STEREOSELECTIVE SYNTHESIS OF HEXAHYDRO-1,4-THIAZEPIN-3-ONE AND DIHYDRO-1,4-THIAZIN-3-ONE DERIVATIVES

Michinori Karikomi, Tohru Yamazaki, Yukiko Abematsu, Kiyoshi Masuzawa, and Takashi Toda*

Department of Applied Chemistry, Faculty of Engineering, Utsunomiya University, Ishiicho Utsunomiya 321-8585, Japan

Abstract - Diastereomerically pure 4-benzyl-7-substituted 6-hydroxy-hexahydro-1,4-thiazepin-3-one and 4-benzyl-1'-hydroxyalkyl-1,4-tetra-hydrothiazin-3-one derivatives are obtained by the reaction of a 2,3-epoxy alkylamine derivative with methyl thioglycolate in good yields. trans-Epoxy amine derivatives gave trans-thiazepinone derivatives and anti-thiazepinone derivatives whilst cis-epoxy amines gave cis-thiazepinone and syn-thiazepinone derivatives. Substituent effects on regioselectivity were also examined with the use of several epoxy amine derivatives.

Recently, several synthetic methods for 2,3-epoxy amine derivatives (1) have been reported due to their unique structure and synthetic utility. As 1 possesses both a nucleophilic amino group and an electrophilic epoxide group, 1 is used for the synthesis of nitrogen-containing heterocycles. Previously, we reported the reaction of 1 with several heterocumulens gave nitrogen containing 1,3-heterocycles. In the course of this study, we envisioned that the reaction of 1 with thioglycolate esters would give sulfur- and nitrogen-containing 1,4-heterocycles.

As part of a program aimed at developing new epoxy amine transformation, we were interested in the general synthetic methods for 1,4-thio-aza-heterocyclic systems. Since 1,4-thio-aza-heterocycles are interesting as biologically active compounds,⁴ 1,4-thiazepinone derivatives are of particular interest as monocyclic analogues of the penicillin antibiotics.⁵ Although extensive work has been reported on penicillin antibiotics, the ring formation of 1,4-thiazepine derivative has been little investigated.⁶ In this communication we describe a new synthetic method for 4,7-substituted 6-hydroxyhexahydro-1,4-thiazepin-3-one derivatives (2) and 1,4-thiazinone derivatives (3).

When a solution of 1 with 1.2 equiv. mol of methyl thioglycolate was refluxed in methanol for 12 h (Method A), cis-4-benzyl-7-methy-6-hydroxyhexahydro-1,4-thiazepin-3-one (cis-2a) and $(6R^*,1'R^*)$ -4-benzyl-1'-hydroxymethyl-1,4-tetrahydrothiazin-3-one derivatives (syn-3a) were

obtained in 54 and 33% yields, respectively (Table 1, Entry 1). Similarly, when the reaction was conducted in the presence of 1.2 equiv. of KOH for 2 h (Method B), 2a and 3a were obtained in 48 and 29% yields, respectively (Entry 2). Several derivatives (1b and 1c) which possess propyl and phenyl groups instead of a methyl group were examined, and 2b and 2c were obtained predominantly (Entries 3-10). On the other hand, trisubstituted derivatives (1d or 1e) gave only 2d or 3e exclusively (Table 2).

As can be seen from Table 1, the reaction of the thiol with epoxides occurred predominantly at the C-3 position and stereospecifically gave the corresponding ring-closed product (2a-2c) in each case. The reaction has the following two features. (i) The configuration of the epoxide affects the regioselectivity, especially in the case of *cis*-1c and *trans*-1c (Entries 7-10). The phenyl substituted *trans*-1c gave *trans*-2c exclusively (Entries 9, 10). On the other hand, *cis*-1c gave mixtures of *cis*-2c and *syn*-3c in a ratio of ca. 6 : 4 (Entries 7, 8). (ii) The presence of KOH in the reaction (Method B) improves the reaction rate but does not affect the regioselectivity.

Table 1. Reaction of 2,3-epoxy amine with methyl thioglycolate

Entry	Substrate	Method (h) a)	Isolated Yields (%) of 2, 3		
1	<i>cis-</i> 1a , R = Me	A (12)	cis-2a (54),	syn- 3a (33)	
2		B (2)	cis-2a (48),	syn- 3a (29)	
3	<i>cis-</i> 1b , R = Pr	A (12)	cis-2b (49),	syn- 3b (21)	
4		B (2)	cis-2b (38),	syn- 3b (24)	
5	trans-1b, R = Pr	A (12)	trans- 2b (52),	anti- 3b (18)	
6		B (2)	trans-2b (47),	anti- 3b (8)	
7	<i>cis-</i> 1c , R = Ph	A (24)	cis-2c (37),	syn- 3c (26)	
8		B (2)	cis-2c (49),	syn- 3c (33)	
9	<i>trans-</i> 1c , R = Ph	A (24)	trans- 2c (68),	anti- 3c (0)	
10		B (2)	trans-2c (81),	anti- 3c (0)	

a) See text.

Substrate	Method (h)	^{a)} Isolated Yield (%)	Substra	ite Method (h)	lsolated Yield (%)
Ph NHBn Me		HO NO O	Me Me	NHBn	Me S N Bn
cis-1d	B (2)	cis- 2d (58)	1e	A (48) B (2)	3e (45) 3e (71)

Table 2. Reaction of trisubstituted 2,3-epoxy amine with methyl thioglycolate

a) See Text.

The structures of these compounds are elucidated on the basis of their NMR, MS, and IR data. The regiochemistry of the reaction can be unequivocally determined by 1 H NMR data. For example, 1 H NMR spectra of the thiazepinone derivatives (**2c**) show a doublet at 4.27 ppm ($J_{vicnal} = 1.5 \text{ Hz}$) due to the benzylic proton, whereas those of the thiazinone derivatives (**3c**) show it at 4.59 ppm ($J_{vicnal} = 5.1 \text{ Hz}$). The 13 C NMR spectrum of **2c** and **3c** was also in agreement with the suggested structure. The stereochemistry of these products was also confirmed by comparing their 1 H NMR and NOESY spectra.

A typical procedure for the reaction of 1 with methyl thioglycolate is as follows. A mixture of *cis-N*-benzyl-2,3-epoxybutylamine (1a) (177 mg, 1.0 mmol) and methyl tioglycolate (127 mg, 1.2 mmol) in methanol (2.5 mL) was refluxed for 12 h. After cooling, the mixture was poured into H_2O (*ca.* 30 mL), and the solid products were filtered and dried *in vacuo*. The solid was recrystallized from toluene to give 135 mg (54%) of pure *cis-*2a as a single diastereomer (mp 175-176 °C). The filtrate was extracted with chloroform (15 mL X 3). The organic phases were dried (Na_2SO_4) and concentrated. The residue was distilled by use of Kugelrohr apparatus to give 84 mg (33%) of pure *syn-*3a as a colorless liquid (bp 200-210 °C/0.08 mmHg). (Method A: Table 1, Entry 1).

Production of diastereomerically pure 2,3-epoxy amine derivatives employing our previously reported method is relatively easy.³ Therefore, our protocol demonstrates a convenient single-step synthesis of diastereomerically pure 1,4-thio-aza heterocyclic systems.

REFERENCES AND NOTES

- 1. M. Karikomi, T. Yamazaki, and T. Toda, *Chem. Lett.*, **1993**, 1787; H. Urabe, Y. Aoyama, and F. Sato, *J. Org. Chem.*, 1992, **57**, 5056.
- V. R. Gaertner, Tetrahedron Lett., 1964, 141; V. R. Gaertner, Tetrahedron, 1967, 23, 2123; L. Pégorier, Y. Petit, and M. Larchevêque, J. Chem., Soc., Chem. Commun., 1994, 633; K. J. M. Beresford, G. P. Howe, and G. Procter, Tetrahedron Lett., 1992, 33, 3355; G. Asensio, R. Mello, C. B-Bernardini, M. E. G-Núñez, and G. Castellano. J. Org. Chem., 1995, 60, 3692; M. Pégorier, M. Haddad, and M. Larchevêque, Synlett, 1996, 585; R. Najime, S. Pilard, and

- Michel Vaulter, *Tetrahedron Lett.*, 1992, **37**, 5351; J-Y. Lai, F-S. Wang, G-Z. Guo, and L-X. Dai, *J. Org. Chem.*, 1993, **58**, 6944.
- 3. M. Karikomi, T. Yamazaki, and T. Toda, Chem. Lett., 1993, 1965.
- 4. M. Botta, A. Crescenza, W. Magara, and F. Corelli, Tetrahedron. Lett., 1997, 38, 2775.
- N. J. Leonard and R. Y. Ning, J. Org. Chem., 1966, 31, 3928; M. F. Semmelhack and B. F. Gilman, J. Chem. Soc., Chem. Commun., 1971, 988; M. F. Semmelhack, S. Kunkes, and C. S. Lee, J. Chem. Soc. Chem. Commun, 1971, 698; W. Ried and H. Bopp, Synthesis, 1978, 211.
- 6. J. T. Sharp, "Comprehensive Heterocyclic Chemistry," ed. by A. R. Katritzky, C. W. Rees, and K. W. Lwowski, Pergamon Press, Oxford, 1984, Vol. 7, pp. 633-636.
- 7. M. Chini, P. Crotti, E. Giovani, F. Macchia, and, M. Pineschi, *Synlett*, **1992**, 303; J. M. Chong and K. B. Sharpless, *J. Org. Chem.*, 1985, **50**, 1560.
- 8. Compound *cis*-**2c**: mp 225-226 °C (chloroform); IR (KBr) 3320, 1610 cm⁻¹; ¹H NMR (DMSO- d_6 , 50 °C) δ 3.19 (1H, d, J = 13.8 Hz), 3.38 (1H, dd, J = 15.6, 6.9 Hz) 3.77 (1H, d, J = 15.6 Hz), 3.87 (1H, d, J = 13.8 Hz), 3.94 (1H, td, J = 6.9, 2.1 Hz), 4.06 (1H, d, J = 15.0 Hz), 4.27 (1H, d, J = 2.1 Hz), 4.81 (1H, d, J = 6.9 Hz), 5.23 (1H, d, J = 15.0 Hz), 7.2-7.4 (10H, m); ¹³C NMR (DMSO- d_6 , 50 °C) δ 34.24, 51.57, 52.10, 54.95, 67.26, 126.84, 127.07, 127.47, 127.87, 128.26, 128.39, 138.01, 139.92, 171.13; MS (70 eV, rel. intensity) 314 (M⁺+1, 23), 313 (M⁺, 100), 91 (43). Compound *syn*-**3c**: mp 157-159 °C (toluene); IR (KBr) 3320, 1610 cm⁻¹; ¹H NMR (CDCl₃) δ 3.26 (1H, d, J = 15.0 Hz), 3.36 (3H, m) 3.49 (1H, d, J = 15.0 Hz), 4.33 (1H, d, J = 14.7 Hz), 4.59 (1H, d, J = 5.1 Hz), 4.80 (1H, d, J = 14.7 Hz), 7.1-7.2 (4H, m), 7.2-7.3 (6H, m); ¹³C NMR (CDCl₃) δ 28.73, 49.17, 49.37, 50.49, 74.22, 126.11, 127.69, 128.12, 128.38, 128.64, 128.68, 136.58, 140.40, 167.91.

Received, 26th May, 1998