HETEROCYCLE FUSED CYCLOHEXYLGLYCINE DERIVATIVES AS NOVEL DIPEPTIDYL PEPTIDASE-IV INHIBITORS

Anthony Mastracchio,* Emma R. Parmee, Barbara Leiting, Frank Marsilio, Reshma Patel, Nancy A. Thornberry, Ann E. Weber, and Scott D. Edmondson

Merck Research Laboratories, Merck & Co., Inc., Rahway, New Jersey 07065, U.S.A.

Abstract – A new class of potent inhibitors of dipeptidyl peptidase IV (DP-IV) for the treatment of type II diabetes are described. Presented herein is the synthesis of indole-fused and thiazole-fused cyclohexylglycines. Pyrrolidine-derived amides of these novel heterocycles led to the discovery of thiazole derivatives (**3f**) and (**11a**), both low nanomolar inhibitors of DP-IV $(IC_{50} = 6 \text{ nM})$.

Inhibition of dipeptidyl peptidase IV (DP-IV) is emerging as a new approach for the treatment of type-II diabetes.¹ Recent research from Merck laboratories led to the discovery of substituted 4-aminocyclohexylglycine derivatives (**1**) as potent DP-IV inhibitors.2 As part of this program, we hoped to improve the *in vitro* profile of these derivatives by incorporating a fused heterocycle onto the cyclohexyl moiety of **1**. In this communication, we present the synthesis and biological evaluation of indoles (**2**) and thiazoles (**3**).

The synthesis of indole derivatives (**2**) is described in Scheme 1. Readily available cyclohexylglycine derivative (**4**) 2b (*cis/trans* mixture) was saponified and the resulting acid coupled to pyrrolidine using standard peptide coupling conditions. The alcohol was then oxidized using Dess-Martin periodinane (DMP) to provide ketone (**5**) in good overall yield (46%, 3 steps). Attempts to use **5** in a Fisher indole synthesis³ resulted in partial thermolysis of the *tert*-butoxycarbonyl (Boc) group and so the protecting

group was converted to the more thermally stable benzyloxycarbonyl (Cbz) group in **6**. Exposure of **6** to commercially available arylhydrazines in the presence of *p*-toluenesulfonic acid (*p*-TsOH) in hot ethanol yielded the desired fused system as a mixture of diastereomers. Catalytic hydrogenolysis of the carboxybenzyl group afforded indoles (**2**) in moderate to good yields (Table 1). In examples where the aryl hydrazines bear a benzoic acid, a mixture of acid and the corresponding ethyl ester were isolated from the same reaction mixture (**2d/e** and **2f/g**).

Conditions: a) LiOH (aq.), MeOH, THF; b) EDC, HOBt, DIEA, DCM, pyrrolidine; c) DMP, CH_2Cl_2 ; d) i) CH₂Cl₂, TFA; ii) NaHCO₃ (aq.), THF, CbzCl; e) ArNHNH₂, *p*-TsOH, EtOH, Δ ; f) Pd(OH)₂, EtOH, H₂.

Scheme 1

The preparation of thiazoles (**3**) relies on the Hantzsch synthesis (Scheme 2). Methyl ester (**4**) was first converted to the fluoropyrrolidide amide (8) using conditions outlined above,⁴ and then treated with phenyltrimethylammonium tribromide (PhNMe₃Br₃) to afford the intermediate α -bromo ketone, which was used without purification due to stability concerns. The α -bromo ketone was heated with thioamides to provide a mixture of the protected diastereomeric thiazoles, which were separable by chiral chromatography.⁵ The protecting group of each diastereomer was removed using iodotrimethylsilane to afford the desired ammonium salts (**3**) after purification.6

Conditions: a) LiOH (aq.), MeOH, THF; b) EDC, HOBt, DIEA, (s)-3-fluoropyrrolidine, CH₂Cl₂; c) DMP, CH_2Cl_2 ; d) i) CH_2Cl_2 , TFA; ii) NaHCO₃ (aq.), THF, CbzCl; e) PhNMe₃Br₃, THF, 0 $^{\circ}C$; f) RCSNH₂, DMF, 120 °C ; g) TMSI, MeCN.

Scheme 2

We next sought to make derivatives of the aminothiazole group (Scheme 3). Since the Boc protected diastereomers of **10** were more readily resolved by chiral chromatography than the corresponding Cbz derivative, a different synthetic protocol using Boc as the protecting group was used to access the

diastereomerically pure aminothiazoles (**10**) (Scheme 3). After deprotection of each pure diastereomer of **10**, the more active isomer was determined to be the isomer $(3d)$.⁷ Acylation of the more active diastereomer of **10** followed by deprotection of the Boc group then afforded pure diastereomers of **11**.

Conditions: a) PhNMe₃Br₃, THF, 0 °C; b) i) SC(NH₂)₂, EtOH, Δ ; ii) Boc₂O, CH₂Cl₂, DIEA; c) RCOCl, $CH₂Cl₂$, DIEA; iii) TFA, $CH₂Cl₂$.

Scheme 3

Table 1. Yields and biological activity of fused heterocyclic inhibitors of DP-IV

	TFA H_2N , N^{\sim} Ĥ. NΗ R $\overline{2}$				$TFA \cdot H_2N$ 'N1 ے۔ F \dot{H} S $\overline{\mathsf{R}}$ 3		
Entry	${\bf R}$	Yield ^a	$DP-IV$	Entry	$\mathbf R$	Yield ^b	$DP-IV$
		$\%$	IC_{50} (nM)			$\%$	IC_{50} (nM)
2a	$5-C1$	76	67	$3a^c$	$-Me$	6	43
2 _b	$5-OCF3$	$30\,$	209	$3b^c$	ξ. CF ₃	$\mathbf{9}$	$\boldsymbol{6}$
2c	$5-CF_3$	60	73	3c	$-NH2$	32	185
2d	$7-CO2H$	13	29	3d	$-NH2$	32	61
2e	$7-CO2Et$	14	13	$11a^d$	$$-NH$ CF ₃	11	6
2f	$5-CO2H$	10	609	$11b^d$	-NHAc	63	13
2g	$5-CO2Et$	36	220				\overline{a} , here \overline{a}

^aYields represent the combined yields of Fisher indole cyclization and deprotection steps (conversion of 6 to 2). ^bYields represent the overall conversion of δ to δ , including the resolution of diastereomers before deprotection (max yield = 50%). ^cOnly active diastereomers are shown.^{9d} Yields are for the conversion of 10 to 11. ¹H NMR spectra of representative examples: 3b (500 MHz, CD3OD) δ 8.08 (d, *J* = 8.2 Hz, 2 H), 7.78 (d, *J* = 8.2 Hz, 2H), 5.39 (m, 1 H, CHF), 4.39 (dd, *J* = 41 Hz and 6.6 Hz), 3.58-4.02 (m, 4 H) 3.02-3.09 (m, 2H), 2.84-2.91 (m, 2 H) 2.08-2.59 (m, 4 H) 1.8-1.9 (m, 1 H); **3c** (500 MHz, CD3OD) δ 5.34 (m, 1 H, CHF) 4.21 (dd, *J* = 35.1 Hz and 6.8 Hz, 1 H), 3.31-3.97 (m, 4 H), 2.53-2.73 (m, 4H), 2.09-2.40 (m, 4H) 1.69-1.75 (m, 1H); **3d** (500 MHz, CD3OD) δ 5.36 (m, 1 H, CHF), 4.22 (dd, *J* = 43.3 Hz and 6.7 Hz, 1 H), 3.50-3.99 (m, 4 H), 2.52-2.66 (m, 4 H) 1.95-2.42 (m, 4 H), 1.65-1.78 (m, 1 H); **11a** (500 MHz, CD3OD) δ 8.35-8.41 (m, 2 H), 7.53 (t, 1 H, *J* = 9.5 Hz), 5.28-5.47 (m, 1 H, CHF), 4.14 (dd, *J* = 245.5 Hz and 5.7 Hz, 1 H), 3.55-4.02 (m, 4 H), 2.82-2.92 (m, 2 H), 2.69-2.78 (m, 2H), 2.02-2.57 (m, 4H), 1.8 (m, 1 H).

Many of the above compounds are potent inhibitors against DP-IV, with IC_{50} 's in the low nanomolar range (Table 1).⁸ For example 3b and 11a were extremely potent, each with a 6 nM inhibition of the enzyme even though lacking a serine trap.¹

In conclusion, this work demonstrates that changing the nature of the cyclohexyl ring in cyclohexylglycine derived DP-IV inhibitors can afford compounds with equal or greater potency than other reported compounds (*i.e.* **1**). Furthermore, we have succeeded in developing simple synthetic methodologies for accessing indole-fused and thiazole-fused cyclohexylglycine derivatives.

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- 4. *(S)-*3-Fluoropyrrolidine was prepared by a modification of the method of G. Giardina, G. Dondio, and M. Grugni, *Synlett*, 1995, 55. The modification is described in Ref. 2 (c)
- 5. Diastereomeric mixtures were resolved using ChiralCel columns. Type OJ (isocratic method, 50% ethanol/hexane) was used for **3a** and type AS (isocratic method, 50% ethanol/hexane) was used for **3b**.
- 6. All final products were purified by reverse phase HPLC to afford TFA salts.
- 7. The two diastereomers of **10** were separated using a ChiralCel column type AD (isocratic method, 75% EtOH/hexane). The most potent deprotected compound (**3d**) is derived from the faster eluting isomer.
- 8. For DP-IV assay conditions, see : B. Leiting, K. D. Pryor, J. K. Wu, F. Marsilio, R. A. Patel, C. S. Craik, J. A. Ellman, R. T. Cummings, and N. A. Thornberry, *Biochem. J.*, 2003, **371**, 525.
- 9. The less active diastereomers of **3a** and **3b** were $IC_{50} = 385$ nM and 171 nM respectively.