LEWIS ACID-MEDIATED RING EXPANSION REACTION OF 2,3-METHANOCHROMANONES WITH SILYL ENOL ETHERS

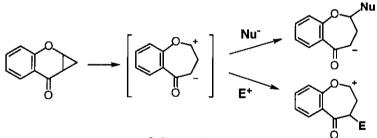
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Abstract - In the presence of trimethylsilyl trifluoromethanesulfonate (TMSOTf), methanochromanones easily reacted with silyl enol ethers to give the [3+2] cycloadducts and the corresponding adducts in good yields.

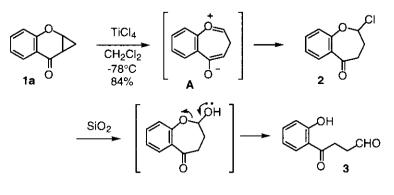
Cyclopropanes having an electron-withdrawing or -donating group are susceptible to ring-opening reactions.¹ Especially, cyclopropanes with donor and acceptor substituents at vicinal positions on the cyclopropane ring are the equivalent of a ring-opened 1,3-zwitterion, which is expected to react with both nucleophiles and electrophiles.² Saigo reported the syntheses of 5-membered carbo-³ and heterocyclic compounds⁴ by the reactions of 2,2-dialkoxycyclopropanecarboxylic acid esters having two electron-donating groups and one electron-withdrawing group with various substrates. Kuwajima also reported Lewis acid-mediated [3+2] cycloaddition reaction of vicinally donor-acceptor-substituted cyclopropane with silyl enol ethers.⁵

We were interested in the reactivity of 2,3-methanochromanones $(1)^6$ readily prepared from the corresponding chromone and dimethyloxosulfonium methylide.⁷ Because methanochromanones have an alkoxy group as a donor and a carbonyl group as an acceptor in the benzopyran ring, formation of a 1,3-zwitterionic intermediate would be expected. We speculated that, upon treatment with various nucleophiles and/or electrophiles in the presence of a Lewis acid, the methanochromanones would undergo ring-opening and provide several 1-benzoxepins as the ring-expanded products⁸ (Scheme 1). We report herein the Lewis acid-promoted reaction of methanochromanones with silyl enol ethers *via* a 1,3-zwitterionic intermediate.



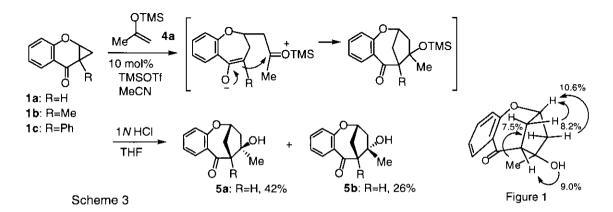
Scheme 1

At first, the reactivity of 2,3-methanochromanone (1a) by the action of a Lewis acid was examined. A solution of a stoichiometric amount of TiCl₄ in CH₂Cl₂ was added to a solution of 1a in CH₂Cl₂ at -78 $^{\circ}$ C to give the 2-chloro-1-benzoxepinone (2)⁹ in 84% yield along with a small amount of its ring-cleaved product (3).^{6c} The obtained benzoxepin (2) is very unstable under acidic conditions and was converted into 3 even when 2 was treated with silica gel (Scheme 2). In this result, we considered that methanochromanone (1a) is the equivalent of a ring-opened zwitterion (A).



Scheme 2

We next examined the reaction of 2,3-methanochromanone (1a) with silyl enol ethers as the nucleophile. When the reaction was performed in the presence of TiCl₄ in CH₂Cl₂ at -78 °C, however, **3** was obtained as a major product and the desired adduct was obtained in low yield. In order to improve the yield of this reaction, we carried out the reaction of **1a** with **4a** under various conditions. Finally, it was found that the best yield of the adduct (**5**) was obtained using trimethylsilyl trifluoromethanesulfonate (TMSOTf) as a Lewis acid in MeCN.¹⁰ Thus, the reaction of **1a** with **4a** in the presence of TMSOTf (10 mol%) in MeCN gave the cycloadducts (**5a** and **5b**)¹¹ in 42% and 26% yields, respectively (Scheme 3). The stereochemistry of the cycloadduct (**5a**) was assigned based on NOE experiments (Figure 1).



A similar reaction of **1a** with **4b** proceeded smoothly to give the corresponding cycloadducts (**6a** and **6b**) in high yield (Table 1, entry 1). The reaction of **1a** with cyclic silyl enol ethers under the same conditions

also gave the cycloadducts in good yields (entries 2-4). 3-Methyl-2,3-methanochromanone (1b) was next treated with 4a under the same conditions as described above to give the cycloadduct (13) and the 1benzoxepin (14) in 35 and 50% yields, respectively (entry 5). It was found that introduction of a bulkier alkyl group at the 3-position of the pyrone ring increased the yield of the non-cyclized adduct. Thus, the adduct (18) was obtained in 68% yield along with a small amount of [3+2] cycloadduct when 1c reacted with enol silyl ether (4a, entry 7). In addition, an enhancement of the *trans/cis* ratio of the non-cyclized adduct was observed when the methyl group at the 3-position of the pyrone ring was replaced with the phenyl group.

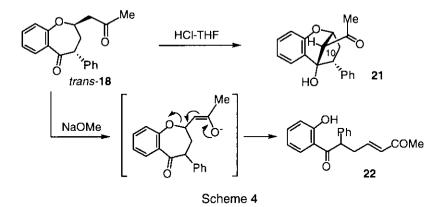
entry	R	Silyl enol ether	temp. (°C)	yield (%) ^a products	yield (%) ^a
1	н	OSiMe ₃ Phr 4b	-40		53
2	н	OSiMe ₃	-40	$\begin{array}{c} \begin{array}{c} \begin{array}{c} \end{array} \\ \end{array} \\ \end{array} \\ \begin{array}{c} \end{array} \\ \end{array} \\ \end{array} \\ \end{array} \\ \begin{array}{c} \end{array} \\ \end{array} $	8 ^b
3	н	OSiMe₃ ↓ 4d	-40	$\begin{array}{c} \begin{array}{c} \begin{array}{c} \begin{array}{c} \end{array} \\ \end{array} \\ \end{array} \\ \begin{array}{c} \end{array} \\ \end{array} \\ \end{array} \\ \begin{array}{c} \end{array} \\ \end{array} $	Od
4	н	OSiMe ₃	-40	$\begin{array}{c} \begin{array}{c} \begin{array}{c} \begin{array}{c} \end{array} \\ \end{array} \\ \end{array} \\ \begin{array}{c} \end{array} \\ \begin{array}{c} \end{array} \\ \end{array} \\ \end{array} \\ \end{array} \\ \begin{array}{c} \end{array} \\ \end{array} $	15 ^b
5	Me	OSiMe₃ Mer 4a	-40	$\begin{array}{c} \bigcirc & \bigcirc & \bigcirc & \bigcirc & H \\ & \bigcirc & Me \\ & \bigcirc & Me \\ & & 13 \end{array} (>10:1)^e \\ & \bigcirc & Me \\ & & 0 \\ \end{array} \begin{array}{c} \bigcirc & \bigcap & Me \\ & & 0 \\ & & Me \\ & & 14 \\ & & 0 \\ \end{array}$	50 ^b
6	Me	OSiMe₃ Phr 4b	-40	$\begin{array}{c} \begin{array}{c} \begin{array}{c} \end{array} \\ \end{array} \\ \end{array} \\ \begin{array}{c} \end{array} \\ \begin{array}{c} \end{array} \\ \begin{array}{c} \end{array} \\ \end{array} \\ \end{array} \\ \begin{array}{c} \end{array} \\ \end{array} \\ \begin{array}{c} \end{array} \\ \end{array} \\ \end{array} \\ \begin{array}{c} \end{array} \\ \end{array} \\ \end{array} \\ \begin{array}{c} \end{array} \\ \end{array} \\ \end{array} \\ \end{array} \\ \begin{array}{c} \end{array} \\ \end{array} $	26 ^b
7	Ph	OSiMe₃ Mer 4a	0	CH 0H 2 0 Ph 17 Ph 18	68 ^b
8	Ph	OSiMe ₃ Ph 4b	0	$\begin{array}{c} \begin{array}{c} \begin{array}{c} \begin{array}{c} \end{array} \\ \end{array} \\ \end{array} \\ \begin{array}{c} \end{array} \\ \end{array} \\ \end{array} \\ \begin{array}{c} \end{array} \\ \end{array} \\ \begin{array}{c} \end{array} \\ \end{array} \\ \end{array} \\ \begin{array}{c} \end{array} \\ \end{array} \\ \begin{array}{c} \end{array} \\ \end{array} \\ \end{array} \\ \begin{array}{c} \end{array} \\ \end{array} \\ \end{array} \\ \end{array} \\ \begin{array}{c} \end{array} \\ \end{array} $	38 ^b

Table 1. Reaction of Methanochromanones (1) with Silyl Enol Ethers

^aIsolated yields. ^bTwo diastereomers were isolated.^cThree diastereomers were isolated.

^dNot detected. ^eRatio of isomers. ^fRatio of *trans-* : *cis*-isomers.

It was also found that when the adduct (*trans*-18) was treated with conc. HCl in THF, the newly formed cycloadduct (21) and its C-10 epimer were obtained in 76% and 20% yields, respectively. On the contrary, treatment of *trans*-18 under basic conditions using NaOMe in ether gave 22 as the sole product in 80% yield (Scheme 4).



In summary, we have demonstrated that the TMSOTf-catalyzed ring-opening addition reactions of 1 with silyl enol ethers proceeded smoothly to afford the corresponding adducts in high yields. We are now investigating the Lewis acid-mediated ring expansion reaction of methanochromanones with various substrates, and the results will be reported in due course.

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- a) K. Saigo, S. Shimada, Y. Hashimoto, and M. Hasegawa, *Chem. Lett.*, 1989, 1293. b) K. Saigo, S. Shimada, and M. Hasegawa, *Chem. Lett.*, 1990, 905. c) K. Saigo, S. Shimada, Y. Hashimoto, T. Nagashima, and M. Hasegawa, *Chem. Lett.*, 1990, 1101. d) S. Shimada, Y. Hashimoto, A. Suda, M. Hasegawa, and K. Saigo, *J. Org. Chem.*, 1992, **57**, 7126. e) S. Shimada, Y. Hashimoto, and K. Saigo, *J. Org. Chem.*, 1993, **58**, 5226. f) S. Shimada, Y. Hashimoto, T. Nagashima, M. Hasegawa, and K. Saigo, *Tetrahedron*, 1993, **49**, 1589.
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- Compound (2): mp 66-68 °C (AcOEt-hexane). IR (CDCl₃) 1690 cm⁻¹. ¹H NMR (500 MHz, CDCl₃) δ
 7.84 (1H, dd, J = 7.6, 1.8 Hz, H-6), 7.59 (1H, ddd, J = 8.0, 7.6, 1.8 Hz, H-8), 7.35 (1H, td, J = 7.6, 1.0 Hz, H-7), 7.19 (1H, dd, J = 8.0, 1.0 Hz, H-9), 6.31 (1H, dd, J = 10.1, 5.2 Hz, H-2),
 2.86-2.76 (2H, m, H-4), 2.60 (1H, m, H-3), 2.36 (1H, m, H-3). High-resolution MS m/z Calcd for C₁₀H₉O₂³⁵Cl (M⁺): 196.0291, Found: 196.0247.
- 10. A typical procedure is as follows: To a stirred solution of 1a (80 mg, 0.5 mmol) and 4a (130 mg, 1.0 mmol) in MeCN (4 mL) was added dropwise a solution of TMSOTF (11 mg, 0.05 mmol) in MeCN (0.5 mL) at -40 ℃ under argon atmosphere. After being stirred for 30 min, the reaction was quenched at the same temperature by adding saturated aqueous NaHCO₃ (2 mL). The mixture was stirred vigorously for 10 min and allowed to warm to room temperature. The mixture was extracted with CH₂Cl₂ (20 mL x 3), the combined organic layers were dried over Na₂SO₄, and the solvent was evaporated under reduced pressure. The residue was dissolved in THF-1N HCl (2 : 1, 6 mL) and the solution was stirred for 1 h at 0 ℃. The mixture was extracted with ether (20 mL x 3), the combined organic layers were dried outer reduced pressure. The residue was evaporated under reduced pressure. The residue was evaporated under reduced pressure. The residue was evaporated under reduced pressure. The residue was extracted with ether (20 mL x 3), the combined organic layers were dried over Na₂SO₄, and the solvent was evaporated under reduced pressure. The residue was evaporated under reduced pressure. The residue was purified by silica gel column chromatography (eluent: hexane-AcOEt = 5 : 1) to give the cycloadducts (5a and 5b) in 42% and 26% yields, respectively.
- 11. Compound (**5a**): mp 93-95 °C (ether-hexane). IR (CDCl₃) 3450, 1660 cm⁻¹. ¹H NMR (500 MHz, CDCl₃) δ 7.99 (1H, dd, J = 8.2, 1.8 Hz, H-7), 7.42 (1H, ddd, J = 8.2, 7.0, 1.8 Hz, H-9), 6.99 (1H, ddd, J = 8.2, 7.0, 1.2 Hz, H-8), 6.90 (1H, dd, J = 8.2, 1.2 Hz, H-10), 5.26 (1H, ddt, J = 7.6, 6.1, 1.8 Hz, H-2), 3.05 (1H, br d, J = 6.1 Hz, H-5), 2.65 (1H, dt, J = 14.7, 6.1 Hz, H-11), 2.36 (1H, ddd, J = 16.2, 7.6, 1.2 Hz, H-3), 2.30 (1H, dd, J = 14.7, 1.8 Hz, H-11), 2.06 (1H, dt, J = 16.2, 1.8 Hz, H-3), 1.64 (1H, br s, OH), 1.36 (3H, s, Me). ¹³C NMR (500 MHz, CDCl₃) δ 199.01, 155.18, 135.20, 131.82, 123.37, 122.40, 121.01, 82.43, 80.95, 65.85, 46.52, 30.59, 25.38. Compound (**5b**): mp 82-83 °C (ether-hexane). IR (CDCl₃) 3400, 1660 cm⁻¹. ¹H NMR (500 MHz, CDCl₃) δ 7.94 (1H, dd, J = 8.2, 1.8 Hz, H-7), 7.42 (1H, ddd, J = 8.2, 7.0, 1.8 Hz, H-9), 7.02 (1H, ddd, J = 8.2, 7.0, 1.2 Hz, H-8), 6.92 (1H, dd, J = 8.2, 1.2 Hz, H-10), 5.08 (1H, m, H-2), 3.04 (1H, dd, J = 6.7, 1.2 Hz, H-5), 2.45 (1H, dd, J = 15.3, 2.2 Hz, H-11), 2.27 (1H, dt, J = 15.8, 2.2 Hz, H-3), 2.20 (1H, ddd, J = 15.3, 6.7, 5.2 Hz, H-11), 2.30-2.20 (1H, m, H-3), 2.12 (1H, br s, OH), 1.48 (3H, s, Me).

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