

REINVESTIGATION OF REACTION OF *CIS*-AZIRIDINYL KETONE SEMICARBAZONES AND TOSYLHYDRAZONES WITH DIETHYL ETHER-BORON TRIFLUORIDE(1/1)

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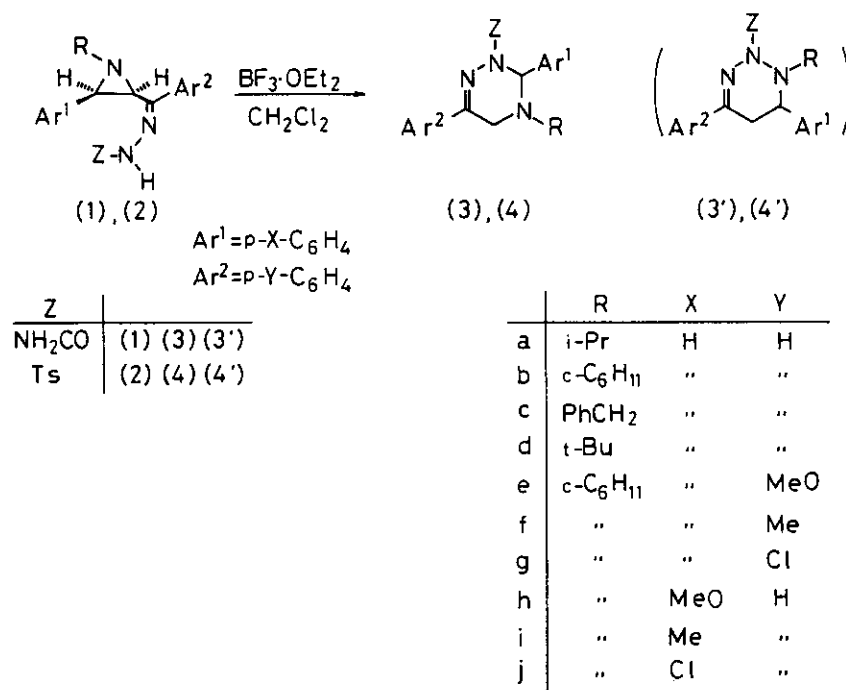
**Abstract**---A previously reported skeleton of reaction products of *cis*-1-alkyl-3-phenyl-aziridin-2-yl phenyl ketone semicarbazones (1) and tosyl hydrazones (2) with diethyl ether-boron trifluoride (1/1) was corrected to 2,3,4,5-tetrahydro-1,2,4-triazine derivatives (3) and (4), based on the results of reinvestigation with the X-Ray crystallography.

**The Structure of 2,3,4,5-tetrahydro-1,2,4-triazine derivatives.** Recently, we reported that 2-carbamoyl derivatives (3')<sup>1</sup> and 2-tosyl (4')<sup>2</sup> of 1-alkyl-4,6-diphenyl-1,2,5,6-tetrahydro-1,2,3-triazine were obtained in moderate yield via the rearrangement shown in Scheme 1. However, as a result of the X-Ray crystallography, it turned out that they were 2,3,4,5-tetrahydro-1,2,4-triazine derivatives (3) and (4) (Figure 1). C5 methylene in the structure (3' and 4') showed previously corresponds to that of 3 and 4, and C6 methine of 3' and 4' to C3 methine of 3 and 4. The IR, <sup>1</sup>H- and <sup>13</sup>C-NMR data presented in the previous papers<sup>1,2</sup> also supported the skeleton. Crystal of compound (4b) was grown by slow evaporation of methanol solutions at room temperature. The bond length between N1 and C6 was 1.300 Å and close to that of C=N double bond, comparing with those (1.28 Å) of oximes and imines reported in the literature.<sup>3</sup> The 1,2,4-triazine ring of 4b was in the half-chair conformation and N2-N1-C6-C5 atoms were coplanar. Ar<sup>1</sup> on C3 and R on N4 were in an axial configuration because of mutual steric hindrance.

#### Experimental

**Materials.** *cis*-1-Alkyl-3-phenylaziridin-2-yl phenyl ketone semicarbazones

(1)<sup>1</sup> and tosylhydrazones (2),<sup>2</sup> 4-alkyl-2-carbamoyl-3,6-diphenyl-2,3,4,5-tetrahydro-1,2,4-triazines (3),<sup>1</sup> and 4-alkyl-3,6-diphenyl-2-tosyl-2,3,4,5-tetrahydro-1,2,4-triazines (4)<sup>2</sup> were prepared according to the method described previously. NMR spectra of 3 and 4 <sup>1</sup>H-NMR data of 3 and 4 reported previously<sup>1,2</sup> were quoted in Tables 1 and 2.



Scheme 1

**X-Ray crystal structure determination of compound (4b).** Crystals of compound 4b were grown by slow evaporation of methanol solutions. A crystal of 0.2 mm X 0.2 mm X 0.3 mm was mounted on a glass fiber on Rigaku AFC7R diffractometer with graphite monochromated Cu-K $\alpha$  radiation and the lattice parameter were obtained by a least-squares refinement of 25 accurately centered reflections in the range 43.26° < 2 $\theta$  < 53.77°. The structure was solved by direct methods using the SHELXS 86 suit of programs and conventional Fourier syntheses. The nonhydrogen atoms were refined anisotropically, and hydrogen atoms isotropically.

**Crystal data** C<sub>28</sub>H<sub>31</sub>N<sub>3</sub>O<sub>2</sub>S, M = 473.63, orthorhombic, space group Pbca(#61), a = 17.248(2) Å, b = 17.380(1) Å, c = 16.956(2) Å, V = 5082.9(8) Å<sup>3</sup>, Z = 8, D<sub>o</sub> = 1.238 g·cm<sup>-3</sup>,  $\mu$ (CuK $\alpha$ ) = 13.59 cm<sup>-1</sup>, F<sub>000</sub> = 2016.00

**Data collection.** The data were collected at a temperature of  $20 \pm 1$  °C using the  $\omega$ - $2\theta$  scan technique to a maximum  $2\theta$  value of  $120.1^\circ$ . Scans of  $(1.57 + 0.30 \tan \theta)^\circ$  were made at a speed of  $16.0^\circ \text{min}^{-1}$  (in omega). Number of data collected was 4242, number with  $I \geq 3.00\sigma(I)$  1817.

**Structure refinement.**

$$R = \sum \| |F_o| - |F_c| \| / \sum |F_o| = 0.043$$

$$R_w = (\sum w (|F_o| - |F_c|)^2 / \sum w F_o^2)^{1/2} = 0.029$$

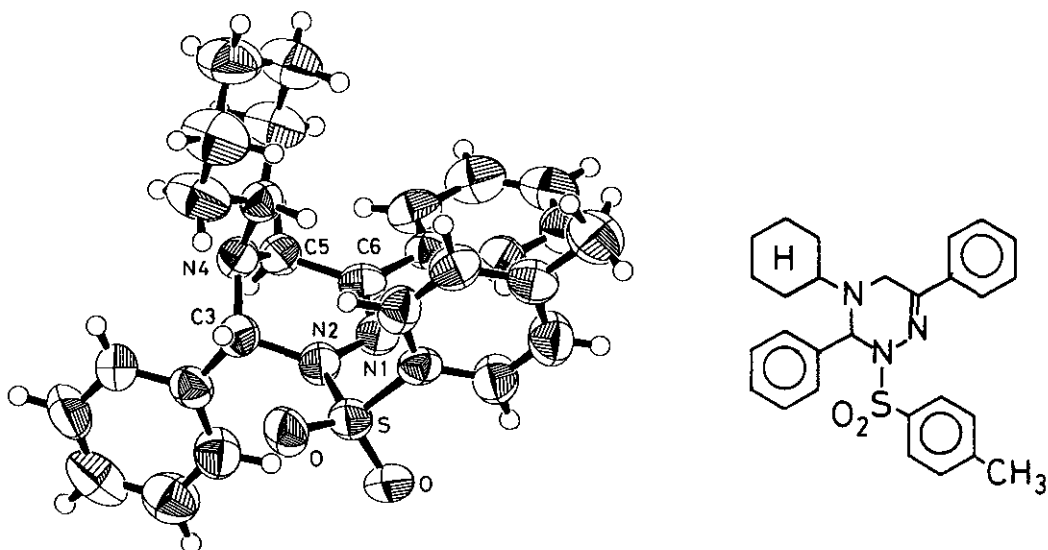


Figure 1

Table 1  $^1\text{H-NMR}$  data of compounds (3)

Compd	$\delta$ , $J$ /Hz(CDCl <sub>3</sub> , TMS)
3a	1.18(3H, d, $J=6.0$ , CH <sub>3</sub> ), 1.25(3H, d, $J=6.0$ , CH <sub>3</sub> ), 3.04(1H, septet, $J=6.0$ , <i>i</i> -propyl CH), 3.38(1H, d, $J=18.7$ , CH <sub>2</sub> ), 3.72(1H, dd, $J=18.7$ , 1.8, CH <sub>2</sub> ), 6.12(2H, br s, NH <sub>2</sub> ), 6.54(1H, m*, CHPh), 7.10-7.64(10H, m, Ph)
3b	1.15-2.05(10H, m, <i>c</i> -hexyl CH <sub>2</sub> ), 2.56-2.80(1H, m, <i>c</i> -hexyl CH), 3.38(1H, d, $J=18.4$ , CH <sub>2</sub> ), 3.74(1H, dd, $J=18.4$ , 1.8, CH <sub>2</sub> ), 6.08(2H, br s, NH <sub>2</sub> ), 6.55(1H, m*, CHPh), 7.10-7.90(10H, m, Ph)
3c	3.45(2H, s, CH <sub>2</sub> Ph), 3.70(1H, d, $J=13.2$ , CH <sub>2</sub> ), 3.97(1H, d, $J=13.2$ , CH <sub>2</sub> ), 5.01(2H, br s, NH <sub>2</sub> ), 6.32(1H, s, CHPh), 7.23-7.70(15H, m, Ph)
3d	1.26(9H, s, CH <sub>3</sub> ), 3.38(1H, d, $J=19.1$ , CH <sub>2</sub> ), 3.83(1H, dd, $J=19.1$ , 2.4, CH <sub>2</sub> ), 5.66(2H, br s, NH <sub>2</sub> ), 6.76(1H, m*, CHPh)**, 7.10-7.65(10H, m, Ph)
3e	1.15-2.05(10H, m, <i>c</i> -hexyl CH <sub>2</sub> ), 2.60-2.67(1H, m, <i>c</i> -hexyl CH), 3.37(1H, d, $J=18.6$ , CH <sub>2</sub> ), 3.68(1H, dd, $J=18.6$ , 2.0, CH <sub>2</sub> ), 3.80(3H, s, OCH <sub>3</sub> ), 5.15(2H, br s,

- NH<sub>2</sub>), 6.51(1H, m\*, CHPh), 6.87-7.56(9H, m, Ph)
- 3f 1.15-2.04(10H, m, c-hexyl CH<sub>2</sub>), 2.35(3H, s, CH<sub>3</sub>), 2.61-2.66(1H, m, c-hexyl CH), 3.39(1H, d, J=18.6, CH<sub>2</sub>), 3.68(1H, dd, J=18.6, 2.0, CH<sub>2</sub>), 5.25(2H, br s, NH<sub>2</sub>), 6.51(1H, m\*, CHPh), 7.15-7.51(9H, m, Ph)
- 3g 1.16-2.03(10H, m, c-hexyl CH<sub>2</sub>), 2.60-2.66(1H, m, c-hexyl CH), 3.39(1H, d, J=18.6, CH<sub>2</sub>), 3.66(1H, dd, J=18.6, 1.5, CH<sub>2</sub>), 5.35(2H, br s, NH<sub>2</sub>), 6.52(1H, m\*, CHPh), 7.22-7.53(9H, m, Ph)
- 3h 1.15-2.04(10H, m, c-hexyl CH<sub>2</sub>), 2.60-2.66(1H, m, c-hexyl CH), 3.43(1H, d, J=18.6, CH<sub>2</sub>), 3.70(1H, dd, J=18.6, 2.0, CH<sub>2</sub>), 3.75(3H, s, OCH<sub>3</sub>), 5.55(2H, br s, NH<sub>2</sub>), 6.47(1H, m\*, CHPh), 6.83-7.61(9H, m, Ph)
- 3i 1.15-2.04(10H, m, c-hexyl CH<sub>2</sub>), 2.29(3H, s, CH<sub>3</sub>), 2.60-2.65(1H, m, c-hexyl CH), 3.43(1H, d, J=18.6, CH<sub>2</sub>), 3.70(1H, dd, J=18.6, 1.5, CH<sub>2</sub>), 5.52(2H, br s, NH<sub>2</sub>), 6.45(1H, m\*, CHPh), 7.10-7.60(9H, m, Ph)
- 3j 1.15-2.02(10H, m, c-hexyl CH<sub>2</sub>), 2.61-2.67(1H, m, c-hexyl CH), 3.38(1H, d, J=18.6, CH<sub>2</sub>), 3.71(1H, dd, J=18.6, 1.5, CH<sub>2</sub>), 5.52(2H, br s, NH<sub>2</sub>), 6.48(1H, m\*, CHPh), 7.23-7.60(9H, m, Ph)

\* The coupling constants were not obtained on account of low resolution.

\*\* The coupling of C3 benzyl protons with one of C5 methylene protons was confirmed with H-H COSY spectroscopy.

Table 2 <sup>1</sup>H-NMR data of compounds (4)

Compd	$\delta$ , J /Hz (CDCl <sub>3</sub> , TMS)
4a	1.04(3H, d, J=6.0, CH <sub>3</sub> ), 1.24(3H, d, J=6.0, CH <sub>3</sub> ), 2.42(3H, s, tolyl CH <sub>3</sub> ), 2.40-2.73(1H, septet, J=6.0, i-propyl CH), 3.21(1H, d, J=18.6, CH <sub>2</sub> ), 3.71(1H, dd, J=18.6, 2.0, CH <sub>2</sub> ), 6.33(1H, m*, CHPh), 7.10-8.00(14H, m, Ph)
4b	0.80-2.30(11H, m, c-hexyl), 2.43(3H, s, tolyl CH <sub>3</sub> ), 3.20(1H, d, J=18.0, CH <sub>2</sub> ), 3.75(1H, dd, J=18.0, 2.0, CH <sub>2</sub> ), 6.34(1H, m*, CHPh), 7.00-8.10(14H, m, Ph)
4c	2.43(3H, s, tolyl CH <sub>3</sub> ), 3.35(2H, s, CH <sub>2</sub> Ph), 3.44(1H, d, J=17.2, CH <sub>2</sub> ), 3.68(1H, dd, J=17.2, 2.0, CH <sub>2</sub> ), 6.04(1H, m*, CHPh), 7.00-8.10(19H, m, Ph)
4d	1.20(9H, s, t-butyl CH <sub>3</sub> ), 2.40(3H, s, tolyl CH <sub>3</sub> ), 3.19(1H, d, J=19.2, CH <sub>2</sub> ), 3.76(1H, dd, J=19.2, 2.0, CH <sub>2</sub> ), 6.55(1H, d, J=2.0, CHPh), 7.16-7.89(14H, m, Ph)
4e	0.80-2.30(11H, m, c-hexyl), 2.40(3H, s, tolyl CH <sub>3</sub> ), 3.12(1H, d, J=17.4, CH <sub>2</sub> ), 3.69(1H, dd, J=17.4, 1.8, CH <sub>2</sub> ), 3.78(3H, s, OCH <sub>3</sub> ), 6.27(1H, m*, CHPh), 6.80-7.91(13H, m, Ph)
4f	0.90-2.30(11H, m, c-hexyl), 2.31(3H, s, tolyl CH <sub>3</sub> ), 2.38(3H, s, tolyl CH <sub>3</sub> ), 3.13(1H, d, J=18.0, CH <sub>2</sub> ), 3.69(1H, dd, J=18.0, 1.8, CH <sub>2</sub> ), 6.24(1H, m*, CHPh), 7.09-7.91(13H, m, Ph)
4g	0.90-2.30(11H, m, c-hexyl), 2.42(3H, s, tolyl CH <sub>3</sub> ), 3.15(1H, d, J=18.0, CH <sub>2</sub> ), 3.66(1H, dd, J=18.0, 2.0, CH <sub>2</sub> ), 6.30(1H, m*, CHPh), 7.24-7.91(13H, m, Ph)
4h	0.90-2.30(11H, m, c-hexyl), 2.42(3H, s, tolyl CH <sub>3</sub> ), 3.21(1H, d, J=18.0, CH <sub>2</sub> ), 3.73(1H, dd, J=18.0, 2.0, CH <sub>2</sub> ), 3.78(3H, s, OCH <sub>3</sub> ), 6.26(1H, m*, CHPh),

6.75-7.92(13H, m, Ph)

4i 0.90-2.30(11H, m, c-hexyl), 2.31(3H, s, tolyl CH<sub>3</sub>), 2.41(3H, s, tolyl CH<sub>3</sub>),  
3.18(1H, d, J=18.0, CH<sub>2</sub>), 3.70(1H, dd, J=18.0, 2.0, CH<sub>2</sub>), 6.22(1H, m\*, CHPh),  
7.02-8.00(13H, m, Ph)

4j 0.90-2.20(11H, m, c-hexyl), 2.43(3H, s, tolyl CH<sub>3</sub>), 3.13(1H, d, J=18.6, CH<sub>2</sub>),  
3.72(1H, dd, J=18.6, 2.0, CH<sub>2</sub>), 6.25(1H, m\*, CHPh), 7.26-7.92(13H, m, Ph)

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\* The coupling constants were not obtained on account of low resolution.

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