HETEROCYCLES, Vol. 52, No. 2, 2000, Received, 21th May, 1999

PHOTOCYCLOADDITION OF 6-CHLORO-1,3-DIMETHYLURACIL TO OLEFINS

Kazue Ohkura, Hideaki Nakamura, Hajime Takahashi, and Koh-ichi Seki* Faculty of Pharmaceutical Sciences, Health Sciences University of Hokkaido, Ishikari-Tobetsu, Hokkaido 061-0293, Japan

Abstract-----Photoreaction of 6-chloro-1,3-dimethyluracil (6-ClDMU) with olefins having alkyl, alkoxy, and vinyl groups in acetone afforded the corresponding cyclobutapyrimidine derivatives with head-to-tail regiochemistry in appreciable yields, while the reaction with olefins bearing electron withdrawing groups gave no cycloadduct. Similar Photolysis of 6-ClDMU with phenylacetylene gave [2+2] cycloadduts, 3,5-diaza[4.2.0]oct-7-enes and diphenylquinazolines.

Since the photolysis of 6-chloro-1, 3-dimethyluracil (6-ClDMU; 1) in benzene in the presence of TFA was found to give a cycloaddition product, 1,3-dimethylcyclooctapyrimidine-2,4-dione in a fair yield, 1 we have studied the photocycloaddition reaction of 6-ClDMU (1) with benzene and its derivatives systematically.²

During the course, we have found that the photoreaction proceeds through the initial *ortho*-cycloaddition followed by the elimination of hydrogen chloride and the subsequent isomerization into various cycloadducts depending on the reaction conditions and the substituents on the benzene ring.^{2e, f}

Meanwhile, the photocycloaddition of 1 to olefins has been reported by Kaneko *et al.*³ However the study was limited to only some simple alkylolefins, and no detail on the reactions has been given. In order to establish the utility and the scope of the photocycloaddition reaction of 1 for the construction of a new ring system including a pyrimidine ring, our attention was focused on the photoreaction of 1 with various olefins. In the present paper, we describe our findings that UV-irradiation of a solution of 1 and olefins in acetone afforded the corresponding cyclobutapyrimidines regioselectively in appreciable yields.

First, we have reinvestigated the photoreaction of 1 with 2,3-dimethyl-2-butene (2a), which had been reported to give the corresponding cyclobutapyrimidine (3a) quantitatively, without any detailed description about the reaction conditions or the analytical method.³ A solution of 1 (0.075 mmol) and 2a (7.5 mmol) in acetone (5 mL) in a degassed sealed tube (Pyrex) was irradiated for two hours with a 500 W high-pressure mercury lamp (>300 nm), and was found to give the desired cycloadduct (3a) in satisfactory yield (60% based on 47% of 1 consumed).

The structure of **3a** was identified by comparison of its melting point (152-153 °C, *lit.*³ 151-153 °C) and

the ¹H-NMR spectroscopic data (CDCl₃) with those reported in the literature.³

We then carried out the photoreaction with other alkylolefins under the similar conditions. 3, 3-Dimethyl-1-butene(**2b**) (2 h) gave the head-to-tail (h-t) cycloadduct (**3b**) in 50% yield. The reaction with cyclohexene (**2c**) (1 h) gave the cycloadduct (**3c**) in 35% yield (based on **1** 49% consumed). Thus, the reactions with alkylolefins were shown to give the corresponding cyclobutapyrimidines in fair yields. The structure of **3b** was deduced by means of MS (M⁺, m/z 222) and NMR spectroscopy. The ¹H-NMR spectrum (C₆D₆) showed C-8 methylene protons at δ 2.19 (*J*=11.2 and 1.2 Hz) and δ 2.37 (*J* = 11.2 and 4.5 Hz), C-7 methine proton at δ 2.41 (*J*=4.5 and 1.2 Hz), and a singlet peak due to *t*-butyl protons at δ 0.57. The ¹³C-NMR spectrum showed two peaks due to C-1 and C-6 *sp*² carbons at δ 108.84 and δ 155.56 ppm, respectively, supporting the presence of cyclobutene ring in the molecule. Further, the cross peak between *N*⁵-methyl protons and the *tert*-butyl protons was observed on the NOESY spectrum, confirming the structural assignment. The structure of **3c** was deduced from the spectral analogy with **3b**.

Hence, our attention was extended to the reaction with olefins having electron donating alkoxy groups (2d-g) and those bearing such electron withdrawing groups as -CN and -CO₂CH₃. In order to investigate an appropriate solvent for the reaction, we have first carried out the photoreaction (1 h) with ethyl vinyl ether (2d, 10 equiv. mol) in various solvents.

Table 1. Photoreaction of 6-Chloro-1, 3-dimethyluracil (1) with Ethyl Vinyl Ether (2d) in Various Solvents.

Solvent	Acetone	MeCN	AcOEt	Benzene	CH ₂ Cl ₂	EtOH	MeO	
							Н	
1 consumed(%)	41	5	4	4	4	5	3	
3 d(%)	42	9	10	0	0	0	0	

As shown in Table 1, no reaction proceeded in benzene, dichloromethane CH_2Cl_2), ethanol (EtOH) or methanol (MeOH). In ethyl acetate (AcOEt), or acetonitrile (MeCN), the desired cycloadduct (**3d**) was obtained with head-to-tail (h-t) regiochemistry, albeit in low yields. The best result (42% yield) was obtained when the reaction was performed in acetone. As the amount of **2d** increased from 10 to 100 equiv. mol, the yields of **3d** increased from 42% (based on **1**, 41% consumed) to 57% (based on **1**, 39% consumed) (Scheme 1). Hence, the following reactions were performed essentially with 100 equiv. molar olefins in acetone, unless cited therein.

The photoreaction with *tert*-butyl vinyl ether (2e) and 2-methoxypropene (2f) gave the corresponding [2+2] cycloadducts (3e, 3f) with h-t regiochemistry⁴ in 30% and 79% yields, respectively (Scheme 1). Similarly, cyclic vinyl ether, 2, 3-dihydropyran (2g), afforded 3g in 30% yield (based on 1, 27% consumed). By contrast, the photoreaction with electron deficient olefins, methyl acrylate and acrylonitrile, gave no cycloadduct.

The structures of the cycloadducts (**3d**, **3e**, **3f**, and **3g**) were deduced from the spectral (¹H-NMR) analogy with **3b**. The nuclear Overhauser enhancement (NOE) experiments confirmed their structural assignments (Figure 1).



a, Irradiated for 2 h.*b*, Irradiated for 1 h.*c*, Irradiated for 24 h.*d*, Yields are given based on **1** consumed. *e. trans* + *cis*-isomer (3:1)

Scheme 1



Figure 1. NOE Correlation for 3d, 3g, and 3i

We then extended the present method to the reaction with conjugated olefins, *trans*-piperylene (2h) and indene (2i). It is well recognized that some conjugated olefins such as piperylene (2h) serve as an efficient triplet quencher,⁵ and hence these olefins were presumed to retard the photoreaction significantly. In fact, the reaction with 2h proceeded inefficiently to consume only 23% of 1 even upon prolonged irradiation (24 h). Nonetheless, the desired [2+2] cycloadduct (3h-*t*) and its *cis*-isomer (3h-*c*) were obtained with h-t regiochemistry in 24% and 8% yields, respectively. The adduct (3h-*c*) may result from the *trans*-isomer (3h-*t*) through its photoisomerization.⁶ Similarly the photoreaction with indene (2i) proceeded slowly (10 h, 11% of 1 consumed), but gave the h-t cycloadduct 3i (Figure 1) in 10% yield.

The above findings that the conjugated olefins afforded the corresponding cycloadduct, though restrained the reaction significantly, encouraged us to investigate the photoreaction with phenylacetylene (4),

which was expected to provide a simple synthesis of cyclobutadiene derivatives. UV-irradiation of a solution of 1 (0.075 mmol) and 4 (3.75 mmol) in acetone (5 mL) in a degassed Pyrex test tube for 24 h resulted in the formation of a complicated mixture of the products. After submission to HPLC, cycloadducts (5 and 6), having a cyclobutane ring with the hydrogen and chlorine atoms at C-5 and C-6 of the pyrimidine ring intact, and the quinazoline derivatives (7 and 8) were isolated (Scheme 2).⁷



Scheme 2

The structures of **5** and **6** were determined by the followings: The MS spectra showed M⁺ m/z 276 (**5**; 33%, **6**; 4.5%) and m/z 278 (**5**; 11.5%, **6**; 1.6%) in the ratio of 3 : 1, respectively. The ¹H-NMR spectrum of **5** showed two doublet peaks due to H-1 and H-8 with a coupling constant of J = 4 Hz, while that of **6** revealed two singlet peaks ascribed to H-1 and H-7. The NOE experiments confirmed the structural assignments. The ¹H-NMR spectra of the quinazolines (**7** and **8**), exhibited no peaks due to aliphatic protons except for two singlet peaks due the *N*-methyl protons, but twelve peaks due to aromatic protons. The ¹³C-NMR spectra and the NOE experiments supported their structures.

The formation of 7 and 8 can be explained by the further [2+2] addition of 4 to the initially produced cycloadducts (5 and 6), followed by the aromatization by eliminating hydrogen chloride (HCl) (Scheme 3). Thus, the formation of cyclobutadiene derivatives (9) failed to be detected. These findings may imply that the formation of 9 through the elimination of HCl from 5 and 6 is energetically unfavorable, or the cyclobutadienes (9) thus formed might be too labile to subsist at room temperature.⁸

Although the reaction mechanism remains unelucidated, from a mechanistic point of view it may be interesting to note that the present cycloaddition reaction takes place with olefins bearing electron donating groups, but not with those having electron withdrawing groups, implying that participation of the formation of polar intermediates could not be excluded from the present reaction.⁹ In addition, the present reaction proceeds most preferentially in a triplet sensitizer, acetone, suggesting that the triplet excited states may be involved.



Scheme 3

EXPERIMENTAL

All melting points are uncorrected. NMR spectra were measured with a JEOL JNM-EX400 (400 MHz) spectrometer, and ¹H-NMR chemical shifts are given on the δ (ppm) scale based on those of the signals of solvents; CDCl₃ (δ 7.26), C₆D₆ (δ 7.15), CD₃CN (δ 1.93), and following abbreviations are used; s = singlet, d = doublet, dd = double doublet, ddd = double doublet, br dd = broad double doublet, t = triplet, dt = double triplet, q = quartet, dq = double quartet, ddq = double double quartet, m = multiplet. ¹³C-NMR chemical shifts were recorded based on those of the signals of solvents; CDCl₃ (δ 77.0) C₆D₆ (d 128.0) CD₃CN (δ 118.2). MS spectra and high-resolution MS (HRMS) spectra were determined on a JEOL JMS-DX303 spectrometer with ionization potential at 70 eV. Short-column chromatography was performed on Kieselgel Si-60 (Merck). HPLC was conducted on a Shim-pac PREP-Sil (H) (25 cm x 20 mm *i.d.*) (silica gel), using a Shimadzu LC-6A apparatus with monitoring at 254 nm. UV-irradiation was carried out externally with a 500 W high-pressure mercury (h.p.Hg) lamp (Eiko-sha) in a degassed Pyrex tube (> 300 nm) on a merry-go-round apparatus at room temperature.

General procedure of the photoreaction ----- A solution of 1 (13.08 mg, 0.075 mmol) and an olefin (100 equiv. molar; 7.5 mmol) in acetone (5 mL) in a degassed Pyrex test tube was irradiated externally for 1 h unless cited therein. The reaction mixture was concentrated *in vacuo*, and analyzed by means of ¹H-NMR spectroscopy with terephthalnitrile as an internal standard.

Typical procedure for the isolation of the cycloadduct ----- After the photoreaction according to the general procedure, the reaction mixtures in several Pyrex tubes were put together, and evaporated *in vacuo*. The residual oil was passed through a short column of silica gel with ethyl acetate. The eluate was submitted to HPLC with following solvent systems; ethyl acetate-hexane for **3b-i**, and ethyl acetate-dichloromethane for **3a**.

3,5,7,7,8,8-Hexamethyl-3,5-diazabicyclo[4.2.0]oct-1(6)-ene-2,4-dione (3a) ----- Color-

less plates, mp 152-153 °C (recrystallized from hexane). ¹H-NMR (CDCl₃) δ : 1.30 (6H, s), 1.32 (6H, s), 3.21 (3H, s), 3.24 (3H, s). MS *m*/*z* (%): 222 (M⁺, 52), 207 (100), 164 (29), 150 (41), 122 (58), 82 (23), 42 (44). HRMS: Calcd for C₁₂H₁₈N₂O₂: 222.1368. Found: 222.1364. *Anal.* Calcd for C₁₂H₁₈N₂O₂: C, 64.84; H, 8.16; N, 12.60. Found: C, 64.82; H, 8.18; N, 12.55.

7-(*tert*-Butyl)-3,5-dimethyl-3,5-diazabicyclo[4.2.0]oct-1(6)-ene-2,4-dione (3b) -----Color-less oil. ¹H-NMR (C₆D₆) δ : 0.57 (9H, s, C(C<u>H</u>₃)₃-7), 2.19 (1H, dd, J = 1.2, 11.2 Hz, H-8), 2.37 (1H, dd, J = 4.5, 11.2 Hz, H-8), 2.41 (1H, dd, J = 4.5, 1.2 Hz, H-7), 2.70 (3H, s, N⁵-CH₃), 3.35 (3H, s, N³-CH₃). ¹³C-NMR (C₆D₆) δ : 26.83 (C-8), 27.51 (C(CH₃)₃-7), 28.01 (N³-CH₃), 31.58 (C(CH₃)₃-7), 32.71 (N⁵-CH₃), 55.73 (C-7), 108.84 (C-1), 153.88 (C-4), 158.64 (C-2). MS *m/z* (%) : 222 (M⁺, 49), 207 (100), 193 (11), 179 (7), 165 (9), 150 (72), 122 (19), 96 (31). HRMS: Calcd for C₁₂H₁₈N₂O₂: 222.1368. Found: 222.1367.

4,6-Dimethyl-4,6-diazatricyclo[**6.4.0.0**^{2,7}]**dodec-2**(7)-**ene-3,5-dione** (**3c**) ----- Colorless plates; mp 74-75 °C (recrystallized from hexane). ¹H-NMR (C₆D₆) & 1.00-1.10 (1H, m, H-9), 1.00-1.12 (2H, m, H-10), 1.14-1.28 (2H, m, H-11), 1.29-1.41 (1H, m, H-9), 1.47-1.58 (1H, m, H-12), 1.64-1.76 (1H, m, H-12), 2.47 (1H, ddd, J = 5.0, 5.2, 5.2 Hz, H-8), 2.54 (3H, s, N⁶-CH₃), 2.89 (1H, ddd, J = 5.0, 4.8, 4.8 Hz, H-1), 3.36 (3H, s, N⁴-CH₃). ¹³C-NMR (C₆D₆) & 17.77 (C-11), 18.28 (C-10), 22.06 (C-9), 23.19 (C-12), 27.85 (N⁴-CH₃), 30.29 (N⁶-CH₃), 37.07 (C-1), 41.13 (C-8), 111.40(C-2), 153.59 (C-5), 158.71 (C-3): MS *m/z* (%): 220 (M⁺, 64), 205 (23), 191 (100), 178 (20), 148 (17), 134 (46), 107 (18), 82 (26). HRMS: Calcd for C₁₂H₁₆N₂O₂: 220.1212. Found: 220.1213. *Anal.* Calcd for C₁₂H₁₆N₂O₂: C, 65.43; H, 7.32; N, 12.72. Found: C, 65.60; H, 7.32; N, 12.68.

7-Ethoxy-3,5-dimethyl-3,5-diazabicyclo[**4.2.0**]oct-1(**6**)-ene-2,4-dione (**3**d) ----- Colorless oil (13.5 mg, 26.3%). ¹H-NMR (C₆D₆) δ : 0.91 (3H, t, J = 7.0 Hz, OCH₂CH₃-7), 2.40 (1H, dd, J = 1.1, 11.0 Hz, H-8), 2.56 (1H, dd, J = 3.5, 11.0 Hz, H-8), 2.70 (3H, s, N⁵-CH₃), 2.93 (2H, m, OCH₂CH₃-7), 3.30 (3H, s, N³-CH₃), 4.05 (1H, dd, J = 1.1, 3.5 Hz, H-7). ¹³C-NMR (C₆D₆) δ : 15.21 (OCH₂CH₃-7), 27.98 (N³-CH₃), 30.63 (N⁵-CH₃), 63.60 (OCH₂CH₃-7), 75.07 (C-7), 107.82 (C-1), 152.88 (C-6), 153.17 (C-4), 159.46 (C-2). MS *m/z* (%): 210 (M⁺, 57), 181 (40), 124 (100), 96 (63), 70 (26), 42 (42). HRMS: Calcd for C₁₀H₁₄N₂O₃: 210.1004. Found: 210.0994.

7-(*tert*-Butoxy) -3,5-dimethyl-3,5-diazabicyclo[4.2.0]oct-1(6)-ene-2,4-dione(3e)Colorless oil. 1 H-NMR (CDCl₃) δ : 1.29 (9H, s, OC(CH₃)₃-7), 2.70 (1H, dd, J = 1.0, 11.1 Hz, H-8),3.13 (1H, dd, J = 3.4, 11.1 Hz, H-8), 3.28 (3H, s, N⁵-CH₃), 3.31 (3H, s, N³-CH₃), 5.06 (1H, dd, J = 1.0, 3.4 Hz, H-7). 13 C-NMR (CDCl₃) δ : 27.94 (OC(CH₃)₃-7), 28.21 (N³-CH₃), 30.87 (N⁵-CH₃), 38.63 (C-8), 69.36 (C-7), 75.73 (QC(CH₃)₃-7), 108.22 (C-1), 153.58 (C-4), 153.85 (C-6), 159.94 (C- 2). MS

80for C₁₂H₁₈N₂O₃: 238.1317. Found: 238.1319.

3,5,7-Trimethyl-7-methoxy-3,5-diazabicyclo[**4.2.0**]oct-1(**6**)-ene-2,4-dione (**3f**) ----- Colorless crystals; mp 91-92 °C (recrystallized from hexane). ¹H-NMR (C₆D₆) δ : 1.12 (3H, s, CH₃-7), 2.28 (1H,d, *J* = 11.3 Hz, H-8), 2.71 (3H, s, N⁵-CH₃), 2.72 (3H, s, OCH₃-7), 2.75 (1H, d, *J* = 11.3Hz, H-8), 3.32 (3H, s, N³-CH₃). ¹³C-NMR (C₆D₆) δ : 22.50 (CH₃-7), 27.98 (N³-CH₃), 30.32 (N⁵-CH₃), 36.74 (C-8), 51.96 (OCH₃-7), 83.29 (C-7), 106.89 (C-1), 153.28 (C-4), 155.64 (C-6), 159.44 (C-2). MS *m/z* (%): 210 (M⁺, 39), 195 (49), 138 (100), 110 (33), 84 (44), 68 (28), 42 (58). HRMS: Calcd for C₁₀H₁₄N₂O₃: 210.1004.Found:210.0988.*Anal*.Calcd for C₁₀H₁₄N₂O₃: C, 57.13; H, 6.71; N, 13.33. Found: C, 57.34; H, 6.75; N, 13.18.

4,6-Dimethyl-9-oxa-4,6-diazatricyclo[6.4.0.02,7]dodec-2(7)-ene-3,5-dione(3g)Colorless oil. 1 H-NMR (CDCl₃) δ : 1.60-1.70 (2H, m, H-11), 1.98-2.14 (2H, m, H-12), 3.32 (3H, s,
N⁶-CH₃), 3.33 (3H, s, N⁴-CH₃), 3.44-3.49 (1H, m, H-1), 3.72-3.87 (2H, m, H-10), 5.01 (1H, d, J =
4.2 Hz, H-8). NOE δ : (irradiation at H-8) 3.44-3.49 (H-1), 3.32 (N⁶-CH₃). 13 C-NMR (CDCl₃) δ : 19.35
(C-11), 22.31 (C-12), 28.23 (N⁴-CH₃), 31.57 (N⁶-CH₃), 38.75 (C-1), 63.07 (C-10), 72.56 (C-8),
113.03 (C-2), 153.48 (C-5), 155.13 (C-7), 159.73 (C-3). MS m/z (%): 222 (M⁺, 87), 193 (100), 166 (56),
136 (25), 108 (37), 81 (72), 43 (31). HRMS: Calcd for C₁₁H₁₄N₂O₃: 222.1044. Found: 222.1002.3,5-Dimethyl-7-trans-prop-1-enyl-3,5-diazabicyclo[4.2.0]oct-1(6)-ene-2,4-dione

----- Colorless oil. ¹H-NMR (C_6D_6) & 1.38 (3H, dd, J = 1.5, 6.4 Hz, CH=CHCH₃-7), 2.18 (1H, dd, J = 1.5, 11.2 Hz, H-8), 2.56 (3H, s, N⁵-CH₃), 2.65 (1H, dd, J = 4.4, 11.2 Hz, H-8), 2.97 (1H, br dd, J = 1.0, 1.5, 4.4, 8.3 Hz, H-7), 3.37 (3H, s, N³-CH₃), 4.93 (1H, ddq, J = 15.1, 8.3, 1.5 Hz, CH=CHCH₃-7), 5.12 (1H, ddq, J = 1.0, 15.1, 6.4 Hz, CH=CHCH₃-7). MS m/z (%): 206 (M⁺, 17), 205 (30), 191 (100), 134 (58), 120 (31), 106 (47). HRMS: Calcd for C₁₁H₁₄N₂O₂: 206.1055. Found: 206.1024.

3,5-Dimethyl-7-*cis*-**prop-1**-**enyl-3,5-diazabicyclo**[**4.2.0**]**oct-1**(**6**)-**ene-2,4-dione** (**3h**-*c*) -----Colorless oil. ¹H-NMR (C₆D₆) δ : 1.24 (3H, dd, J = 1.7, 7.0 Hz, CH=CHC<u>H</u>₃-7), 2.18 (1H, dd, J = 2.0, 11.2 Hz, H-8), 2.45 (3H, s, N⁵-CH₃), 2.68 (1H, dd, J = 4.2, 11.2 Hz, H-8), 3.32 (1H, br dd, J = 1.1, 2.0, 4.2, 8.6 Hz, H-7), 3.38 (3H, s, N³-CH₃), 4.92 (1H, ddq, J = 10.7, 8.6, 1.7 Hz, C<u>H</u>=CHCH₃-7), 5.26 (1H, ddq, J = 1.1, 10.7, 7.0 Hz, CH=C<u>H</u>CH₃-7).

12,14-Dimethyl-12,14-diazatetracyclo[7.6.0.0.², ⁷0.¹⁰, ¹⁵]pentadeca-2(3),4,6,10(15)-tetraene-11,13------ Colorless 175-176 °C (recrystallized dione(3i) crystals; mp from methanol). ¹H-NMR (C₆D₆) δ : 2.58 (3H, s, N¹⁴-CH₃), 2.70 (1H, dd, J = 9.2, 17.2 Hz, H-8), 3.00 (1H, dd, J = 2.2, 17.2 Hz, H-8), 3.23 (3H, s, N¹²-CH₃), 3.41 (1H, ddd, J = 2.2, 3.7, 9.2 Hz, H-9), 3.83 (1H, d, J = 3.7 Hz, H-1), 6.78 (1H, br d, *J* = 7.1 Hz, H-3), 6.90 (1H, br d, *J* = 7.1 Hz, H-6), 6.94-7.04 (2H, m, H-4), 6.94-7.04 (2H, m, H-5). NOE δ: (irradiation at N¹⁴-CH₃) 6.78 (H-3), 3.83 (H-1), (irradiation at H-3) 6.94-7.04 (H-4), 2.58 (N¹⁴-CH₃), 3.83 (H-1), (irradiation at H-8) 3.00 (H-8), 3.41 (H-9), (irradiation at H-8) 2.70 (H-8), 6.90 (H-6). ¹³C-NMR (C₆D₆) δ : 27.79 (N¹²-CH₃), 30.08 (N¹⁴-CH₃), 32.62 (C-8), 41.26 (C-9), 54.93 (C-1), 110.30 (C-10), 124.98 (C-3), 126.70 (C-4), 127.33 (C-6), 128.40 (C-5), 138.07 (C-7), 145.46 (C-2), 152.92 (C-13), 157.79 (C-15), 158.53 (C-11). MS *m/z* (%): 254 (M⁺, 100), 197 (35), 169 (59), 142 (35), 128 (71), 115 (87), 89 (18), 63 (28). HRMS: Calcd for C₁₅H₁₄N₂O₂: 254.1055. Found: 254.1061. Anal. Calcd for C₁₅H₁₄N₂O₂: C, 70.85; H, 5.55; N, 11.02. Found: C, 70.78; H, 5.71; N, 10.97.

Photoreaction of 1 with phenylacetylene (4) ----- Eight tubes of a solution of 1 (13.08 mg, 0.075 mmol) and an olefin (50 equiv. mol; 3.75 mmol, 0.412 mL) in acetone (5 mL) were irradiated externally for 1 h. The reaction mixtures were put together and evaporated *in vacuo*. The residual oil was passed through a short column of silica gel (20 g) with ethyl acetate. The eluate was submitted to HPLC with 3% ethyl acetate in dichloromethane, and separated into a fraction (F1) consisting of **5** and **6**, and a fraction (F2) of **7** and **8**, together with recovered **1** (67.5%). F1 was submitted to HPLC with 5% ethyl acetate in dichloromethane to give **5** (0.6 mg, 0.9%) and **6** (1.0 mg, 1.7%), and submission of the F2 onto HPLC with 1% ethyl acetate in dichloromethane afforded **7** (6.2 mg, 8.5%) and **8** (5.8 mg, 7.2%).⁷

6-Chloro-3,5-dimethyl-7-phenyl-3,5-diazabicyclo[**4.2.0**]**oct-7-ene-2,4-dione** (**5**): Color-less oil.¹H-NMR (C₆D₆) δ : 2.30 or 3.29 (3H, s, N³-CH₃), 2.30 or 3.29 (3H, s, N⁵-CH₃), 3.76 (1H, d, J = 4.0, H-1), 4.92 (1H, d, J = 4.0, H-8), 6.71-6.68 (2H, m, C₆H₅-7), 7.10-7.06 (3H, m, C₆H₅-7). MS m/z (%): 278 (M⁺, 2), 276 (M⁺, 5),

241 (80), 184 (67), 156 (100), 116 (49), 89 (22), 43 (44). HRMS : Calcd for $C_{14}H_{13}N_2O_2Cl$: 276.0666. Found: 276.0677.

6-Chloro-3,5-dimethyl-8-phenyl-3,5-diazabicyclo[**4.2.0**]**oct-7-ene-2,4-dione** (**6**): Color-less oil. ¹H-NMR (C₆D₆) δ: 2.66 or 3.30 (3H, s, N³-CH₃), 2.66 or 3.30 (3H, s, N⁵-CH₃), 5.65 or 5.91 (1H, s, H-1), 5.65 or 5.91 (1H, s, H-7), 6.67-6.97 (2H, m, C₆H₅-8), 6.88-7.00 (3H, m, C₆H₅-8). MS *m/z* (%): 278 (M⁺, 11), 276 (M⁺, 33), 241 (64), 184 (70), 156 (61), 115 (28), 82 (68), 43 (100). HRMS: Calcd for C₁₄H₁₃N₂O₂Cl: 276.0666. Found: 276.0689.

1,3-Dimethyl-7,8-diphenylquinazoline-2,4-dione (**7**): Colorless crystals, mp 223-224 °C (recrystallized from methanol). ¹H-NMR (CDCl₃) & 2.85 (3H, s, N¹-CH₃), 3.51 (3H, s, N³-CH₃), 7.00-7.04 (3H, m, C₆H₅-8), 6.93-6.89 (2H, m, C₆H₅-7), 7.10-7.24 (3H, m, C₆H₅-7), 7.10-7.24 (3H, m, C₆H₅-8), 7.28 (1H, d, J = 8.1 Hz, H-6), 8.28 (1H, d, J = 8.1 Hz, H-5). NOE: irradiation at N¹-CH₃, enhancement observed for C₆H₅-8; irradiation at H-5, enhancement observed for H-6. MS m/z (%): 342 (M⁺, 100), 284 (14), 256 (35), 241 (14), 180 (8), 128 (14), 91 (66). HRMS: Calcd for C₂2H₁₈N₂O₂: 342.1368. Found: 342.1347.

1,3-Dimethyl-5,7-diphenylquinazoline-2,4-dione (**8**): Colorless crystals, mp 269-270 °C (recrystallized from methanol). ¹H-NMR (CDCl₃) δ : 3.53 (3H, s, N³-CH₃), 3.66 (3H, s, N¹-CH₃), 7.22 (1H, s, H-8), 7.30-7.10 (5H, m, C₆H₅-5), 7.30-7.10 (5H, m, C₆H₅-7), 8.29 (1H, s, H-6). NOE: irradiation at N¹-CH₃, enhancement observed for H-8; irradiation at H-6, enhancement observed for C₆H₅-5 and C₆H₅-7. MS m/z (%): 342 (M⁺, 100), 285 (9), 256 (29), 229 (15), 215 (15), 143 (6), 114 (9) HRMS: Calcd for C₂₂H₁₈N₂O₂ : 342.1368. Found: 342.1394.

REFERENCES AND NOTES

- 1. K. Seki, N. Kanazashi, and K. Ohkura, *Heterocycles*, 1991, **32**, 229.
- a) K. Ohkura, N. Kanazashi, and K. Seki, *Chem. Pharm. Bull.*, 1993, **41**, 239; b) *idem, Chem. Lett.*, **1993**, 667; c) K. Seki, K. Ohkura, H. Hiramatsu, K. Aoe, and M. Terashima, *Heterocycles*, 1997, **44**, 467; d) K. Ohkura, Y. Noguchi, and K. Seki, *Chem. Lett.*, **1997**, 99; e) *idem, Heterocycles*, 1997, **46**, 141; f) *idem, ibid.*, 1998, **47**, 429; g) *idem, ibid.*, 1998, **49**, 59.
- 3. C. Kaneko and N. Shimomura, *Tetrahedron Lett.*, 1982, 23, 2571. There are distinct differences in the yield reported for 3 (quantitative) and ours (60%). However, we have not made any more effort to elucidate the reason, since details on the analytical procedure or the reaction conditions were not reported in the literature.
- 4. E. J. Corey, J. D. Bass, R. LeMahieu, and R. B. Mitra, J. Am. Chem. Soc., 1964, 86, 5570; T. Suishu, T. Shimo, and K. Somekawa, *Tetrahedron.*, 1997, 53, 3545, and references therein.
- 5. A. A. Lamola and N. J. Turro, "Technique of Organic Chemistry", Vol. 14, ed. by P. A. Leemakers and A. Weisberger, Wiley-Interscience 1969, p102.
- 6. W. M. Horspool, "Aspects of Organic Photochemistry", Academic Press, New York, 1976. Chapter 4.
- 7. Isolated yields, estimated based on 1 consumed.
- 8. M. P. Cava and M. J. Mitchell," Cyclobutadiene and Related Compounds", Academic Press, New York, 1967.
- 9. D. Andrew, D. J. Hastings, and A. C. Weedon, J. Am. Chem. Soc., 1994, 116, 10870.