

PRENYLATED ISOFLAVONES FROM *DERRIS SCANDENS*

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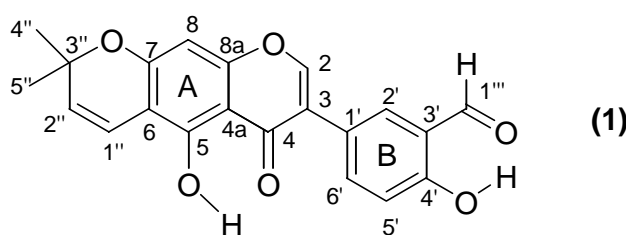
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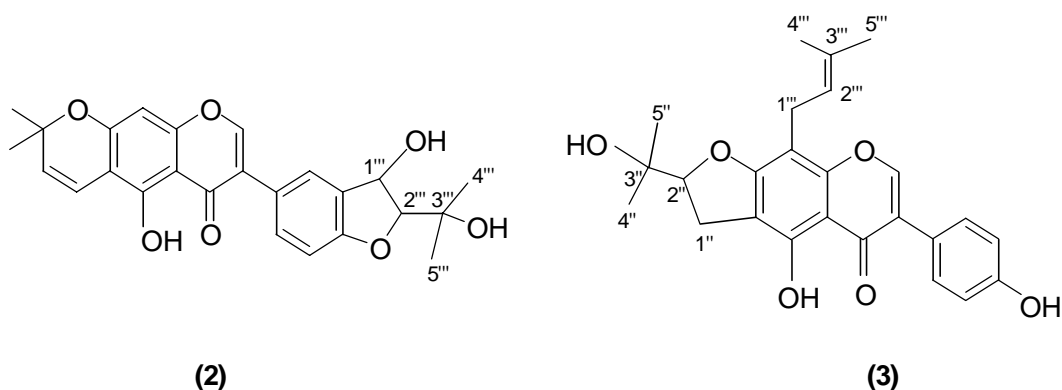
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**Abstract-** 3'-Formylalpinumisoflavone (**1**), and 2-(1-hydroxy-1-methylethyl)-3-hydroxy-2,3-dihydrofuranoalpinumisoflavone (**2**) were isolated from the stem of *Derris scandens* along with five known isoflavones. The two new compounds were characterised by extensive use of high resolution NMR spectroscopy. Senegalensin (**3**) was reported here for the first time in *D. scandens*

*D. scandens* (Leguminosae) stem has been widely used as Thai traditional medicine for relieving muscular pain and diuretic.<sup>1</sup> Hypotensive,<sup>2</sup> immunostimulating activities<sup>1</sup> and smooth muscle stimulant<sup>3</sup> were also evaluated. Investigations of the plant from various sources obtained numerous isoflavones.<sup>4-8</sup> However, rotenoids were isolated from our earlier investigation of the chemical constituents of the *Derris* species.<sup>9</sup>

We recently examined *D. scandens* stem collected from Chantaburi Province, Thailand during 1997 and reported here two new prenylated isoflavone, 3'-formylalpinumisoflavone (**1**), 2-(1-hydroxy-1-methylethyl)-3-hydroxy-2,3-dihydrofuranoalpinumisoflavone (**2**) along with revised chemical shifts senegalensin (**3**). Structures of the known compounds were principally deduced from NMR spectroscopic data and by comparison with those found in *Lupinus albus* roots<sup>11</sup> and previous reports.<sup>7,8,10</sup> They were identified as lupalbigenin,<sup>8</sup> lupinisoflavone G,<sup>8</sup> lupinisol A<sup>8</sup> and 5,7,4'-trihydroxy-6,8-diprenylisoflavone.<sup>8</sup>





**3'-Formylalpinumisoflavone (1)** was a colorless solid (mp 205-208°C) with molecular formula  $C_{21}H_{16}O_6$  determined by HREIMS ( $m/z = 364.0945$ ). The  $^1H$  NMR spectrum revealed a characteristic singlet signal ( $\delta$  7.84, H-2) of isoflavone,<sup>12</sup> 2,2-dimethylpyrano ring ( $\delta$  1.48, s, 6H, 2xMe; two olefinic protons at  $\delta$  5.64 d,  $J = 10$  Hz and 6.73 d,  $J = 10$  Hz) associated with a singlet H-8 at  $\delta$  6.36. Chemical shifts of H-2' ( $\delta$  7.82, d,  $J = 1.2$  Hz), H-6' ( $\delta$  7.66, dd,  $J = 8.8, 1.2$  Hz) and H-5' ( $\delta$  7.08, d,  $J = 8.8$  Hz) were deduced as three aromatic protons having *m*-, *o/m*- and *o*-coupling of B-ring. Moreover, the location of two hydroxy protons at C-5 and C-4' which nearby carbonyl groups suggested magnetically anisotropic down field shifted signals at  $\delta$  12.97 and  $\delta$  11.10 respectively, indicating intramolecular hydrogen bondings in both cases. The IR spectrum displayed strong absorption at  $1650\text{ cm}^{-1}$  which indicated the presence of  $\alpha,\beta$ -unsaturated carbonyl group along with downward shifted absorption at  $3200\text{ cm}^{-1}$  due to the chelated hydroxy group. An intense MS ion at  $m/z$  349 belongs to ion  $[M-15]^+$  which was typically found in pyranoisoflavone due to loss of methyl group.<sup>10</sup> The  $^{13}C$  NMR spectrum indicated the presence of 15 distinct carbon resonances of the isoflavone moiety ( $\delta$  95.0-161.6) including a carbonyl carbon (C-4) at  $\delta$  180.4. Carbon resonances of 2,2-dimethylpyrano ring were also apparent from olefinic  $^{13}C$  NMR signals ( $\delta$  115.3, C-1'' and  $\delta$  128.4, C-2'') associated with a quaternary carbon ( $\delta$  78.2, C-3'') and two methyl carbons ( $\delta$  28.3, C-4'', 5''). The location of the ring was assigned at C-6 and C-7 by HMBC experiment which revealed correlations of olefinic H-1'' ( $\delta$  6.73) with C-5 ( $\delta$  156.8), C-6 ( $\delta$  105.8) and C-7 ( $\delta$  159.8) (Figure 1). Olefinic and aromatic carbons of the isoflavone moiety contained a total of 14 resonances according to the HMQC and DEPT spectra. These were 5 tertiary carbons ( $\delta$  95.0, C-8;  $\delta$  118.1, C-5',  $\delta$  137.2, C-6';  $\delta$  134.2, C-2'  $\delta$  152.6, C-2) and 9 quaternary carbons ( $\delta$  122.2, C-1';  $\delta$  107.5, C-4a;  $\delta$  156.8, C-5;  $\delta$  105.8, C-6;  $\delta$  159.8, C-7; 157.2, C-8a; 122.6, C-3;  $\delta$  120.5, C-3';  $\delta$  161.6, C-4'). The placement of the formyl group at C-3' was confirmed by the HMBC spectrum, in which the formyl proton H-1''' ( $\delta$  9.96) showed long range correlation with C-2' ( $\delta$  134.2) and C-4' ( $\delta$  161.6) in agreement with correlation

of H-2' ( $\delta$ 7.82) with intense signal of the formyl carbonyl C-1''' ( $\delta$  196.5) (Figure 1). Therefore, the structure of **1** was deduced as 3'-formylalpinumisoflavone.

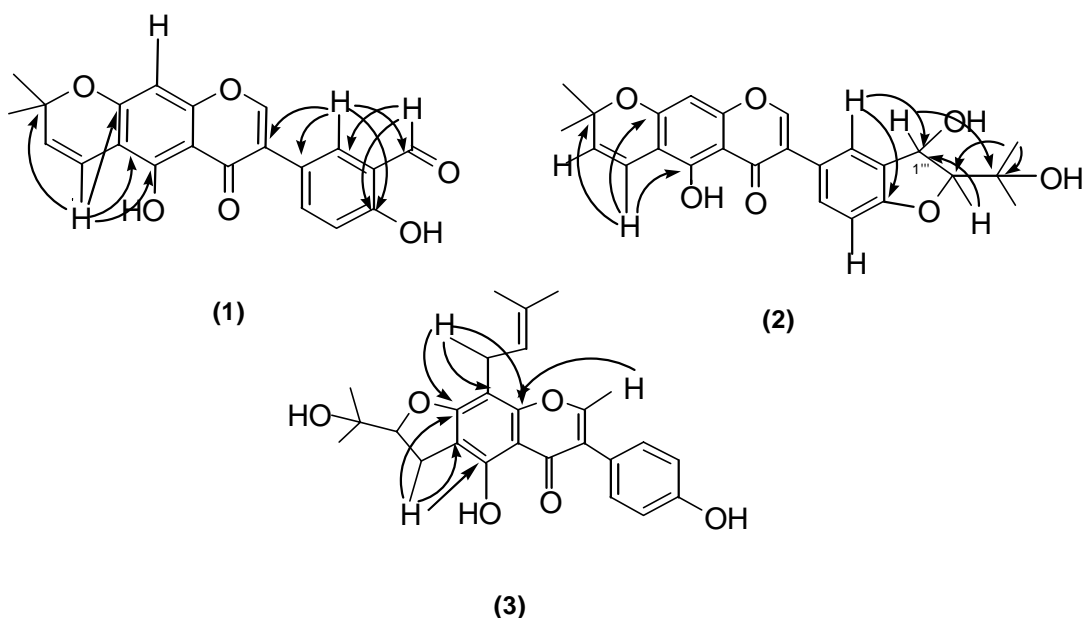


Figure 1. Important H-C correlations observed in HMBC spectra of (1), (2) and (3)

**2-(1-Hydroxy-1-methylethyl)-3-hydroxy-2,3-dihydrofuranoalpinumisoflavone (2)** was found as colorless gum and showed the location of 2,2-dimethylpyranoisoflavone moiety in  $^1\text{H}$  NMR,  $^{13}\text{C}$  NMR, DEPT and HMQC spectra like those of compound (1) thus, only B-ring substitution remain to be established. Three aromatic protons of the B-ring assembled the same pattern *m*-, *o/m* and *o*-coupling as compound (1) with coupling constants 8.3 (*ortho*) and 1.9 (*meta*) Hz in  $^1\text{H}$  NMR spectrum. Chemical shifts of H-1''' ( $\delta$  5.42, br d,  $J = 4.6$  Hz), H-2''' ( $\delta$  4.34, d,  $J = 4.8$  Hz), HO-1''' ( $\delta$  2.50, br s) and H-4''' ,5''' ( $\delta$  1.31, 1.37, s, 6H) were deduced as dihydrofuran ring with 2-(1-hydroxy-1-methylethyl) and 3-hydroxy substituted groups. The position of H-1''' and H-2''' signals were confirmed from the COSY spectrum in which they demonstrated vicinal couplings of the H-1''' to both H-2''' and HO-1''' protons along with 3-bonds correlation of H-1''' with C-3''' carbon in HMBC spectrum. As evidenced from the 3-bond correlation between aromatic H-2' ( $\delta$  7.54) proton of the B-ring and the C-1''' ( $\delta$  73.4) of dihydrofuran ring indicated the location of the dihydrofuran ring at 3' and 4' positions of the B-ring (Figure 1).

**Senegalensin (3)** showed most of the  $^1\text{H}$  and  $^{13}\text{C}$  signals very similar to previous report.<sup>13</sup> Careful examination of the COSY, HMQC and HMBC spectra led to some revision of the assignments as showed in the EXPERIMENTAL.

## EXPERIMENTAL

**General :** Mps were uncorrected. Analytical thin-layer separation was carried out on Merck pre-coated silica gel plates (F-254; layer thickness 0.25 mm). Silica gel 60, particle size (0.063-0.260 mm) and less than 0.063 mm were used in column chromatography. UV spectra were taken on a JASCO Uvidex-650 spectrophotometer. IR spectra were recorded on a JASCO FT-IR 700 spectrophotometer.  $^1\text{H}$  and  $^{13}\text{C}$  NMR spectra were recorded on a Bruker Avance 400 spectrometer at 400 or 100 MHz. EIMS, HREIMS and HRFABMS spectra were determined on a Finnigan MAT 8200 mass spectrometer.

**Plant material:** The white stems of *D. scandens* were collected from Chantaburi Province, Thailand during summer of 1997, the plant was identified by comparison with the herbarium specimen kept at the Forest Herbarium, Royal Forest Department, Bangkok, Thailand. A voucher specimen (MCDS/1997) is kept at the Department of Chemistry, Faculty of Science, Ramkhamhaeng University.

**Extraction and Isolation:** Dried and pulverized white stems (1.6 kg) were extracted successively with n-hexane,  $\text{CHCl}_3$  and MeOH at refluxing temperature in a Soxhlet extractor (5 days each). After evaporation of the solvents under reduced pressure, the n-hexane (42 g),  $\text{CHCl}_3$  (35 g) and MeOH (40 g) extracts were obtained. The extracts were each chromatographed on silica gel column eluting with a gradient of n-hexane,  $\text{CHCl}_3$  and finally MeOH giving combined 3'-formylalpinumisoflavone (**1**) (120 mg), 2-(1-hydroxy-1-methylethyl)-3-hydroxy-2,3-dihydrofuranoalpinumisoflavone (**2**) (32 mg) and senegalensin (**3**) (56 mg).

**3'-Formylalpinumisoflavone (1):** pale yellow solid, mp 205-208°C; IR ( $\text{CHCl}_3$  solution)  $\nu_{\text{max}}$ : 3200 (chelated OH), 1654 (conjugated C=O), 1589, 1488, 1463, 1266  $\text{cm}^{-1}$ ; UV  $\lambda_{\text{max}}$  (MeOH) [ $\log \epsilon$ ]: 289 (2.2) nm;  $^1\text{H}$  NMR ( $\text{CDCl}_3$ , 400 MHz)  $\delta$  1.48 (6H, s, H-4'', 5''), 5.64 (1H, d,  $J=10$  Hz, H-2''), 6.36 (1H, s, H-8), 6.73 (1H, d,  $J=10$  Hz, H-1''), 7.08 (1H, d,  $J=8.8$  Hz, H-5'), 7.66 (1H, dd,  $J=8.8, 1.2$  Hz, H-6'), 7.82 (1H, d,  $J=1.2$  Hz, H-2'), 7.84 (1H, s, H-2), 9.96 (1H, s, formyl H-1'''), 11.10 (1H, s, HO-4') and 12.97 (1H, s, HO-5);  $^{13}\text{C}$  NMR ( $\text{CDCl}_3$ , 100 MHz):  $\delta$  28.3, 28.3 (C-4'', 5''), 78.2 (C-3'''), 95.0 (C-8), 105.8 (C-6), 107.5 (C-4a), 115.3 (C-1''), 118.1 (C-5'), 122.2\* (C-1'), 122.6\* (C-3), 120.5 (C-3'), 128.4 (C-2''), 134.2 (C-2'), 137.2 (C-6'), 152.6 (C-2), 156.8 (C-5), 157.2 (C-8a), 159.8 (C-7), 161.6 (C-4'), 180.4 (C-4), 196.5 (C-1''') \*interchangeable assignments; EIMS  $m/z$  (rel.int.) 364 [ $\text{M}$ ] $^+$  (17), 349 [ $\text{M}-15$ ] $^+$  (100), 350 (20), 321 (2), 203 (3), 174 (3), 61 (2.4); HREIMS  $m/z$  364.0945 [ $\text{M}$ ] $^+$  (calcd for  $\text{C}_{21}\text{H}_{16}\text{O}_6$ , 364.0947).

**2-(1-Hydroxy-1-methylethyl)-3-hydroxy-2,3-dihydrofuranoalpinumisoflavone (2)** colorless gum; IR ( $\text{CHCl}_3$  solution)  $\nu_{\text{max}}$ : 3574 (free OH), 2968, 2909, 1653 (conjugated C=O), 1616, 1583, 1494, 1456, 1249  $\text{cm}^{-1}$ ; UV  $\lambda_{\text{max}}$  ( $\text{CHCl}_3$ ) [ $\log \epsilon$ ]: 285 (6.6) nm;  $^1\text{H}$  NMR ( $\text{CDCl}_3$ , 400 MHz):  $\delta$  1.31, 1.37 (6H, s, H-4''', 5'''), 1.47 (6H, s, H-4'', 5''), 2.50 (1H, br s, HO-1'''), 5.42 (1H, br d,  $J=4.6$  Hz, H-1'''), 4.34 (1H, d,

$J=4.8$  Hz, H-2'''), 5.63 (1H, d,  $J=10$  Hz, H-2''), 6.34 (1H, s, H-8), 6.72 (1H, d,  $J=10$  Hz, H-1''), 6.93 (1H, d,  $J=8.3$  Hz, H-5'), 7.37 (1H, dd,  $J=8.3, 1.9$  Hz, H-6'), 7.54 (1H, d,  $J=1.9$  Hz, H-2'), 7.83 (1H, s, H-2), 13.08 (1H, s, HO-5);  $^{13}\text{C}$  NMR( $\text{CDCl}_3$ , 100 MHz):  $\delta$  24.5, 25.7 (C-4''', 5'''), 28.3, 28.3 (C-4'', 5''), 71.3 (C-3'''), 73.4 (C-1'''), 94.9 (C-8), 97.3 (C-2'''), 105.5 $^\circ$  (C-6), 105.6 $^\circ$  (C-4a), 110.3 (C-5'), 115.4 (C-1''), 123.5\* (C-3), 123.6\*(C-1'), 126.1 (C-2'), 128.2 (C-2''), 129.1 (C-3'), 131.3 (C-6'), 152.6 (C-2), 156.0 (C-5), 156.8 (C-8a), 159.6 (C-7), 160 (C-4'), 180.8 (C-4) \* \* interchangable assignments; EIMS  $m/z$  (rel.int.): 400 (8), 385 (27), 360 (17), 346 (21), 345 (100), 203 (25); HRFABMS  $m/z$ : 437.1604  $[\text{M}+\text{H}]^+$  (calcd for  $\text{C}_{25}\text{H}_{24}\text{O}_7$ , 437.1600).

**Senegalensin (3):** yellow amorphous solid, IR (KBr)  $\nu_{\text{max}}$ : 3357 (free OH), 1651 (conjugated C=O), 1576, 1515, 1215  $\text{cm}^{-1}$ ; UV  $\lambda_{\text{max}}$  (MeOH) [ $\log \epsilon$ ]: 273 (3.9) nm;  $^1\text{H}$ -NMR ( $\text{CDCl}_3$ , 400 MHz):  $\delta$  1.21, 1.31 (6H, s, H-4'', 5''),  $\delta$  1.67, 1.77 (6H, s, H-4''', 5'''),  $\delta$  3.11, 3.25 (2H, dd,  $J_{\text{gem}}=15.7, J_{\text{vic}}=9.4, 7.8$  Hz, H-1''),  $\delta$  3.37 (2H, d,  $J=6.9$  Hz, H-1'''),  $\delta$  4.77 (1H, dd,  $J=9.4, 7.8$  Hz, H-2''),  $\delta$  5.20 (1H, t-like,  $J=6.9$ , H-2'''),  $\delta$  6.83 (2H, d,  $J=8.5$  Hz, H-3', 5'),  $\delta$  7.38 (2H, d,  $J=8.5$  Hz, H-2', 6'),  $\delta$  7.84 (1H, s, H-2),  $\delta$  12.95 (1H, s, HO-5);  $^{13}\text{C}$  NMR ( $\text{CDCl}_3+\text{CD}_3\text{OD}$ , 100 MHz):  $\delta$  17.7, 25.6 (C-4''', 5'''), 21.9 (C-1'''), 24.2, 25.0 (C-4'', 5''), 27.0 (C-1''), 71.7 (C-3'''), 91.2 (C-2''), 102.5 (C-8), 106.7 (C-4a), 108.5 (C-6), 110.6 (C-3', 5'), 121.6 (C-2'''), 122.2 (C-1'), 123.8 (C-3), 130.5 (C-2', 6'), 132.3 (C-3'''), 152.5 (C-2), 153.4 (C-5), 154.0 (C-8a), 157.0 (C-4'), 164.2 (C-7), 181.1 (C-4); EIMS  $m/z$  (rel.int.): 422  $[\text{M}]^+$  (100), 420 (35), 407  $[\text{M}-\text{Me}]^+$  (21), 405  $[\text{M}-\text{OH}]^+$  (24), 389  $[\text{M}-\text{H}_2\text{O}-\text{Me}]^+$  (34), 363  $[\text{M}-59]^+$  (19), 349 (71), 335 (35); HRFABMS  $m/z$ : 423.1808  $[\text{M}+\text{H}]^+$  (calcd for  $\text{C}_{25}\text{H}_{27}\text{O}_6$ , 423.1808).

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#### REFERENCES

1. B. Sriwanthana and P. Chavalittumrong, *J. Ethnopharmacology*, 2001, 125 and references cited.
2. C. Jansakul, A. Srichanbarn, and A. Saelee, *J. Sci. Soc. Thailand*, 1997, **23**, 323.
3. M. Mokkaasmit, W. Ngarmwathana, K. Sawasdimongkol, and U. Permpiphat, *J. Med. Ass. Thailand*, 1971, **54**, 490.
4. C. P. Falshaw, R. A. Harmer, W. D. Ollis, R. E. Wheeler, V. R. Lalitha, and N. V. Subba Rao, *J. Chem. Soc. (C)*, 1969, 374.
5. A. Pelter and P. Stainton, *J. Chem. Soc. (C)*, 1966, 701.

6. A. P. Johnson, A. Pelter, and P. Stainton, *J. Chem. Soc. (C)*, 1966, 192.
7. M. N. Rao, G. L. D. Krupadanam, and G. Srimannarayana, *Phytochemistry*, 1994, **37**, 267 and references cited.
8. T. Sekine, M. Inagaki, F. Ikegami, Y. Fujii, and N. Ruangrunsi, *Phytochemistry*, 1999, **52**, 87.
9. N. Thasana, M. Chuankamnerdkarn, and S. Ruchirawat, *Heterocycles*, 2001, **55**, 1121.
10. A. K. Singhal, R. P. Sharma, G. Thyagarajan, W. Herz, and S. V. Govindan, *Phytochemistry*, 1980, **19**, 929.
11. S. Tahara, S. Orihara, J. L. Ingham, and J. Mizutani, *Phytochemistry*, 1989, **28**, 901.
12. K. R. Markham and T. J. Mabry, "The Flavanoids" ed. by J. B. Harborne, T. J. Mabry, and H. Mabry, Chapman & Hall, London, 1975, p. 62.
13. J. Wandji, E. Augustin, Z. Nkengfack, T. Fomum, R. Ubillas, K. B. Killday, and M. S. Tempesta, *J. Nat. Prod.*, 1990, **53**, 1425.