SYNTHESIS OF EGONOL AND (±)-MACHICENDIOL

Yutaka Aoyagi, Toshihiko Mizusaki, Akiko Hatori, Taichi Asakura, Takashi Aihara, Sakae Inaba, Ken Hayatsu, and Akihiro Ohta*

School of Pharmacy, Tokyo University of Pharmacy and Life Science, 1432-1 Horinouchi, Hachioji, Tokyo 192-03, Japan

Abstract-Two norneolignans, egonol (1) and (\pm)-machicendiol (2), were synthesized by using the palladium-catalyzed cross-coupling reaction as a key step. This paper describes a new strategy for the preparation of norneolignans having a benzo[b] furan skeleton.

Lignans and neolignans, which are contained in most plants, have attracted much attention in medicinal chemistry for their various biological activities such as insecticidal, fungicidal, antimicrobial and antioxidant properties. Egonol (1) was first isolated from the seed oil of $Syrax\ japonicum^2$ and (\pm)-machicendiol (2) from the leaf extracts of $Machilus\ glaucescens$, which are used for the treatment of asthema, rheumatism, and ulcers.

Compound (1) and (2) are normeolignans with a benzo[b] furan skeleton and Hirano et al. reported that they show cytosatic activity against human leukemic HL-60 cells.⁴

Synthesis of benzo[b] furans has been reported by several groups. For example, egonol (1)⁵ and (1)-machicendiol (1)⁶ were prepared by Castro's method, 1 via the coupling reaction of a cuprous acetylide with an 10-halophenol. However, since the coupling reactions of this type often give 1,4-bisaryl-1,3-dienes as a by-product, the yield of the desired benzo[10] furan is generally low. Watanabe 10 reported a more efficient synthesis of 2-arylbenzo[10] furans 11 lithiation of 12-tetramethylphosphorodiamidates. Mann and co-workers describe a total synthesis of moracin M, 12-halobenzo[13] furans with 5-metalloresorcinol tricarbonyl chromium(13) complexes or 2-metallobenzo[13] furans with 5-haloresorcinol tricarbonyl chromium(13) complexes. We have already reported an efficient palladium-catalyzed cross-coupling reaction of benzo[13] benzo[14] bromides.

In this paper, we describe a total synthesis of egonol ($\underline{1}$) and ($\underline{+}$)-machicendiol ($\underline{2}$) from 5-bromo-7-methoxybenzo[b]furan ($\underline{4}$) by application of our coupling reaction. The bromobenzo[b]furan ($\underline{4}$) was prepared by cyclization of 4-bromo-2-formyl-6-methoxyphenylacetic acid ($\underline{3}$)¹¹ according to the procedures for the synthesis of coumarones described by Burgstahler. ¹²

Introduction of alkyl substituents to 5-position of $\underline{4}$ was achieved by the treatment of a lithiated $\underline{4}$ with 3-bromopropyl tetrahydropyranyl ether. ¹³ The palladium-catalyzed cross-coupling reaction with 4-bromo-1,2-methylenedioxybenzene did not occur, when non-metallated $\underline{5}$ was used. When the zinc salt of $\underline{5}$, prepared by treating $\underline{5}$ with ⁿBuLi / ZnCl₂ at 0°C, was employed as a substrate, the coupling reaction proceeded to give the corresponding coupling product ($\underline{6}$) in 19 % yield. The deprotection of the tetrahydropyranyl group in $\underline{6}$ gave egonol ($\underline{1}$) in 75 % yield (Scheme 2).

i) Ac₂O, AcONa, AcOH, reflux ii) ⁿBuLi, -78 °C then Br(CH₂)₃OTHP iii) a) ⁿBuLi, 0°C b) ZnCl₂ c) 4-Bromo-1,2-methylenedioxybenzene, Pd(PPh₃)₄, reflux iv) *p*-TsOH, MeOH

Scheme 2. Synthesis of Egonol (1)

(\pm)-Machicendiol ($\underline{2}$) was also synthesized from $\underline{4}$ by the combination of the following reactions: introduction of an alcoholic function using 3- t butyldiphenylsilyloxypropan-1-al, 14 protection of the hydroxyl group of $\underline{7}$, palladium-catalyzed arylation of $\underline{8}$, and desilylation (Scheme 3). Application of the synthetic strategy described above to other natural lignans with a benzo[b] furan skeleton are now in progress.

ACKNOWLEGEMENT

The authors thank the Ministry of Education, Science and Culture, Japan, for Grant-in-Aid (No. 03771697).

i) ⁿBuLi, -78 °C then TBDPSO(CH₂)₂CHO ii) TBDMSCI, imidazole, DMF iii) a) ⁿBuLi, 0°C b) ZnCl₂ c) 4-Bromo-1,2-methylenedioxybenzene, Pd(PPh₃)₄, reflux iv) ⁿBu₄NF, THF

Scheme 3. Synthesis of (\pm) -Machicendiol $(\underline{2})$

EXPERIMENTAL

No correction was made to melting and boiling points. ¹H-Nmr spectral data were obtained with a Varian Gemini-300 (300 MHz) or a Brucker AM-400 (400 MHz) in CDCl3 using TMS as an internal standard. Medium pressure liquid chromatography (MPLC) was conducted using a UVILOG 5III as a UV detector (Oyo Bunko Kiki Co., Ltd. Tokyo) and Kiesel gel 60 (Merck AG, Darmstadt) as the packing material. Other spectral data were obtained using the following instruments. Ir spectra: Japan Spectroscopic Co. A-100 (Nihon Bunko Kiki Co., Ltd.); Ms: Hitachi M-80B (Hitachi Co., Ltd.) or Auto spec spectrometer (Faisons Co., Ltd.).

Preparation of 4

A mixture of 3 (30.0 g, 104 mmol), Ac₂O (98 ml, 1.04 mol), anhydrous powdered AcONa (37.7 g, 468 mmol), and AcOH (95.0 ml, 1.66 mol) was refluxed for 8 h. After the reaction mixture was poured into ice-cooled H₂O (500 ml), the aqueous layer was extracted with Et₂O (300 ml x 3). The combined organic phase was washed with cold 5 % NaOH (300 ml x 4), H₂O (300 ml x 3), and sat. NaCl (300 ml x 3), successively. The Et₂O was removed *in vacuo* and the oily residue was distilled using a bulb to bulb distillation apparatus (180 -190 °C / 18 torr) to give 4 in 26 % yield.

5-Bromo-7-methoxybenzo[b]furan ($\underline{4}$): colorless oil, bp 180-190 °C / 18 torr. Ms: m/z 226 (M⁺). ¹H-Nmr (300 MHz): 4.00 (3H, s), 6.71 (1H, d, J = 2.1 Hz), 6.92 (1H, d, J = 1.7 Hz), 7.34 (1H, d, J = 1.7 Hz), 7.62 (1H, d, J = 2.1 Hz). Anal. Calcd for C9H7O2Br: C, 47.61; H, 3.11. Found: C, 47.57; H, 3.13.

Alkylation of Compound 4: Synthesis of 5 and 7

1.6 M ⁿBuLi in hexane (4 ml, 6.4 mmol) was added to a solution of $\underline{4}$ (0.66 g, 3.0 mmol) in dry THF (30 ml) at -78 °C. After the resulting mixture was stirred at this temperature for 15 min, 3-bromopropyl tetrahydropyranyl ether ¹³ (1.34 g, 6 mmol) or 3-^tbutyldimetylsilyloxypropan-1-al ¹⁴ (1.87 g, 6 mmol) in dry THF (5 ml) was added dropwise. The mixture was stirred at -78 °C for 1.5 h and at 0 °C for 30 min, and then poured into sat. NH4Cl (50 ml). After the usual treatment, MPLC (hexane:Et₂O = 4:1) gave 5-alkylated benzo[b]furan ($\underline{5}$ or $\underline{7}$).

7-Methoxy-5-(3-tetrahydropyranyloxypropyl)benzo[b] furan (\underline{s}): Yield 26 %, colorless oil, bp 190-200 °C / 1 torr (bath temp.). Ms: m/z 290 (M+). ¹H-Nmr (300 MHz): 1.51-2.02 (8H, m), 2.78 (2H, td, J = 7.8 and 3.0 Hz), 3.39-3.54 (2H, m), 3.77-3.89 (2H, m), 4.00 (3H, s), 4.40 (1H, dd, J = 4.4 and 2.9 Hz),6.66 (1H, d, J = 1.3 Hz), 6.68 (1H, d, J = 2.1 Hz), 7.01 (1H, d, J = 1.4 Hz), 7.59 (1H, d, J = 2.1 Hz). High-resolution ms Calcd for C17H22O4: 290.151809. Found: 290.149528.

5-(1-Hydroxy-3-^tbutyldiphenylsilyloxypropyl)-7-methoxybenzo[b]furan (7): Yield 19%, colorless oil. Ir (neat): 3450 cm⁻¹. CIms: m/z 461 (M++1). ¹H-Nmr (300 MHz): 1.10 (9H, s), 1.91-2.15 (2H, m), 3.88

(2H, m), 4.01 (3H, s), 5.10 (1H, dd, J = 8.2 and 3.8 Hz), 6.74 (1H, d, J = 2.1 Hz), 6.89 (1H, d, J = 1.4 Hz), 7.19 (1H, d, J = 0.9 Hz), 7.35-7.47 (6H, m), 7.62 (1H, d, J = 2.1 Hz), 7.66-7.71 (4H, m). High-resolution ms Calcd for $C_{28}H_{32}O_{4}S_{i}$: 460.206988. Found: 460.203930.

Silvlation of 7

A DMF (5 ml) solution of 7 (0.246 g, 0.53 mmol), butyldimethylsilyl chloride (0.121 g, 0.80 mmol), and imidazole (0.075 g, 1.1 mmol) was stirred overnight under Ar atmosphere. The mixture was poured into H₂O (30 ml) and the aqueous layer was extracted with Et₂O (30 ml x 3). The combined organic layer was washed with H₂O (20 ml x 2), and dried over Na₂SO₄. The solvent was evaporated *in vacuo* to give an oily residue, which was purified by MPLC (hexane:AcOEt= 4:1) to give 8 (0.287 g, 94 %) as a colorless oil.

5-(1-*Butyldimethylsilyloxy-3-*butyldiphenylsilyloxypropyl)-7-methoxybenzo[b] furan ($\underline{8}$): Colorless oil. Ms: m/z 574 (M+). 1 H-Nmr (300 MHz): 0.14 (3H, s), 0.16 (3H, s), 1.00 (9H, s), 1.21 (9H, s), 1.92-2.18 (2H, m), 3.71-3.79 (1H, m), 3.91-3.99 (1H, m), 4.11 (3H, s), 5.14 (1H, dd, J = 7.7 and 5.3 Hz), 6.85 (1H, d, J = 2.2 Hz), 6.96 (1H, d, J = 1.4 Hz), 7.19 (1H, d, J = 1.0 Hz), 7.45-7.56 (6H, m), 7.74-7.82 (5H, m). High-resolution ms Calcd for C34H46O4Si2: 574.293467. Found: 574.292038.

Arvlation of 5 and 8

A THF (10 ml) solution of 5 (0.21 g, 0.72 mmol) was added to a THF (5 ml) solution of 1.6 M ⁿBuLi in hexane (0.5 ml, 0.79 mmol) at 0 °C under Ar atmosphere and the mixture was stirred at 0 °C for 1 h. A 1.0 M solution of ZnCl₂ in Et₂O (0.8 ml, 0.8 mmol) was added dropwise and the mixture was stirred for further 20 min at this temperature. The resulting solution was added to a THF (5 ml) solution of 4-bromo-1,2-methylenedioxybenzene (0.1 ml, 0.86 mmol) and Pd(PPh₃)₄ (0.084 g, 0.072 mmol) at 0 °C through a double-chipped needle and the mixture was refluxed for 12 h under Ar atmosphere. After addition of sat. NH₄Cl (20 ml), the mixture was extracted with Et₂O (20 ml x 3), and the organic layer was dried over Na₂SO₄ and evaporated to give an oily residue, which was purified by MPLC (hexane:Et₂O= 9:1) to give

the corresponding coupling product ($\underline{\mathbf{6}}$) (0.057 g, 19 %) as an oil. By the same procedure, in which $\underline{\mathbf{8}}$ (0.413 g, 0.72 mmol) was used in place of $\underline{\mathbf{5}}$, $\underline{\mathbf{9}}$ (0.075 g, 15 %) was obtained as an oil.

7-Methoxy-2-(3,4-methylenedioxy)phenyl-5-(3-tetrahydropyranyloxypropyl)benzo[b]furan ($\underline{6}$): Colorless oil. Ms: m/z 410 (M+). ¹H-Nmr (400 MHz): 1.51-1.74 (6H, m), 1.97 (2H, t, J = 7.6 Hz), 2.77 (2H, m), 3.41-3.53 (2H, m), 3.77-3.91 (2H, m), 4.03 (3H, s), 4.59 (1H, dd, J = 4.4 and 2.9 Hz), 6.00 (2H, s), 6.64 (1H, d, J = 1.3 Hz), 6.78 (1H, s), 6.86 (1H, d, J = 8.1 Hz), 6.97 (1H, d, J = 1.4 Hz), 7.32 (1H, d, J = 1.6 Hz), 7.40 (1H, dd, J = 8.1 and 1.7 Hz). High-resolution ms Calcd for C₂4H₂6O₆: 410.172939. Found: 410.171669.

7-Methoxy-2-(3,4-methylenedioxyphenyl)-5-(1- f butyldimethylsilyloxy-3- f butyldiphenylsilyloxypropyl)-benzo[b]furan ($\underline{9}$): Colorless oil. Ms: m/z 694 (M+). 1 H-Nmr (300 MHz): 0.13 (3H, s), 0.16 (3H, s), 0.87 (9H, s), 1.26 (9H, s), 1.90-2.17 (2H, m), 3.71-3.79 (1H, m), 3.91-3.90 (1H, m), 4.12 (3H, s), 5.12 (1H, dd, J = 7.6 and 5.3 Hz), 6.14 (2H, s), 6.93 (1H, d, J = 1.2 Hz), 6.93 (1H, s), 7.00 (1H, d, J = 8.0 Hz), 7.14 (1H, s), 7.44-7.55 (8H, m), 7.76-7.82 (4H, m). High-resolution ms Calcd for C41H50O6Si2: 694.314597. Found: 694.316345.

Removal of the THP Group in 6: Synthesis of Egonol (1)

A mixture of 6 (0.057 g, 0.014 mmol), p-TsOH (0.027 g, 0.16 mmol), and MeOH (5 ml) was treated in a sonic bath at room temperature for 1 h. The solvent was evaporated under reduced pressure and sat. NaHCO3 (10 ml) was added to the residue. The mixture was extracted with Et2O (10 ml x 3) and the combined organic phase was washed with H2O (15 ml x 3). The solvent was evaporated *in vacuo*, and the residue was purified by MPLC (hexane:AcOEt= 1:1) to give 1 (0.034 g, 75 %) as a colorless solid, whose physical data were identical with those reported previously.²

Desilylation of 9: Synthesis of (±)-Machicendiol (2)

A mixture of 2 (0.044 g, 0.063 mmol), 1.0 M ⁿBu₄NF in THF (0.5 ml, 0.5 mmol), and THF (5 ml) was stirred at room temperature overnight. Sat. NH₄Cl (5 ml) was added to the mixture and the aqueous layer

was extracted with Et₂O (10 ml x 3). The combined organic layer was washed with sat. NaCl (10 ml x 3) and dried over Na₂SO₄. The solvent was evaporated *in vacuo* and the residue was purified by MPLC (CHCl₃:MeOH= 20:1) to give $\underline{2}$ (0.021 g, 100 %) as a coloress solid, whose physical data were identical with those reported previously.³

REFERENCES

- 1. D. C. Ayres and J. Loike, 'Lignans,' Cambridge University Press, Inc., Cambridge, 1990.
- 2. H. Okada, J. Pharm. Soc. Japan, 1915, 657.
- 3. B. Talaparta, T. Ray, and S. Talaparta, Indian, J. Chem., 1976, 14B, 613.
- 4. T. Hirano, M. Goto, and K. Oka, Life Sci., 1994, 55, 1061.
- 5. F. G. Schreiber and R. Stevenson, J. Chem. Soc., Perkin Trans. I, 1976, 1514.
- 6. G. E. Schneiders and R. Stevenson, J. Org. Chem., 1979, 44, 4710.
- 7. C. E. Castro and R. D. Stephens, J. Org. Chem., 1963, 28, 2163.
- 8. M. Date, K. Kawanishi, T. Torii, M. Watanabe, and S. Furukawa, *Chem. Pharm. Bull.*, 1989, 37, 2884.
- 9. a) J. M. Clough, I. S. Mann, and D. A. Widdowson, *Tetrahedron Lett.*, 1987, 28, 2645. b) I. S. Mann and D. A. Widdowson, *Tetrahedron*, 1991, 47, 7981.
- 10. A. Ohta, (the late) Y. Akita, T. Ohkuwa, M. Chiba, R. Fukunaga, A. Miyafuji, T. Nakata, N. Tani, and Y. Aoyagi, *Heterocycles*, 1990, 31, 1951.
- 11. V. S. Salvi and S. Sethena, J. Ind. Chem. Soc., 1968, 45, 433.
- 12. A. W. Burgstahler and L. R. Worden, Org. Synth., 1966, 46, 28.
- 13. L. A. Paquette and M. K. Scott, J. Am. Chem. Soc., 1972, 94, 6751.
- 14. M. A. Blanchette, M. S. Malamas, M. H. Nanz, J. C. Roberts, P. Somfai, D. C. Whritenour, and S. Masamune, J. Org. Chem., 1989, 54, 2817.