

CONSTITUENTS OF LEAVES OF *PHELLODENDRON* *CHINENSE* VAR. *GLABRIUSCULUM*

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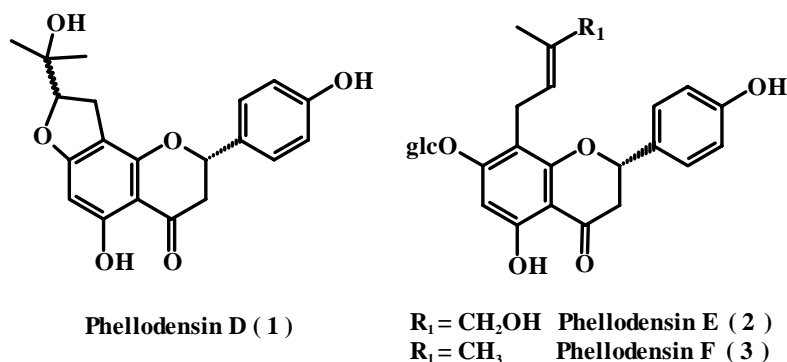
Abstract —The leaves of *Phellodendron chinense* var. *glabriusculum* yielded three new flavanones, phellodensin D (**1**), phellodensin E (**2**), and phellodensin F (**3**) along with five known compounds, amurensin (**4**), phellamurin (**5**), methyl caffeate (**6**), β -sitosterol (**7**), and pheophytin-a (**8**). The structural assignment of new compounds was based on spectroscopic studies.

INTRODUCTION

Phellodendron chinense var. *glabriusculum* (Rutaceae) is a deciduous tree that occurred widely in southwestern China.¹ The bark of the plants of genus *Phellodendron* has found application in Chinese traditional medicine for various diseases like meningitis, bacillary dysentery, pneumonia, tuberculosis, and liver cirrhosis.¹⁻³ Examination of a sample of the leaves of this plant collected in Yunnan, mainland China, has now led to the isolation and characterization of three new flavanones, phellodensin D (**1**), phellodensin E (**2**), and phellodensin F (**3**), as well as five known compounds.

RESULTS AND DISCUSSION

A methanol extract of leaves of *P. chinense* var. *glabriusculum* was partitioned between chloroform and water. Chloroform solubles were column chromatographed on silica gel to give phellodensin D (**1**). Aqueous layer on column chromatography over silica gel afforded phellodensin E (**2**) and phellodensin F (**3**).



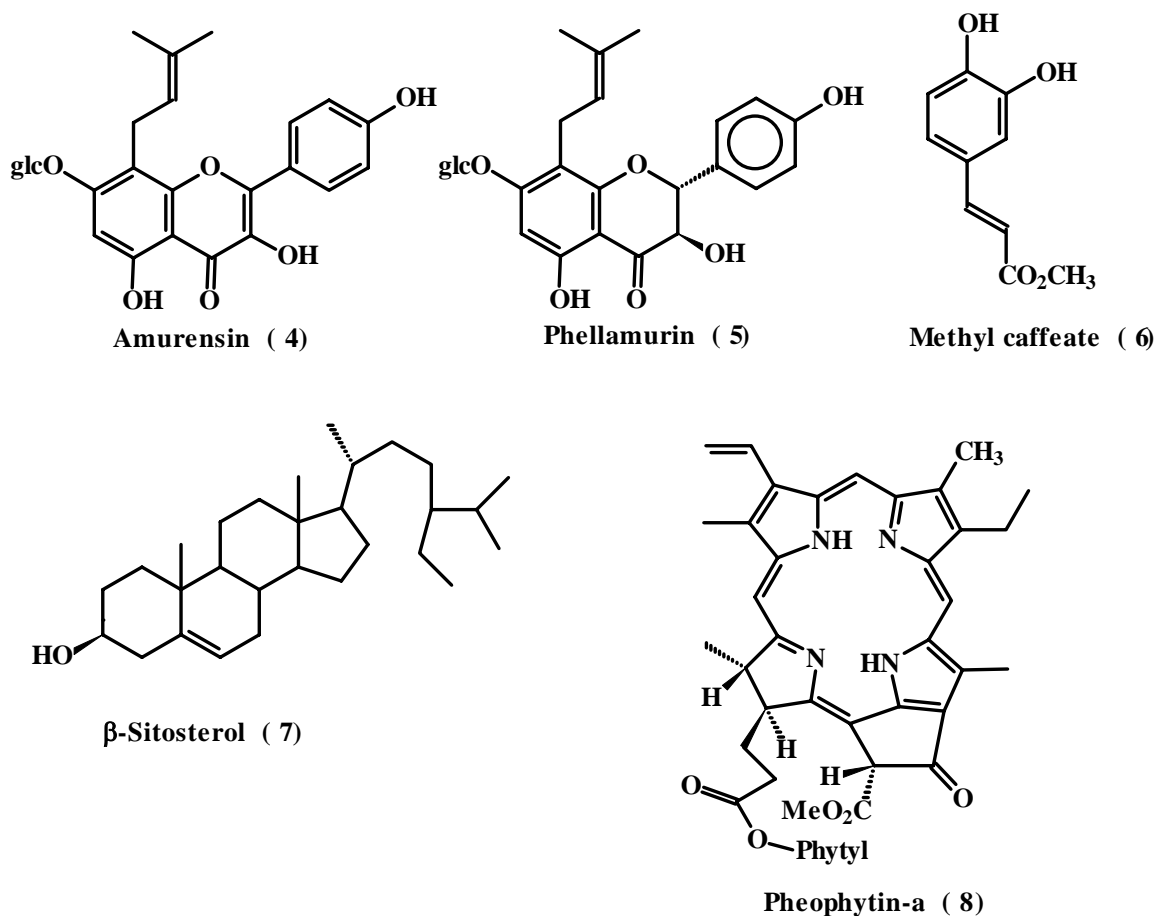
Phellodensin D (**1**) was isolated as yellow amorphous powder and was assigned a molecular formula of C₂₀H₂₀O₆ from its HRFABMS. The IR spectrum showed absorption bands at 3381 and 1647 cm⁻¹, consistent with the presence of a hydroxyl group and a conjugated carbonyl group in the molecule. The UV spectrum exhibited characteristic absorption peaks of the flavanone moiety at 243, 294, and 336 nm.⁴ The presence of a flavanone structure was also indicated by an AMX system with resonances at δ 5.49 (1H, dd, *J* = 12.7, 2.9 Hz, H-2), 3.18 (1H, dd, *J* = 17.2, 12.7 Hz, H-3 α), and 2.75 (1H, dd, *J* = 17.2, 2.9 Hz, H-3 β) and the resonance occurring at δ 196.8 for C-4. In the ¹H NMR and ¹H-¹H COSY spectra, a pair of A₂B₂ doublets at δ 7.39 (2H, *J* = 8.4 Hz, H-2', -6') and 6.89 (2H, *J* = 8.4 Hz, H-3', -5'), and a broad singlet at 8.55 (1H, 4'-OH) were typical of a *p*-hydroxy substituted B-ring. A downfield signal at δ 12.43, exchangeable with D₂O, was attributed to the chelated hydroxyl at C-5. A sharp singlet at δ 5.87 was assigned to H-6 as it showed HMQC correlation with a carbon at δ 91.5 and HMBC correlation with C-5 at δ 165.8. The ¹H NMR signals of **1** at δ 1.18 (3H, s), 1.25 (3H, s), 2.97 (1H, dd, *J* = 15.2, 9.6 Hz), 3.05 (1H, dd, *J* = 15.2, 7.7 Hz), 4.74 (1H, dd, *J* = 9.6, 7.7 Hz), and 3.76 (1H, br s) revealed the presence of an α -(1-hydroxy-1-methylethyl) dihydrofuran moiety in **1**.⁵ The fusion of this group at C-7 and C-8 was confirmed by the ²*J* and ³*J* correlations observed between H-1''/C-7, -8, and -9 in its HMBC spectrum. In the CD spectrum of **1**, characteristic Cotton effects based on n- π^* transition of flavanone, a positive absorption at 313 nm and a negative absorption at 292 nm were observed, from which the absolute

configuration of **1** was deduced as *2S*.⁶ Thus, the structure of phellodensin D was elucidated as **1**.

Phellodensin E (**2**), obtained as yellow amorphous powder, showed $[M+H]^+$ at m/z 519.1865 in its HR-FABMS corresponding to the pseudomolecular formula $C_{26}H_{31}O_{11}$ and a prominent fragment at m/z 341 ($[M+H-162]^+$) indicating the presence of an hexosyl moiety. The IR spectrum of **2** displayed a hydroxyl absorption band at 3368 cm^{-1} and a conjugated carbonyl absorption band at 1630 cm^{-1} . The UV absorption maxima at 226, 287, and 340 nm were typical of a flavanone derivative.⁴ The ^1H NMR spectrum revealed a D_2O exchangeable downfield signal at δ 12.08 (5-OH), a broad singlet at δ 8.49 (4'-OH), an A_2B_2 system of proton signals at δ 7.46 (d, $J = 8.6$ Hz, H-2', -6') and 6.89 (d, $J = 8.6$ Hz, H-3', -5'), a sharp singlet at δ 6.29 (H-6), and an AMX system of protons at δ 5.48 (1H, dd, $J = 12.6, 2.8$ Hz, H-2), 3.17 (1H, dd, $J = 17.1, 12.6$ Hz, H-3 α), and 2.78 (1H, m, H-3 β). Additionally, characteristic signals for 4-hydroxy-3-methyl-2-butenyl group at δ 3.23 (1H, dd, $J = 14.4, 7.6$ Hz, H-1''), 3.41 (1H, d, $J = 14.4, 7.6$ Hz, H-1''), 5.43 (1H, m, H-2''), 3.86 (2H, br s, H-4''), 4.29 (1H, br s, 4''-OH), and 1.64 (3H, s, 5''-CH₃) and an anomeric proton doublet at δ 5.06 ($J = 7.6$ Hz) were also observed. Considering these informations, **2** was thus identified as a flavanone, phellodensin derivative with glucosyl and 4-hydroxy-3-methyl-2-butenyl substituents. The placement of glucose moiety was determined by analysis of HMBC spectrum in which the anomeric proton signal showed long range correlation with the carbon at δ 164.5 (C-7) indicating that the glucosyl moiety is linked to C-7 of **2**. The coupling constant ($J = 7.6$ Hz) of the anomeric proton signal inferred the β -configuration of the glucopyranosyl moiety. The position of 4-hydroxy-3-methyl-2-butenyl group whose *E*-configuration was inferred by the NOESY correlations between H-2'' / H-4'' was deduced to be at C-8 from the HMBC correlations of H-1'' with C-7, 8, and 9. The absolute configuration at C-2 was confirmed as *S* by a positive Cotton effect at 336 nm and a negative Cotton effect at 289 nm in CD spectrum.⁶ Thus the structure of phellodensin E was established as shown in **2**.

Phellodensin F (**3**) was obtained as yellow amorphous powder that was analyzed for the molecular formula $C_{26}H_{30}O_{10}$ from its HRFABMS spectrum. The IR spectrum suggested the presence of hydroxyl group (3416 cm^{-1}) and conjugated carbonyl group (1641 cm^{-1}). The UV spectrum exhibited absorption

maxima at 288 and 344 nm characteristic of a flavanone derivative.⁴ In its ¹H and ¹³C NMR spectra, data corresponding to flavanone and glucosyl moieties were almost superimposable to those of phellodensin E. Also, characteristic prenyl proton signals were observed at δ 3.18 (1H, dd, $J = 13.9, 7.5$ Hz, H-1''), 3.36 (1H, dd, $J = 13.9, 7.5$ Hz, H-1''), 5.20 (1H, t, $J = 7.5$ Hz, H-2''), 1.60 (3H, s, 4''-CH₃), and 1.64 (3H, s, 5''-CH₃). These results indicated that **3** has a flavanone skeleton with two hydroxyl groups, one glucose unit, and one prenyl subunit. The position of these functional groups were determined unambiguously as C-5 and C-4' for two hydroxyls, C-7 for glucosyl, and C-8 for prenyl, respectively, using the HMBC NMR technique. The absolute configuration at C-2 was assigned as *S* by CD spectral data comparison with literature values for a group of flavanones.⁶ Accordingly, the structure of the phellodensin F was elucidated as shown in **3**.



Five known compounds, amurensin (**4**),⁷ phellamurin (**5**),⁷ methyl caffeate (**6**),⁸ β -sitosterol (**7**),⁹ and pheophytin-a (**8**)¹⁰ were also isolated from the leaves of *P. chinense* var. *glabriusculum*. These compounds were identified by comparison with the authentic samples.

EXPERIMENTAL

General Experimental Procedures. Melting points were measured on Yanaco MP-S3 melting point apparatus without correction. UV spectra were recorded on a Hitachi UV-3210 spectrophotometer.

Optical rotations were measured on a Jasco DIP-370 digital polarimeter. IR spectra were recorded on a Shimadzu FT-IR DR-8011 spectrophotometer with KBr discs. ^1H and ^{13}C NMR spectra were determined on Varian Unity plus 400 spectrometer. Chemical shifts are shown in δ values (ppm) with tetramethylsilane (TMS) as internal standard. FAB and HRFABMS spectra were measured on a VG-70-250S spectrometer by a direct inlet system. CD spectra were recorded with Jasco J-720 spectropolarimeter.

Plant Material. The leaves of *P. chinense* var. *glabriusculum* were collected from Kunming, Yunnan, in 1997. A herbarium specimen (Wu 19970009) is deposited at the National Cheng Kung University, Tainan, Taiwan.

Extraction and Isolation. The air-dried leaves of *P. chinense* var. *glabriusculum* (75 g) was ground and extracted with methanol (500 mL \times 6) under reflux for 8 hours to give an extract (8.6 g), which as partitioned between chloroform and water. The chloroform soluble fraction (4 g) was subjected to silica gel column chromatography using chloroform-methanol gradients to give 6 fractions. Fraction 5 was subjected to column chromatography with chloroform to get **7** (1.5 mg) and **8** (2.5 mg). Similarly, fraction 6 gave **1** (8.3 mg) with chloroform-methanol (19:1) solvent system. The water soluble fraction (4.6 g) was chromatographed over reversed-phase Diaion HP-20 using water-methanol gradients which afforded 8 fractions. Fractions 7 and 8 were separately subjected to column chromatography over silica gel with chloroform-acetone-methanol (4:1:1) and chloroform-methanol (9:1) solvent systems to give **2** (13.7 mg), **5** (3.6 mg), and **6** (2.2 mg), and **3** (237.8 mg) and **4** (5.4 mg), respectively.

Phellodensin D (1) $\text{C}_{20}\text{H}_{20}\text{O}_6$: Yellow powder; mp 88-89°C (MeOH); $[\alpha]_{\text{D}}^{25} +35.6^\circ$ (c 0.083, MeOH); UV $\lambda_{\text{max}}^{\text{MeOH}}$ nm (log ϵ): 219 (4.35), 243 (3.93), 294 (4.15), 336 (3.56); IR ν_{max} cm^{-1} : 3381, 1647, 1611, 1384, 1252. ^1H -NMR (400 MHz, Acetone- d_6): δ 1.18 (3H, s, CH_3), 1.25 (3H, s, CH_3), 2.75 (1H, dd, $J = 17.2, 2.9$ Hz, H-3 β), 2.97 (1H, dd, $J = 15.2, 9.6$ Hz, H-1''), 3.05 (1H, dd, $J = 15.2, 7.7$ Hz, H-1''), 3.18 (1H, dd, $J = 17.2, 12.7$ Hz, H-3 α), 3.76 (1H, br s, 3''-OH), 4.74 (1H, dd, $J = 9.6, 7.7$ Hz, H-2''), 5.49 (1H,

dd, $J = 12.7, 2.9$ Hz, H-2), 5.87 (1H, s, H-6), 6.89 (2H, d, $J = 8.4$ Hz, H-3', -5'), 7.39 (2H, d, $J = 8.4$ Hz, H-2', -6'), 8.55 (1H, br s, 4'-OH), and 12.43 (1H, s, 5-OH). $^{13}\text{C-NMR}$ (100 MHz, Acetone- d_6): δ 25.5 (CH₃), 25.8 (CH₃), 27.0 (C-1''), 43.4 (C-3), 71.4 (C-3''), 79.9 (C-2), 91.5 (C-6), 92.7 (C-2''), 103.4 (C-10), 105.8 (C-8), 116.2 (C-3', -5'), 129.0 (C-2', -6'), 130.7 (C-1'), 158.2 (C-4'), 158.7 (C-9), 165.8 (C-5), 169.9 (C-7), and 196.8 (C-4). CD (MeOH: $c = 0.00005$): $[\theta]_{313} +8458$, $[\theta]_{306} 0$, $[\theta]_{292} -33350$, $[\theta]_{268} 0$, $[\theta]_{240} +10110$, $[\theta]_{217} +45230$. FAB-MS m/z (*rel. int.* %): 357 ([M+H]⁺, 50), 297 (17), 237 (26). HRFAB-MS m/z 357.1337 [M+H]⁺ (Calcd for C₂₀H₂₁O₆: 357.1338).

Phellodensin E (2) C₂₆H₃₀O₁₁ : Yellow powder; mp 235-236°C (MeOH); $[\alpha]_{\text{D}} +35.8^\circ$ (c 0.094, MeOH); UV $\lambda_{\text{max}}^{\text{MeOH}}$ nm (log ϵ): 226 (4.36), 287 (4.15), 327 (3.50); IR ν_{max} cm⁻¹: 3368, 2922, 1630, 1520, 1177. $^1\text{H-NMR}$ (400 MHz, Acetone- d_6): δ 1.64 (3H, s, CH₃), 2.78 ~ 2.84 (1H, m, H-3 β), 3.17 (1H, dd, $J = 17.1, 12.6$ Hz, H-3 α), 3.23 (1H, dd, $J = 14.4, 7.6$ Hz, H-1''), 3.41 (1H, dd, $J = 14.4, 7.6$ Hz, H-1''), 3.43 ~ 3.65 (4H, m, H-2''' ~ H-5'''), 3.65 ~ 3.75 (1H, m, H-6'''), 3.84 ~ 3.90 (1H, m, H-6'''), 3.86 (2H, br s, H-4''), 4.29 (1H, br s, OH), 4.39 (1H, br s, OH), 4.51 (1H, br s, OH), 5.06 (1H, d, $J = 7.6$ Hz, H-1'''), 5.43 ~ 5.48 (1H, m, H-2''), 5.48 (1H, dd, $J = 12.6, 2.8$ Hz, H-2), 6.29 (1H, s, H-6), 6.89 (2H, d, $J = 8.6$ Hz, H-3', -5'), 7.40 (2H, d, $J = 8.6$ Hz, H-2', -6'), 8.49 (1H, br s, 4'-OH), and 12.08 (1H, s, 5-OH). $^{13}\text{C-NMR}$ (100 MHz, Acetone- d_6): δ 13.9 (CH₃), 22.0 (C-1''), 43.6 (C-3), 62.2 (C-6'''), 68.4 (C-4''), 71.2 (C-4'''), 74.7 (C-2'''), 78.0&78.1 (C-3''' & -5'''), 79.8 (C-2), 96.3 (C-6), 101.5 (C-1'''), 104.4 (C-10), 110.0 (C-8), 116.2 (C-3', -5'), 123.6 (C-2''), 128.9 (C-2', -6'), 130.8 (C-1'), 135.6 (C-3''), 158.7 (C-4'), 160.3 (C-9), 163.1 (C-5), 164.5 (C-7), and 198.3 (C-4). CD (MeOH: $c = 0.00005$): $[\theta]_{336} +7766$, $[\theta]_{311} 0$, $[\theta]_{298} -39860$, $[\theta]_{264} 0$, $[\theta]_{253} +5232$, $[\theta]_{218} +52690$. FAB-MS m/z (*rel. int.* %): 519 ([M+H]⁺, 7), 462 (23), 397 (13), 357 (11), 338 (24), 321 (22), 320 (27), 319 (26), 312 (35). HRFAB-MS m/z 519.1865 [M+H]⁺ (Calcd for C₂₆H₃₁O₁₁: 519.1866).

Phellodensin F (3) C₂₆H₃₀O₁₀ : Yellow powder; mp 221-222°C (MeOH); $[\alpha]_{\text{D}} +18.2^\circ$ (c 0.036, MeOH); UV $\lambda_{\text{max}}^{\text{MeOH}}$ nm (log ϵ): 288 (4.37), 344 (3.76); IR ν_{max} cm⁻¹: 3416, 2918, 1641, 1599, 1364. $^1\text{H-NMR}$ (400 MHz, Acetone- d_6): δ 1.60 (6H, s, 2 \times CH₃), 2.70 ~ 2.90 (1H, m, H-3 β), 3.18 (1H, dd, $J = 17.0, 12.6$ Hz, H-3 α), 3.18 (1H, dd, $J = 13.9, 7.5$ Hz, H-1''), 3.36 (1H, dd, $J = 13.9, 7.5$ Hz, H-1''), 3.44 ~ 3.51 (1H,

m, H-4'''), 3.51 ~ 3.57 (2H, m, H-2''', -3'''), 3.57 ~ 3.62 (1H, m, H-5'''), 3.71 (1H, m, H-6'''), 3.90 (1H, m, H-6'''), 4.27 (1H, br s, OH), 4.37 (1H, br s, OH), 4.45 (1H, br s, OH), 5.05 (1H, d, $J = 7.5$ Hz, H-1'''), 5.20 (1H, t, $J = 7.3$ Hz, H-2''), 5.47 (1H, dd, $J = 12.6, 3.0$ Hz, H-2), 6.29 (1H, s, H-6), 6.89 (2H, d, $J = 8.6$ Hz, H-3', -5'), 7.40 (2H, d, $J = 8.6$ Hz, H-2', -6'), 8.49 (1H, br s, 4'-OH), and 12.08 (1H, s, 5-OH).

^{13}C -NMR (100 MHz, Acetone- d_6): δ 17.9 (CH₃), 22.5 (C-1''), 25.9 (CH₃), 43.6 (C-3), 62.6 (C-6'''), 71.3 (C-4'''), 74.7 (C-2'''), 78.0&78.1 (C-3'''&-5'''), 79.8 (C-2), 96.3 (C-6), 101.5 (C-1'''), 104.4 (C-10), 110.3 (C-8), 116.2 (C-3', -5'), 123.8 (C-2''), 128.9 (C-2', -6'), 130.9 (C-1'), 131.2 (C-3''), 158.6 (C-4'), 160.3 (C-9), 163.0 (C-5), 164.4 (C-7), and 198.3 (C-4). CD (MeOH: $c = 0.00005$): $[\theta]_{336} +12300$, $[\theta]_{308} 0$, $[\theta]_{289} -68250$, $[\theta]_{262} 0$, $[\theta]_{254} +7161$, $[\theta]_{217} +83570$. FAB-MS m/z (*rel. int.* %): 503 ([M+H]⁺, 8), 428 (11), 341 (21), 340 (14), 339 (13), 326 (15), 289 (13), 285 (10). HRFAB-MS m/z 503.1916 [M+H]⁺ (Calcd for C₂₆H₃₁O₁₀: 503.1917).

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