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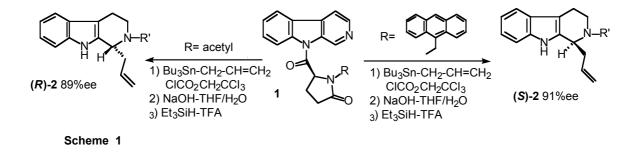
SYNTHESES OF 1,2,3,4,6,7,12,12b-OCTAHYDROINDOLO[2,3-*a*]-QUINOLIZINE AND HARMICINE USING A CHIRAL 1-ALLYL-1,2,3,4-TETRAHYDRO-**β**-CARBOLINE AS THE STARTING MATERIAL

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<u>Abstract</u> – Total syntheses of (S)-1,2,3,4,6,7,12,12b-octahydroindolo[2,3-a]quinolizine and (S)-harmicine were carried out using chiral 1-allyl-1,2,3,4-tetrahydro- β -carboline as the starting material.

In recent years, we have developed a general method for the formation of a chiral center at the C-1 position of β -carboline nucleus using chiral auxiliaries derived from readily available amino acids and their derivatives.¹ Using the method, both enantiomers of the chiral 1-allyl derivatives (2) were obtained in a highly enantioselective manner by simply changing the *N*-protecting group of the chiral auxiliary derived from pyroglutamic acid (Scheme 1).

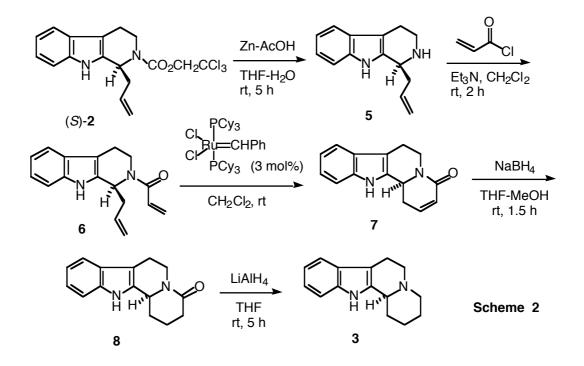


Allyl group is a versatile tool for syntheses of other functional groups,² and there are several reports concerning the application of racemic allyl group at C-1 position of β -carboline to the synthesis of various indole alkaloids.³ However, it has been well known that 1-substituted 1,2,3,4-tetrahydro- β -carboline derivatives are readily racemized under acidic conditions,⁴ therefore, we decided to investigate the stability of the allyl adduct by transforming it to chiral natural products in rather straightforward ways. 1,2,3,4,6,7,12,12b-Octahydroindolo[2,3-*a*]quinolizine (**3**) was selected as the first target molecule because the synthesis of **3** was considered as a standard procedure to evaluate chiral synthetic

methods for various indole alkaloids.⁵ Thereafter, the chiral precursor was applied to the synthesis of a natural product harmicine (4),⁶ whose absolute configuration was not determined. This paper describes these results.⁷

1,2,3,4,6,7,12,12b-Octahydroindolo[2,3-*a*]quinolizine (**3**) is an alkaloid which was isolated from *Dragontomelum mangiferum*,⁸ and has been a synthetic target as a model compound in connection with the preparation of structurally more complicated alkaloids. There have been several reports⁵ concerning the asymmetric synthesis of the compound.

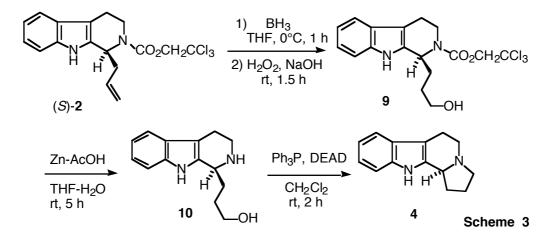
Our synthesis of **3** was commenced with the reaction of (*S*)-**2** with Zn-CH₃CO₂H in THF/H₂O to remove the trichloroethoxycarbonyl group (Scheme 2). The starting material ((*S*)-**2**) of 91% was obtained from the method reported earlier (overall yield 71% from β -carboline *via* 4 steps).^{1d} The amine (**5**) thus obtained was allowed to react with acryloyl chloride in the presence of triethylamine to give the *N*-acryloyl derivative (**6**) in 88% yield. The compound (**6**) was treated with the Grubbs reagent to afford a ring-closing product (**7**). The C=C double bond in **7** was reduced with sodium borohydride to give a saturated amide (**8**) in an excellent yield. Single recrystallization of **8** raised the ee of the compound from 91% to 97%. Although the transformation of **8** to the target (**3**) was carried out according to the reported method,^{5e} slight racemization occurred to give the product (**3**) in 85% yield and 90% ee.⁹



Through the above transformation, it was found that the chiral allyl adduct (2) and its derivatives are stable under several reaction conditions. Thus, we next applied the compound (2) to the synthesis of an alkaloid whose stereochemistry was not determined.

Although 2,3,5,6,11,11b-hexahydro-1*H*-indolizino[8,7-*b*]indole (4) has been used as a starting material for racemic total synthesis of many alkaloids such as tubifoline, condyfoline, fluorocurarine *etc.*,¹⁰ the compound had been thought to be a non-natural product. It was recently isolated, however, from the leaf extract of *Kopsia griffithii*, and named as harmicine. The absolute configuration of the C-11b

chiral center was assigned as (*S*), but the definitive evidence was not shown.⁶ Therefore, we applied the allyl adduct (2) to the synthesis of harmicine in order to determine the absolute configuration. The synthetic scheme is shown below (Scheme 3).



Since the report⁶ suggested the structure of **4** as the (*S*) isomer, we selected (*S*)-**2** as a starting material. A standard hydroboration-oxidation procedure afforded a 1-(3-hydroxypropyl)- β -carboline (**9**) without any noticeable racemization in 79% yield. At first, we tried the hydroboration-oxidation procedure by the use of the amine (**5**) as a substrate expecting the shortening of the synthetic procedures. In this case, however, the expected product (**10**) was not obtained. The reduction of trichloroethoxycarbonyl group by zinc/acetic acid readily eliminated the group to give 1-(3-hydroxypropyl)-1,2,3,4-tetrahydro- β -carboline (**10**) in 77% yield. The compound (**10**) was recrystallized from hexane-AcOEt to give an optically pure form. Then a Mitsunobu procedure¹¹ was applied to **10** and the ring-closed product (**4**) was obtained in good yield with high enantiopurity (84% yield, 98% ee). All the values of the ee were obtained from the chiral HPLC analysis. We started the synthesis using the (*S*) isomer, and the racemization did not occur in the reaction process. The value of the specific rotation was, however, showed [α]¹⁷_D=-108.31° (c 0.10, CHCl₃), which has an opposite sign for the reported value of (+)-(*S*)-harmicine. Therefore, the present study suggested that the naturally occurring harmicine has an (*R*) configuration.

In this paper, we described the application of the chiral allyl adducts derived from β -carboline to the syntheses of 1,2,3,4,6,7,12,12b-octahydroindolo[2,3-*a*]quinolizine and harmicine in short steps. The results showed that the chiral allyl adduct (2) was of use as a starting material for asymmetric synthesis of indole alkaloids, and that the reported configuration of (+)-harmicine was reversed.¹² Application of the allyl adducts to syntheses of other chiral alkaloids is now under investigation.

EXPERIMENTAL

General Remarks. Melting points are uncorrected. ¹H and ¹³C NMR spectra of CDCl₃ and CD₃OD solutions were recorded at 500 and 125 MHz, respectively, with tetramethylsilane (TMS) as an internal standard. FAB-HRMS spectra were measured using *p*-nitrobenzyl alcohol as a matrix.

Synthesis of 1-allyl-1,2,3,4-tetrahydro- β -carboline (5). To the solution of (*S*)-2 (40 mg, 0.1 mmol), acetic acid (30 µL) in THF (0.5 mL)/H₂O (0.5 mL) was added zinc powder (38 mg, 0.5 mmol), and the mixture was allowed to react for 1 h at rt. Then the excess zinc was removed by filtration, and the filtrate was diluted with H₂O. The solution was basified with K₂CO₃, and extracted with CH₂Cl₂. The organic layer was dried over MgSO₄, and evaporated off to leave the product (5).

1-Allyl-1,2,3,4-tetrahydro-β-carboline (5) Yield 85%; Colorless granules from hexane-AcOEt; mp 118-119°C; $[\alpha]^{20}_{D}$ = -116.40° (c 0.64, CHCl₃); ee=91% (chiralcel OD, λ=254 nm, *i*PrOH/hexane=1/2); ¹H-NMR (CDCl₃) δ: 1.72-1.85 (1H, brs), 2.54-2.59 (2H, m), 2.68-2.81 (2H, m), 3.03 (1H, ddd, *J*=12.8, 8.6, 5.3 Hz), 3.37 (1H, ddd, *J*=12.8, 5.0, 3.7 Hz) 4.16 (1H, tt, *J*=6.3, 1.8 Hz), 5.20-5.29 (2H, m), 5.92 (1H, ddt, *J*=17.1, 10.2, 7.0 Hz), 7.09 (1H, td, *J*=7.0, 1.1 Hz), 7.15 (1H, td, *J*=7.0, 1.3 Hz), 7.29 (1H, d, *J*=7.0 Hz), 7.48 (1H, d, *J*=7.0 Hz), 7.96 (1H, brs); ¹³C-NMR (CDCl₃) δ: 22.7, 39.5, 42.9, 51.6, 109.1, 110.6, 117.9, 118.4, 119.2, 121.4, 127.1, 134.6, 135.4, 135.6. Anal. Calcd for C₁₄H₁₆N₂: C, 79.21; H, 7.60; N, 13.20. Found: C, 79.01; H, 7.71; N, 13.21.

Synthesis of 1-allyl-1,2,3,4-tetrahydro-2-propenoyl- β -carboline (6). To the CH₂Cl₂ solution (5 mL) of the compound (5) (492 mg, 2.3 mmol) and triethylamine (644 μ L, 4.6 mmol) was added acryloyl chloride (196 μ l, 2.4 mmol), and the mixture was allowed to react for 2 h at rt. Then the solution was washed with H₂O and brine, dried over MgSO₄, and evaporated off. The residue was chromatographed on silica gel (hexane/AcOEt = 4) to give the product (6).

1-Allyl-1,2,3,4-tetrahydro-2-propenoyl-β-carboline (6) Yield 88%; Pale yellow granules from CH₂Cl₂-*i*Pr₂O; mp 139-141°C; $[\alpha]^{18}_{D}$ = +105.41° (c 0.56, CHCl₃); ee=91% (chiralcel OJ-R, λ=254 nm, H₂O:CH₃CN=4:3). The compound was obtained as a mixture of two conformational isomers, and the spectral data of the major one are shown below. ¹H-NMR (CDCl₃) δ: 2.60-2.71 (2H, m), 2.76-2.88 (2H, m), 3.49-3.55 (1H, m), 4.16-4.20 (1H, m), 5.07-5.12 (2H, m), 5.76 (1H, dd, *J*=10.7, 1.8 Hz), 5.84-5.96 (2H, m), 6.35 (1H, dd, *J*=7.1, 1.8 Hz), 6.70 (1H, dd, *J*=16.8, 10.7 Hz), 7.06-7.10 (1H, m), 7.12-7.17 (1H, m), 7.30 (1H, d, *J*=8.2 Hz), 7.44 (1H, d, *J*=7.6 Hz), 8.63 (1H, s); ¹³C-NMR (CDCl₃) δ: 22.3, 38.8, 41.0, 49.4, 107.6, 111.1, 117.9, 118.3, 119.4, 121.7, 126.5, 128.0, 128.2, 133.7, 134.3, 136.1, 166.1. Anal. Calcd for C₁₇H₁₈N₂O: C, 76.66; H, 6.81; N, 10.52. Found: C, 76.81; H, 7.18; N, 10.26.

Metathesis of 1-allyl-1,2,3,4-tetrahydro-2-propenoyl- β -carboline. To the CH₂Cl₂ solution (5 mL) of the compound (6) (266 mg, 1.0 mmol) was added the Grubbs catalyst (2.5 mg, 0.03 mmol), and the mixture was allowed to react for 1 h at rt. Then the solution was diluted with ether, and the precipitate thus formed was collected by filtration. The solid was washed with a small amount of CH₂Cl₂ to give the desired product (7) in a pure form (50 mg). The filtrate was evaporated off to leave a residue, which was chromatographed on silica gel (AcOEt) to give the product (7) (140 mg).

6,7,12,12b-Tetrahydro-1*H***-indolo**[**2,3***-a*]**quinolizin-4-one (7).** Yield 71%; A colorless powder from CH₂Cl₂-*i*Pr₂O; mp 229-231°C; $[\alpha]^{18}{}_{D}$ = -425.61° (c 0.60, THF); ee=91% (chiralcel OJ-R, λ =254 nm, H₂O:CH₃CN=7:3). ¹H-NMR (DMSO-*d*₆) : δ 2.22-2.30 (1H, m), 2.64-2.71 (1H, m), 2.79-2.84 (2H, m), 3.03 (1H, dt, *J*=17.7, 5.8 Hz), 4.77-4.80 (1H, m), 4.87 (1H, dd, *J*=13.4, 4.6 Hz), 5.92 (1H, dd, *J*=9.8, 2.7 Hz), 6.80 (1H, ddd, *J*=9.2, 6.1, 1.8 Hz), 6.99 (1H, t, *J*=7.6 Hz), 7.07 (1H, t, *J*=7.6 Hz), 7.33 (1H, d, *J*=7.6 Hz), 7.44 (1H, d, *J*=7.6 Hz), 10.96 (1H, s); ¹³C-NMR (DMSO-*d*₆): δ 20.5, 30.3, 38.2, 51.2, 107.0,

111.1, 117.8, 118.6, 121.1, 124.7, 126.1, 133.6, 136.3, 139.4, 164.1. Anal. Calcd for $C_{15}H_{14}N_2O$: C, 75.61; H, 5.92; N, 11.76. Found: C, 75.22; H, 5.92; N, 11.68.

Synthesis of 6,7,12,12b-tetrahydro-1*H*-indolo[2,3-*a*]quinolizin-4-one (8). To the THF/MeOH(10:1) solution (1 mL) of the starting material (7) (24 mg, 0.1 mmol) was added sodium borohydride (4 mg, 0.1 mmol), and the mixture was allowed to react for 2.5 h at rt. Then H_2O was added to the mixture, which was extracted with AcOEt. The organic layer was dried over MgSO₄ and evaporated off to leave a residue, which was chromatographed on silica gel (AcOEt) to give the product.

6,7,12,12b-Tetrahydro-1*H***-indolo[2,3-***a***]quinolizin-4-one (8). Yield 93%; A colorless powder from EtOH-H₂O ; mp 250-255°C; [\alpha]^{17}_{D}= -234.1° (c 0.20, CHCl₃); ee=97% (chiralcel OJ-H, \lambda=254 nm, hexane:***i***PrOH=1:1). ¹H-NMR (CDCl₃) : \delta 1.73-2.00 (3H, m), 2.36-2.49 (2H, m), 2.56-2.61 (1H, m), 2.74-2.92 (3H, m), 4.76-4.79 (1H, m), 5.14-5.22 (1H, m), 7.12 (1H, td,** *J***=7.4, 1.1 Hz), 7.18 (1H, td,** *J***=7.2, 1.2 Hz) 7.34 (1H, d,** *J***=7.9 Hz), 7.51 (1H, d,** *J***=7.7 Hz), 8.04 (1H, br s); ¹³C-NMR (CDCl₃): \delta 19.4, 21.0, 29.1, 32.4, 40.1, 54.4, 109.6, 110.9, 118.4, 119.8, 122.1, 126.9, 133.3, 136.2, 169.2.**

Synthesis of 1,2,3,4,6,7,12,12b-octahydroindolo[2,3-*a***]quinolizine (3). To the THF solution (2 mL) of the starting material (8) (48 mg, 0.2 mmol) was added LiAlH₄ (30 mg, 0.8 mmol) under Ar atmosphere, and the mixture was allowed to react for 5 h at rt. The mixture was cooled to -0^{\circ}C, and H₂O was added to quench the excess LiAlH₄. Then aqueous 1 M NaOH solution (4.7 mL) and CH₂Cl₂ (24 mL) were added to the mixture, and the precipitate thus formed was filtered off using Celite. The filtrate was extracted with CH₂Cl₂, which was dried over MgSO₄, and evaporated off. The residue was chromatographed on alumina (CH₂Cl₂) to give the product (3**) (41 mg).

1,2,3,4,6,7,12,12b-Octahydroindolo[**2,3***-a*]**quinolizine (3)** Yield 85%; Pale yellow granules from EtOH-H₂O; mp 149-152°C; $[\alpha]^{16}_{D}$ = -73.97° (c 1.99, MeOH); ee=90% (chiralcel OD, λ =254 nm, hexane:*i*PrOH=2:1). ¹H-NMR and ¹³C-NMR (CDCl₃) spectra were consistent with the reported ones. ¹H-NMR (CDCl₃): δ 1.37-1.53 (2H, m), 1.64-1.75 (2H, m), 1.81 (1H, d, *J*=12.5 Hz), 1.97 (1H, dd, *J*=12.5, 2.7 Hz), 2.31 (1H, td, *J*=11.1, 3.9 Hz), 2.52-2.67 (2H, m), 2.90-3.02 (3H, m), 3.16 (1H, d, *J*=11.0 Hz), 7.00 (1H, td, *J*=7.3, 1.2 Hz), 7.05 (1H, td, *J*=7.5, 1.5 Hz), 7.21 (1H, d, *J*=7.3 Hz), 7.39 (1H, d, *J*=7.5 Hz), 7.69 (1H, bs).; ¹³C-NMR (CDCl₃): δ 21.5, 24.3, 25.7, 29.9, 53.5, 55.7, 60.2, 108.0, 110.7, 118.1, 119.3, 121.2, 127.4, 135.0, 135.9.

Synthesis of 1,2,3,4-tetrahydro-1-(3-hydroxypropyl)-2-(2,2,2-trichloroethoxycarbonyl)- β -carboline (9). To the THF solution (10 mL) of (*S*)-2 (1.95 g, 5 mmol) was added 1M BH₃-THF solution (7.5 mL, 7.5 mmol) at 0°C, and the mixture was allowed to react for 1 h at 0°C. Then H₂O (10 mL) was added to quench excess BH₃, and 3M aqueous NaOH solution (2 mL) and 30% aqueous H₂O₂ solution (2 mL) were successively added to the mixture, which was allowed to react another 30 min at rt. Thereafter, saturated aqueous NaCl solution was added, and the mixture was extracted with CH₂Cl₂. The organic layer thus obtained was dried over MgSO₄, and evaporated off to leave a residue, which was chromatographed on silica gel (AcOEt: CH₂Cl₂=1:1) to give the alcohol (9) (1.596 g).

1,2,3,4-Tetrahydro-1-(3-hydroxypropyl)-2-(2,2,2-trichloroethoxycarbonyl)-\beta-carboline (9) Yield 79%; Colorless oil; [\alpha]^{14}{}_{D}= +77.36° (c 0.97, CHCl₃); ee=91% (chiralcel OD, \lambda=254 nm, hexane:,PrOH=1:3). The compound was obtained as a mixture of two conformational isomers, and the

spectral data of the major one are shown below. ¹H-NMR (CDCl₃): δ 1.74-1.80 (2H, m), 1.83-2.00 (1H, m), 2.02-2.13 (1H, m), 2.25 (1H, brs), 2.76 (1H, dd, *J*=15.4, 2.9 Hz), 2.90 (1H, td, *J*=15.5, 5.5 Hz), 3.23-3.36 (1H, m), 3.73-3.89 (2H, m), 4.49 (1H, dd, *J*=13.6, 4.5 Hz), 4.79 (1H, d, *J*=12.0 Hz), 4.83 (1H, d, *J*=12.0 Hz), 5.42-5.48 (1H, m), 7.07-7.17 (2H, m), 7.28 (1H, d, *J*=7.7 Hz), 7.47 (1H, d, *J*=7.3 Hz), 8.75 (1H, s); ¹³C-NMR (CDCl₃): 21.6, 28.2, 31.4, 38.8, 51.7, 62.5, 75.1, 95.6, 107.7, 111.0, 118.0, 119.4, 121.8, 126.6, 133.7, 135.9, 154.4. Anal. Calcd for C₁₇H₁₉N₂O₃ Cl₃: C, 50.33; H, 4.72; N, 6.90. Found: C, 50.47; H, 4.71; N, 6.55.

Synthesis of 1,2,3,4-tetrahydro-1-(3-hydroxypropyl)- β -carboline (10). To the THF/H₂O (1:1) solution (1 mL) of the alcohol (9) (40 mg, 0.1 mmol) and AcOH (30 µL) was added zinc powder (38 mg, 0.5 mmol), and the mixture was allowed to react for 1 h at rt. Then, excess Zn was removed by filtration, and the filtrate was diluted with H₂O. The solution was basified with saturated aqueous K₂CO₃ solution, and extracted with CH₂Cl₂. The organic layer was dried over MgSO₄, and evaporated off to leave a residue (18 mg, 77%), which was an almost pure product. The ee of the compound was confirmed as 96% using HPLC (Chiralcel OD-R, MeOH). A crystallization with AcOEt-hexane raised the ee to 99.6%.

1,2,3,4-Tetrahydro-1-(3-hydroxypropyl)-β-carboline (10) Yield 77%; Colorless powder from AcOEthexane; mp 133.2-133.8°C ; $[\alpha]^{17}_{D}$ = -74.72° (c 0.81, CH₃OH); ee=99.6% (chiralcel OD-R, λ=254 nm, MeOH). ¹H-NMR (CD₃OD): δ 1.60-1.78 (3H, m), 1.99-2.06 (1H, m), 2.64 (1H, dtd, *J*=15.9, 4.4, 1.6Hz), 2.69-2.75 (1H, m), 2.89 (1H, ddd, *J*=12.5, 8.8, 4.8 Hz), 3.22 (1H, dt, *J*=12.5, 5.6 Hz) 3.59 (2H, td, *J*=6.2, 1.0 Hz) 3.97-3.99 (1H, m) 6.96 (1H, ddd, *J*= 7.9, 7.0, 1.0 Hz) 7.03 (1H, ddd, *J*=8.0, 7.0, 1.0 Hz), 7.27 (1H, dt, *J*=8.0, 1.0 Hz), 7.37 (1H, dt, *J*=7.9, 1.0 Hz); ¹³C-NMR (CD₃OD): δ 22.8, 29.4, 32.4, 43.2, 53.7, 63.0, 108.8, 111.8, 118.5, 119.6, 121.9, 128.6, 136.6, 137.7. Anal. Calcd for C₁₄H₁₈N₂O: C, 73.01; H, 7.88; N, 12.16. Found: C, 72.90; H, 8.04; N, 12.02.

Synthesis of harmicine (4). To the CH_2Cl_2 solution (10 mL) of the compound (10) (198 mg, 0.86 mmol) was added triphenyl phosphine (270 mg, 1.03 mmol) and diethyl azodicarboxylate (179 mg, 1.03 mmol), and the mixture was allowed to react for 3 h at rt. Then CH_2Cl_2 was added to the mixture, which was extracted with 1M aqueous HCl (12 mL x 4). The aqueous layer was basified with K_2CO_3 , and extracted with CH_2Cl_2 . The organic layer was dried over MgSO₄, and evaporated off to leave a residue, which was chromatographed on NH silica gel (AcOEt) to give the product. The spectral data were consistent with those of the reported one except the value of the specific rotation.

(*ent*)-Harmicine (4) Yield 84%; Colorless powder from CH₂Cl₂-hexane; mp 161.5-164.5°C ; $[\alpha]^{17}_{D}$ = -108.31° (c 0.10, CHCl₃); ee=98% (chiralcel OD, λ =254 nm, hexane:*i*PrOH=1:3). ¹H-NMR (CDCl₃): δ 1.80-1.96 (3H, m), 2.21-2.32 (1H, m), 2.65 (1H, ddt, *J*=15.4, 4.7, 2.0 Hz), 2.83-2.99 (3H, m), 3.08 (1H, ddd, *J*=12.8, 10.6, 4.6 Hz), 3.32 (1H, ddd, *J*=12.8, 5.1, 2.0 Hz), 4.21-4.24 (1H, m), 7.09 (1H, td, *J*=7.1, 1.3 Hz), 7.13 (1H, td, *J*=7.1, 1.3 Hz), 7.29 (1H, d, *J*=7.1 Hz), 7.49 (1H, d, *J*=7.1 Hz), 7.92 (1H, br s); ¹³C-NMR (CDCl₃): δ 17.9, 23.5, 29.5, 46.0, 49.4, 57.0, 107.7, 110.6, 118.0, 119.3, 121.3, 127.2, 135.1, 135.8.

REFERENCES AND NOTES

- a) T. Itoh, Y. Matsuya, Y. Enomoto, K. Nagata, M. Miyazaki, and A. Ohsawa, *Synlett*, 1999, 1799.
 b) Y. Matsuya, T. Itoh, Y. Enomoto, and A. Ohsawa, *Heterocycles*, 2000, 53, 2357. c) T. Itoh, Y. Matsuya, Y. Enomoto, and A. Ohsawa, *Tetrahedron*, 2001, 57, 7277. d) T. Itoh, M. Miyazaki, S. Ikeda, K. Nagata, M. Yokoya, Y. Matsuya, Y. Enomoto, and A. Ohsawa, *Tetrahedron*, 2003, 59, 3527.
- 2. Y. Yamamoto and N. Asao, Chem. Rev., 1993, 93, 2207, and references cited therein.
- a) S. F. Martin, B. Benage, and J. E. Hunter, *J. Am. Chem. Soc.*, 1988, **110**, 5925. b) S. F. Martin, B. Benage, L. S. Geraci, J. E. Hunter, and M. Mortimore, *J. Am. Chem. Soc.*, 1991, **113**, 6161. c) R. Yamaguchi, T. Hamasaki, T. Sasaki, T. Ohta, K. Utimoto, S. Kozima, and H. Takaya, *J. Org. Chem.*, 1993, **58**, 1136. d) V. B. Birman and V. H. Rawal, *Tetrahedron Lett.*, 1998, **39**, 7219. e) V. B. Virman and V. H. Rawal, *J. Org. Chem.*, 1998, **63**, 9146.
- 4. a) L. H. Zhang and J. M. Cook, *Heterocycles*, 1988, 27, 1357. b) L. H. Zhang, A. K. Gupta, and J. M. Cook, *J. Org. Chem.*, 1989, 54, 4708. c) E. D. Cox, L. K. Hamaker, J. L. Peng, K. M. Czerwunski, L. Deng, D. W. Bannett, and J. M. Cook, *J. Org. Chem.*, 1997, 62, 44.
- a) H. Akimoto, K. Okamura, M. Yui, T. Shioiri, M. Kuramoto, Y. Kikugawa, and S. Yamada, *Chem. Pharm. Bull.*, 1974, 22, 2614. b) J. E. Johansen, B. D. Christie, and H. Rapoport, *J. Org. Chem.*, 1981, 46, 4914. c) M. F. Loewe and A. I. Meyers, *Tetrahedron Lett.*, 1985, 26, 3291. d) D. H. Hua, S. N. Bharathi, F. Takusagawa, A. Tsujimoto, J. A. K. Panangadan, M.-H. Hung, A. A. Bravo, and A. Erpelding, *J. Org. Chem.*, 1989, 54, 5659. e) D. H. Hua, S. N. Bharathi, J. A. K. Panangadan, and A. Tsujimoto, *J. Org. Chem.*, 1991, 56, 6998. f) H. Waldmann, M. Braun, M. Weymann, and M. Gewehr, *Tetrahedron*, 1993, 49, 397.
- 6. T.-S. Kam and K.-M. Sim, *Phytochemistry*, 1998, 47, 145.
- 7. A part of this work was reported in a preliminary communication; see, T. Itoh, M. Miyazaki, K. Nagata, M. Yokoya, S. Nakamura, and A. Ohsawa, *Heterocycles*, 2002, **58**, 115.
- 8. S. R. Johns, J. A. Lamberton, and J. L. Occolowitz, Aust. J. Chem., 1966, 19, 1951.
- 9. In the reference 5e, the racemization of 8 under the same conditions was not reported.
- a) B. A. Dadson, J. Harler-Mason, and G. H. Foster, J. Chem. Soc., Chem. Commun., 1968, 1233.
 b) J. Harley-Mason and C. G. Tayler, J. Chem. Soc., Chem. Commun., 1970, 812. c) G. C. Crawley and J. Harley-Mason, J. Chem. Soc., Chem. Commun., 1971, 685. d) W. R. Ashcroft, S. J. Martinez, and J. A. Joule, Tetrahedron, 1981, 37, 3005.
- 11. R. C. Bernotas and R. V. Cube, *Tetrahedron Lett.*, 1991, **32**, 161.
- 12. There is only one discrepancy between the reported data of (+)-harmicine and those of our synthetic compound: the chemical shift of C-11a carbon appeared at 135.1 ppm in the synthetic compound, whereas the reported one was at 133.5 ppm. Since we did not have opportunity to compare our spectral data with the natural product, we could not definitely confirm that our synthetic sample is identical to the natural one.