

SYNTHESIS OF TRIFLUOROMETHYLPYRIMIDINES FROM FLUORINATED ENAMINODIKETONES

Mustapha Soufyane,^b Sarah van den Broek,^a Layachi Khamliche,^b and Catherine Mirand^{a*}

^a Laboratoire de Transformations et Synthèse de Substances Naturelles, UPRESA/CNRS 6013, Université de Reims Champagne Ardenne, Faculté de Pharmacie, 51 rue Cognacq-Jay, F-51096 Reims Cedex, France

^b Département de Chimie, Faculté des Sciences, 24000 El Jadida, Maroc

Abstract - 4-Trifluoromethylpyrimidines (**2**, **7**) were easily synthesized by reaction of enamino diketones (**1**, **6**) with *N,N*-dimethylguanidine, *O*-methylisourea or guanidine. The 4,6-bistrifluoromethylpyrimidine (**11**) was obtained from **10** and *O*-methylisourea. Hydrolysis of the 2-methoxypyrimidines (**2c**, **7c**, **12**) gave the 4-hydroxypyrimidones (**3**, **8**, **13**).

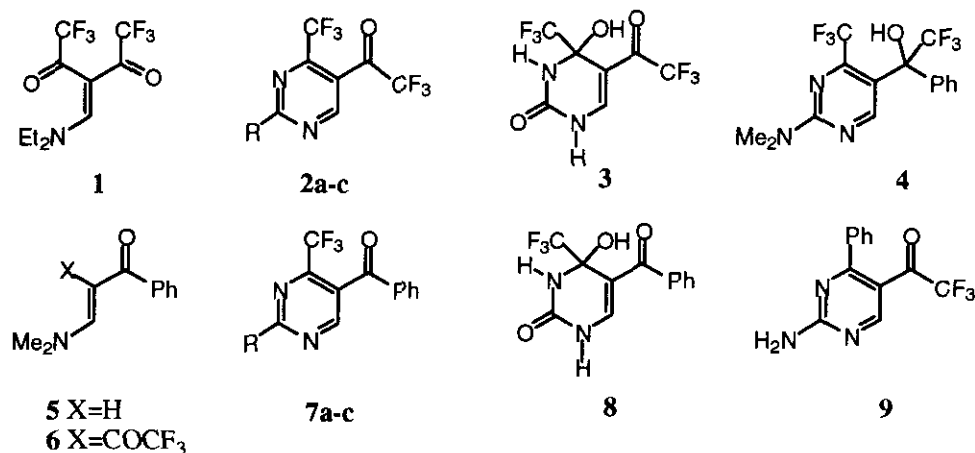
In a previous paper, we have described a simple and efficient synthesis of various trifluoromethylated nitrogen heterocycles from *N,N*-diethylaminomethylene-1,1,1,5,5,5-hexafluoroacetylacetone (**1**) (DAMFA), an enamino diketone easily obtained by reacting trifluoroacetic anhydride with triethylamine.¹⁻³ Especially, DAMFA undergoes facile cyclocondensation with *N,N*-dimethylguanidine and *O*-methylisourea to give the 5-trifluoroacetyl-4-trifluoromethylpyrimidines (**2a**) and (**2c**) with good yields.²

The increasing interest for fluoroheterocyclic compounds in medicinal and agricultural fields⁴ and recent developments of new synthetic methodologies towards such compounds⁵ prompted us to apply our approach to other N-C-N nucleophiles and other CF₃ containing enamino ketones.

As expected, condensation of DAMFA (**1**) with guanidine yielded the 2-aminopyrimidine (**2b**) (82%). However attempts at reacting **1** with urea were fruitless (acidic or alkaline conditions) and we envisioned therefore an indirect method to access to 2-hydroxypyrimidines. Thus *O*-demethylation of the pyrimidine (**2c**) was achieved by treatment with hydrobromic acid/acetic acid, but hydration of the 3,4-double bond occurred, leading to the 4-hydroxydihydropyrimidone (**3**). The result is not unexpected as fluoro imines are known to easily react with nucleophiles to form stable adducts.⁶ Compound (**3**) was also obtained through deamination of the 2-aminopyrimidine (**2b**) with nitrous acid. The structure of **3** was clearly established on the basis of its spectral data: the ¹H NMR spectrum showed four singlets at 10.63, 8.91,

7.86, 7.68 ppm and the C-4 signal on the ^{13}C NMR spectrum was shielded to 82.1 ppm, disclosing the lack of aromaticity.

In order to prepare benzylpyrimidine derivatives, **2a** was reacted with phenyllithium to give compound (**4**) (29%).

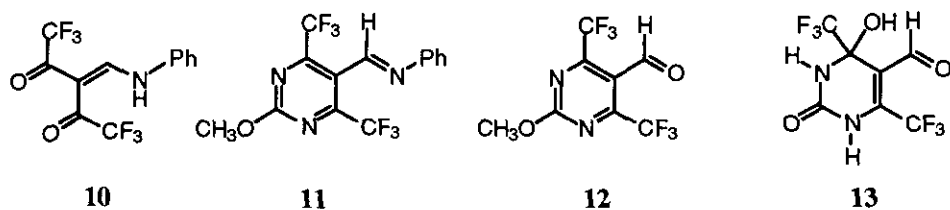


Scheme 1 a: R = NMe₂; b: R = NH₂; c: R = OMe

As an extension of this work, we investigated the synthesis of 5-aryl-4-trifluoromethylpyrimidines from enamino diketone (**6**), which was easily prepared in two steps: condensation of *N,N*-dimethylformamide dimethylacetal with acetophenone led to enaminone (**5**) whose trifluoroacetylation provided **6**.⁸ Similarly to DAMFA enamino diketone (**6**) reacted with *N,N*-dimethylguanidine, guanidine and *O*-methylisourea to give the desired pyrimidines (**7a-c**) respectively, with moderate yields (60-70%). With guanidine, the phenylpyrimidine (**9**) was isolated as a by product (30%). The structural distinction between **7b** and **9** was established as follows:

- the ^{13}C NMR spectra showed the carbonyl carbon as a singlet at 192.2 ppm for **7b** and a quadruplet at 179.5 ppm ($^2J_{\text{CF}}=37.4$) for **9**;
- on the MS spectra, the base peaks were seen at m/z 105 (loss of COPh) for **7b** and at m/z 170 (loss of COCF₃) for **9**.

Treatment of **7c** with HBr/AcOH afforded the 4-hydroxydihydropyrimidone (**8**), analogous to **3**.



Scheme 2

Synthesis of bistrifluoromethylpyrimidines was also performed from DAMFA (1) by replacing the diethylamino group by a less labile group, in order to favor the nucleophilic addition to the two carbonyl groups. Thus, cyclocondensation of **10** (available by nucleophilic substitution of DAMFA with aniline²) with *O*-methylisourea led to the imine (**11**) whose hydrolysis gave aldehyde (**12**). The transformation of **12** to dihydropyrimidone (**13**) was carried out under the conditions used for **2c**.

Thus, the fluorinated enamino ketones (**1**, **6** and **10**) constitute useful synthetic blocks for efficient syntheses of pyrimidines bearing one or two trifluoromethyl groups at C-4 and C-6 positions.

EXPERIMENTAL

Melting points were determined on a Reichert melting point apparatus and are uncorrected. IR spectra (ν cm^{-1}) were recorded on a BOMEM MB series apparatus. ^1H (300 MHz) and ^{13}C (75 MHz)-NMR spectra were obtained on a Bruker AC 300 instrument with TMS as internal standard; the chemical shifts are expressed in ppm and coupling constants in Hz. MS spectra were recorded on a UG Autospec spectrometer. Elemental analyses were performed with a Perkin Elmer CHN 2400 analyzer.

Monitoring of reactions was carried out using Merck TLC aluminium sheets (Kieselgel 60 PF 254). Preparative TLC (Kieselgel 60 PF 254) and column chromatographies (Kieselgel 60 70-230 mesh) were performed with indicated eluents.

2-Amino-5-trifluoroacetyl-4-trifluoromethylpyrimidine (**2b**):

To a solution of **1** (1 g, 3.44 mmol) in MeCN (7 mL) were added guanidinium nitrate (430 mg, 3.52 mmol) and K_2CO_3 (486 mg, 3.52 mmol). The reaction mixture was stirred at 65°C for 20 h. After filtration of the precipitate and evaporation of the solvent, the crude product was chromatographed on silica gel ($\text{CH}_2\text{Cl}_2/\text{MeOH}$ 5:1) to give **2b** (730 mg; 82%): mp 138-139°C (hexane); IR (KBr) 3358, 1734, 1665, 1591; ^1H NMR ($\text{DMSO}-d_6$) 8.90 (s, 1H), 8.65 (br s, 1H) 8.66 (br s, 1H); ^{13}C NMR ($\text{DMSO}-d_6$) 176.9 (q, $J_{\text{CF}}=35.6$), 163.7, 162.4, 156.5 (q, $J_{\text{CF}}=36.4$), 119.9 (q, $J_{\text{CF}}=275.8$), 116.2 (q, $J_{\text{CF}}=290.2$), 110.9; EIMS: m/z (%) = 259 (M^+ , 62), 190 (98), 163 (100), 121 (52), 69 (47). Anal. Calcd for $\text{C}_7\text{H}_3\text{N}_3\text{OF}_6$ C 32.43, H 1.17, N 16.22. Found C 32.41, H 0.90, N 15.84.

5-Trifluoroacetyl-4-trifluoromethyl-4-hydroxy-1,2,3,4-tetrahydropyrimidin-2-one (**3**):

from 2c: A solution of pyrimidine (**2c**) (900 mg, 3.28 mmol) and 33% HBr in acetic acid (15 mL) was left at rt for 24 h. The reaction mixture was then concentrated under reduced pressure to give a solid residue which was recrystallized from acetone to give **3** (660 mg, 72%): mp 145-146°C; IR (KBr) 3358, 3229, 3128, 1726, 1676; ^1H NMR ($\text{DMSO}-d_6$) 10.63 (d, 1H, $J=4.5$), 8.91 (br s, 1H), 7.86 (br s, 1H), 7.68 (d, 1H, $J=4.5$); ^{13}C NMR ($\text{DMSO}-d_6$) 173.7 (q, $J_{\text{CF}}=32.3$), 148.6, 147.1, 123.8 (q, $J_{\text{CF}}=298.8$), 116.6 (q,

$J_{CF}=292.2$), 101.3, 82.1 (q, $J_{CF}=33.8$); CIMS: m/z (%) = 279 (MH⁺, 38), 261 (100), 241 (12), 209 (35), 191 (10), 139 (10), 83 (30); HRMS calcd for [(M+H)] C₇H₅N₂O₃F₆ 279.0204, found 279.0168.

from 2b: **2b** (150 mg, 0.58 mmol) was dissolved in acetone/water/acetic acid(1 : 1 : 1) (4.5 mL) and NaNO₂ (870 mg, 12.61 mmol) was added. This mixture was stirred at 80°C for 24 h. After cooling to rt, the reaction solution was quenched with 10% NaHCO₃ (40 mL), extracted with EtOAc (3x40 mL). The combined organic layers were washed with water (20 mL), dried (Na₂SO₄), and concentrated under reduced pressure. The residue was chromatographed on silica gel (CH₂Cl₂/MeOH 4:1) affording **3** (120 mg, 74%).

1-Phenyl-1-(2-*N,N*-dimethylamino-4-trifluoromethylpyrimidin-5-yl)-2,2,2-trifluoroethanol (**4**):

A solution of **2a** (287 mg, 1 mmol) in dry THF was cooled to -80°C, and 1.8 M phenyllithium (0.83 mL, 1.5 mmol) in cyclohexane/Et₂O was added. The reaction mixture was stirred at -80°C for 2 h. Then H₂O was added and the mixture was extracted three times with CH₂Cl₂. The combined organic phases were washed (brine), dried (Na₂SO₄), filtered and concentrated under reduced pressure. The residue (289 mg) was purified by column chromatography (hexane/40-60% CH₂Cl₂) to yield **4** (107 mg, 29%): white amorphous solid; IR (KBr) 3235, 1601; ¹H NMR (CDCl₃) 8.59 (s, 1H), 7.46-7.27 (m, 5H), 3.64 (s, 1H), 3.20 (s, 6H); ¹³C NMR (CDCl₃) 160.3, 159.5, 154.3 (q, $J_{CF}=37$), 138.5, 128.9, 128.2, 127.0, 124.9 (q, $J_{CF}=286$), 120.5 (q, $J_{CF}=275$), 116.7, 78.4 (q, $J_{CF}=29$), 36.8; EIMS: m/z (%) = 365 (M⁺, 41), 296 (100), 276 (58), 218 (44), 121 (8), 105 (58); HRMS calcd for C₁₅H₁₃N₃O₃F₆ 365.0963, found 365.0948.

3-Benzoyl-1,1,1-trifluoro-4-*N,N*-dimethylaminobut-3-en-2-one (**6**):

Trifluoroacetic anhydride (5 mL, 35.5 mmol) was added dropwise to a solution of enamino ketone (**5**) (1.5 g, 8.57 mmol) in CH₂Cl₂ (12 mL). The resulting solution was kept at rt for 3 days. The reaction mixture was quenched with saturated NaHCO₃ and extracted with CH₂Cl₂. The organic phase was washed (H₂O), dried (Na₂SO₄) and concentrated under reduced pressure. The residue was chromatographed on silica gel (CH₂Cl₂) to give **6** (1.84 g, 79%): mp 68°C (hexane); IR (KBr) 1672, 1614; ¹H-NMR (CDCl₃) 7.85 (d, 2H, J=7.5), 7.75 (s, 1H), 7.55 (t, 1H, J=7.5), 7.44 (t, 2H, J=7.5), 3.29 (s, 3H), 2.70 (s, 3H); ¹³C-NMR (CDCl₃) 194.2, 176.9 (q, $J_{CF}=32.2$), 157.2, 138.9, 132.9, 129.1, 128.5, 117.3 (q, $J_{CF}=290.6$), 105.8, 47.9, 41.5; EIMS: m/z (%) = 271 (M⁺, 22), 202 (40), 194 (10), 174 (8), 129 (9), 105 (100); HRMS calcd for C₁₃H₁₂NO₂F₃ 271.0820, found 271.0817.

5-Benzoyl-4-trifluoromethyl-2-*N,N*-dimethylaminopyrimidine (**7a**):

A solution of enamino diketone (**6**) (1457 mg, 5.38 mmol) and *N,N*-dimethylguanidine (475 mg, 5.46 mmol) in acetonitrile (20 mL) was left at rt for 16 h. After evaporation of the solvent, the crude product was chromatographed on silica gel (CH₂Cl₂) to give **7a** (1108 mg, 70%): oil; IR (film) 1678, 1590; ¹H NMR (CDCl₃) 8.48 (s, 1H), 7.80 (d, 2H, J=7.5), 7.58 (t, 1H, J=7.5), 7.46 (t, 2H, J=7.5), 3.27 (s, 6H); ¹³C

NMR (CDCl₃) 192.3, 161.3, 160.1, 154.3 (q, $J_{CF}=35.6$), 137.1, 133.5, 129.7, 128.5, 120.2 (q, $J_{CF}=275.8$), 117.7, 36.9; EIMS m/z (%) = 295 (M⁺, 94), 280 (24), 266 (27), 226 (16), 218 (100), 105 (44); HRMS calcd for C₁₄H₁₂N₃OF₃ 295.0932, found: 295.0933.

2-Aminopyrimidines (**7b** and **9**):

To a solution of enamino diketone (**7**) (600 mg, 2.21 mmol) in MeCN (6 mL) were added guanidinium nitrate (280 mg, 2.30 mmol) and K₂CO₃ (317 mg, 2.30 mmol). The reaction mixture was heated to 65°C for 14 h. After filtration of the precipitate and evaporation of the solvent, the crude product was purified by TLC (98:2 CH₂Cl₂/MeOH) to afford **7b** (355 mg, 60%) and **9** (177 mg, 30%).

2-Amino-5-benzoyl-4-trifluoromethylpyrimidine (7b): mp 136-137°C (CHCl₃); IR (KBr) 3360, 1675, 1601; ¹H NMR (DMSO-d₆) 8.56 (s, 1H), 7.88 (br s, 2H), 7.82 (d, 2H, $J=7.5$), 7.73 (t, 1H, $J=7.5$), 7.57 (t, 2H, $J=7.5$); ¹³C NMR (DMSO-d₆) 192.2, 163.4, 161.1, 153.3 (q, $J_{CF}=35.2$), 136.8, 134.1, 129.9, 129.1, 120.5 (q, $J_{CF}=275.3$), 118.7; EIMS: m/z (%) = 267 (M⁺, 60), 198 (7), 190 (92), 163 (11), 121 (7), 105 (100); HRMS calcd for C₁₂H₈N₃OF₃ 267.0619, found: 267.0554.

2-Amino-5-trifluoroacetyl-4-phenylpyrimidine (9): mp 153-154°C (CHCl₃); IR (KBr) 3378, 1668, 1597; ¹H NMR (DMSO-d₆) 8.75 (s, 1H), 8.15 (br s, 2H), 7.57-7.41 (m, 5H); ¹³C NMR (DMSO-d₆) 179.5 (q, $J_{CF}=37.4$), 170.8, 163.6, 161.6, 138.3, 130.0, 128.4, 128.3, 116.4 (q, $J_{CF}=292.1$), 113.0; EIMS: m/z (%) = 267 (M⁺, 72), 198 (100), 170 (13), 156 (12), 143 (10), 129 (49), 116 (27), 104 (63). Anal. Calcd for C₁₂H₈N₃OF₃ C 53.92, H 3.02, N 15.73. Found C 53.77, H 2.67, N 15.58.

5-Benzoyl-4-trifluoromethyl-2-methoxypyrimidine (**7c**):

To a solution of enamino diketone (**6**) (1.5 g, 5.54 mmol) in MeCN (12 mL) were added *O*-methylisourea hydrogen sulfate (964 mg, 5.60 mmol) and K₂CO₃ (773 mg, 5.60 mmol). The reaction mixture was stirred at rt for 20 h. After filtration of the precipitate and evaporation of the solvent, the crude product was chromatographed on silica gel (CH₂Cl₂/hexane 4:1) to give **7c** (985 mg; 63%): oil; IR (film) 1674, 1586; ¹H NMR (CDCl₃) 8.71 (s, 1H), 7.81 (d, 2H, $J=7.5$), 7.68 (t, 1H, $J=7.5$), 7.52 (t, 2H, $J=7.5$), 4.18 (s, 3H); ¹³C NMR (CDCl₃) 191.2, 165.6, 161.1, 155.0 (q, $J_{CF}=36.8$), 136.1, 134.5, 129.8, 128.8, 124.7, 119.8 (q, $J_{CF}=275.1$), 55.9; EIMS: m/z (%) = 282 (M⁺, 70), 262 (7), 252 (7), 213 (12), 205 (54), 110 (21), 105 (100). Anal. Calcd for C₁₃H₉N₂O₂F₃ C 55.30, H 3.21, N 9.93. Found C 55.35, H 2.78, N 9.72.

5-Benzoyl-4-trifluoromethyl-4-hydroxy-1,2,3,4-tetrahydropyrimidin-2-one (**8**):

A solution of pyrimidine (**7c**) (200 mg, 0.68 mmol) and 33% HBr in acetic acid (15 mL) was left at rt for 24 h. The reaction mixture was then concentrated under reduced pressure to give a solid residue which was recrystallized from acetone to give **8** (148 mg, 76%): mp 172-173 °C; IR (KBr) 3225, 3123, 1709,

1655, 1613; ^1H NMR (DMSO- d_6) 10.08 (br s, 1H), 8.79 (br s, 1H), 7.78 (s, 1H), 7.65-7.50 (m, 5H), 7.14 (d, 1H, $J=4.5$); ^{13}C NMR (DMSO- d_6) 193.9, 149.3, 144.9, 138.6, 132.2, 128.9, 128.8, 124.1 (q, $J_{\text{CF}}=288.2$), 104.6, 82.7 (q, $J_{\text{CF}}=32.8$); EIMS: m/z (%) = 268 (M^+ , -18, 13), 217 (80), 191 (7), 139 (100), 105 (88); HRMS calcd for [($\text{M}-\text{H}_2\text{O}$)] $\text{C}_{12}\text{H}_7\text{N}_2\text{O}_2\text{F}_3$ 268.0459, found 268.0450.

4,6-Bis(trifluoromethyl)-2-methoxy-5-(phenyliminomethyl)pyrimidine (**11**):

Compound (**10**) (900 mg, 2.89 mmol) was added to a mixture of *O*-methylisourea hydrogen sulfate (500 mg, 2.91 mmol) and potassium carbonate (402 mg, 2.91 mmol) in MeCN (6 mL). After stirring at rt for 20 h, the precipitate was filtered off and the solvent was evaporated. The resulting product was recrystallized from hexane/chloroform to give **11** (730 mg; 72%): mp 116°C; IR (KBr) 1647, 1589; ^1H NMR (CDCl_3) 8.67 (m, 1H), 7.44 (t, 2H, $J=7.5$), 7.31 (t, 1H, $J=7.5$), 7.18 (d, 2H, $J=7.5$), 4.21 (s, 3H); ^{13}C NMR (CDCl_3) 164.6, 158.4 (q, $J_{\text{CF}}=36.2$), 150.8, 150.6, 129.4, 127.3, 120.8, 120.6, 118.0 (q, $J_{\text{CF}}=284.6$), 56.5; EIMS: m/z (%) = 349 (M^+ , 10), 348 (70), 104 (35), 77 (100). Anal. Calcd for $\text{C}_{14}\text{H}_9\text{N}_3\text{OF}_6$ C 48.13, H 2.59, N 12.03. Found C 48.03, H 2.19, N 11.72.

4,6-Bis(trifluoromethyl)-2-methoxypyrimidine-5-carbaldehyde (**12**):

Pyrimidine (**11**) (600 mg, 1.72 mmol) was stirred with 10% HCl (15 mL) in THF (25 mL) at rt for 48 h. The reaction mixture was extracted twice with ethyl acetate. The combined organic layer were washed with brine, dried (Na_2SO_4), filtered and concentrated under reduced pressure. Purification of the residue by column chromatography on silica gel (CH_2Cl_2 /hexane 2:1) yielded **12** (460 mg, 98%): oil; IR (film) 1721, 1586; ^1H NMR (CDCl_3) 10.42 (m, 1H), 4.23 (s, 3H); ^{13}C NMR (CDCl_3) 184.9, 165.3, 159.0 (q, $J_{\text{CF}}=36.6$), 120.9, 119.5 (q, $J_{\text{CF}}=286.2$), 57.0; EIMS: m/z (%) = 274 (M^+ , 65), 273 (48), 244 (35), 147 (22), 110 (100), 69 (48). Anal. Calcd for $\text{C}_8\text{H}_4\text{N}_2\text{O}_2\text{F}_6$ C 35.03, H 1.47, N 10.22. Found C 35.21, H 1.32; N 10.30.

4,6-Bis(trifluoromethyl)-4-hydroxy-1,2,3,4-tetrahydropyrimidin-2-one-5-carbaldehyde (**13**):

This product was obtained from pyrimidine (**12**) using the procedure described for tetrahydropyrimidone (**3**) (from **2c**): yield 92%; mp 127-128 °C (acetone); IR (KBr) 3443, 3262, 3204, 3162, 1726, 1668; ^1H NMR (DMSO- d_6) 10.99 (br s, 1H), 9.68 (s, 1H), 9.24 (s, 1H), 8.36 (br s, 1H); ^{13}C NMR (DMSO- d_6) 185.6, 148.7, 138.3 (q, $J_{\text{CF}}=36.2$), 123.5 (q, $J_{\text{CF}}=286.6$), 119.4 (q, $J_{\text{CF}}=276.6$), 107.1, 81.9 (q, $J_{\text{CF}}=33.8$); EIMS: m/z (%) = 279 ($\text{M}^+ + 1$, 7), 261 (33), 241 (10), 223 (45), 216 (30), 209 (100), 180 (20), 166 (785), 138 (94); HRMS calcd for [(MH^+)] $\text{C}_7\text{H}_5\text{N}_2\text{O}_3\text{F}_6$; 279.0204, found 279.0205. Anal. Calcd for $\text{C}_7\text{H}_4\text{N}_2\text{O}_3\text{F}_6$ C 30.21, H 1.45, N 10.07. Found C 30.48, H 1.21, N 9.88.

ACKNOWLEDGEMENTS

We thank C. Peterman and P. Sigaut for recording the NMR and MS spectra.

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Received, 21st May, 1999