 5H, ArH); ¹³C nmr (CDCl₃, 75 MHz) δ 12.1 (C-18), 22.9 (C-19), 24.7 (C-16), 26.7 (C-17), 47.3 (C-6), 50.2 (NCH₂Ph), 50.3 (C-7), 67.2 (C-2), 106.5 (C-8), 116.1 (C-20), 117.6 (C-10), 121.7 (C-9), 127.1 (C-11), 127.1 128.6 (5 Ar CH), 132.6 (C-8), 135.2 (C-21), 138.7 (C-1'), 150.2 (C-13), 175.5 (CO); ms (*m/z*, %) 344 (M^{+*}, 35), 305 (75), 258 (65), 253 (40), 220 (40), 130 (70), 91 (100), 81 (40), 71 (35), 69 (45), 57 (50); exact mass calcd for C₂₃H₂₄N₂O 344.1883, found 344.1884. Anal. Calcd for C₂₃H₂₄N₂O: C, 80.20; H, 7.02; N, 8.13. Found: C, 79.79; H, 6.91; N, 7.83.

20-Ethyl-5-oxo-4-allyl-1-benzylhexahydropyrrolocarbazole (7)

This compound was prepared from **17a,b** (30:70) (321 mg, 0.8 mmol), **16a** (241 mg, 0.6 mmol) or **16b** (201 mg, 0.5 mmol) following the conditions described for the preparation of **6** (method A from **17a,b**; method B from **16a** or **16b**). A quick filtration on a neutral alumina pad with 70:30 hexane-AcOEt as the eluent, gave **7** in 89% yield: white crystals; mp 88-90° (hexane); ir (CCl₄) 1720, 1675 (CO), 1595; ¹H nmr (CDCl₃, 300 MHz) δ 0.97 (t, 3H, Me-18, J = 7.5), 1.50-2.32 (m, 4H, 2H-16, 2H-17), 2.13 (q, 2H, 2H-19, J = 7.5), 2.61 (AB, 2H, 2H-6, J = 14.0, $\Delta v = 14.3$), 3.80 (t, 1H, H-2, J = 4.0), 4.20-4.58 (m, 2H, *NCH*₂-CH=CH₂), 4.37 (AB, 2H, NCH₂Ph, J = 16.0, $\Delta v = 50.0$), 5.25 (m, 2H, NCH₂-CH=CH₂), 5.90 (m, 1H, NCH₂-CH=CH₂), 6.36 (d, 1H, H-12, J = 8.0), 6.58 (t, 1H, H-10, J = 8.0), 6.97 (d, 1H, H-9, J = 8.0), 7.03 (t, 1H, H-11, J = 8.0), 7.32 (m, 5H, ArH); ¹³C nmr (CDCl₃, 75 MHz) δ 13.2 (CH₃-18), 25.1 (C-19), 25.5 (C-16), 26.6 (C-17), 45.1 (NCH₂-CH=CH₂), 46.9 (C-6), 50.3 (C-7), 50.4 (NCH₂Ph), 67.3 (C-2), 106.6 (C-12), 117.0 (NCH₂-CH=CH₂), 117.7 (C-10), 118.8 (C-20), 121.7 (C-9), 128.6 (C-11), 127.2-128.9 (5 Ar CH), 133.2 (NCH₂-CH=CH₂), 134.5 (C-21), 135.1 (C-8), 138.8 (C-1'), 150.6 (C-13), 173.5 (C-5); ms (*m*/*z*, %) 384 (M^{+*}, 12), 356 (3), 293 (14), 233 (6), 202 (6), 183 (6), 174 (6), 144 (3), 130 (10), 105 (6), 91 (100), 77 (10), 65 (8), 55 (7), 41 (30); exact mass calcd for C₂₆H₂₈N₂O 384.2200, found 384.2203. Anal. Calcd for C₂₆H₂₈N₂O: C, 81.21; H, 7.34; N, 7.29. Found: C, 81.10; H, 7.29; N, 7.27.

Imine (9), Enamine (11) and Hydroperoxyimine (19)

A solution of enamide (6) (516 mg, 1.5 mmol) in THF (30 ml) was added to a suspension of LiAlH₄ (171 mg, 4.5 mmol) in THF (60 ml). The mixture was refluxed for 16 h, cooled and water (1.5 ml) was added. After filtration, the inorganic salts were washed with CH₂Cl₂. The organic layers were combined, dried and evaporated. The residue was partially dissolved in AcOEt. Filtration of the mixture provided a first fraction of **19** (190 mg, 35%). The filtrate was then evaporated, and the residue was chromatographed on a short column of neutral alumina to afford a second fraction of **19** (65 mg, 12%) (elution with 70:30 hexane-AcOEt) and a mixture of imine (**9**) and enamine (**11**) (163 mg, 33%) (elution with AcOEt).

9 and **11**: ir (CHCi₃) 3400 (br, NH enamine), 1690 (N-C=C enamine), 1640 (C=N imine), 1600; ¹H nmr (CDCl₃, 60 MHz) δ 0.93 (t, 3H, Me-18, *J* = 7.5), 1.3-2.8 (m, 9H, 2H-19, 2H-17, 2H-16, 2H-6, H-20 imine), 3.4-3.8 (m, 1H, H-2), 3.85-4.15 (m, 2H, 2H-5), 4.40 (AB, 2H, NCH₂Ph, *J* = 16.5, $\Delta v = 14.0$), 6.4-7.5 (m, 9H, ArH); ms (*m/z*, %) 330 (M^{+•}, 5), 263 (21), 234 (13), 144 (10), 130 (4), 117 (7), 91 (100), 77 (11), 65 (20), 55 (20), 43 (41), 41 (43), 29 (54); exact mass calcd for C_{23H26}N₂ 330.2096, found 330.2092.

19: white crystals; mp 147-150°C (AcOEt); ir (CHCl₃) 3620 (OH), 1630 (C=N), 1600; ¹H nmr COSY ¹H-¹H (CDCl₃, 300 MHz) δ 0.95 (t, 3H, Me-18, *J* = 7.5), 1.45 (ddd, 1H, H-17ax, *J* = 15.0, 9.5, 4.0), 1.56-1.80 (m, 2H, 2H-16), 1.75 (dq, 1H, H-19B, *J* = 14.0, 7.0), 1.91 (ddd, 1H, H-17eq, *J* = 15.0, 7.0, 4.0), 2.07 (ddd, 1H, H-6ax, *J* = 12.5, 10.0, 9.5), 2.34 (ddd, 1H, H-6eq, *J* = 12.5, 6.5, 1.0), 2.39 (dq, 1H, H-19A, *J* = 14.0, 7.0), 3.57 (dd, 1H, H-2, *J* = 8.0, 5.0), 3.92 (ddd, 1H, H-5ax, *J* = 15.5, 10.0, 6.5), 4.08 (ddd, 1H, H-5eq, *J* = 15.5, 9.5, 1.0), 4.37 (AB, 2H, NCH₂Ph, *J* = 15.5, Δv = 68.0), 6.49 (d, 1H, H-12, *J* = 15.5, Δv = 68.0), 6.49 (d, 1H, H-12, *J* = 15.5, Δv = 68.0), 6.49 (d, 1H, H-12, *J* = 15.5, Δv = 68.0), 6.49 (d, 1H, H-12), Δv = 68.0), 68.0), 68.0 (d, 1H, H-12), Δv = 68.0), 68.0 (d, 1

J = 7.5), 6.67 (t, 1H, H-10, J = 7.5), 6.96 (d, 1H, H-9, J = 7.5), 7.08 (br s, 1H, OH), 7.11 (t, 1H, H-11, J = 7.5), 7.37 (m, 5H, ArH); ¹³C nmr (CDCl₃, 75 MHz) δ 7.0 (CH₃-18), 21.7 (C-19), 26.0 (C-16), 28.9 (C-17), 42.8 (C-6), 49.8 (NCH₂Ph), 57.8 (C-5), 61.1 (C-7), 71.9 (C-2), 83.3 (C-20), 107.8 (C-12), 117.7 (C-10), 121.1 (C-9), 127.3 (C-11), 127.6-128.9 (5 Ar CH), 132.7 (C-8), 138.3 (C-1'), 150.0 (C-13), 177.4 (C-21); ms (m/z, %) 362 (M^{+*}, 3), 346 (14), 330 (3), 289 (6), 278 (6), 234 (21), 220 (9), 144 (8), 91 (100), 77 (6), 65 (9), 57 (6), 28 (84); exact mass calcd for C₂₃H₂₆N₂O₂ 362.1994, found 362.2000. Anal. Calcd for C₂₃H₂₆N₂O₂: C, 76.21; H, 7.23; N, 7.73. Found: C, 76.09; H, 7.21; N, 7.76.

Enamine (12) and 4a-Oxoethylhexahydrocarbazolones (18a,b)

Those compounds were prepared from enamide (7) (384 mg, 1.0 mmol) following the conditions described for the preparation of 9, 11 and 19. The crude material was filtered on a short column of neutral alumina to give hexahydrocarbazolones (18a,b) (elution with 80:20 hexane-AcOEt), and enamine (12) (100 mg, 27%) (elution with AcOEt). A flash chromatography of the diastereoisomeric mixture (18a,b), on silica gel (elution with 80:20 hexane:AcOEt), afforded pure 18a (55 mg, 16%) and 18b (160 mg, 46%).

12: oil (unstable); ir (CHCl₃) (N-C=C), 1600; ¹H nmr (CDCl₃, 60 MHz) δ 0.90 (t, 3H, Me-18, *J* = 7.5), 1.3-3.2 (m, 8H, 2H-19, 2H-17, 2H-16, 2H-6), 3.5-4.7 (m, 5H, H-2, 2H-5, N*CH*₂-CH=CH₂), 4.35 (AB, 2H, NCH₂Ph, *J* = 17.0, Δv = 12.0), 5.0-6.0 (m, 3H, NCH₂-*CH*=*CH*₂), 6.4-7.4 (m, 9H, ArH); ¹³C nmr (CDCl₃, 15 MHz) δ 13.3 (CH₃-18), 24.5 (C-19), 25.6 (C-16), 30.0 (C-17), 38.8 (C-6), 50.1 (NCH₂Ph), 52.8 (C-7), 57.6 (N*CH*₂-CH=CH₂), 62.6 (C-5), 68.5 (C-2), 106.5 (C-12), 115.6 (NCH₂-CH=*CH*₂), 116.8 (C-10), 117.7 (C-20), 123.2 (C-9), 127.5 (C-11), 127.3-128.9 (5 Ar CH), 136.6 (NCH₂-*CH*=CH₂), 137.0 (C-8), 139.1 (C-1'), 141.4 (C-21), 150.0 (C-13).

18a: amorphous solid; ir (CCl₄) 2820 and 2725 (CH aldehyde), 1725 (CO aldehyde), 1705 (CO ketone), 1605; ¹H nmr (CDCl₃, 300 MHz) δ 0.84 (t, 3H, Me-11, *J* = 7.5), 1.36 (ddq, 1H, H-10B, *J* = 14.0, 7.5, 7.0), 1.40-1.55 (m, 2H, H-1ax, H-2ax), 1.75-1.90 (m, 2H, H-1eq, H-10A), 1.93-2.10 (m, 2H, H-2eq, H-3), 2.96 (AB, 2H, CH₂CO, *J* = 18.0, $\Delta v = 86.4$), 3.72 (dd, 1H, H-9a, *J* = 10.0, 5.0), 4.27 (AB, 2H, NCH₂Ph, *J* = 16.0, $\Delta v = 78.4$), 6.37 (d, 1H, H-8, *J* = 8.0), 6.62 (t, 1H, H-6, *J* = 8.0), 6.84 (d, 1H, H-5, *J* = 8.0), 7.04 (t, 1H, H-7, *J* = 8.0), 7.20-7.30 (m, 5H, ArH), 9.56 (s, 1H, CHO); ¹³C nmr (CDCl₃, 75 MHz) δ 11.3 (CH₃-11), 22.3 (C-10), 24.4 (C-1), 25.4 (C-2), 48.8 (C-3), 49.0 (NCH₂Ph), 50.3 (CH₂CO), 57.8 (C-4a), 70.7 (C-9a), 107.7 (C-8), 118.3 (C-6), 123.4 (C-5), 126.2 (C-4b), 127.5 (C-7), 127.6-129.3 (5 Ar CH), 137.9 (C-1), 149.6 (C-8a), 200.9 (CHO), 210.0 (CO); ms (*m*/z, %) 347 (M^{+*}, 27), 319 (3), 290 (9), 276 (4), 256 (6), 234 (5), 228 (13), 220 (12), 200 (7), 144 (4), 130 (4), 91 (100), 65 (7), 55 (4), 41 (7); exact mass calcd for C₂₃H₂₅NO₂ 347.1885, found 347.1886.

18b: white solid; mp 86-88°C (hexane); ir (CCl₄) 2820 and 2725 (CH aldehyde), 1725 (CO aldehyde), 1700 (CO ketone), 1600; ¹H nmr (CDCl₃, 300 MHz) δ 0.83 (t, 3H, Me-11, J = 7.5), 1.19 (ddq, 1H, H-10B, J = 14.0, 7.5, 7.0), 1.59-1.74 (m, 2H, H-2ax, H-10A), 1.84-1.96 (m, 3H, H-2eq, 2H-1), 2.34 (dddd, 1H, H-3, J = 9.0, 7.0, 7.0, 5.0), 3.08 (AB, 2H, CH₂CO, $J = 18.0, \Delta v = 187.0$), 3.83 (dd, 1H, H-9a, J = 4.0, 3.0), 4.40 (AB, 2H, NCH₂Ph, $J = 16.0, \Delta v = 70.0$), 6.43 (d, 1H, H-8, J = 8.0), 6.62 (t, 1H, H-6, J = 8.0), 6.91 (d, 1H, H-5, J = 8.0), 7.09 (t, 1H, H-7, J = 8.0), 7.30-7.40 (m, 5H, ArH), 9.61 (s, 1H, CHO); ¹C nmr (CDCl₃, 75 MHz) δ 11.8 (CH₃-11), 22.7 (C-10), 24.4 (C-1), 26.6 (C-2), 49.4 (NCH₂Ph), 50.1 (C-3), 53.3 (*CH₂*CO), 57.1 (C-4a), 69.3 (C-9a), 106.5 (C-8), 117.4 (C-6), 124.3 (C-5), 126.2 (C-4b), 127.4 (C-7), 127.3-129.5 (5 Ar CH), 138.3 (C-1'), 151.2 (C-8a), 200.4 (CHO), 211.0 (CO); ms (*m*/z, %) 347 (M⁺⁺, 23), 319 (3), 290 (9), 276 (4), 256 (3), 234 (5), 228 (11), 220 (14), 200 (6), 144 (5), 130 (5), 115 (3), 91 (100), 65 (10), 55 (6), 41 (10); exact mass calcd for C₂₃H₂₅NO₂: C, 79.50; H, 7.25; N, 4.03. Found: C, 79.57; H, 7.63; N, 3.60.

AKNOWLEDGEMENTS

The authors wish to thank Mr. Michael Dickman for correcting the English version of the article.

REFERENCES

- J.E. Saxton, "Heterocyclic Compounds, the Monoterperoid Indole Alkaloids", J. Wiley, New York, 1983, Vol. 25, part 4, p. 38 and chs 8 and 11. P.L. Feldman and H. Rapoport, J. Am. Chem. Soc., 1987, 109, 1603 and references cited therein. P. Magnus, M. Ladlow, and J. Elliot, J. Am. Chem. Soc., 1987, 109, 7929. A. Brossi and M. Suffness, "The Alkaloids, Antitumor Bisindole Alkaloids from Catharanthus Roseus", Academic Press : San Diego, 1990, Vol. 37.
- 2. J.-C. Gramain, H.-P. Husson, and Y. Troin, J. Org. Chem., 1985, 50, 5517.
- 3. D. Gardette, J.-C. Gramain, and M.-E. Sinibaldi, Heterocycles, 1990, 31, 1439.
- The biogenetic numbering is used for tetra- and pentacyclic derivatives: J. Le Men and W.I. Taylor, *Experientia*, 1965, 21, 508. The absolute configuration depicted is that of natural (+)-aspidospermidine.
- 5. D. Dugat, J.-C. Gramain, and G. Dauphin, J. Chem. Soc., Perkin Trans. 2, 1990, 605.
- 6. N. Benchekroun-Mounir, D. Dugat, and J.-C. Gramain, Tetrahedron Lett., 1992, 33, 4001.
- 7. J.-C. Gramain, Y. Troin, and H.-P. Husson, J. Heterocycl. Chem., 1988, 25, 201.
- 8. N. Benchekroun-Mounir, D. Dugat, J.-C. Gramain, and H.-P. Husson, J. Org. Chem., 1993, 58, 6457.
- 9. M. Langlois, C. Guillonneau, J. Meingan, and J. Maillard, Tetrahedron, 1971, 27, 5641.
- F. Weygand, G. Eberhardt, H. Linden, F. Schäfer, and I. Eigen, Angew. Chem., 1953, 65, 525. F. Weygand and H. Linden, Angew. Chem., 1954, 66, 174.
- 11. J. Zabicky, "The Chemistry of Amides", ed. by S. Pataï, J. Wiley and Sons, London, New York, 1970, pp. 795-800.
- 12. D. Swern, "Organic Peroxides", Wiley Interscience, New York, 1971, Vol. 2, pp. 19-21.
- 13. A.G. Cook, "Enamines, Synthesis, Structure and Reactions", 2nd edition, Marcel Dekker, Valparaiso, 1988, p. 485.
- (a) L.A. Cohen and B. Witkop, *J. Am. Chem. Soc.*, 1955, **77**, 6595. (b) B. Witkop and J.B. Patrick, *J. Am. Chem. Soc.*, 1951, **73**, 2196. (c) R.F. Parcell and F.P. Hauck, *J. Org. Chem.*, 1963, **28**, 3468. (d) C.O. Bendler and R. Bonnett, *Chem. Commun.*, 1966, 198.
- 15. C.L. Stevens and R.J. Gasser, J. Am. Chem. Soc., 1957, 79, 6057.

Received, 11th April, 1994