

## SYNTHESIS AND POLYBROMINATION OF AZULENO[6,5-*d*]-THIAZOLE AND ITS 2-METHYL DERIVATIVES

Tian-Chyuan Huang<sup>a\*</sup>, Yun-Shan Lin<sup>b\*</sup>, and Tetsuo Nozoe<sup>c</sup>

<sup>a</sup>National Taipei College of Nursing, Shih-Pai, Taipei, Taiwan

<sup>b</sup>Department of Chemistry, Tamkang University, Tamsui, Taipei, Taiwan

<sup>c</sup>Tokyo Research Laboratories, Kao Corporation, 2-1-3, Bunka, Sumida-Ku, Tokyo 131, Japan

**Abstract**-Azuleno[6,5-*d*]thiazole (**5a**) and its 2-methyl derivative (**8**) were obtained by the deethoxycarbonylation of the corresponding diethyl azuleno[6,5-*d*]thiazole-5,7-dicarboxylate (**4a**) and 2-methyl derivative (**6b**) which were synthesized from diethyl 2-acetylamino-6-aminoazulene-1,3-dicarboxylate (**1**) in a few steps. Polybromination of **5a** and **8** in 10% aqueous THF led to the corresponding 5-bromo-7,9-azulenequinono[6,5-*d*]thiazole (**9a**) and 2-methyl derivative (**9b**), respectively.

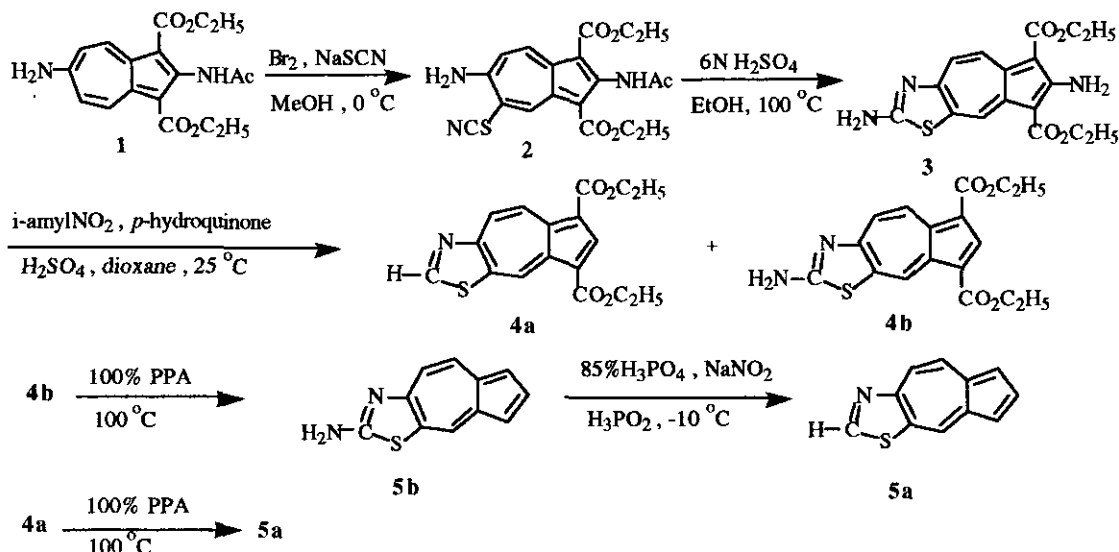
The azulenoid heterocyclic compounds which consist of an azulene ring condensed with heterocyclic aromatics are of interest not only in their physical properties<sup>1,2</sup> and chemical behaviors,<sup>3</sup> but also in their physiological activities. Heterocyclic compounds fused with the five-membered ring of azulene, such as furan,<sup>4</sup> thiophene,<sup>5</sup> pyrrole,<sup>4</sup> pyrazole,<sup>6</sup> imidazole,<sup>7</sup> thiazole,<sup>8</sup> pyridine,<sup>9</sup> pyridazine,<sup>10</sup> pyrimidine,<sup>7</sup> pyrazine,<sup>11</sup> thiapyran,<sup>12</sup> and quinoxaline<sup>13</sup> are known. However, no azuleno[6,5-*d*]thiazoles and its 2-methyl derivatives are known. We describe herein a facile method for the synthesis of these novel heterocyclic compounds, azuleno[6,5-*d*]thiazole (**5a**) and its 2-methyl derivative (**8**), in which a heterocycle was fused with the seven-membered ring of the azulene, and we report an application of azulenequinone synthesis<sup>14</sup> to prepare two azulenequinone derivatives (**9a**) and (**9b**) which have a fused-thiazole on the seven-membered moiety.

The reaction of diethyl 2-acetylamino-6-aminoazulene-1,3-dicarboxylate (**1**)<sup>15</sup> with thiocyanogen bromide

gave the 5-thiocyano compound (2) in 98% yield. Deacetylation of compound (2), on treatment with an aqueous 6N sulfuric acid solution in ethanol at 100 °C, formed the thiazole ring by annelation to give diethyl 6-amino-2-aminoazuleno[6,5-*d*]thiazole-5,7-dicarboxylate (3) in 98% yield. The structure of 3 was confirmed by its ir spectrum which revealed no absorption of thiocyno group. Reaction of 3 with isoamyl nitrite in the presence of concentrated sulfuric acid and *p*-hydroquinone in dioxane at 25 °C<sup>16</sup> produced diethyl azuleno[6,5-*d*]thiazole-5,7-dicarboxylate (4a) and diethyl 2-aminoazuleno[6,5-*d*]thiazole-5,7-dicarboxylate (4b) in 62% and 25% yields, respectively. Heating of 4b in 100% phosphoric acid at 100 °C resulted in deethoxycarbonylation to give 2-aminoazuleno[6,5-*d*]thiazole (5b) in 92% yield. The spectral data (<sup>1</sup>H Nmr and ir) have shown that 5b exists in the amino form.

The replacement of the 2-amino functionality of 5b with hydrogen was achieved by use of the following procedure:<sup>17</sup> Compound 5b was diazotized with a concentrated aqueous sodium nitrite solution in 85% phosphoric acid at -10 °C and followed by addition of hypophosphorous acid to give azuleno[6,5-*d*]thiazole (5a) in 23% yield. The same compound was also obtained by the treatment of 4a, with 100% phosphoric acid at 100 °C in 90% yield (Scheme 1).

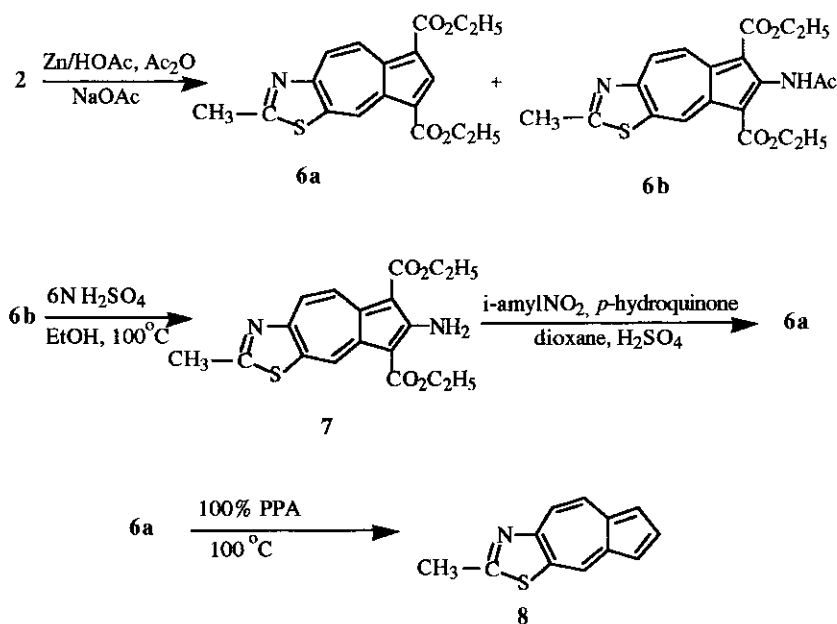
Scheme 1.



Reductive acetylation of compound (2), on treatment with zinc dust in acetic acid and acetic anhydride at room temperature, underwent ring closure to give a mixture of diethyl 2-methylazuleno[6,5-*d*]thiazole-5,7-dicarboxylate (6a) and diethyl 6-acetylamino-2-methylazuleno[6,5-*d*]thiazole-5,7-dicarboxylate (6b) in

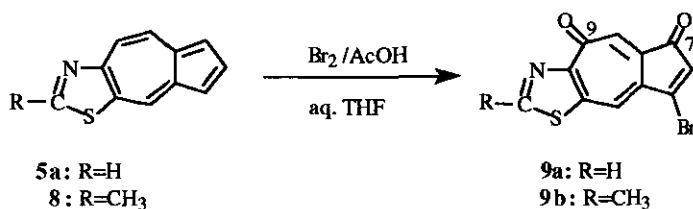
15% and 53% yields.  $^1\text{H-Nmr}$  spectrum of **6a** revealed that there is a singlet at  $\delta$  8.80, which is corresponding to the H-6 proton on the five-membered ring. Hydrolysis of **6b**, on heating with the 6N  $\text{H}_2\text{SO}_4$  in ethanol at  $100^\circ\text{C}$  gave diethyl 6-amino-2-methylazuleno[6,5-*d*]thiazole-5,7-dicarboxylate (**7**) in 98% yield, followed by deamination to obtain **6a** in 86% yield. Furthermore, the deethoxycarbonylation of **6a** by heating with 100% phosphoric acid at  $100^\circ\text{C}$  gave 2-methylazuleno[6,5-*d*]thiazole (**8**) in 95% yield (Scheme 2).

Scheme 2



Polybrominations of **5a** and **8** were carried out in accordance with the procedure described in the paper.<sup>14</sup> Reaction of **5a** and **8** in 10% aqueous THF with bromine in acetic acid at  $5\text{-}10^\circ\text{C}$  gave 5-bromo-7,9-azulenequinono[6,5-*d*]thiazole (**9a**) and 2-methyl-5-bromo-7,9-azulenequinono[6,5-*d*]thiazole (**9b**) in 83% and 87% yield, respectively (Scheme 3).

Scheme 3



## EXPERIMENTAL

All melting points are uncorrected. The instruments to record spectra were, for ultraviolet-visible Shimadzu UV-202 and UV-160, for infrared Perkin-Elmer IR-983G spectrophotometers, for mass spectra Finnigan TSQ-46C, for high-resolution mass spectra JEOL JMS-HX110, for nuclear magnetic resonance (<sup>1</sup>H nmr) Bruker AC-300. Chemical shifts (δ) and coupling constants (Hz) were measured with respect to TMS.

### Diethyl 2-acetyl-amino-5-thiocyano-6-aminoazulene-1,3-dicarboxylate (2)

To a solution of **1** (500 mg, 1.45 mmol) and NaSCN (235 mg, 2.9 mmol) in 25 ml of MeOH, a solution of bromine (348 mg, 2.2 mmol) in 5 ml of MeOH which was saturated with NaBr was added at 0 °C. After the reaction mixture was stirred for 3 h, the mixture was concentrated under reduce pressure, diluted with water, extracted with EtOAc, and dried over anhydrous Na<sub>2</sub>SO<sub>4</sub>. Removal of EtOAc in *vacuo* gave the residue which was purified by column chromatography (silica gel, EtOAc:n-hexane = 5:1) to obtain a yellow solid. It was crystallized from EtOAc and CH<sub>2</sub>Cl<sub>2</sub> (1: 2) to afford **2** (570 mg, 98%) as a yellow prisms; mp 282 °C (decomp.).

**2:** <sup>1</sup>H-Nmr (300 MHz, DMSO-d<sub>6</sub>) δ 1.33 (6H, t, J=7.0 Hz, CO<sub>2</sub>CH<sub>2</sub>CH<sub>3</sub> x 2), 2.13 (3H, s, NHCO-CH<sub>3</sub>), 4.27 (4H, q, J=7.0 Hz, CO<sub>2</sub>CH<sub>2</sub>CH<sub>3</sub> x 2), 7.84 (1H, d, J=11.3 Hz, H-7), 8.63 (2H, br s, NH<sub>2</sub>), 9.04 (1H, d, J=11.3 Hz, H-8), 9.56 (1H, s, H-4), 9.89 (1H, s, NHCOCH<sub>3</sub>); ir (KBr) 3402, 3394, 3300, 2159, 1685 cm<sup>-1</sup>; ms (40 ev) m/z 401 (M<sup>+</sup>), 359, 313, 241; HRms m/z Calcd for C<sub>19</sub>H<sub>19</sub>N<sub>3</sub>O<sub>5</sub>S 401.1045, Found 401.1031.

### Diethyl 2-amino-6-aminoazuleno[6,5-d]thiazole-5,7-dicarboxylate (3)

A solution of **2** (450 mg, 1.1 mmol) in 30 ml of EtOH was added 6N H<sub>2</sub>SO<sub>4</sub> (5 ml), the mixture was stirred at reflux for 3 h. After cooling, neutralization with 6M NaOH, the yellow precipitate was collected, washed with water and dissolved in EtOAc. After drying over anhydrous Na<sub>2</sub>SO<sub>4</sub>, the solvent was removed in *vacuo*, 387 mg of pure **3** was obtained as yellow solid (98%); mp 298 °C (decomp.).

**3:**  $^1\text{H-Nmr}$  (300 MHz,  $\text{DMSO-d}_6$ )  $\delta$  1.37 (6H, t,  $J=7.0$  Hz,  $\text{CO}_2\text{CH}_2\text{CH}_3 \times 2$ ), 4.35 (4H, q,  $J=7.0$  Hz,  $\text{CO}_2\text{CH}_2\text{CH}_3 \times 2$ ), 7.45 (2H, br s,  $\text{NH}_2$ ), 7.78 (1H, d,  $J=11.4$  Hz, H-9), 8.23 (2H, br s,  $\text{NH}_2$ ), 9.03 (1H, d,  $J=11.4$  Hz, H-8), 9.45 (1H, s, H-4);  $^{13}\text{C-nmr}$  (75.0 MHz,  $\text{DMSO-d}_6$ ) 14.51, 59.11, 59.34, 95.45, 99.84, 122.85, 123.83, 128.38, 137.86, 140.78, 157.96, 158.99, 165.50, 165.67, 168.61;  $\nu$  (KBr) 3490, 3338, 2980, 1652, 1589  $\text{cm}^{-1}$ ;  $m/z$  (40 ev) 359 ( $\text{M}^+$ ), 313, 268, 241; HRms  $m/z$  Calcd for  $\text{C}_{17}\text{H}_{17}\text{N}_3\text{O}_4\text{S}$  359.0939, Found 359.0925.

**Diethyl azuleno[6,5-*d*]thiazole-5,7-dicarboxylate (4a) and its 2-amino derivative (4b)**

To a stirred solution of **3** (500 mg, 1.4 mmol), hydroquinone (2.31 g, 21 mmol) and conc.  $\text{H}_2\text{SO}_4$  (137 mg, 1.4 mmol) in dioxane (125 ml) at 25  $^\circ\text{C}$  was added isopentyl nitrite (2.45 g, 21 mmol) in dioxane (10 ml) during 20 min. After stirring for 3 h, the reaction mixture was quenched by adding 1M  $\text{Na}_2\text{SO}_3$  (100 ml) and extracted with EtOAc, the organic layer was washed with saturated  $\text{NaHSO}_3$ , dried over anhydrous  $\text{Na}_2\text{SO}_4$  and concentrated under reduce pressure. The crude residue was subjected to column chromatography (silica gel, EtOAc:n-hexane = 1:1) to give **4a** (285 mg, 62%) as a pink needles (from acetone), mp 148-150  $^\circ\text{C}$ , and **4b** (120 mg, 25%) as a dark red needles (from EtOAc), mp > 300  $^\circ\text{C}$ , respectively.

**4a:**  $^1\text{H-Nmr}$  (300 MHz,  $\text{CDCl}_3$ )  $\delta$  1.45 (6H, t,  $J=7.2$  Hz,  $\text{CO}_2\text{CH}_2\text{CH}_3 \times 2$ ), 4.46 (4H, q,  $J=7.1$  Hz,  $\text{CO}_2\text{CH}_2\text{CH}_3 \times 2$ ), 8.50 (1H, d,  $J=11.1$  Hz, H-9), 8.79 (1H, s, H-6), 9.33 (1H, s, H-2), 9.78 (1H, d,  $J=11.1$  Hz, H-8), 10.46 (1H, s, H-4);  $^{13}\text{C-nmr}$  (75.0 MHz,  $\text{CDCl}_3$ )  $\delta$  14.54, 60.18, 60.25, 115.63, 117.97, 127.01, 132.97, 133.21, 137.02, 138.73, 141.54, 142.76, 159.08, 159.36, 164.99, 165.05;  $\nu$  (KBr) 2988, 1675, 1426, 1384  $\text{cm}^{-1}$ ;  $m/z$  (40 ev) 329 ( $\text{M}^+$ ), 299, 271, 241, 200; HRMS  $m/z$  Calcd for  $\text{C}_{17}\text{H}_{15}\text{NO}_4\text{S}$  329.0827, Found 329.0831.

**4b:**  $^1\text{H-Nmr}$  (300 MHz,  $\text{DMSO-d}_6$ )  $\delta$  1.35 (6H, t,  $J=7.1$  Hz,  $\text{CO}_2\text{CH}_2\text{CH}_3 \times 2$ ), 4.31 (4H, q,  $J=7.1$  Hz,  $\text{CO}_2\text{CH}_2\text{CH}_3 \times 2$ ), 7.92 (1H, d,  $J=11.1$  Hz, H-9), 8.28 (1H, s, H-6), 8.87 (2H, br s,  $\text{NH}_2$ ), 9.47 (1H, d,  $J=11.0$  Hz, H-8), 9.91 (1H, s, H-4);  $^{13}\text{C-nmr}$  (75.0 MHz,  $\text{DMSO-d}_6$ )  $\delta$  14.43, 59.38, 113.07, 114.51, 128.50, 130.35, 134.73, 135.30, 135.77, 136.73, 137.92, 163.64, 164.42, 172.62;  $\nu$  (KBr) 3330, 3328, 2986, 1660, 1515  $\text{cm}^{-1}$ ;  $m/z$  (40 ev) 344 ( $\text{M}^+$ ), 299, 271, 241, 200; HRms  $m/z$  Calcd for  $\text{C}_{17}\text{H}_{16}\text{N}_2\text{O}_4\text{S}$  344.0827, Found 344.0831.

**Azuleno[6,5-*d*]thiazole (5a) and its 2-amino derivative (5b)**

To a vigorously stirred hot 100% phosphoric acid (5 ml) was quickly added **4a** or **4b** (100 mg) at 100  $^\circ\text{C}$ . After only 15 min stirring, crushed ice (50 g) was added. The mixture was extracted with  $\text{CH}_2\text{Cl}_2$ . The organic extract was washed with saturated  $\text{NaHCO}_3$  and brine, then dried over anhydrous  $\text{Na}_2\text{SO}_4$ . After the solvent was removed, the residue was chromatographed on silica gel with  $\text{CH}_2\text{Cl}_2$  to give **5a** (56 mg,

90% ) as blue prisms (from EtOAc), mp 128-130 °C, and **5b** (58 mg, 92%) as purplish blue prisms (from EtOAc), mp 232 °C (decomp.), respectively.

**5a**:  $^1\text{H-Nmr}$  (300 MHz,  $\text{CDCl}_3$ )  $\delta$  7.35 (1H, d,  $J=3.7$  Hz, H-7), 7.42 (1H, d,  $J=3.7$  Hz, H-5), 7.81 (1H, t,  $J=3.7$  Hz, H-6), 7.90 (1H, d,  $J=10.9$  Hz, H-9), 8.27 (1H, d,  $J=10.9$  Hz, H-8), 8.80 (1H, s, H-4), 9.10 (1H, s, H-2);  $^{13}\text{C-nmr}$  (75.0 MHz,  $\text{CDCl}_3$ )  $\delta$  118.29, 119.11, 121.17, 125.02, 128.67, 132.01, 134.32, 136.04, 154.86, 155.66, 160.10; ir (KBr) 2958, 2920, 1656, 1560, 1384  $\text{cm}^{-1}$ ; ms (40 ev)  $m/z$  185 ( $\text{M}^+$ ), 158, 149; HRms  $m/z$  Calcd for  $\text{C}_{11}\text{H}_7\text{NS}$  185.0299, Found 185.0312.

**5b**:  $^1\text{H-Nmr}$  (300 MHz, Acetone- $d_6$ )  $\delta$  7.17 (1H, d,  $J=3.7$  Hz, H-7), 7.24 (1H, d,  $J=3.7$  Hz, H-5), 7.33 (2H, br s,  $\text{NH}_2$ ), 7.38 (1H, d,  $J=10.6$  Hz, H-9), 7.53 (1H, t,  $J=3.7$  Hz, H-6), 8.19 (1H, d,  $J=10.7$  Hz, H-8), 8.64 (1H, s, H-4);  $^{13}\text{C-nmr}$  (75.0 MHz, Acetone- $d_6$ )  $\delta$  116.35, 117.37, 119.30, 127.67, 129.47, 133.14, 134.02, 135.42, 136.21, 160.28, 170.30; ir (KBr) 3389, 3280, 1633, 1590, 1496  $\text{cm}^{-1}$ ; ms (40 ev)  $m/z$  200 ( $\text{M}^+$ ), 173, 158, 146; HRms  $m/z$  Calcd for  $\text{C}_{11}\text{H}_8\text{N}_2\text{S}$  200.0410, Found 200.0408.

**Diethyl 2-methylazuleno[6,5-*d*]thiazole-5,7-dicarboxylate (6a) and its 6-acetylamino derivative (6b)**

To a stirred solution of **2** (500 mg, 1.24 mmol) and NaOAc (580 mg, 4.3 mmol) in glacial acetic acid (10 ml) and acetic anhydride (10 ml) at 25 °C was added zinc dust (1.5 g, 23 mmol) over a period of 10 min. The color of the solution changed to brown as the zinc was added. After stirring for 5 h the suspension was poured into water and the whole extracted with  $\text{CH}_2\text{Cl}_2$ . The organic extract was washed with saturated  $\text{NaHCO}_3$  and brine, then dried over anhydrous  $\text{Na}_2\text{SO}_4$ . After the solvent was removed, the residue was chromatographed on silica gel with EtOAc and n-hexane (1:1) to give two eluates. The first eluate afforded **6a** (64 mg, 15%) as pink prisms (from  $\text{CH}_2\text{Cl}_2$ ); mp 146-148 °C. The second eluate gave **6b** (263 mg, 53%) as brownish yellow prisms (from EtOAc); mp 175-177 °C.

**6a**:  $^1\text{H-nmr}$  (300 MHz,  $\text{CDCl}_3$ )  $\delta$  1.47 (6H, t,  $J=7.0$  Hz,  $\text{CO}_2\text{CH}_2\text{CH}_3 \times 2$ ), 2.97 (3H, s), 4.45 (4H, q,  $J=7.0$  Hz,  $\text{CO}_2\text{CH}_2\text{CH}_3 \times 2$ ), 8.40 (1H, d,  $J=11.3$  Hz, H-9), 8.80 (1H, s, H-6), 9.80 (1H, d,  $J=11.3$  Hz, H-8), 10.40 (1H, s, H-4); ir (KBr) 2935, 1701, 1686, 1471  $\text{cm}^{-1}$ ; ms (40 ev)  $m/z$  343 ( $\text{M}^+$ ), 298, 270, 199; HRms  $m/z$  Calcd for  $\text{C}_{18}\text{H}_{17}\text{NO}_4\text{S}$  343.3973, Found 343.3975.

**6b**:  $^1\text{H-nmr}$  (300 MHz,  $\text{CDCl}_3$ )  $\delta$  1.46 (6H, t,  $J=7.0$  Hz,  $\text{CO}_2\text{CH}_2\text{CH}_3 \times 2$ ), 2.30 (3H, s,  $\text{NHCOCH}_3$ ), 2.93 (3H, s), 4.46 (4H, q,  $J=7.0$  Hz,  $\text{CO}_2\text{CH}_2\text{CH}_3 \times 2$ ), 8.29 (1H, d,  $J=11.3$  Hz, H-9), 9.22 (1H, d,  $J=11.3$  Hz, H-8), 9.87 (1H, s, H-4), 10.45 (1H, br s,  $\text{NHCOCH}_3$ ); ir (KBr) 3259, 2979, 2909, 1711, 1693, 1652  $\text{cm}^{-1}$ ; ms (40 ev)  $m/z$  400 ( $\text{M}^+$ ), 358, 314, 240; HRms  $m/z$  Calcd for  $\text{C}_{20}\text{H}_{20}\text{N}_2\text{O}_5\text{S}$  400.4492, Found 400.4489.

**Diethyl 6-amino-2-methylazuleno[6,5-*d*]thiazole-5,7-dicarboxylate (7)**

Compound (**7**) was synthesized from **6b** using the same procedure as that described for **3** (98%); mp 268 °C (decomp.).

$^1\text{H-Nmr}$  (300 MHz,  $\text{CDCl}_3$ )  $\delta$  1.48 (6H, t,  $J=7.0$  Hz,  $\text{CO}_2\text{CH}_2\text{CH}_3 \times 2$ ), 2.87 (3H, s), 4.49 (4H, q,  $J=7.0$  Hz,  $\text{CO}_2\text{CH}_2\text{CH}_3 \times 2$ ), 7.70 (2H, br s,  $\text{NH}_2$ ), 8.20 (1H, d,  $J=11.4$  Hz, H-9), 9.21 (1H, d,  $J=11.4$  Hz, H-8), 9.73 (1H, s, H-4); ir (KBr) 3490; 2935, 1683, 1475  $\text{cm}^{-1}$ ; ms (40 ev)  $m/z$  358 ( $\text{M}^+$ ), 313; HRms  $m/z$  Calcd for  $\text{C}_{18}\text{H}_{18}\text{N}_2\text{O}_4\text{S}$  358.4119, Found 358.4114.

### 2-Methylazuleno[6,5-*d*]thiazole (**8**)

Compound (**8**) was synthesized from **6a** using the same procedure as that described for **5a-b** (95%); mp 92-94 °C.

$^1\text{H-Nmr}$  (300 MHz,  $\text{CDCl}_3$ )  $\delta$  2.90 (3H, s), 7.34 (1H, d,  $J=3.45$  Hz, H-7), 7.40 (1H, d,  $J=3.45$  Hz, H-5), 7.81 (1H, t,  $J=3.45$  Hz, H-6), 7.80 (1H, d,  $J=10.9$  Hz, H-9), 8.29 (1H, d,  $J=10.9$  Hz, H-8), 8.72 (1H, s, H-4);  $^{13}\text{C-nmr}$  (75.0 MHz,  $\text{CDCl}_3$ )  $\delta$  20.25, 117.83, 118.29, 120.47, 128.03, 128.60, 131.52, 132.34, 134.82, 135.53, 155.66, 169.40; ir (KBr) 2948, 2915, 1653, 1545, 1378  $\text{cm}^{-1}$ ; ms (40 ev)  $m/z$  199 ( $\text{M}^+$ ), 158, 117; HRms  $m/z$  Calcd for  $\text{C}_{12}\text{H}_9\text{NS}$  199.2701, Found 199.2706.

### 5-Bromo-7,9-azulenequinono[6,5-*d*]thiazole (**9a**) and its 2-methyl derivative (**9b**)

To a stirred solution of **5a** or **8** (100 mg) in 10 % aqueous THF (20 ml) was added 4.3 equiv. of bromine (0.37 g) in acetic acid (4 ml) during 3 min at 5-10 °C. After stirring for 1 h water (20 ml) was added, the solution was kept at room temperature overnight and extracted with  $\text{CH}_2\text{Cl}_2$ . The combined organic layers were dried over anhydrous  $\text{Na}_2\text{SO}_4$ , evaporated in *vacuo*, and the residue was purified by column chromatography (silica gel,  $\text{CH}_2\text{Cl}_2$ ) to give **9a** (83%) as pale yellow needles (from  $\text{CH}_2\text{Cl}_2$ ); mp 162 °C (decomp.), **9b** (87%) as a pale yellow needles (from  $\text{CH}_2\text{Cl}_2$ ); mp 176 °C (decomp.).

**9a**:  $^1\text{H-Nmr}$  (300 MHz,  $\text{CDCl}_3$ )  $\delta$  6.92 (1H, s, H-6), 7.34 (1H, s, H-4), 7.95 (1H, s, H-8), 9.15 (1H, s, H-2); ir (KBr) 2958, 2854, 1730, 1701, 1636, 1465  $\text{cm}^{-1}$ ; ms (40 ev)  $m/z$  295 ( $\text{M}^+ + 2$ ), 293 ( $\text{M}^+$ ), 267, 265, 186, 158.

**9b**:  $^1\text{H-Nmr}$  (300 MHz,  $\text{CDCl}_3$ )  $\delta$  2.90 (3H, s), 6.89 (1H, s, H-6), 7.33 (1H, s, H-4), 7.85 (1H, s, H-8); ir (KBr) 2922, 2854, 1724, 1702, 1639, 1471  $\text{cm}^{-1}$ ; ms (40 ev)  $m/z$  309 ( $\text{M}^+ + 2$ ), 307 ( $\text{M}^+$ ), 295, 293, 229, 200.

### ACKNOWLEDGMENT

The authors thank the National Science Council of Republic of China and Chen Shui-Chin Foundation for financial support and the Japan Academy for partial support of reagents (to T. N.).

## REFERENCES

1. N. Tsukahara, H. Yamaguchi, M. Higashi, K. Fujimori, and T. Takei, *Spectrochimica Acta*, 1995, **51A**, 729.
2. H. Yamaguchi, M. Higashi, K. Fujimori, and T. Kobayashi, *J. Chem. Soc., Faraday Trans. 2*, 1989, **85**(2), 157.
3. K. Yamane, K. Fujimori, S. Ichikawa, S. Miyoshi, and K. Hashizume, *Heterocycles*, 1983, **20**, 1263; S. Nikolic, A. Juric, and N. Trinajstic, *Heterocycles*, 1987, **26**, 2025.
4. T. Nozoe, S. Seto, and S. Matsumura, *Chem. and Ind.*, 1961, 1715.
5. K. Matsui and T. Nozoe, *Chem. and Ind.*, 1960, 1302.
6. T. Nozoe, K. Takase, and M. Tada, *Bull. Chem. Soc. Jpn.*, 1969, **36**, 1016.
7. T. Nozoe, T. Asao, and M. Kobayashi, *Bull. Chem. Soc. Jpn.*, 1973, **46**, 3161.
8. K. Fujimori, H. Kitahashi, S. Koyama, and K. Yamane, *Bull. Chem. Soc. Jpn.*, 1986, **59**, 3320.
9. T. Nozoe and K. Kikuchi, *Chem. and Ind.*, 1962, 358.
10. K. Hafner, H. J. Linder, and W. Wassem, *Heterocycles*, 1978, **11**, 387.
11. T. Nozoe, P. W. Yang, H. Ogawa, and T. Toda, *Bull. Chem. Soc. Jpn.*, 1968, **41**, 2095.
12. L. L. Replogle, K. Katsumoto, T. C. Morrill, and C. A. Minor, *Tetrahedron Lett.*, 1965, 1877.
13. T. Morita, M. Karasawa, and K. Takase, *Chem. Lett.*, 1980, 197.
14. T. Nozoe, H. Wakabayashi, K. Shindo, and M. Yasunami, *Chem. Lett.*, 1995, 1877.
15. L.-J. Wang, Ms. Thesis, 1995; M. Tada, *Bull. Chem. Soc. Jpn.*, 1966, **39**, 1954.
16. R. N. MacDonald, and J. M. Richmond, *J. Chem. Soc. Chem. Comm.*, 1976, 605.
17. I. A. Ismail, D. E. Sharp, and M. R. Chedekel, *J. Org. Chem.*, 1980, **45**, 2243.

Received, 9th February, 1996