

**A FACILE SYNTHESIS OF NINE- AND TEN- MEMBERED  
CYCLIC ETHERS UTILIZING INTRAMOLECULAR *ENDO*-MODE  
CYCLIZATION OF THE CONJUGATED ALLENYL KETONES**

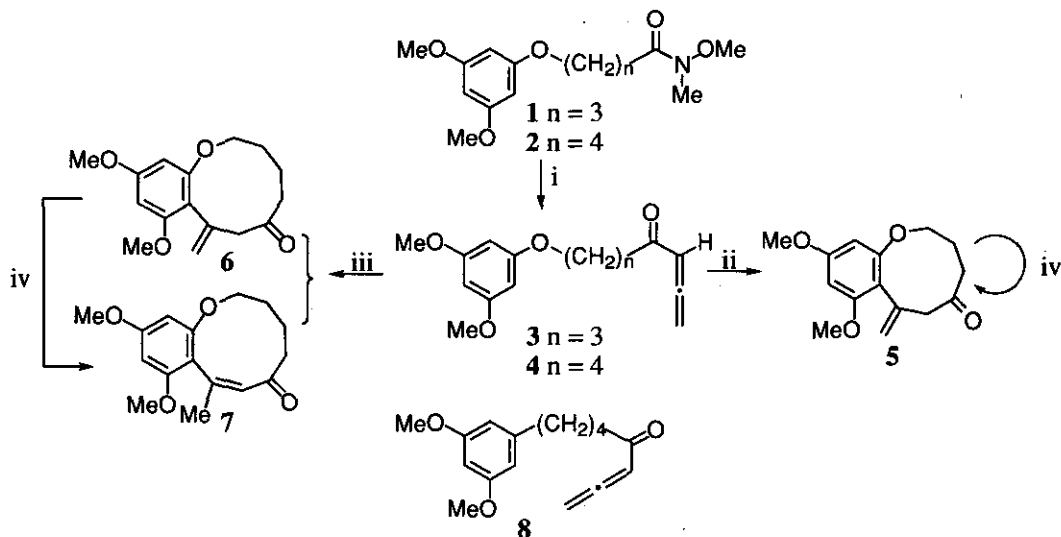
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**Abstract** - Intramolecular *endo*-mode cyclization reactions of the conjugated allenyl ketones (3) and (4) were carried out in the presence of BF<sub>3</sub>·OEt<sub>2</sub>, B(CF<sub>3</sub>SO<sub>3</sub>)<sub>3</sub>, and CF<sub>3</sub>SO<sub>3</sub>H in CH<sub>2</sub>Cl<sub>2</sub> to give the corresponding nine- and ten-membered cyclic ethers (5 - 7) as crystalline compounds in satisfactory yields. The structure of 6 was confirmed by its X-ray crystallographic analysis.

Six- to nine-membered cyclic ethers have often appeared in marine natural products<sup>1</sup> such as brevetoxins<sup>2</sup> and ciguatoxin.<sup>3</sup> In general, an annulation reaction toward the medium-sized eight- to eleven-membered cyclic compounds seems to be difficult from the viewpoint of unfavorable entropy factors.<sup>4</sup> Recently, we have disclosed various cyclization reactions for five- to eight-membered carbocyclic compounds,<sup>5a,b</sup> spiro[4.5]decatriene and [5.5]undecatriene diones,<sup>5c</sup> and seven-membered cyclic ethers<sup>5d</sup> and amides<sup>5e</sup> on the basis of intramolecular *endo*-mode ring closure at the sp carbon atom of the conjugated allenyl ketone moiety. In the earlier studies on remarkable *endo*-mode cyclization reactions,<sup>5</sup> we realized that the conjugated allenyl ketone system bearing a substituted phenyl group should be suitable enough for the intramolecular cyclization being difficult in the conjugated alkenyl and alkynyl ketones. Here, we describe a facile construction method of nine- and ten-membered cyclic ethers.



**Scheme 1** i, Propargylmagnesium bromide (1.5 mol eq.), THF, 0 °C, (**3** : 53%, **4** : 62%); ii,  $\text{CF}_3\text{SO}_3\text{H}$  (10 mol %),  $\text{CH}_2\text{Cl}_2$ , 0 °C (63%); iii,  $\text{B}(\text{CF}_3\text{SO}_3)_3$  (2.4 mol eq.),  $\text{CH}_2\text{Cl}_2$ , -78 °C (54%); iv, NaH (1.2 mol eq.), THF, reflux (**7** : 56% with 40% recovery of **6**, 84% recovery of **5**)

Conjugated allenyl ketones (**3**) [colorless prisms, mp 39 - 40 °C ( $\text{CH}_2\text{Cl}_2$  - hexane)] and (**4**) (pale yellow oil) were readily prepared by treatment of the corresponding *N*-methoxy-*N*-methylamides (**1**) and (**2**)<sup>6,7</sup> with 1.5 mol eq. of propargylmagnesium bromide (1M solution in  $\text{Et}_2\text{O}$ )<sup>5a</sup> in THF at 0 °C. Treatment of **3** and **4** with 1.2 mol eq. of  $\text{BF}_3 \cdot \text{OEt}_2$  in  $\text{CH}_2\text{Cl}_2$  at -78 °C for 5 min gave the desired nine-membered cyclic ether (**5**) [colorless prisms, mp 69 - 70 °C ( $\text{CH}_2\text{Cl}_2$  - hexane)] in 30% yield and ten-membered one (**6**) [colorless prisms, mp 92 °C ( $\text{CH}_2\text{Cl}_2$  - hexane)] in 16% yield, respectively. Although treatment of **3** with 2.4 mol eq. of  $\text{B}(\text{CF}_3\text{SO}_3)_3$ <sup>8</sup> at -78 °C gave the compound (**5**) in only 27% yield, the same reaction of **4** afforded a mixture of exo-methylene cyclic ether (**6**) and endo-olefinic one (**7**) in 54% total yield. The mixture could be separated on a silica gel plate (Merck Kieselgel 60 F254) with ether - hexane (2 : 1) to give each pure compound (**6**) or (**7**) [colorless prisms, mp 78 °C ( $\text{CH}_2\text{Cl}_2$  - hexane)]. Both compounds (**3**) and (**4**) were allowed to react with a catalytic amount (10 mol %) of  $\text{CF}_3\text{SO}_3\text{H}$  in  $\text{CH}_2\text{Cl}_2$  at 0 °C for 5 min to furnish the corresponding cyclic ether (**5**) in 63% yield and the mixture of **6** and **7** in a ratio of 1.3 : 1 and in 52% total yield. All experimental results for cyclization reactions as described above are summarized in Table 1. Interestingly, in contrast to successful cyclization of the ether (**3**), similar reactions of the methylene derivative (**8**) employing various Lewis acids and  $\text{CF}_3\text{SO}_3\text{H}$  resulted in production of a complex mixture.

The structure of exo-methylene product (**6**) was established by its X-ray crystallographic analysis<sup>9</sup> as shown in Figure 1. The structure of endo-olefinic product (**7**) was confirmed by identification with all

**Table 1.** Intramolecular Cyclization of Conjugated Allenyl Ketones (**3**) and (**4**).

Compd	Reaction Conditions		Product	Yield <sup>b)</sup> /%
	LA <sup>a)</sup> or A <sup>a)</sup>	Temp/°C		
n = 3				
<b>3</b>	B	-78	5	30
"	BT	"	"	27
"	A	0	"	63
n = 4				
<b>4</b>	B	-78	<b>6</b>	16
"	BT	"	<b>6 + 7</b>	54 <sup>c)</sup>
"	A	0	"	52 <sup>d)</sup>

a) LA : Lewis acid. B =  $\text{BF}_3 \cdot \text{OEt}_2$  (1.2 mol eq.), BT =  $\text{B}(\text{CF}_3\text{SO}_3)_3$  (2.4 mol eq.), A =  $\text{CF}_3\text{SO}_3\text{H}$  (10 mol %). b) Isolation yield. c) Yield of a mixture of **6** and **7** (3.6 : 1). d) Yield of a mixture of **6** and **7** (1.3 : 1).

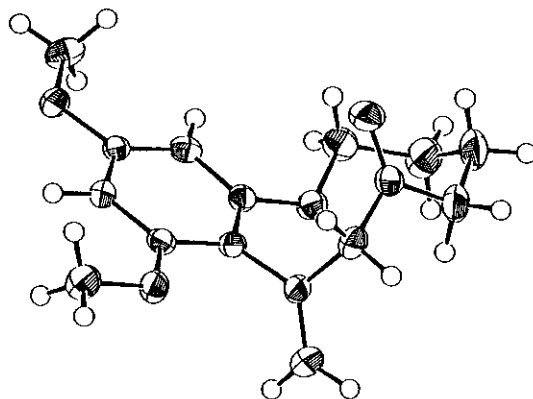


Figure 1. Perspective view of the crystallographic structure of exo - **6**.

spectroscopic data [HRms calcd for  $\text{C}_{16}\text{H}_{20}\text{O}_4$  MW 276.1362, found  $m/z$  276.1360 ( $\text{M}^+$ ); ir (KBr)  $\nu$  1734  $\text{cm}^{-1}$  ( $\alpha, \beta$ -unsaturated carbonyl);  $^1\text{H}$  nmr (400 MHz,  $\text{CDCl}_3$ )  $\delta$  2.13 (d, 3H,  $J = 1.5$  Hz, olefinic Me), 3.75 (t, 2H,  $J = 4.9$  Hz,  $-\text{OCH}_2\text{CH}_2-$ ), 6.05 and 6.16 (each d, each 1H,  $J = 2.5$  Hz, aromatic H), 6.23 (d, 1H,  $J = 1.5$  Hz, olefinic H)] of the compound (**6**) under the basic conditions (NaH, THF, reflux for 3 h) as shown in Scheme 1. The structure of **5** was determined by its similar characteristic spectroscopic data [HRms calcd for  $\text{C}_{15}\text{H}_{18}\text{O}_4$  MW 262.1205, found  $m/z$  262.1185 ( $\text{M}^+$ ); ir (KBr)  $\nu$  1704  $\text{cm}^{-1}$  (carbonyl), 1643  $\text{cm}^{-1}$  (exo-methylene);  $^1\text{H}$  nmr (400 MHz,  $\text{CDCl}_3$ )  $\delta$  3.19 (s, 2H, allylic H), 4.11 (t, 2H,  $J = 5.4$  Hz,  $-\text{OCH}_2\text{CH}_2-$ ), 5.35 and 5.56 (each d, each 1H,  $J = 2.0$  Hz, olefinic H), 6.14 and 6.22 (each d, each 1H,  $J = 2.5$  Hz, aromatic H)] to

those [HRms calcd for  $C_{16}H_{20}O_4$  MW 276.1362, found  $m/z$  276.1353 ( $M^+$ ); ir (KBr)  $\nu$  1706  $cm^{-1}$  (carbonyl), 1646  $cm^{-1}$  (exo-methylene);  $^1H$  nmr (400 MHz,  $CDCl_3$ )  $\delta$  3.20 (s, 2H, allylic H), 3.86 (t, 2H,  $J = 2.4$  Hz,  $-OCH_2CH_2-$ ), 5.08 and 5.52 (each d, each 1H,  $J = 2.0$  Hz, olefinic H), 5.93 and 6.13 (each d, each 1H,  $J = 2.4$  Hz, aromatic H)] of the compound (6). The double bond shift (exo $\rightarrow$ endo) of 5 was not recognized in the presence of NaH in THF under reflux for 4 h, instead the starting compound (5) was obtained in 84% recovery. This remarkable localization of the exo-methylene double bond in the nine-membered cyclic ether (5) under basic or acidic conditions has never been observed in other similar cyclic compounds such as seven- and eight-membered carbocyclic compounds,<sup>5b</sup> seven-membered cyclic ethers and amides,<sup>5d,e</sup> and ten-membered cyclic ether (6).

## REFERENCES AND NOTES

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7. Amides (1) and (2) were readily synthesized by starting from commercially available 3,5-dimethoxyphenol via the several reaction steps.
8. A solution of  $B(CF_3SO_3)_3$  was prepared by treatment of  $BH_3 \cdot THF$  (1M solution in THF) with 3.0 mol eq. of  $CF_3SO_3H$  in  $CH_2Cl_2$  at 0  $^\circ C$  for 10 min.<sup>5e</sup>
9. The crystallographic data of 6 :  $C_{16}H_{20}O_4$ , FW = 276.33, trigonal,  $R\bar{3}$  (#148),  $a = 32.093(3)$   $\text{\AA}$ ,  $c = 7.204(3)$   $\text{\AA}$ ,  $V = 6425(2)$   $\text{\AA}^3$ ,  $z = 18$ ,  $D_{calc} = 1.285$   $g/cm^3$ ,  $R = 0.033$ .