

TOWARDS THE SYNTHESIS OF SCYPHOSTATIN

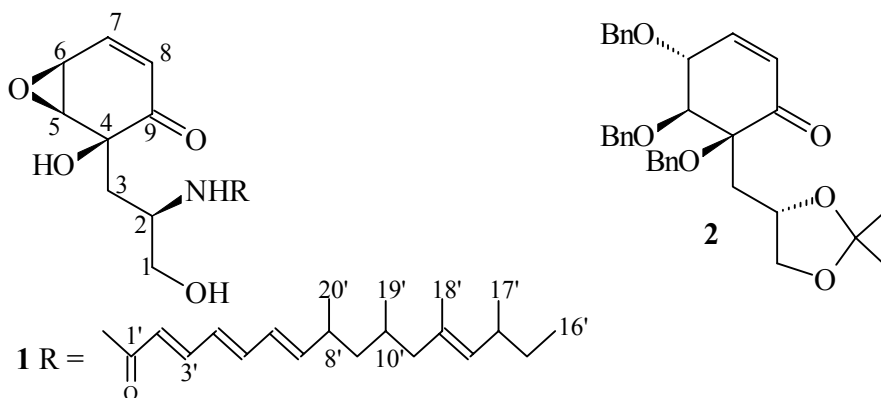
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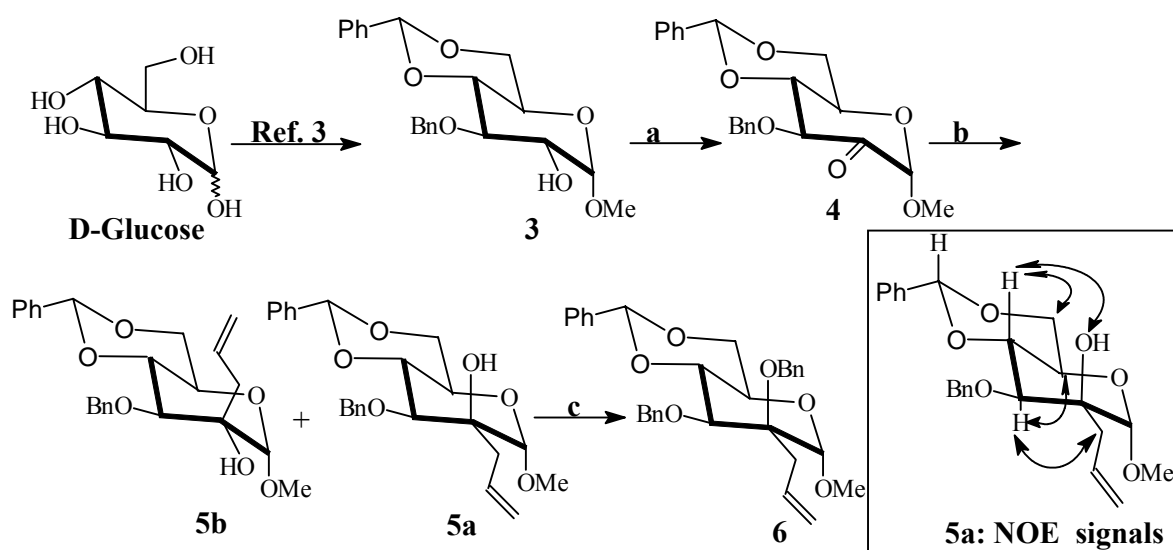
Abstract - Stereocontrolled synthesis of the substituted cyclohexenone segment (2) of scyph-ostatin (1) is described starting from D-glucose.

Scyphostatin (1) is a novel neutral sphingomyelinase (N-Smase) inhibitor isolated¹ during the search for N-Smase inhibitors from the fermentation broth of *Dascysphus mollissimus* SANK-13892. N-Smase inhibitors are useful in the regulation of ceramide level and thus are of immense need in the therapy of autoimmune diseases and inflammation.² Scyphostatin (1) is a structurally unique natural product containing an unprecedented cyclohexenone epoxide moiety in which the *tertiary* chiral center at C-4 is linked to an *n*-propylamino alcohol and a lipophilic side chain[C(1')-C(20')]. The structure of scyphostatin (1) was determined by NMR and MS spectral studies. Chemical and spectroscopic analysis, coupled with modified Mosher's method, provided stereo-



chemical configuration of the cyclohexenone epoxide ring structure of scyphostatin. However, the initial report¹ did not confirm the absolute configuration of the lipophilic side chain. Later Saito *et al.* has confirmed³ the absolute and relative configurations by degradation of 1 and chemical correlation to synthetic fragments. Hoyer *et al.*⁴ have synthesized the lipophilic side chain and till date there is no single report on synthesis of polar cyclohexenone moiety. As part of our interest, we have embarked on developing a carbohydrate based strategy to prepare the suitably substituted cyclohexenone derivative (2) present in the scyphostatin (1). D-Glucose was converted⁵ into methyl 3-*O*-benzyl-4,6-benzylidene- α -D-glucopyranoside (3) whose oxidation at C-2 under

Swern conditions, gave the 2-ulose derivative (**4**) (Scheme 1).⁶ Subsequent C-allylation of **4** was studied⁷ under various conditions. However, we observed that **4** with allylmagnesium bromide in anhydrous ether/CH₂Cl₂ at -78 °C gave a 9:1 diastereomeric mixture of *tertiary* alcohols (**5a**) and (**5b**) which were separated by silica gel column chromatography. The major product was subjected to NOE studies which unequivocally proved the structure as **5a** while the structure of the minor product (**5b**) was unambiguously assigned based on interesting chemical transformations.^{7b}

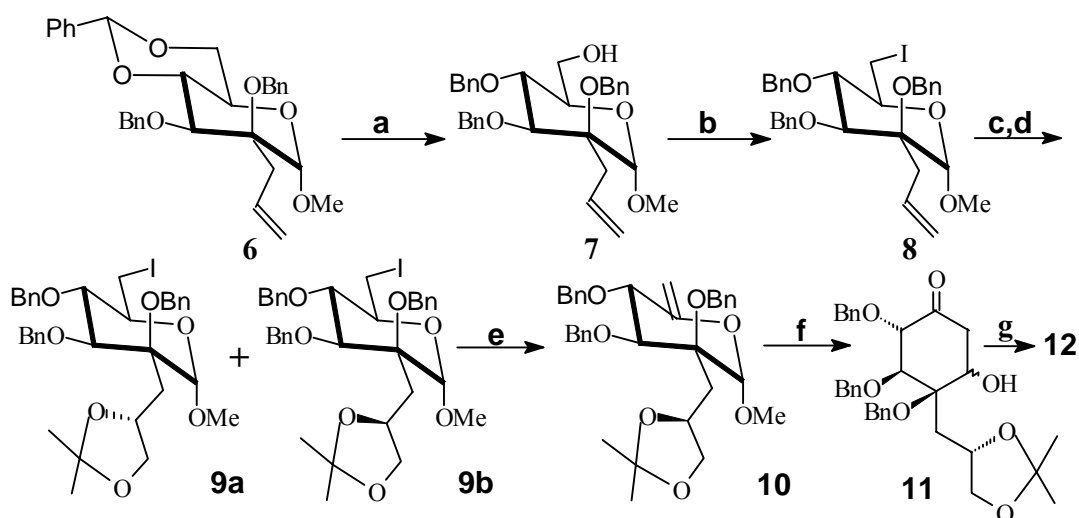


Scheme 1: a) (COCl)₂, DMSO, Et₃N, CH₂Cl₂, -78 °C, 89 %; b) allylmagnesium bromide, Et₂O:CH₂Cl₂ (4:1), -78 °C, 68 %; c) NaH, BnBr, DMF, n-Bu₄NI, 0 °C-rt, 97 %.

The derived *O*-benzyl derivative (**6**) was reduced⁸ with LiAlH₄-AlCl₃ to give rise to the 4-*O*-benzyl derivative (**7**) containing a primary hydroxy group at C-6. Treatment of **7** with Ph₃P-I₂-imidazole in toluene gave⁹ the 6-deoxy-6-iodo derivative (**8**) (Scheme 2). Catalytic osmylation of **8** in *t*-butanol-water gave a 4:1 mixture of diastereomeric products whose acetonide derivatives (**9a/9b**) could be separated by silica gel column chromatography. In accordance with Sharpless modifications,¹⁰ use of AD-MIX- α provided predominantly the diastereomer (**9b**) whose stereochemical assignments were based on hypothesis reported by Sharpless *et al.*¹⁰ Compound (**9b**) was treated with DBU in DMF at 100 °C for 12 h to give the 5,6-ene derivative (**10**). The Ferrier rearrangement¹¹ of **10** in the presence of Hg(OAc)₂ in aqueous acetone gave the carbocyclic derivative (**11**) whose dehydration in the presence of MsCl-Et₃N-CH₂Cl₂ provided the α , β -unsaturated derivative (**12**) (Scheme 3). In order to rearrange **12**, into the target structure (**2**), an approach based on Barton's radical opening reaction¹² of α,β -epoxythioimidazolide was investigated (Scheme 3).

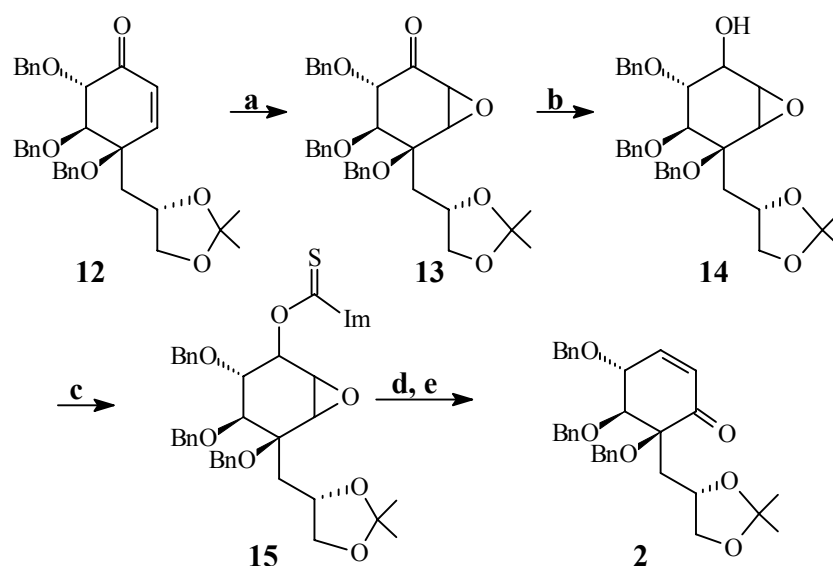
Compound (**12**) was epoxidised with 30 % hydrogen peroxide-K₂CO₃ in THF-H₂O to give **13** which was consequently reduced with NaBH₄ in methanol to give the epoxy alcohol (**14**). In the ¹H-NMR spectrum of **14**, epoxy protons were clearly visible at 3.23 (doublet, J = 3.8 Hz) and 3.35 ppm (triplet, J = 3.8 Hz). Conversion of **14** into the thiocarbonate derivative (**15**) was accomplished with *N,N'*-thiocarbonyldiimidazole in

refluxing C₆H₆. The ¹H-NMR spectrum of **15** showed characteristic downfield shift of the proton carrying



Scheme 2: a) LiAlH₄, AlCl₃, Et₂O, 15 min, 90 %; b) PPh₃, Im, I₂, C₆H₅Me, reflux, 2 h, 95 %; c) AD-mix- α , *t*-BuOH-H₂O (1:1), 0 °C, 24 h, 89 %; d) DMP, PTSA, CH₂Cl₂, rt, 30 min, 98 %; e) DBU, DMF, 100 °C, 12 h, 48 %; f) Hg(OAc)₂, MeCOMe, H₂O (3:1), reflux, 5 h, 80 %; g) MsCl, Et₃N, CH₂Cl₂, rt, 3 h, 85 %

thiocarbonate group. Treatment of **15** with Bu₃SnH and cat. AIBN in refluxing benzene gave a mixture of products from which the major product was isolated (30 %) by silica gel chromatography. Subsequent oxidation with MnO₂ in refluxing CH₂Cl₂ provided the target cyclohexenone derivative (**2**). In the ¹H-NMR spectrum of **2**, the characteristic¹³ signals due to olefinic protons were located at 5.96 (dd, J = 2.04, 10.39 Hz) and 6.85 ppm (dd, J = 1.89, 10.39 Hz). Rest of the spectral data was in conformity with the assigned structure (**2**).¹⁴



Scheme 3: a) H₂O₂, K₂CO₃, H₂O, THF, 0 °C, 6 h, 65 %; b) NaBH₄, MeOH, -40 °C, 15 min, 93 %; c) Im-C(=S)Im, C₆H₆, 80 °C, 12 h, 98 %; d) Bu₃SnH, AIBN, C₆H₆, 80 °C, 4 h, 30 %; e) MnO₂, CH₂Cl₂, 45 °C, 87%.

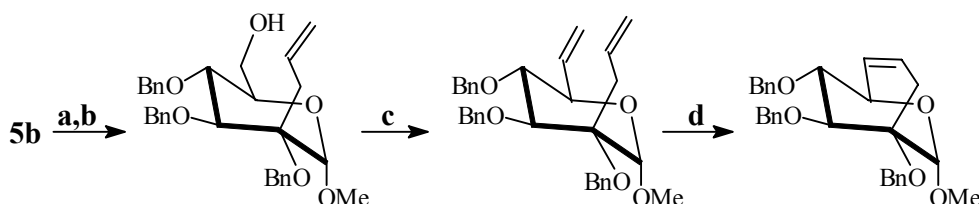
In conclusion we have reported for the first time the stereocontrolled synthesis of a novel and highly substituted cyclohexenone ring system of scyphostatin.

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a) NaH, BnBr, DMF, rt, 90%; b) LiAlH₄, AlCl₃, Et₂O, CH₂Cl₂, 15 min, 95%; c) (i) (COCl)₂, DMSO, Et₃N, CH₂Cl₂, -78°C, 80%; (ii) Ph₃P=CH₂, THF, 65%; d) Grubbs catalyst, C₆H₅Me, reflux, 3 days, 45%.

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14. Structures of new compounds were established by ^1H NMR, FABMS/HRMS or microanalysis. The ^1H NMR spectral data of some key intermediates are given below:

5a (400 MHz, CDCl_3): 2.29 (dd, 1 H, $J = 7.6, 15.3$ Hz), 2.57 (dd, 1 H, $J = 3.8, 15.3$ Hz), 3.35 (s, 3 H), 3.65 (d, 1 H, $J = 7.7$ Hz), 3.83 (m, 2 H), 4.10 (m, 1 H), 4.32 (m, 1 H), 4.40 (s, 1 H), 4.65, 4.95 (ABq, 2 H, $J = 15.6$ Hz), 5.10 (m, 2 H), 5.60 (s, 1 H), 5.85 (m, 1 H), 7.30 (m, 10 H); **8** (200 MHz, CDCl_3): 2.27 (dd, 1 H, $J = 9.0, 14.5$ Hz), 3.09 (dd, 1 H, $J = 4.5, 14.5$ Hz), 3.27 (dd, 1 H, $J = 6.8, 10.0$ Hz), 3.38 (s, 3 H), 3.45 (m, 2 H), 3.75 (d, 2 H, $J = 4.5$ Hz), 4.52, 4.73 (ABq, 2 H, $J = 10.2$ Hz), 4.61, 4.79 (ABq, 2 H, $J = 11.4$ Hz), 4.65, 4.88 (ABq, 2 H, $J = 10.9$ Hz), 4.72 (s, 1 H), 5.11 (m, 2 H), 5.81 (m, 1 H), 7.30 (m, 15 H); **9** (200 MHz, CDCl_3): 1.29, 1.36 (2s, 6 H), 1.70 (dd, 1 H, $J = 9.0, 15.9$ Hz), 2.77 (dd, 1 H, $J = 4.3, 15.9$ Hz), 3.29 (dd, 1 H, $J = 8.1, 11.3$ Hz), 3.40 (s, 3 H), 3.47 (t, 3 H, $J = 9.0$ Hz), 3.75 (m, 2 H), 3.95 (dd, 1 H, $J = 6.8$ Hz), 4.20 (m, 1 H), 4.45 (s, 1 H), 4.63, 4.79 (ABq, 2 H, $J = 11.5$ Hz), 4.70, 4.90 (ABq, 2 H, $J = 11.2$ Hz), 4.75, 5.06 (ABq, 2 H, $J = 11.5$ Hz), 7.30 (m, 15 H); **12** (200 MHz, CDCl_3): 1.15, 1.31 (2s, 6 H), 2.04 (m, 1H), 2.18 (dd, 1 H, $J = 2.3, 15.9$ Hz), 3.25 (m, 1 H), 3.56 (m, 2 H), 3.88 (d, 1 H, $J = 9.0$ Hz), 4.54, 4.77 (ABq, 2 H, $J = 10.5$ Hz), 4.61, 5.00 (ABq, 2 H, $J = 11.4$ Hz), 4.67, 5.11 (ABq, 2 H, $J = 11.4$ Hz), 4.68 (d, 1 H, $J = 9.0$ Hz), 6.04 (d, 1 H, $J = 9.0$ Hz), 6.90 (d, 1 H, $J = 9.0$ Hz), 7.30 (m, 15 H); **2** (500 MHz, CDCl_3): 1.22, 1.30 (2s, 6 H), 2.05 (dd, 1 H, $J = 5.0, 13.8$ Hz), 2.69 (dd, 1 H, $J = 7.5, 13.7$ Hz), 3.53 (t, 1 H, $J = 7.7$ Hz), 3.98 (d, 1 H, $J = 7.4$ Hz), 4.00 (t, 1 H, $J = 7.6$ Hz), 4.14 (m, 1 H), 4.42, 4.60 (ABq, 2 H, $J = 11.6$ Hz), 4.78 (s, 2 H), 4.75, 5.13 (ABq, 2 H, $J = 11.6$ Hz), 4.77 (ddd, 1 H, $J = 1.9, 2.1, 7.7$ Hz), 5.96 (dd, 1 H, $J = 2.1, 10.4$ Hz), 6.85 (dd, 1H, $J = 1.89, 10.4$ Hz), 7.3 (m, 15 H).