STUDIES ON THE STEREOCHEMISTRY OF THEONEZOLIDES A-C¹: ELUCIDATION OF THE RELATIVE CONFIGURATIONS OF 1,3-DIOL MOIETIES OF THE C-4 ~ C-17 FRAGMENT

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Abstract- Four model compounds having *syn* and *anti* 1,3-diol type moieties corresponding to C-8/C-10 and C-14/C-16 positions contained in the C-4 ~ C-17 fragment of theonezolides A-C were prepared. Comparison of their spectral data suggested that the 1,3-diol at C-8/C-10 and the OH/OMe groups at C-14/C-16 positions of theonezolides A-C were both *syn*.

Theonezolides A (1), B (2), and C (3) are novel cytotoxic 37-membered macrolides consisting of two principal fatty acid chains isolated from the Okinawan marine sponge *Theonella* sp., 1,2 which have been revealed to possess unique bioactivity of induction of rabbit platelet shape change and aggregation. Theonezolides A ~ C (1 ~ 3) contain 23 chiral centers, among which the absolute configuration of one chiral center at the terminal position bearing a primary amine and secondary methyl groups (C-75, C-73, and C-77 of 1 ~ 3, respectively) was determined as all R on the basis of synthesis of their ozonolysis products. Here we describe our study on the stereochemistry of the theonezolides as to the C-4 ~ C-17 fragment (4), 4 which was commonly obtained by ozonolysis of the three macrolides, theonezolides A ~ C (1 ~ 3). 1,2 This fragment (4) contains 4 chiral centers which comprise two 1,3-diol type moieties (14-OAc/16-OMe and 8-OAc/10-OAc). Their relative configurations were investigated by preparation of four model compounds (5/6 and 7/8) corresponding to *syn* and *anti* diastereomers for the 14-OAc/16-OMe and 8-OAc/10-OAc moieties, respectively. As a result, the two 1,3-diol type moieties were both suggested as *syn* as described below.

Synthesis of the four model compounds (syn: 5 and 7; anti: 6 and 8) began with the diastereoisomeric homoallyl alcohols [9 (syn) and 10 (anti)], respectively, which were both prepared from (-)-(S)-malic acid by literature procedures.⁵ Preparation of the diastereomer (5) was outlined in Scheme 1. The secondary hydroxyl group of the syn homoallyl alcohol (9) was protected as a benzyl ether, and the

[†]Dedicated to Dr. Bernhard Witkop on the occasion of his 80th birthday

acetonide group was deprotected to give the 1,2-diol (11). The primary hydroxyl group of 11 was protected as triphenylmethyl (Tr) group, and the remaining hydroxyl group was then methylated to furnish the methyl ether (12). Hydrogenolysis of both benzyl and trityl groups of 12 followed by acetylation afforded the *syn* diastereomer (5).⁶ The *anti* diastereomer (6)⁶ was obtained by the same procedures for 5 (Scheme 1), starting from the *anti* homoallyl alcohol (10).

The syn diacetate (7) was prepared as shown in Scheme 2. After removal of the acetonide group of the syn homoallyl alcohol (9), the primary hydroxyl group was selectively protected as t-butyldiphenylsilyl (TBDPS) group to give the 1,3-diol (13). Protection of the 1,3-diol group of 13 as isopropylidene ketal followed by desilylation afforded the acetonide (14), which was oxidized with pyridinium chlorochromate and the resulting aldehyde was subjected to Wittig reaction with n-hexyltriphenylphosphonium bromide to give the Z-olefin (15) predominantly. Hydrogenation of 15 followed by hydrolysis of the acetonide and acetylation furnished the syn diastereomer (7). Starting from the anti homoallyl alcohol (10), the anti diacetate (8) was also prepared by the same methods as shown in Scheme 2.

Comparisons of the ¹H and ¹³C NMR data of synthetic syn and anti diastereomers (5/6 and 7/8) with

Scheme 1. (a) 1) BnBr, NaH, *n*-Bu4N⁺I⁻, DMF (79%); 2) 3N HCl, THF (99%). (b) 1) Ph₃CCl, DMAP, pyridine (64%); 2) MeI, KH, THF (92%). (c) 1) H₂, 20% Pd(OH)₂/C, EtOH (66%); 2) Ac₂O, pyridine (85%).

Scheme 2. (a) 1) 3N HCl, THF (80%); 2) t-BuPh₂SiCl, imidazole, CH₂Cl₂ (53%). (b) 1) (Me)₂C(OMe)₂, PPTS, CH₂Cl₂ (72%); 2) 2N NaOH, EtOH (50%). (c) 1) PCC, MS 3Å, CH₂Cl₂ (72%); 2) n-BuLi, Me(CH₂)₅P+Ph₃Br⁻, THF (32%, 2 steps). (d) 1) H₂, Pd/C (10%), EtOH (84%); 2) 3N HCl, THF (99%); 3) Ac₂O, pyridine (65%).

those of the C-4 ~ C-17 fragment (4) derived from the natural specimens were outlined in Tables 1 and 2, which apparently showed that the ${}^{1}H$ and ${}^{13}C$ chemical shifts of syn diastereomers (5 and 7) corresponded to those of 4 quite better than those of anti diastereomers (6 and 8), as to both of the C-14 ~ C-17 positions and the C-8 ~ C-10 positions, respectively. Particularly, the ${}^{1}H$ NMR signals for

Table 1. Comparison of the ¹H and ¹³C NMR Data of 4 (natural), 5 (synthetic, syn), and 6 (synthetic, anti)⁴

position	4	5	6	position	4	5	6
-		δ_{C}				δ _Η	
14	70.9	71.0	70.7	14	5.24 m	5.21 m	5.40 m
15	36.4	36.5	37.6	15	1.90 m	1.89 m	1.60 (2H) m
	2011				1.64 m	1.66 m	
16	76.5	76.6	76.0	16	3.40 m	3.38 m	3.34 m
17	64.8	64.8	65.5	17	4.24 dd	4.25 dd	4.17 m
• •	00	0	55.5		J = 4.1, 11.8	J = 4.0, 11.7	
					4.08 dd	4.07 dd	3.99 m
				J = 5.0, 11.8	J = 4.1, 11.5		
OMe	56.8	56.8	57.9	OMe	3.20 s	3.17 s	3.28 s

Table 2. Comparison of the ¹H and ¹³C NMR Data of 4 (natural), 7 (synthetic, syn), and 8 (synthetic, anti)⁴

position	4	7	8	position	4	7	8
8,10 9	70.8,70.4 39.3	δ _C 71.2,70.9 39.3	69.8,69.5 39.1	8,10 9	1.90 m	δ _H 5.22 m 1.90 m 1.65 m	5.24 m 1.80 (2H) m

methylene protons located between the two oxymethines were characteristic: the two methylene protons of syn diastereomers (5) (H₂-15) and (7) (H₂-9) were magnetically non-equivalent, while those of anti diastereomers (6) (H₂-15) and (8) (H₂-9) resonated equivalently.⁷ The methylene proton signals of 4 (H₂-15 and H₂-9) were both magnetically non-equivalent. Thus, the two 1,3-diol type systems of the C-4 \sim C-17 fragment (4), the 14-OAc/16-OMe and 8-OAc/10-OAc moieties, were both suggested as syn. Consequently, out of sixteen possibilities, four feasible structures (4a and 4b and their enantiomers) now remain for the C-4 \sim C-17 fragment (4) of theonezolides A \sim C (1 \sim 3). Further investigation to establish the absolute stereochemistry of 4 is currently in progress by us on the basis of synthesis of 4a and 4b.

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- 6. **5**: $[\alpha]D^{31}$ -12.8° (*c* 1.3, CHCl₃); IR (neat) v_{max} 1740, 1360, and 1240 cm⁻¹; EIMS m/z 247 (M+H)+, 215 (M-OMe)+, 187 (M-CH₃CO₂)+, and 173 (M-CH₃CO₂CH₂)+. **6**: $[\alpha]D^{31}$ -2.9° (*c* 2.0, CHCl₃); IR (neat) v_{max} 1740, 1380, and 1240 cm⁻¹; EIMS m/z 247 (M+H)+, 215 (M-OMe)+, 187 (M-CH₃CO₂)+, and 173 (M-CH₃CO₂CH₂)+. 7: $[\alpha]D^{27}$ -5.1° (*c* 2.3, CHCl₃); IR (neat) v_{max} 1740, 1370, and 1240 cm⁻¹; FABMS (positive, matrix: 3-nitrobenzyl alcohol) m/z 301 (M+H)+. **8**: $[\alpha]D^{27}$ -4.7° (*c* 3.4, CHCl₃); IR (neat) v_{max} 1740, 1370, and 1240 cm⁻¹; FABMS (positive, matrix: 3-nitrobenzyl alcohol) m/z 301 (M+H)+.
- 7. For related arguments, see: J. S. Mynderse and R. E. Moore, *Phytochemistry*, 1979, **18**, 1181.