## **A VERSATILE APPROACH TO TRANS-1,3-DISUBSTITUTED TETRAHYDRO-B-CARBOLINES USING OXAZINANES**

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Abstract - Under the conditions of thermodynamic control, the diastereoselectivity of the reaction between tryptophan esters and oxazinanes (carbonyl equivalents) can be controlled by incorporating  $N<sub>b</sub>$ -benzyl substituents; the reaction proceeds in essentially quantitative yield and with **rrons-diastereoselectivity.** 

The Pictet-Spengler reaction<sup>1</sup> has long been an important reaction for the synthesis of numerous naturally occurring alkaloids, embodying tetrahydro-ß-carbolines or tetrahydroisoquinoline framework, mediating pharmacologically useful effects.' Therefore, the synthesis of these systems in a synthetically useful manner is of widespread interest to both organic synthesis and medicinal chemistry. Conventional Pictet-Spengler reaction quite often lacks practicability owing to handling and non-availability of desired functionalized aldehydes. Perhydrooxazines (oxazinanes), which have been synthesized from a variety of reagents other than aldehydes, $3$  have amply demonstrated synthetic superiority<sup>4</sup> over conventional carbonyl compounds owing to carbonyl character<sup>3</sup> of the C-2. However, a generally applicable methodology for the execution of title protocol is currently not available. We have seen, in accordance with the carbonyl compounds, by using conditions of thermodynamic control, the Pictet-Spengler reaction between (L)-tryptophan ester and appropriately substituted oxazinanes can be controlled to give the title compounds selectively.

The oxazinanes which were used as reagents in the diastereoselective synthesis are readily available from a variety of cheaply available combinations. If the secondary amines  $(1)$  are treated with oxazinanes  $(2)$  existing mainly in tautomeric iminium form,<sup>5</sup> at 80 $^{\circ}$ C in anhydrous acetonitrile as solvent in the presence of 2 to 3 equivalents of trifluoroacetic acid, the iminium intermediate (Scheme) formed in situ cyclises spontaneously by intramolecular attack of C-2 of indole nucleus on iminium functionality to deliver 1,3-disubstituted tetrahydro-β-carbolines (3) and (4) in excellent yields and with very high diastereomeric ratios (Table). The *trans-N*<sub>b</sub>- substituted diastereomers are thermodynamically more stable than their cis-congeners especially where the reactions are catalysed by TFA and conversion of cisdiastereomer into the more stable trans-diastereomers is believed to occur under acidic conditions by cleavage of the carbon (C-I)-nitrogen (N-2) bond with complete retention of configuration at C-3



Table : Diastereoselectivity in the Pictet-Spengler reaction between oxazinanes **(2)** and **(L)-TrpOR**<sup>3</sup>.



stereocentre.<sup>2</sup> However, like the conventional Pictet-Spengler reaction<sup>7</sup> the diastereoselectivity is not controlled by varying the size of the ester group where  $N_b$ -benzyl substituents<sup>8</sup> are present. The

predominantly formed stereoisomer can be isolated in a straightfoward way using simple chromatography and recrystallisation. The diastereomers were identified by analysing 200 MHz- 'H NMR and 50.3 MHz-  $<sup>13</sup>C$  NMR and the stereochemistry was unambiguously assigned by comparison with the literature NMR</sup> data.9

This method of preparation of the title compounds may be considered a variant of the classical Pictet-Spengler reaction in that the same iminium intermediate is deemed to be involved in the reaction. This approach will be preferred in those cases where the aldehydes required for a Pictet-Spengler reaction arc unstable or difficult to access as the oxazinanes can be readily functionalized at  $C-2<sup>3</sup>$ . The utility of this approach using oxazinanes bearing aliphatic substituents (equivalents of aliphatic aldehydes) at C-2 to effect synthesis of some target alkaloids is in progress.

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## **REFERENCES AND NOTES**

- 1. E. D. Cox and J. M. Cook, *Chem. Rev.,* 1995,95, 1797 and references cited therein
- 2. E. D. Cox, L. K. Hamaker, J. Li, P. Yu, K. M. Czenvinski, L. Deng, D. W. Bennett, J. M. Cook, W. H. Watson, and M. Krawiec, *J Org Chem.,* 1997, 62,44 and references cited therein.
- 3. A. I. Meyers, A. Nabeya, H. W. Adickes, I. R. Politzer, G. R. Malone, A. C. Kovelesky, R. L. Nolen, and R. C. Portnoy, J. *Org Chem.,* 1973,38,36.
- 4. K. Singh, J. Singh, and H. Singh, *Tetrahedron,* 1996,52, 14273 and references cited therein.
- 5. L. Lazar, A. G. Lakatos, F. Fulop, G, Bernath, and F. G. Riddell, *Tetrahedron,* 1997,53, 1081 and references cited therein.
- *6.* Using our standard conditions of reflux at 80°C, the yields refer to that of isolated compounds.
- 7. P. D. Bailey. M. H. Moore, K. M. Morgan, D. I. Smith, and J. M. Vernon, *Tetrahedron Letts.,*  1994, *35,* 3587.
- 8. The (L)-tryptophan methyl or iso-propyl esters lacking  $N_b$ -benzyl substituents yielded equilibrated mixtures of corresponding **3** and 4 .
- 9. F. Ungemach, D. Soerens, R. Weher, M. DiPierro, 0. Campos, P. Mokry, J. M. Cook, and J. V. Silverton, **L** *Am. Chem.* Soc., 1980, 102, 6976. The compounds reported herein exhibit satisfactory spectroscopic and microanalytical data.

Spectroscopic data of selected **3** 

Entry 1: IR (KBr) 3337 (s, NH), 1722 (s, ester) cm<sup>-1</sup>; <sup>1</sup>H NMR (CDCl<sub>3</sub>)  $\delta$  3.22 (d, J=4.5 Hz, 2H, CH<sub>2</sub>), 3.62 (s, 3H, OCH<sub>3</sub>), 3.86 (d, J=2.3 Hz, 2H, CH<sub>2</sub>), 3.92-3.96 (m, 1H, CH), 5.46 (s, 1H, CH), 7.07-7.52 (m, 15H, Ar-H & NH); <sup>13</sup>C NMR (CDCl<sub>3</sub>)  $\delta$  24.36 (CH<sub>2</sub>), 51.31(OCH<sub>3</sub>)\*, 54.32  $(CH_2C_6H_5)$ , 56.06  $(C-3)*$ , 60.84  $(C-1)*$ , 106.31, 110.83, 118.16, 119.24, 121.54, 127.05, 127.53, 128.00, 128.31, 128.55, 128.67, 128.89, 134.88, 136.50, 139.41, 142.17, 173.56 (CO). Entry 4: IR (CHCl<sub>3</sub>) 3310 (s, NH), 1714 (s, ester) cm<sup>-1</sup>; <sup>1</sup>H NMR (CDCl<sub>3</sub>)  $\delta$  1.11 (d, J=6.3 Hz, 3H, CH<sub>3</sub>), 1.21 (d, J=6.2 Hz, 3H, CH<sub>3</sub>), 3.17 (d, J=4.2 Hz, 2H, CH<sub>2</sub>), 3.79 (m, 3H, OCH<sub>3</sub>, merged with multiplet of CH<sub>2</sub>Ph ), 3.78-3.82 (m, 2H, CH<sub>2</sub>Ph), 3.86-3.91 (m, 1H, CH), 4.91-5.00 (heptet, J=6.2 Hz, 1H, CH), 5.43 (s, 1H, C-1 H), 6.83-7.52 (m, 14H, Ar-H & NH). <sup>13</sup>C NMR (CDCl<sub>3</sub>)  $\delta$  21.78  $(CH_3)$ , 21.90 (CH<sub>3</sub>), 24.35 (CH<sub>2</sub>), 53.72 (CH), 55.92 (C-3)<sup>\*</sup>, 60.95 (C-1)<sup>\*</sup>, 67.59 (OCH<sub>3</sub>), 106.40, 110.73, 113.72, 118.11, 119.22, 121.45, 127.12, 127.93, 128.65, 128.84, 129.77, 131.44, 134.94, 136.50, 142.42, 158.76, 172.69 (CO).

\* characteristic signals

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