## **SPECTAMINES A AND B, POSSIBLE INHIBITORS OF SUPEROXIDE ANION PRODUCTION OF MACROPHAGES FROM** *CASSIA SPECTABILIS*

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**Abstract** – Two novel piperidine alkaloids were isolated from an African legume, *Cassia spectabilis*, and identified as the *O*-benzoyl (**1**, named spectamine A) and *O*-acetyl (**2**, named spectamine B) derivatives of (+)-iso-6-cassine (**3**). The absolute configurations of **1**-**3** were established to be (2*R*,3*R*,6*R*) using the modified Mosher's method. Compound (**1**) inhibited the superoxide anion production of macrophages, while it did not quench the superoxide anion which is produced by xanthine oxidase at a concentration of  $25 \mu M$ .

Superoxide anion  $(O_2^-)$  is one of active oxygens which are produced in human body. The over-production of  $O_2$ <sup>-</sup> causes several diseases, such as inflammation, cancer, and hypertension.<sup>1</sup> Scavenging of the over-produced superoxide or depression of the superoxide production would be useful for maintaining a healthy human body condition. There are many reports that radical scavengers such as polyphenols are able to quench the superoxide anion,<sup>2</sup> however, there are a few reports on the inhibitors of superoxide production.<sup>3</sup> We searched for an inhibitor of the superoxide anion production from African plants using the macrophage test and found that the methanolic extract of *Cassia spectabilis,* a Leguminosae plant, suppressed the superoxide production of macrophages. The bioassay-guided isolation from the methanolic extract of this plant by silica gel column chromatography led to two novel piperidine alkaloids, called spectamines A (**1**) and B (**2**).

The <sup>1</sup>H NMR spectrum of **1** was similar to that of  $(+)$ -iso-6-cassine (3),<sup>4</sup> except for the presence of signals assignable to a phenyl group (5H,  $\delta$  7.45-8.06 ppm; Table 1). In the <sup>13</sup>C NMR spectrum of **1**, signals were observed at  $\delta$  128.4, 129.6, 130.6 and 132.9 ppm (C<sub>6</sub>H<sub>5</sub>) and 165.9 ppm (C=O) in addition to those of **3**. These spectra suggested that **1** was an *O*-benzoyl derivative of **3**, which was supported by a molecular ion peak at  $m/z$  401.2963 (C<sub>25</sub>H<sub>39</sub>NO<sub>3</sub>) in the HREIMS of **1**. Compound (**1**) was converted to **3** by hydrolysis, confirming the assumption (Figure 1). The absolute configuration of **1**, either (2*R*,3*R*,6*R*) or (2*S*,3*S*,6*S*), could not be established by the optical rotation value, since the absolute configuration and the optical rotation values had been contradictory in past studies on  $3^{4,5}$  We determined the absolute configuration of **3** using the modified Mosher's method.6 The (*R*)- and  $(S)$ -MTPA esters of Boc-3 were prepared from 3 and then their chemical shift values in the  ${}^{1}H$  NMR spectra were compared (Figure 2). The δ values at C-2, C-2-Me, C-4 and C-5 revealed that the absolute configuration at C-3 was (*R*), meaning that the absolute configuration of **3** was (2*R*,3*R*,6*R*). Thus, **1** was identified as a novel alkaloid, (2*R*,3*R*,6*R*)-(+)-3-benzoyloxy-2-methyl-6-(11''-oxododecyl)piperidine (spectamine A).



**Figure 1.** Chemical conversion of **1** and **2** to **3**

Spectamine $A(1)$			Spectamine B $(2)$	
position	$^{13}$ C	$\rm ^1H$	$^{13}$ C	$\rm ^1H$
2	48.8 (CH)	3.41 (1H, qd, $J = 6.8$ , 3.9 Hz)	48.4 (CH)	3.27 (1H, qd, $J = 6.8$ , 3.5 Hz)
3		72.8 (CH) 5.11 (1H, ddd, $J = 4.4$ , 4.1, 3.9 Hz)	72.1 (CH)	4.85 (1H, ddd, $J = 4.3$ , 3.8, 3.5 Hz)
$\overline{4}$	24.5 $(CH_2)$	$1.85$ (1H, m)	24.4 $(CH_2)$	$1.72$ (1H, m)
		$1.92$ (1H, m)		$1.80$ (1H, m)
5	26.5 (CH <sub>2</sub> )	$1.30$ (1H, m)	26.6 $(CH2)$	$1.30$ (1H, m)
		$1.35$ (1H, m)		$1.36$ (1H, m)
6		49.0 (CH) $2.88$ (1H, m)	49.3 (CH)	$2.84$ (1H, m)
$2$ -CH <sub>3</sub>		14.8 (CH <sub>3</sub> ) 1.21 (3H, d, $J = 6.8$ Hz)		14.9 (CH <sub>3</sub> ) 1.10 (3H, d, $J = 6.8$ Hz)
	$3-O-C(=O) - 165.9(C)$		170.6 $(C)$	
$1^{\prime}$	130.6 $(C)$			$21.3$ (CH <sub>3</sub> ) $2.07$ (3H, s)
2', 6'	129.6 $(CH)$	8.06 (2H, d, $J = 7.3$ Hz)		
3', 5'		128.4 (CH) 7.45 (2H, dd, $J = 7.4$ , 7.3 Hz)		
4'		132.9 (CH) 7.56 (1H, t, $J = 7.4$ Hz)		
1"		34.6 (CH <sub>2</sub> ) 1.30 (1H, m)		34.1 $(CH_2)$ 1.30 $(1H, m)$
		$1.58$ (1H, m)		$1.44$ (1H, m)
$2" - 8"$	29.4-29.8 (CH <sub>2</sub> ) 1.27 (14H, m)		29.2-29.7 (CH <sub>2</sub> ) 1.27 (14H, m)	
9"		$23.9$ (CH <sub>2</sub> ) 1.57 (2H, m)		$23.9$ (CH <sub>2</sub> ) 1.56 (2H, m)
10"		43.8 (CH <sub>2</sub> ) 2.41 (2H, t, $J = 7.5$ Hz)		43.8 (CH <sub>2</sub> ) 2.41 (2H, t, $J = 7.4$ Hz)
11"	209.4(C)		$209.3$ (C)	
12"		$29.9$ (CH <sub>3</sub> ) 2.13 (3H, s)		29.8 (CH <sub>3</sub> ) 2.13 (3H, s)

**Table 1.** <sup>1</sup>H and <sup>13</sup>C NMR spectral data for **1** and **2** (500 MHz for <sup>1</sup>H and 125 MHz for <sup>13</sup>C, CDCl<sub>3</sub>)



**Figure 2.** Differences in proton chemical shift (ppm) between the (*S*)- and (*R*)-MTPA esters of Boc-**3**. Underlined values: (δ value from 1 H NMR spectrum of the (*S*)-MTPA ester of Boc-3) - ( $\delta$  value from <sup>1</sup>H NMR spectrum of the (*R*)-MTPA ester of Boc-3)

Absolute configuration	$\lceil \alpha \rceil_D$	Method used for determination
(2R, 3R, 6R)	$-3.3^{\circ}$ (c 0.26, CHCl <sub>3</sub> )	Horeau's process
(2R, 3R, 6R)	$+1.5^{\circ}$ (c 1.22, CHCl <sub>3</sub> )	the modified Mosher's method
(2S, 3S, 6S)	$-1.5^{\circ}$ (c 1.50, CHCl <sub>3</sub> )	asymmetric synthesis

**Table 2.** The  $\alpha|_D$  values and the absolute configurations of natural and synthesized 3

The <sup>1</sup>H NMR spectrum of **2** was similar to that of **1**, except for another singlet (3H,  $\delta$  2.07 ppm; Table 1) assignable to an acetyl group instead of the benzoyl group. The 13C NMR spectrum of **2** showed signals at  $\delta$  21.3 ppm (CH<sub>3</sub>) and 170.6 ppm (C=O) in addition to those of **3**, suggesting that **2** was an *O*-acetyl derivative of **3**. A molecular ion peak at *m/z* 399.2762 in the HREIMS of **2**, suggesting a molecular formula of  $C_{20}H_{37}NO_3$ , supported the structure. Hydrolysis of 2 gave 3 whose optical rotation value, [ $α$ ]<sup>30</sup><sub>D</sub> +1.4° (*c* 1.97, CHCl<sub>3</sub>), was almost equal to that of **3** prepared from **1**,  $[α]$ <sup>30</sup><sub>D</sub> +1.5° (*c* 1.22, CHCl<sub>3</sub>), indicating that the absolute configuration of **2** was also (2*R*,3*R*,6*R*). These observations proved that **2** was a novel alkaloid,  $(2R,3R,6R)$ -(+)-3-acetoxy-2-methyl-6-(11"-oxododecyl)piperidine (spectamine B). Christofidis *et al.* reported the isolation and identification of **3**,  $[\alpha]_D^{25}$  -3.3° (*c* 0.26, CHCl<sub>3</sub>), from *C*. *spectabilis*, and determined its absolute configuration to be  $(2R,3R,6R)$  by Horeau's process (Table 2).<sup>4</sup> The asymmetric synthesis of (2*S*,3*S*,6*S*)-**3**, which corresponded to the enantiomer of the natural **3**, was achieved by Toyooka *et al.*<sup>5</sup> The  $[\alpha]_D^{25}$  value of the synthesized (2*S*,3*S*,6*S*)-3 was -1.5° (*c* 1.50, CHCl<sub>3</sub>), which should have been  $+3.3^{\circ}$  if that of natural **3** was correct. In the present paper, we described the isolation of **3** as the *O*-benzoyl and *O*-acetyl derivatives (**1** and **2**, respectively), and the identification of its absolute configuration as  $(2R,3R,6R)$  using the modified Mosher's method. The  $\lceil \alpha \rceil_D$  values and the absolute configuration we described were well correlated with those of Toyooka *et al.*,<sup>5</sup> suggesting that the  $\lceil \alpha \rceil_D$  value of the previously reported natural **3** was incorrect.<sup>4</sup> A trace of impurity might have produced a bad effect on the optical rotation value of **3** in their experiment.

The inhibitory activities of the superoxide production of **1**-**3** were evaluated using both the xanthine oxidase (XOD) test and the macrophage test.<sup>7</sup> As shown in Table 3, a radical scavenger, quercetin,

quenched 30.9% and 72.3% of the superoxide which was produced by XOD, at 25 and 125 µM, respectively, while **1**-**3** showed slight activities at 25 and 125 µM during the XOD test. For the macrophage test, **1** showed the strongest activity and suppressed 46.7% of the superoxide production at 25 µM, although its activity to quench the superoxide anion was weak. Compound (**1**) might be a specific inhibitor of the superoxide production of macrophages.

		$%$ Inhibitory activity ( $\pm$ Standard deviation)			
	XOD test		Macrophage test		
Tested compound	$25 \mu M$	125 µM	$25 \mu M$		
spectamine $A(1)$	8.7 $(\pm 3.4)$	15.1 $(\pm 6.0)$	46.7 $(\pm 13.2)$		
spectamine B $(2)$	2.5 $(\pm 4.7)$	$-1.9 \ (\pm 7.4)$	5.0 $(\pm 9.9)$		
$(+)$ -iso-6-cassine (3)	$-0.7 \ (\pm 3.5)$	4.8 $(\pm 15.5)$	7.0 $(\pm 5.4)$		
quercetin*	30.9 $(\pm 8.4)$	72.3 $(\pm 1.6)$	19.0 $(\pm 7.5)$		

**Table 3.** Inhibitory activities of **1**-**3** against the superoxide production by xanthine oxidase (XOD) and macrophage

\*A positive control in the XOD and the macrophage tests.

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