HETEROCYCLES, Vol. 59, No. 1, 2003, pp. 97 - 100, Received, 31st July, 2002 REGIOCONTROL IN Pd-CATALYZED ALLYLIC SUBSTITUTION WITH P,N-LIGAND

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Dedicated to Professor Yuichi Kanaoka in recognition of his outstanding contributions to the area of heterocyclic chemistry.

Abstract – A racemic ligand, (\pm) -*N*-[2-(diphenylphosphino)naphthyl]-2-(pyrrolidinylmethyl)piperidine, has been found to exhibit better regioselectivity than the racemic Pfaltz ligand, BINAP and dppp in the palladium-catalyzed regioselective allylic substitution with the use of (*E*)-1-phenyl-4-alkoxy-2-butenyl acetate as the substrate.

In the course of our studies in relation to synthetic organic chemistry based on the N-Ar axis, ^{1,2} we have developed a novel chiral ligand (1) mimicking N-Ar axial chirality on the following concept: ³ our chiral ligand (1) may have the two diasteromers due to the N-Ar axis and the chiral carbon in solution (Figure 1). But, if the N-Ar axis in (1) is configurationally flexible, and the complexation of (1) and a metal is largely reflected by the asymmetric center of pyrrolidinyl group A in (1), one of the two diastereomer complexes due to the N-Ar axis, is expected to be selectively formed (2 in Figure 1), leading to the creation of favorable asymmetric environment.



This designed ligand (1) has been found to exhibit 99% ee with the use of (*E*)-1,3-diphenyl-2-propenyl acetate as the standard substrate, and has achieved slightly better results than the representative Pfaltz⁴ and Trost⁵ ligands with the use of (*E*)-1-phenyl-3-trimethylsilyl-2-propenyl acetate as the substrate, in the

palladium-catalyzed asymmetric allylic substitution with dimethyl malonate as the pronucleophile.³ In order to explore further application of the ligand (1), we planned to search for a suitable achiral reaction with the racemic ligand ((\pm)-1) as the initial goal with the aim of application to asymmetric reaction. We concentrated on a palladium-catalyzed regioselective allylic substitution, because the lack of regioselectivity is often a problem⁶. In this paper, we would like to report our investigation of this reaction with the racemic ligand ((\pm)-1).

(*E*)-1-Phenyl-4-methoxy-2-butenyl acetate (**3a**) and (*E*)-1-phenyl-4-silyloxy-2-butenyl acetates (**3b-d**) were chosen as substrates. The substrate (**3a**) was prepared as shown in Scheme 1. Alkynylation of benzaldehyde (**4**) with the lithium acetylide (**5**) in THF at 0 $^{\circ}$ C afforded the coupling product (**6**). The reduction⁷ of the triple bond in (**6**) with LiAlH₄ and NaOMe in THF at 0 $^{\circ}$ C and subsequent acetylation of the OH group afforded the substrate (**3a**) in 59% yield (2 steps). Likewise, other substrates (**3b-d**) were prepared. The representative preparation of (**3d**) is shown in Scheme 2. Alkynylation of aldehyde (**4**) with the lithium acetylide (**7**) and subsequent deprotection of the *tert*-butyldiphenylsilyl (TBDPS) group afforded the coupling product (**8**) in 75% yield. After reduction of the triple bond in (**8**), chemoselective silylation and subsequent acetylation afforded the substrate (**3d**) in 66% yield (**3** steps).



With the four substrates available, the regioselective allylic substitution was examined. The results are shown in Table 1. As can be seen in entry 1, treatment of the substrate (**3a**) with $[PdCl(\eta^3-C_3H_5)]_2$ (2.5 mol%), the racemic ligand ((±)-1) (5 mol%), dimethyl malonate (3 equiv), *N*, *O*-bis(trimethylsilyl)acetamide⁸ (BSA) (3 equiv) and KOAc (5 mol%) in DMF at rt afforded (**9a**) in 60% yield together with (**10a**) (39%).^{9,10} Nucleophilic attack took place preferentially at the phenyl-substituted side. Reducing the reaction temperature from room temperature to 0 °C increased slightly the regioselectivity (**9a**: 65%, **10a**: 24%, entry 2). Contrary to this result, in the case of the racemic Pfaltz ligand (**11**), BINAP¹¹ (**12**) and dppp (**13**), no regioselectivity was observed (entries 3-5). This better regioselectivity with our ligand ((±)-1) might be due to low reactivity¹² of the π -allyl intermediate, compared with the ligands (**11-13**), although further investigation is needed to understand the effects of

these ligands. Furthermore, reaction of the substrates (**3b-d**) bearing a sterically bulky silyl group, afforded better regioselectivity than that of the substrate (**3a**) (enrties 6-8). Among them, the substrate (**3d**) bearing a TBDPS group, gave the best selectivity (entry 8).

PI	OAc 3a: R=M 3b: R=TI 3c: R=TE	OAc OR a: R=Me o: R=TMS c: R=TBDMS		$[PdCl(\eta^{3}-C_{3}H_{5})]_{2}$ (2.5 mol%) Ligand (5 mol%) CH ₂ (COOMe) ₂ KOAc, BSA, DMF		CH(CO ₂ Me) ₂ Ph 9a: R=Me 9b: R=TMS 9c: R=TBDMS			(10a: F 10b: F 10c: F	CH(CO ₂ Me) ₂ OR R=Me R=TMS R=TBDMS
_	JU. KEIBUPS				9	u . R=160	F 3		10d: F	
	Entry Suba		strate R=)	Ligand	Tem	ıp (h	ne Y 1) 9	íeld ((%) 10	
_	1	1 3a (R:		1	rt	1	6	0	39	
	2	3a (R:	=Me)	1	0 °C	; 3	3 6	5	24	
	3	3a (R:	=Me)	11	rt	1	4	8	47	
	4	3a (R:	=Me)	12	rt	2	<u>2</u> 4	4	48	
	5	3a (R:	=Me)	13	rt	48	3 4	3	41	
	6	3b (R=	TMS)	1	rt	2	2 6	0	20	
	7	3c (R=T	BDMS)	1	rt	2	2 6	6	22	
	8	3d (R=T	BD PS)	1	rt	48	3 7	2	15	
<i>i</i> -Pr					PPh ₂ PPh ₂				PPh ₂ PPh ₂	
	racemic	a cemic Pfaltz ligand 11			racemic BINAP 12			dppp		

Table 1. Regioselective Allylic Substitution

In conclusion, we have found that our racemic ligand $((\pm)-1)$ exhibits better regioselectivity than the racemic Pfaltz ligand (11), BINAP (12) and dppp (13). Asymmetric regioselective allylic substitution with (6a-d) catalyzed by the Pd/chiral ligand¹³ (1) complex is now in progress in our laboratory.

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REFERENCES AND NOTES

- 1. For a review, see: K. Kondo and Y. Murakami, J. Synth. Org. Chem., Jpn., 2001, 59, 866.
- 2. Y. Murakami, K. Kondo, K. Miki, Y. Akiyama, T. Watanabe, and Y. Yokoyama, Tetrahedron Lett.,

1997, **38**, 3751; K. Kondo, T. Kurosaki, and Y. Murakami, *Synlett*, 1998, 725; K. Kondo, H. Fujita, T. Suzuki, and Y. Murakami, *Tetrahedron Lett.*, 1999, **40**, 5577; K. Kondo, E. Sekimoto, J. Nakao, and Y. Murakami, *Tetrahedron*, 2000, **56**, 5843; K. Kondo, T. Iida, H. Fujita, T. Suzuki, K. Yamaguchi, and Y. Murakami, *Tetrahedron*, 2001, **57**, 4115, and references cited therein.

- 3. K. Kondo, K. Kazuta, H. Fujita, Y. Sakamoto, and Y. Murakami, *Tetrahedron*, 2002, 58, 5209.
- 4. P. von Matt and A. Pfaltz, Angew. Chem., Int. Ed. Engl., 1993, 32, 566.
- 5. B.M. Trost, D.L. Van Vranken, and C. Bingel, J. Am. Chem. Soc., 1992, 114, 9327.
- 6. B.M. Trost and D.L. Van Vranken, *Chem. Rev.*, 1996, **96**, 395.
- 7. E.J. Corey, J.A. Katzenellenbogen, and G.H. Posner, J. Am. Chem. Soc., 1967, 89, 4245.
- 8. B.M. Trost and S.J. Bricker, J. Am. Chem. Soc., 1983, 105, 568.
- 9. The use of toluene and α, α, α -trifluorotoluene, less polar solvents, gave much lower chemical yields of the alkylated products (9) and (10), due to competitive β -hydrogen elimination from the π -allyl inertmediate.
- 10. The use of diisopropyl malonate in place of dimethyl malonate gave similar regioselectivity.
- 11. R. Noyori and H. Takaya, Acc. Chem. Res., 1990, 23, 345.
- 12. In the case of the racemic ligand ((±)-1), other nucleophiles such as acetylacetone and benzylamine gave low yields of the product. On the other hand, in the case of the Pfaltz ligand (11) and BINAP (12), these nucleophiles gave satisfactory yields.
- For chiral P,N(sp³)-ligands, see: (a) Y. Uozumi and K. Shibatomi, J. Am. Chem. Soc., 2001, 123, 2919. (b) T. Mino, S. Hata, K. Ohtaka, M. Sakamoto, and T. Fujita, *Tetrahedron Lett.*, 2001, 42, 4837. (c) S. Pellet-Rostaing, C. Saluzzo, R.T. Halle, J. Breuzard, L. Vial, F.L. Guyader, and M. Lemaire, *Tetrahedron: Asymmetry*, 2001, 12, 1983. (d) Y. Okuyama, H. Nakano, and H. Hongo, *Tetrahedron: Asymmetry*, 2000, 11, 1193. (e) M.J. Jin, J.-A. Jung, and S-H. Kim, *Tetrahedron Lett.*, 1999, 40, 5197. (f) J.P. Cahill, D. Cunneen and P.J. Guiry, *Tetrahedron: Asymmetry*, 1999, 10, 4157. (g) Y. Suzuki, I. Abe, and K. Hiroi, *Heterocycles*, 1999, 50, 89. (h) M. Bourghida and Widhalm, *Tetrahedron: Asymmetry*, 1998, 9, 1073. (i) T. Hattori, Y. Komuro, N. Hayashizaka, H. Takahashi, and S. Miyano, *Enantiomer*, 1997, 2, 203. (j) P. Wimmer and Widhalm, M. *Tetrahedron: Asymmetry*, 1995, 6, 657. (k) H. Kubota, M. Nakajima, and K. Koga, *Tetrahedron Lett.*, 1993, 34, 8135. (l) B. Vriesema and R. Kellog, *Tetrahedron Lett.*, 1986, 27, 2049. (m) T. Hayashi, M. Fukushima, M. Konishi, and M. Kumada, *Tetrahedron Lett.*, 1980, 21, 79, and references cited therein.