

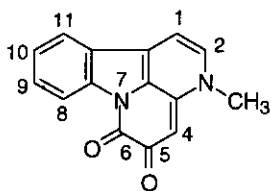
## SYNTHESIS AND ANTITUMOR ACTIVITY OF CANTHIN-5,6-DIONE DERIVATIVES

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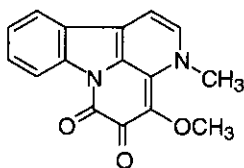
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**Abstracts-** Syntheses of canthin-5,6-dione derivatives have been achieved *via* one step route starting from their respective  $\beta$ -carbolines. Their synthetic compounds showed antitumor activities against P-388 murine leukemia cells and PC-6 human lung carcinoma cells.

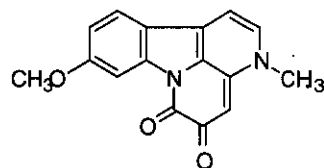
The picrasidine L (**1**),<sup>1,2</sup> picrasidine O (**2**)<sup>3</sup> and eurycomine E (**3**)<sup>4</sup> are novel canthin-5,6-dione alkaloids isolated from the Simaroubaceae plants, *Picrasma quassioides* Bennet and *Eurycoma longifolia* Jack. Their structures have been proposed on the basis of interpretation of spectral data and partial synthesis of their parent 5-hydroxycanthin-6-one derivatives.<sup>2,3</sup> In order to search for new lead compounds having potential for antitumor activity, we are interested in development of a facile synthesis of new canthin-5,6-diones which are a subclass of  $\beta$ -carboline alkaloid with diketonic D-ring. In this paper, we describe the one-pot synthesis and antitumor activity *in vitro* of canthin-5,6-dione derivatives.



Picrasidine L (**1**)



Picrasidine O (**2**)

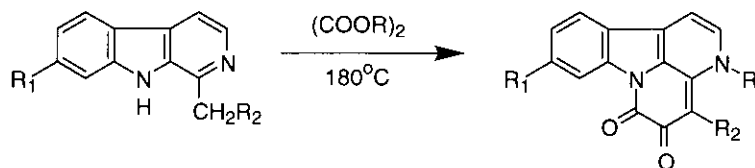


Eurycomine E (**3**)

### Synthesis

Canthin-5,6-diones (**1-9**) were synthesized by the route shown in Scheme 1. Cycloaddition of  $\beta$ -carbolines (**10-12**) with alkyl oxalates for 20-60 min at 180°C yielded *N*<sub>3</sub>-alkylcanthin-5,6-dione derivatives (**1-9**). The cycloaddition proceeded easily to give only canthin-5,6-dione derivatives without side reactions. However, the yield decreased with the increase in the carbon number of alkyl group of the alkyl oxalates. To confirm the cycloadducts, we conducted X-Ray crystallographic analysis of the synthetic sample of **2**. The crystal structure of **2** established the canthin-5,6-dione formation (Figure 1). The UV-VIS spectra of **2** showed a hypsochromic shift on the addition of acid, but was unchanged by

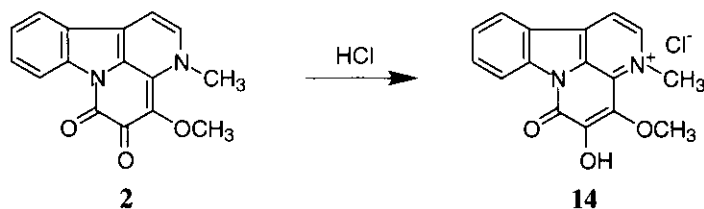
base. The hypsochromic shift of absorption maxima was very similar to that of 4-methoxy-5-hydroxycanthin-6-one (**13**).<sup>5</sup> In the <sup>1</sup>H-NMR spectrum of **2** in the presence of TFA, the chemical shifts were also similar to those of **13** in TFA.<sup>5</sup> All the spectral data indicated that protonation converted the 5-keto form (**2**) to the 5-enol form (**14**) in the acid condition (Scheme 2). Finally, the structure of the enol formation of **2** in an acidic medium was established by X-Ray crystallographic analysis of the hydrochloric acid salt of **2** (Figure 2).



	R <sub>1</sub>	R <sub>2</sub>		R	R <sub>1</sub>	R <sub>2</sub>
<b>10</b>	H	H	<b>1</b>	CH <sub>3</sub>	H	H
<b>11</b>	H	OCH <sub>3</sub>	<b>4</b>	CH <sub>2</sub> CH <sub>3</sub>	H	H
<b>12</b>	OCH <sub>3</sub>	H	<b>5</b>	CH <sub>2</sub> CH <sub>2</sub> CH <sub>2</sub> CH <sub>3</sub>	H	H
			<b>2</b>	CH <sub>3</sub>	H	OCH <sub>3</sub>
			<b>6</b>	CH <sub>2</sub> CH <sub>3</sub>	H	OCH <sub>3</sub>
			<b>7</b>	CH <sub>2</sub> CH <sub>2</sub> CH <sub>2</sub> CH <sub>3</sub>	H	OCH <sub>3</sub>
			<b>3</b>	CH <sub>3</sub>	OCH <sub>3</sub>	H
			<b>8</b>	CH <sub>2</sub> CH <sub>3</sub>	OCH <sub>3</sub>	H
			<b>9</b>	CH <sub>2</sub> CH <sub>2</sub> CH <sub>2</sub> CH <sub>3</sub>	OCH <sub>3</sub>	H

Reagents: (COOR)<sub>2</sub>: R=CH<sub>3</sub>, CH<sub>2</sub>CH<sub>3</sub>, CH<sub>2</sub>CH<sub>2</sub>CH<sub>2</sub>CH<sub>3</sub>.

Scheme 1.



Scheme 2.

Table 1.  $^1\text{H-NMR}$  spectral data for compound (2) and (13) in  $\text{DMSO-}d_6$  (323 K)

H	2	2+TFA	13	13+TFA
1	7.43 (d, 7.0)	8.60 (d, 6.6)	7.98 (d, 5.0)	8.31 (d, 5.5)
2	7.91 (d, 7.0)	8.92 (d, 6.6)	8.71 (d, 5.0)	8.80 (d, 5.5)
8	8.44 (d, 8.2)	8.55 (d, 8.2)	8.38 (d, 7.7)	8.51 (d, 8.2)
9	7.66 (t, 8.2)	7.93 (t, 8.2)	7.67 (t, 7.7)	7.81 (t, 8.2)
10	7.50 (t, 8.2)	7.70 (t, 8.2)	7.49 (t, 7.7)	7.61 (t, 8.2)
11	8.15 (d, 8.2)	8.50 (d, 8.2)	8.14 (d, 7.7)	8.40 (d, 8.2)
3- $\text{CH}_3$	3.82 (s)	4.68 (s)		
4- $\text{OCH}_3$	4.26 (s)	4.11 (s)	4.28 (s)	4.22 (s)
5-OH			5.77 (s)*	

\*Disappeared with the addition of  $\text{D}_2\text{O}$  or TFA. Coupling constant in Hz.

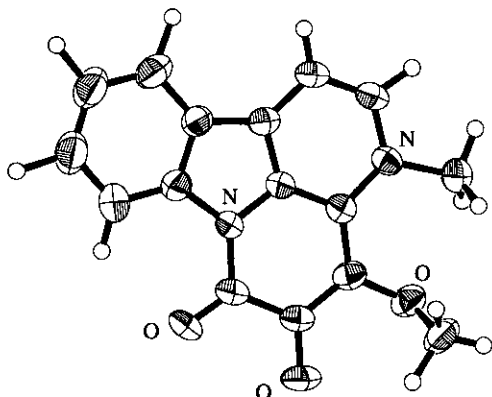


Figure 1. An ORTEP drawing of 2.

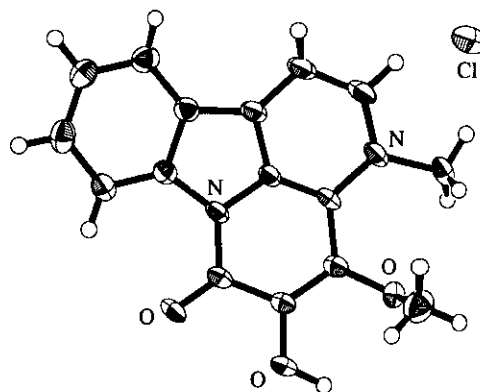


Figure 2. An ORTEP drawing of 14.

### Biological Results

A study of the biological properties of nine canthin-5,6-dione derivatives (1-9) was carried out in P-388 murine leukemia cells and PC-6 human lung carcinoma cells *in vitro*. All compounds tested exhibited weaker antitumor activity than cisplatin. However, the antitumor activity increased as the carbon number of the alkyl chain was increased from the activity-structure relationship, indicating that the alkyl chain was a requisite functional group for the antitumor activities against P-388 and PC-6 cell lines *in vitro*. Further synthetic work aimed at structural modification is in progress to find compounds which show more potent antitumor activity than cisplatin.

Table 2. Antitumor activities against P-388 and PC-6 cell lines (GI<sub>50</sub> μg/mL)

compound	P-388	PC-6	compound	P-388	PC-6
<b>1</b>	27.9	27.2	<b>3</b>	>50	>50
<b>4</b>	10.6	16.2	<b>8</b>	10.3	30.7
<b>5</b>	5.9	8.5	<b>9</b>	4.8	13.8
<b>2</b>	42.6	38.5	cisplatin	0.02	0.28
<b>6</b>	33.8	28.9			
<b>7</b>	9.5	16.0			

## EXPERIMENTAL

**General Procedures.** All melting points were measured on a Yanagimoto micromelting point hot-stage type apparatus without being uncorrected. UV-VIS spectra were recorded with a Hitachi 340 spectrophotometer. IR spectra as KBr pellets were recorded with a JASCO D300 FT-IR spectrophotometer. The <sup>1</sup>H-NMR spectra were recorded with a JEOL EX-400 (<sup>1</sup>H: 400 MHz) spectrometer. Chemical shifts are given on the δ scale (ppm) with tetramethylsilane as an internal standard, and coupling constants are given in Hz. EIMS and HRMS spectra were conducted using JEOL D-300 and DX-303 mass spectrometers, respectively. Column chromatography was performed on silica gel (Merck).

**General Procedure of Preparation for β-Carbolines.** These compounds were prepared from tryptamine or *dl*-tryptophan by the Pictet-Spengler reaction with the appropriate aldehyde in an acidic medium, according to the procedure in the literature.<sup>6</sup> 7-Methoxy-1-methyl-β-carboline (**12**) was purchased from Sigma.

**1-Methyl-β-carboline (10)**<sup>7</sup>: Yield 88.7%, colorless needles, mp 232°C (acetone). IR (KBr) cm<sup>-1</sup>: 3300, 1530, 1500, 1450, 1380, 1330, 1310, 1240. MS *m/z*: 182 (M<sup>+</sup>). <sup>1</sup>H-NMR (DMSO-*d*<sub>6</sub>, 323K) δ: 2.76 (3H, s, CH<sub>3</sub>), 7.23 (1H, t, *J*=8.2 Hz, H-6), 7.53 (1H, t, *J*=8.2 Hz, H-7), 7.60 (1H, d, *J*=8.2 Hz, H-8), 7.92 (1H, d, *J*=5.3 Hz, H-4), 8.19 (1H, d, *J*=8.2 Hz, H-5), 8.20 (1H, d, *J*=5.3 Hz, H-3), 11.53 (1H, brs, NH). MS *m/z*: 182 (M<sup>+</sup>). *Anal.* Calcd for C<sub>12</sub>H<sub>10</sub>N<sub>2</sub>: C 79.10, H 5.53, N 15.37, Found: C 79.07, H 5.48, N 15.18.

**1-Methoxymethyl-β-carboline (11)**<sup>8</sup>: Yield 65.3%, colorless needles, mp 123-125°C (acetone). IR (KBr) cm<sup>-1</sup>: 3310, 1610, 1550, 1225, 1150, 1070. <sup>1</sup>H-NMR (CDCl<sub>3</sub>) δ: 3.49 (3H, s, CH<sub>2</sub>OCH<sub>3</sub>), 5.04 (2H, s, CH<sub>2</sub>OCH<sub>3</sub>), 7.23 (1H, t, *J*=7.2 Hz, H-6), 7.45 (1H, t, *J*=7.2 Hz, H-7), 7.64 (1H, d, *J*=7.2 Hz, H-8), 8.05 (1H, d, *J*=6.6 Hz, H-4), 8.08 (1H, d, *J*=6.6 Hz, H-3), 8.12 (1H, d, *J*=7.2 Hz, H-5), 11.33 (1H, brs, NH). MS *m/z*: 212 (M<sup>+</sup>). *Anal.* Calcd for C<sub>13</sub>H<sub>12</sub>N<sub>2</sub>O: C 73.56, H 5.70, N 13.20, Found: C 73.79, H 5.56, N 13.19.

**3-Methylcanthin-5,6-dione (picrasidine L, 1):** A mixture of 1-methyl- $\beta$ -carboline (**10**, 0.8 g, 4.4 mmol) and dimethyl oxalate (1.0 g, 8.5 mmol) was heated for 20 min at 180°C. The red reactant was recrystallized from MeOH to give red needles. Yield 0.82 g (75.0%), mp >300°C. IR (KBr)  $\text{cm}^{-1}$ : 1688, 1651, 1541, 1508, 1486, 1447, 1409, 1337, 1310, 1218. UV  $\lambda$  max (MeOH) nm (log  $\epsilon$ ): 225 (4.24), 245 (4.33), 250 (4.31), 273 (3.98), 288 (4.07), 353 (sh, 3.63), 363 (sh, 3.60), 450 (4.08). UV  $\lambda$  max (MeOH+NaOH) nm (log  $\epsilon$ ): 225 (4.25), 245 (4.34), 250 (4.33), 273 (3.98), 288 (4.01), 353 (3.67), 450 (4.10). UV  $\lambda$  max (MeOH+HCl) nm (log  $\epsilon$ ): 254 (4.30), 324 (3.90), 332 (sh, 3.92), 380 (4.04), 397 (4.04), 450 (3.40).  $^1\text{H-NMR}$  (DMSO- $d_6$ , 323K)  $\delta$ : 3.91 (3H, s,  $\text{CH}_3$ -3), 6.00 (1H, s, H-4), 7.47 (1H, d,  $J=7.0$  Hz, H-1), 7.54 (1H, t,  $J=8.1$  Hz, H-10), 7.7 (1H, t,  $J=8.1$  Hz, H-8), 8.03 (1H, d,  $J=7.0$  Hz, H-2), 8.22 (1H, d,  $J=8.1$  Hz, H-8), 8.47 (1H, d,  $J=8.1$  Hz, H-11). EIMS  $m/z$  (%): 250 ( $\text{M}^+$ , 48), 236 (5), 222 (69), 193 (100). HRMS  $m/z$ : Calcd for  $\text{C}_{15}\text{H}_{10}\text{N}_2\text{O}_2$ , 250.0740. Found: 250.0733 [ $\text{M}^+$ ]. *Anal.* Calcd for  $\text{C}_{15}\text{H}_{10}\text{N}_2\text{O}_2 \cdot \text{H}_2\text{O}$ : C, 67.16; H, 4.51; N, 10.44. Found: C, 67.62; H, 4.25; N, 10.10.

**3-Ethylcanthin-5,6-dione (4):** A mixture of 1-methyl- $\beta$ -carboline (**10**, 0.5 g, 2.7 mmol) and diethyl oxalate (2.0 g, 13.7 mmol) was heated for 30 min at 180°C. The red reactant was chromatographed on silica gel eluting with  $\text{CHCl}_3$ -MeOH (9:1). The product was recrystallized from  $\text{CHCl}_3$ -MeOH (7:3) to give red needles. Yield 0.20 g (28.2%), mp >300°C. IR (KBr)  $\text{cm}^{-1}$ : 1697, 1648, 1614, 1591, 1548, 1520, 1455, 1368, 1338, 1325, 1311, 1208, 1183. UV  $\lambda$  max (MeOH) nm (log  $\epsilon$ ): 216 (3.98), 224 (4.06), 227 (4.05), 268 (3.69), 290 (3.82), 352 (sh, 3.36), 364 (sh, 3.34), 451 (3.82). UV  $\lambda$  max (MeOH+NaOH) nm (log  $\epsilon$ ): 216 (3.98), 224 (4.06), 227 (4.05), 268 (3.69), 290 (3.82), 352 (sh, 3.36), 364 (sh, 3.34), 451 (3.82). UV  $\lambda$  max (MeOH+HCl) nm (log  $\epsilon$ ): 249 (4.03), 292 (3.37), 323 (3.70), 335 (sh, 3.67), 381 (3.78), 400 (3.78), 450 (3.00).  $^1\text{H-NMR}$  (DMSO- $d_6$ , 323K)  $\delta$ : 1.44 (3H, t,  $J=7.2$  Hz,  $\text{CH}_2\text{CH}_3$ -3), 4.34 (2H, q,  $J=7.2$  Hz,  $\text{CH}_2\text{CH}_3$ -3), 6.11 (1H, s, H-4), 7.53 (1H, d,  $J=7.0$  Hz, H-1), 7.55 (1H, t,  $J=7.3$  Hz, H-10), 7.72 (1H, t,  $J=7.3$  Hz, H-9), 8.06 (1H, d,  $J=7.0$  Hz, H-2), 8.23 (1H, d,  $J=7.3$  Hz, H-8), 8.47 (1H, d,  $J=7.3$  Hz, H-11). EIMS  $m/z$  (%): 264 ( $\text{M}^+$ , 64), 250 (3), 236 (57), 207 (100). HRMS  $m/z$ : Calcd for  $\text{C}_{16}\text{H}_{12}\text{N}_2\text{O}_2$ , 264.0896. Found: 264.0845 [ $\text{M}^+$ ]. *Anal.* Calcd for  $\text{C}_{16}\text{H}_{12}\text{N}_2\text{O}_2$ : C, 72.72; H, 4.58; N, 10.60. Found: C, 72.58; H, 4.57; N, 10.52.

**3-Butylcanthin-5,6-dione (5):** A mixture of 1-methyl- $\beta$ -carboline (**10**, 0.5 g, 2.7 mmol) and dibutyl oxalate (1.0 g, 5 mmol) was heated for 30 min at 180°C. The red reactant was chromatographed on silica gel eluting with  $\text{CHCl}_3$ -MeOH (9:1). The product was recrystallized from MeOH to give yellow needles. Yield 0.05 g (6.3%), mp 278-280°C. IR (KBr)  $\text{cm}^{-1}$ : 1695, 1646, 1614, 1595, 1551, 1364, 1272. UV  $\lambda$  max (MeOH) nm (log  $\epsilon$ ): 228 (4.29), 245 (4.37), 248 (4.36), 273 (4.01), 287 (4.13), 348 (sh, 3.70), 360 (sh, 3.65), 450 (4.12). UV  $\lambda$  max (MeOH+NaOH) nm (log  $\epsilon$ ): 228 (4.26), 244 (4.33), 252 (4.32), 270 (3.99), 285 (4.09), 348 (sh, 3.68), 360 (sh, 3.63), 450 (4.09). UV  $\lambda$  max (MeOH+HCl) nm (log  $\epsilon$ ): 250 (4.35), 320 (4.04), 225 (sh, 4.00), 275 (4.07), 294 (4.07), 450 (3.42).  $^1\text{H-NMR}$  (DMSO- $d_6$ , 323K)  $\delta$ : 0.95 (3H, t,  $J=7.5$  Hz,  $\text{CH}_2\text{CH}_2\text{CH}_2\text{CH}_3$ -3), 1.40 (2H, sextet,  $J=7.5$  Hz,  $\text{CH}_2\text{CH}_2\text{CH}_2\text{CH}_3$ -3), 1.79 (2H, quintet,  $J=7.5$  Hz,  $\text{CH}_2\text{CH}_2\text{CH}_2\text{CH}_3$ -3), 4.28 (2H, t,  $J=7.5$  Hz,  $\text{H-CH}_2\text{CH}_2\text{CH}_2\text{CH}_3$ -3), 7.49 (1H, d,  $J=7.0$  Hz, H-1), 7.53 (1H, t,  $J=8.4$  Hz, H-10), 7.69 (1H, t,  $J=8.4$  Hz, H-9), 8.04 (1H, d,  $J=7.0$  Hz, H-2), 8.21 (1H, d,  $J=8.4$  Hz, H-11), 8.27 (1H, s, H-4), 8.46 (1H, d,  $J=8.4$  Hz, H-8). EIMS  $m/z$  (%): 292

( $M^+$ , 75), 264 (13), 250 (8), 235 (13), 222 (100). HRMS  $m/z$ : Calcd for  $C_{18}H_{16}N_2O_2$ , 292.1208. Found: 292.1205 [ $M^+$ ]. *Anal.* Calcd for  $C_{18}H_{16}N_2O_2 \cdot 1.5H_2O$ : C, 73.95; H, 5.52; N, 9.58. Found: C, 73.48; H, 5.43; N, 9.24.

**3-Methyl-4-methoxycantin-5,6-dione (picrasidine O, 2):** A mixture of 1-methoxymethyl- $\beta$ -carboline (**11**, 1.5 g, 7.1 mmol) and dimethyl oxalate (1.4 g, 12 mmol) was heated for 20 min at 180°C. The red reactant was recrystallized from MeOH to give red needles. Yield 1.51 g (75%). mp 274°C (decomp). IR (KBr)  $cm^{-1}$ : 1682, 1642, 1549, 1500, 1450, 1282, 1108. UV  $\lambda$  max (MeOH) nm (log  $\epsilon$ ): 242 (4.28), 250 (sh, 4.27), 290 (4.20), 360 (3.54), 480 (4.04), 500 (4.00). UV  $\lambda$  max (MeOH+NaOH) nm (log  $\epsilon$ ): 242 (4.28), 250 (sh, 4.27), 290 (4.20), 360 (3.54), 480 (4.04), 500 (4.00). UV  $\lambda$  max (MeOH+HCl) nm (log  $\epsilon$ ): 242 (4.14), 258 (sh, 4.11), 320 (3.84), 328 (sh, 3.81), 378 (3.76), 416 (sh, 3.52).  $^1H$ -NMR: Table I. EIMS  $m/z$  (%): 280 ( $M^+$ , 54), 273 (3), 265 (70), 250 (6), 235 (4), 209 (39), 193 (4), 181 (100). HRMS  $m/z$ : Calcd for  $C_{16}H_{12}N_2O_3$ , 280.0845. Found: 280.0838 [ $M^+$ ]. *Anal.* Calcd for  $C_{16}H_{12}N_2O_3$ : C, 68.59; H, 4.22; N, 9.99. Found: C, 68.46; H, 4.21; N, 9.93.

**3-Ethyl-4-methoxycanthin-5,6-dione (6):** A mixture of 1-methoxymethyl- $\beta$ -carboline (**11**, 0.8 g, 3.8 mmol) and diethyl oxalate (1.0 g, 7 mmol) was heated for 1 h at 180°C. The red reactant was chromatographed on silica gel eluting with  $CHCl_3$ -MeOH (9:1). The product was recrystallized from  $CHCl_3$ -MeOH (7:3) to give red powder. Yield 0.32 g (28.6%), mp 248-250°C. IR (KBr)  $cm^{-1}$ : 1694, 1636, 1541, 1505, 1468, 1455, 1357, 1116. UV  $\lambda$  max (MeOH) nm (log  $\epsilon$ ): 247 (4.46), 294 (4.34), 350 (sh, 3.86), 361 (sh, 3.80), 481 (4.14). UV  $\lambda$  max (MeOH+NaOH) nm (log  $\epsilon$ ): 247 (4.43), 294 (4.30), 350 (sh, 3.85), 361 (sh, 3.79), 481 (4.11). UV  $\lambda$  max (MeOH+HCl) nm (log  $\epsilon$ ): 246 (4.41), 257 (4.38), 325 (4.18), 335 (sh, 4.16), 380 (4.11).  $^1H$ -NMR (DMSO- $d_6$ , 323K)  $\delta$ : 1.47 (3H, t,  $J=7.2$  Hz,  $CH_2CH_3$ -3), 3.87 (3H, s,  $OCH_3$ -4), 4.67 (2H, q,  $J=7.2$  Hz,  $CH_2CH_3$ -3), 7.51 (1H,  $J=7.0$  Hz, H-1), 7.51 (1H, t,  $J=8.3$  Hz, H-10), 7.67 (1H, t,  $J=8.3$  Hz, H-9), 8.02 (1H, d,  $J=7.0$  Hz, H-2), 8.18 (1H, d,  $J=8.3$  Hz, H-11), 8.44 (1H, d,  $J=8.3$  Hz, H-8). EIMS  $m/z$  (%): 294 ( $M^+$ , 53), 279 (85), 265 (67), 251 (7), 236 (13), 223 (47), 207 (3), 195 (63), 179 (6), 167 (100). HRMS  $m/z$ : Calcd for  $C_{17}H_{14}N_2O_3$ , 294.1001. Found: 294.0983 [ $M^+$ ]. *Anal.* Calcd for  $C_{17}H_{14}N_2O_3$ : C, 69.38; H, 4.79; N, 9.52. Found: C, 69.68; H, 4.91; N, 9.22.

**3-Butyl-4-methoxycanthin-5,6-dione (7):** A mixture of 1-methoxymethyl- $\beta$ -carboline (**11**, 1.0 g, 4.7 mmol) and dibutyl oxalate (2.0 g, 9.9 mmol) was heated for 1 h at 180°C. The red reactant was chromatographed on silica gel eluting with  $CHCl_3$ -MeOH (9:1). The product was recrystallized from  $CHCl_3$ -MeOH (7:3) to give red powder. Yield 0.051 g (3.4%), mp 248-249°C. IR (KBr)  $cm^{-1}$ : 1685, 1635, 1609, 1585, 1352, 1275, 1149, 1123. UV  $\lambda$  max (MeOH) nm (log  $\epsilon$ ): 246 (4.56), 296 (4.45), 350 (sh, 3.98), 364 (sh, 3.93), 488 (4.24). UV  $\lambda$  max (MeOH+NaOH) nm (log  $\epsilon$ ): 245 (4.52), 294 (4.40), 350 (sh, 3.95), 364 (sh, 3.90), 481 (4.19). UV  $\lambda$  max (MeOH+HCl) nm (log  $\epsilon$ ): 244 (4.53), 255 (4.50), 321 (4.29), 379 (4.21).  $^1H$ -NMR (DMSO- $d_6$ , 323K)  $\delta$ : 0.95 (3H, t,  $J=7.3$  Hz,  $CH_2CH_2CH_2CH_3$ -3), 1.40 (2H, sextet,  $J=7.3$  Hz,  $CH_2CH_2CH_2CH_3$ -3), 1.84 (2H, quintet,  $J=7.3$  Hz,  $CH_2CH_2CH_2CH_3$ -3), 3.86 (3H, s,  $OCH_3$ -4), 4.61 (2H, t,  $J=7.3$  Hz,  $CH_2CH_2CH_2CH_3$ -3), 7.50 (1H, d,  $J=7.0$  Hz, H-1), 7.51 (1H, t,

$J=8.3$  Hz, H-10), 7.67 (1H, t,  $J=8.3$  Hz, H-9), 8.01 (1H, d,  $J=7.0$  Hz, H-2), 8.18 (1H, d,  $J=8.3$  Hz, H-11), 8.44 (1H, d,  $J=8.3$  Hz, H-8). EIMS  $m/z$  (%): 322 ( $M^+$ , 88), 307 (100), 293 (17), 275 (5), 265 (94), 251 (96), 237 (16), 220 (11). HRMS  $m/z$ : Calcd for  $C_{19}H_{18}N_2O_3$ , 322.1313. Found: 322.1347 [ $M^+$ ]. Anal. Calcd for  $C_{19}H_{18}N_2O_3$ : C, 70.79; H, 5.63; N, 8.69. Found: C, 70.97; H, 5.64; N, 8.64.

**3-Methyl-9-methoxycanthin-5,6-dione (eurycomine E, 3):** A mixture of 1-methyl-7-methoxy- $\beta$ -carboline (**12**, 0.8 g, 3.7 mmol) and dimethyl oxalate (1.0 g, 8.5 mmol) was heated for 30 min at 180°C. The red reactant was recrystallized from MeOH to give red needles. Yield 0.83 g (79.8%), mp >300°C. IR (KBr)  $cm^{-1}$ : 1692, 1649, 1616, 1591, 1553, 1272, 1232, 1217, 1175, 1153, 1099, 1073. UV  $\lambda$  max (MeOH) nm (log  $\epsilon$ ): 450 (4.19), 347 (4.26), 279 (4.39), 255 (4.24), 230 (4.45). UV  $\lambda$  max (MeOH+NaOH) nm (log  $\epsilon$ ): 450 (4.17), 347 (4.22), 279 (4.35), 255 (4.20), 230 (4.40). UV  $\lambda$  max (MeOH+HCl) nm (log  $\epsilon$ ): 400 (4.05), 358 (4.17), 303 (4.35), 260 (4.24), 223 (4.46).  $^1H$ -NMR (DMSO- $d_6$ , 323K)  $\delta$ : 3.89 (3H, s,  $CH_3$ -3), 3.93 (3H, s,  $OCH_3$ -9), 7.14 (1H, dd,  $J=2.6$  and 8.4 Hz, H-10), 7.40 (1H, d,  $J=7.0$  Hz, H-1), 8.00 (1H, d,  $J=7.0$  Hz, H-2), 8.01 (1H, d,  $J=2.6$  Hz, H-8), 8.10 (1H, d,  $J=8.4$  Hz, H-11). EIMS  $m/z$  (%): 280 ( $M^+$ , 60), 266 (4), 252 (100), 237 (11), 223 (74), 209 (29). HRMS  $m/z$ : Calcd for  $C_{16}H_{12}N_2O_3$ , 280.0845. Found: 280.0837 [ $M^+$ ]. Anal. Calcd for  $C_{16}H_{12}N_2O_3$ : C, 68.56; H, 4.32; N, 9.99. Found: C, 68.38; H, 4.40; N, 9.84.

**3-Ethyl-9-methoxycanthin-5,6-dione (8):** A mixture of 1-methyl-7-methoxy- $\beta$ -carboline (**12**, 0.8 g, 3.7 mmol) and diethyl oxalate (1.25 g, 8.5 mmol) was heated for 1 h at 180°C. The red reactant was recrystallized from MeOH to give red needles. Yield 0.30 g (52.6%), mp >300°C. IR (KBr)  $cm^{-1}$ : 1696, 1646, 1544, 1517, 1464, 1439, 1277. UV  $\lambda$  max (MeOH) nm (log  $\epsilon$ ): 235 (4.18), 264 (3.95), 284 (4.11), 345 (3.99), 425 (3.91). UV  $\lambda$  max (MeOH+NaOH) nm (log  $\epsilon$ ): 235 (4.26), 263 (4.04), 284 (4.20), 350 (4.08), 450 (4.01). UV (MeOH+HCl) nm (log  $\epsilon$ ): 227 (4.21), 265 (3.92), 285 (4.04), 362 (4.05), 402 (3.73).  $^1H$ -NMR (DMSO- $d_6$ , 323K)  $\delta$ : 1.42 (3H, t,  $J=7.2$  Hz,  $CH_2CH_3$ -3), 3.93 (3H, s,  $OCH_3$ -9), 4.30 (2H, q,  $J=7.2$  Hz,  $CH_2CH_3$ -3), 6.02 (1H, s, H-4), 7.13 (1H, dd,  $J=2.6$  and 8.8 Hz, H-10), 7.39 (1H, d,  $J=7.0$  Hz, H-1), 7.98 (1H, d,  $J=7.0$  Hz, H-2), 8.02 (1H, d,  $J=2.6$  Hz, H-8), 8.08 (1H, d,  $J=8.8$  Hz, H-11). EIMS  $m/z$  (%): 294 ( $M^+$ , 100), 280 (11), 266 (78), 252 (14), 237 (97), 223 (37), 210 (24). HRMS  $m/z$ : Calcd for  $C_{17}H_{14}N_2O_3$ , 294.1001. Found: 294.0978 [ $M^+$ ]. Anal. Calcd for  $C_{17}H_{14}N_2O_3$ : C, 69.38; H, 4.79; N, 9.52. Found: C, 68.85; H, 4.81; N, 9.46.

**3-Butyl-9-methoxycanthin-5,6-dione (9):** A mixture of 1-methyl-7-methoxy- $\beta$ -carboline (**12**, 0.5 g, 2.3 mmol) and dibutyl oxalate (1.0 g, 4.9 mmol) was heated for 1 h at 180°C. The red reactant was chromatographed on silica gel eluting with  $CHCl_3$ -MeOH (9:1). The product was recrystallized from MeOH to give orange needles. Yield 0.031 g (4.2%), mp 270-272°C. IR (KBr)  $cm^{-1}$ : 1695, 1647, 1582, 1543, 1509, 1490, 1453, 1359, 1340, 1206. UV  $\lambda$  max (MeOH) nm (log  $\epsilon$ ): 235 (3.88), 260 (3.65), 277 (3.80), 350 (3.69), 452 (3.60). UV  $\lambda$  max (MeOH+NaOH) nm (log  $\epsilon$ ): 234 (3.94), 260 (3.74), 282 (3.87), 348 (3.74), 450 (3.65). UV  $\lambda$  max (MeOH+HCl) nm (log  $\epsilon$ ): 227 (3.92), 263 (3.65), 287 (3.75), 360 (3.77), 400 (3.45).  $^1H$ -NMR (DMSO- $d_6$ , 323K)  $\delta$ : 0.94 (3H, t,  $J=7.3$  Hz,  $CH_2CH_2CH_2CH_3$ -3), 1.39 (2H, sextet,  $J=7.3$  Hz,  $CH_2CH_2CH_2CH_3$ -3), 1.79 (2H, quintet,  $J=7.3$  Hz,  $CH_2CH_2CH_2CH_3$ -3), 3.93

(3H, s, OCH<sub>3</sub>-9), 4.28 (2H, t,  $J=7.3$  Hz, CH<sub>2</sub>CH<sub>2</sub>CH<sub>2</sub>CH<sub>3</sub>-3), 6.04 (1H, s, H-4), 7.14 (1H, dd,  $J=2.6$  and 8.43 Hz, H-10), 7.43 (1H, d,  $J=7.0$  Hz, H-1), 8.01 (1H, d,  $J=2.6$  Hz, H-8), 8.02 (1H, d,  $J=7.0$  Hz, H-2), 8.11 (1H, d,  $J=8.4$  Hz, H-11). EIMS  $m/z$  (%): 322 (M<sup>+</sup>, 86), 294 (14), 280 (11), 266 (16), 252 (100), 237 (32), 224 (14), 210 (41). HRMS  $m/z$ : Calcd for C<sub>19</sub>H<sub>18</sub>N<sub>2</sub>O<sub>3</sub>, 322.1313. Found: 322.1359 [M<sup>+</sup>]. Anal. Calcd for C<sub>19</sub>H<sub>18</sub>N<sub>2</sub>O<sub>3</sub>: C, 70.79; H, 5.63; N, 8.69. Found: C, 70.84; H, 5.64; N, 8.66.

**Crystal Data of 3:** Crystallized from methanol and belonging to monoclinic space group *P*1 (#1). Lattice constants and intensity data were measured on a Rigaku AFC-5R diffractometer with a device for graphite-monochromated CuK $\alpha$  radiation. Crystal data: C<sub>16</sub>H<sub>12</sub>N<sub>2</sub>O<sub>3</sub>,  $a=6.1954(28)$ ,  $b=11.113(1)$ ,  $c=4.9527(5)$  Å,  $Z=1$ ,  $\alpha=97.600(7)$ ,  $\beta=105.674(8)$ ,  $\gamma=101.419(9)^\circ$ ,  $D_{\text{calc}}=1.475$  g/cm<sup>3</sup>, CuK $\alpha$   $\lambda=1.54178$  Å. A total of 925 independent reflections with  $I>3\sigma(I)$  was used for structure analysis. The structure was solved by the direct method (MULTAN88)<sup>9</sup> and expanded using Fourier techniques.<sup>10</sup> The structure was then refined by full-matrix least squares with anisotropic temperature factors for non-hydrogen and isotropic atoms for hydrogen atoms to an *R* factor of 0.028 ( $R_w=0.038$ ).

**Crystal Data of 14:** Crystallized from methanol and belonging to monoclinic space group *P* $\bar{1}$  (#2). Lattice constants and intensity data were measured on a Rigaku AFC-7R diffractometer with a device for graphite-monochromated CuK $\alpha$  radiation. Crystal data: C<sub>18</sub>H<sub>13</sub>N<sub>2</sub>O<sub>3</sub>Cl,  $a=8.807(2)$ ,  $b=12.446(3)$ ,  $c=7.437(4)$  Å,  $\alpha=100.77(3)$ ,  $\beta=95.65(3)$ ,  $\gamma=99.38(2)^\circ$ ,  $Z=2$ ,  $D_{\text{calc}}=1.445$  g/cm<sup>3</sup>, CuK $\alpha$   $\lambda=1.54178$  Å. A total of 2215 independent reflections with  $I>3\sigma(I)$  was used for structure analysis. The structure was solved by the direct method (SIR88)<sup>11</sup> and expanded using Fourier techniques.<sup>10</sup> The structure was then refined by full-matrix least squares with anisotropic temperature factors for non-hydrogen and isotropic atoms for hydrogen atoms to an *R* factor of 0.062 ( $R_w=0.103$ ).

**In Vitro Cytotoxicity:** To examine the direct growth-inhibitory effects of test compounds against P-388 against murine leukemia cells and PC-6 human lung carcinoma cells, the 3-(4,5-dimethylthiazol-2-yl)-2,5-diphenyltetrazolium bromide (MTT) assay was performed and the concentration giving a growth inhibition of 50% (GI<sub>50</sub>) was calculated according to a published procedure.<sup>12</sup>

## ACKNOWLEDGMENT

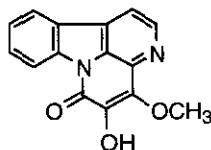
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 4-Methoxy-5-hydroxycanthin-6-one (nigakinone, **13**): UV  $\lambda$  max (MeOH) nm (log  $\epsilon$ ): 246 (4.60), 262 (sh, 4.50), 286 (4.25), 340 (4.01), 350 (4.00). UV  $\lambda$  max (MeOH+NaOH) nm (log  $\epsilon$ ): 254 (4.53), 286 (4.44), 318 (3.95), 4.20 (4.19). UV  $\lambda$  max (MeOH+HCl) nm (log  $\epsilon$ ): 248 (3.65), 322 (4.00), 360 (4.25), 274 (4.25), 234 (3.94).

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