# HETEROCYCLES, Vol. 65, No. 4, 2005, pp. 871 - 877 Received, 6th January, 2005, Accepted, 4th February, 2005, Published online, 4th February, 2005 TWO NEW ISOFLAVONOIDS AND A NEW 2-ARYLBENZOFURAN FROM THE ROOTS OF *ERYTHRINA VARIEGATA*

Hitoshi Tanaka,<sup>\*,a</sup> Masaru Sudo,<sup>a</sup> Miyuki Hirata,<sup>a</sup> Magoichi Sako,<sup>b</sup> Masaru Sato,<sup>c</sup> Ih-Sheng Chen,<sup>d</sup> and Toshio Fukai<sup>e</sup>

<sup>a</sup>Faculty of Pharmacy, Meijo University, Yagoto, Tempaku-ku, Nagoya 468-8503, Japan, <sup>b</sup>Gifu Pharmaceutical University, Mitahora-higashi, Gifu 502-8585, Japan, <sup>c</sup>Department of Oral Pathology, Asahi University School of Dentistry, 1851-Hozumi, Mizuho, Gifu 501-0296, Japan, <sup>d</sup>School of Pharmacy, Kaohsiung Medical University, Kaohsiung, Taiwan, *R.O.C.* <sup>e</sup>School of Pharmaceutical Sciences, Toho University, Miyama, Funabashi, Chiba 274-8510, Japan E-mail address: hitoshi@ccmfs.meijo-u.ac.jp

**Abstract** – Two new isoflavonoids, eryvarins S (1) and T (2), and a new 2arylbenzofuran, eryvarin U (3), together with nine known compounds, were isolated from the roots of *Erythrina variegata*. Their structures were established on the basis of spectroscopic analysis. Eryvarin U is a rare naturally-occurring 2arylbenzofuran. The antibacterial activity of these three new compounds against 13 strains of methicillin-resistant *Staphylococcus aureus* (MRSA) was examined in which eryvarin U showed the highest antibacterial activity.

### **INTRODUCTION**

*Erythrina variegata* (Leguminosae) has been used as folk medicine in southern parts of Japan and China,<sup>1</sup> and comprises bioactive secondary metabolites with antimicrobial activity.<sup>2</sup> We previously reported the isolation of anti-MRSA isoflavonoids (erycristagallin, eryvarin Q and orientanol B) from the roots of this plant collected in Pakistan and Indonesia.<sup>3,4</sup> In continuation of our screening of anti-MRSA compounds from *Erythrina* plants, we describe the isolation and structural elucidation of a new isoflavone, eryvarin S (1), a new isoflavan, eryvarin T (2), and a new 2-arylbenzofuran, eryvarin U (3), together with nine known isoflavonoids from the roots of Taiwanese *E. variegata*. We also report the antibacterial activities of the new compounds (1–3) against MRSA. The nine known compounds were



identified as auriculatin,<sup>5</sup> bidwillols  $A^6$  and B,<sup>6</sup> erystagallin A,<sup>7</sup> erysubin E,<sup>8</sup> erythrabyssin II (8),<sup>9</sup> eryvarin K,<sup>10</sup> phaseollidin<sup>7</sup> and phaseollin<sup>11</sup> by comparing spectroscopic data with those of authentic samples or reported values.

# **RESULTS AND DISCUSSION**

Eryvarin S (1) was assigned a molecular formula of  $C_{25}H_{26}O_4$  ([M]<sup>+</sup> m/z 390.1823) from the HREIMS spectrum. The IR spectrum showed the presence of conjugated carbonyl (1620 cm<sup>-1</sup>) and hydroxyl (3380 cm<sup>-1</sup>) groups. The UV spectrum and the typical singlet signal assigned to H-2 ( $\delta$  7.97) in the <sup>1</sup>H NMR spectrum revealed that compound (1) is an isoflavone derivative.<sup>12,13</sup> The <sup>1</sup>H NMR spectrum exhibited a singlet aromatic proton ( $\delta$  7.96) and two  $\gamma$ , $\gamma$ -dimethylallyl (prenyl) groups ( $\delta$  1.80, 3.45 and 5.33, and 1.77, 1.86, 3.61 and 5.27), as well as a set of AA'BB'-type aromatic protons (8 6.84 and 7.37) on a 4-hydroxyphenyl group. The placement of one of the prenyl groups at the C-6 position was assigned by NOESY experiment which revealed NOE interactions between H-1"/H-5 and H-2"/H-5 (H-5 was assigned with the HMBC spectrum; the cross-peak between H-5 and C-4). Further support for the assignment of the prenyl group at the C-6 position was obtained from the HMBC spectrum which indicated correlations between H-1"/C-5, H-1"/C-6, H-1"/C-7 and H-5/C-1". The other prenyl group was located at the C-8 position as showed from the HMBC spectrum, indicating correlations between H-1"'/C-7, H-1"'/C-8 and H-1"'/C-9. Therefore, the structure of eryvarin S is represented by 1. Eryvarin T (2) was obtained in racemic form and its molecular formula was determined as  $C_{17}H_{18}O_5$  $([M]^+ m/z 302.1161)$  from the HREIMS spectrum. This compound was found to be an isoflavan on the basis of its characteristic spectral data:  $\lambda_{max}$  230 and 289 nm in the UV spectrum and a set of aliphatic proton signals (§ 2.77, 2.96, 3.47, 3.96 and 4.17) in the <sup>1</sup>H NMR spectrum. <sup>14</sup> The <sup>1</sup>H NMR spectrum showed three aromatic protons in an AMX system ( $\delta$  6.28, 6.36 and 6.89), and two singlet aromatic protons (\$ 6.57 and 6.85) and two methoxyl groups (\$ 3.77 and 3.78) on a 1,2,4,5-tetrasubstituted benzene moiety. The placement of the C-5 position in the AMX-type was confirmed from both the

Table 1. C INVIX spectral data for $1-3$				-
position	1 <sup>a</sup>	<b>2</b> <sup>b</sup>	<b>3</b> <sup>a</sup>	HMBC spectrum (correlations: H-5/C-4, H-5/C-9 and H-
2	152.3	70.6	151.7	5/C-7). The positions of the methoxyl groups at the C-2'
3	124.5	32.5	103.5	and C-5' were assigned from the NOESY spectrum which
4	176.8	31.3	120.9	displayed NOE interactions between OMe-2'/H-3' and
5	124.4	131.0	111.7	$C_{\rm M}$
6	126.3	108.7	153.3	OMe-5/H-6. The further assignments of the methoxyl
7	157.7	157.4	98.0	groups were obtained by the HMBC experiment,
8	114.6	103.6	154.7	revealing correlations between OMe-2'/C-2' and OMe-
9	154.2	156.1	123.5	5'/C-5'. The attachment of the B-ring to the isoflavan
10	118.0	114.3		mojety at the C-3 position was established from both the
1'	124.0	120.6	116.7	NOESN 14 CHOE : ( ) H CHI 2 1 H CHI
2'	130.3	152.9	153.1	NOESY data (NOE interactions: H-6/H-2a and H-6/H-
3'	115.7	101.0	115.4	4a) and the HMBC experiment (correlation: H-6 <sup>'</sup> /C-3).
4'	155.9	146.9	154.1	Therefore, the structure of eryvarin T is represented by <b>2</b> .
5'	115.7	142.2	112.9	Ervyarin U (3) was assigned a molecular formula of
6'	130.3	113.0	127.1	$C_{1}H_{1}O_{2}$ ([M] <sup>+</sup> m/z 222 1214) from the HPEIMS
1"	29.8			$C_{20}H_{18}O_4$ ([M] $m/2$ 522.1214) from the fixed method
2"	121.0		76.3	spectrum. This compound was found to be a 2-
3"	135.8		130.8	arylbenzofuran derivative on the basis of the UV spectral
4"	18.0		116.8	data and the characteristic olefinic proton signal ( $\delta$ 7.10)
5"	25.9		27.9	in the <sup>1</sup> U NMD supertrain <sup>15,16</sup> The <sup>1</sup> U NMD supertrain
6"			27.9	In the H NMR spectrum. The H NMR spectrum
1'''	22.3			showed three aromatic protons in an AMX system (\delta
2'''	120.7			6.76, 6.99 and 7.40), a set of ortho-coupled aromatic
3'"	135.1			protons ( $\delta$ 6 69 and 7 71) and a methoxyl group ( $\delta$ 3 78)
4'''	18.0			
5'''	25.8			as well as two methyl groups ( $\delta$ 1.46) and two olefinic
2'-OMe		56.4	61.1	protons (& 5.70 and 6.68) on a 2,2-dimethylpyran ring.
5'-OMe		57.3		- The location of the C-4 position in the AMX-type was
	-			

Table 1 <sup>13</sup>C NIMD anastrol data for 1 - 2

-	
<b>3</b> <sup>a</sup>	HMBC spectrum (correlations: H-5/C-4, H-5/C-9 and H-
151.7	5/C-7). The positions of the methoxyl groups at the C-2'
103.5	and C-5' were assigned from the NOESY spectrum which
120.9	displayed NOE interactions between OMe-2'/H-3' and
111.7	OMe-5'/H-6' The further assignments of the methoxyl
153.3	source static design the state state in the interiory
98.0	groups were obtained by the HMBC experiment,
154.7	revealing correlations between OMe-2'/C-2' and OMe-
123.5	5'/C-5'. The attachment of the B-ring to the isoflavan
116.7	moiety at the C-3 position was established from both the
153.1	NOESY data (NOE interactions: H-6'/H-2a and H-6'/H-
115.4	4a) and the HMBC experiment (correlation: H-6'/C-3).
154.1	Therefore, the structure of eryvarin T is represented by <b>2</b> .
112.9	Eryvarin U (3) was assigned a molecular formula of
127.1	$C_{20}H_{18}O_4$ ([M] <sup>+</sup> <i>m/z</i> 322.1214) from the HREIMS
76.3	spectrum. This compound was found to be a 2-
130.8	arylbenzofuran derivative on the basis of the UV spectral
116.8	data and the characteristic olefinic proton signal ( $\delta$ 7 10)
27.9	
27.9	in the <sup>1</sup> H NMR spectrum. <sup>13,10</sup> The <sup>1</sup> H NMR spectrum
	showed three aromatic protons in an AMX system ( $\delta$
	6.76, 6.99 and 7.40), a set of ortho-coupled aromatic
	protons ( $\delta$ 6.69 and 7.71) and a methoxyl group ( $\delta$ 3.78),

NOESY data (NOE interaction: H-5/H-4b) and the

determined from the HMBC spectrum which displayed

a: CDCl<sub>3</sub>. b: acetone- $d_6$ .

correlations between H-4/C-3 and H-4/C-8. The assignment of the methoxyl group at the C-2' position was confirmed from both the NOESY spectrum (NOE interactions: MeO-2'/H-4" and MeO-2'/H-3) and the HMBC spectrum (correlation: MeO-2'/C-2'). The presence of the 2,2-dimethylpyran moiety was evidenced from the EIMS spectrum that revealed the characteristic fragment ion at m/z 307 [M-CH<sub>3</sub>]<sup>+,17</sup> The placement of the 2,2-dimethylpyran moiety fused to the C-3' and C-4' positions was decided from the HMBC spectrum, indicating correlations between H-3"/C-3' and H-5'/C-4'. Therefore, the structure of eryvarin U is represented by **3**.

Antibacterial activity of the three new compounds (1-3) against 13 strains of MRSA was evaluated by determining minimum inhibitory concentration (MIC) and minimum bactericidal concentration (MBC). The MIC and MBC were determined by a broth dilution method as previously reported.<sup>4</sup> Eryvarin U inhibited the growth of MRSA strains with MIC values of  $6.25-12.5 \ \mu g \ mL^{-1}$ , and both MIC<sub>50</sub> and MIC<sub>90</sub> (minimum concentration needed to inhibit the growth of 50% and 90% of the tested strains, respectively) were 6.25 µg mL<sup>-1</sup>. Eryvarin S also showed growth inhibitory potency (MIC values of 6.25–25  $\mu$ g mL<sup>-1</sup>), but failed to inhibit 2 strains at 25  $\mu$ g mL<sup>-1</sup> (the maximum concentration in the present study). Eryvarin T did not show anti-MRSA activity at 25 µg mL<sup>-1</sup>. The MIC values of mupirocin, an authentic antibiotic for MRSA, against tested strains ranged from 0.20 to 3.13  $\mu$ g mL<sup>-1</sup> (MIC<sub>50</sub> and MIC<sub>90</sub>: both 0.39  $\mu$ g mL<sup>-1</sup>). Although the growth inhibitory potency of eryvarin U was lower than that of mupirocin, it inhibited the recovery of MRSA cells at  $6.25-12.5 \ \mu g \ mL^{-1}$  (MBC<sub>50</sub>: 12.5  $\mu$ g mL<sup>-1</sup>). This value was lower than that of mupirocin (MBC<sub>50</sub>: 25  $\mu$ g mL<sup>-1</sup>). Thus, eryvarin U was revealed to have strong bactericidal potency against MRSA. The high bactericidal activity of eryvarin U is expected to have a great advantage in treating MRSA infections, because it would reduce a risk of development of resistant mutants. The new arylbenzofuran could be a potent leading compound for the development of phytotherapeutic agents against MRSA infections.

## **EXPERIMENTAL**

**General Experimental Procedures.** Optical rotation was measured using a JASCO DIP-370 digital polarimeter, and CD spectrum was recorded on a JASCO J-725 spectropolarimeter. IR spectra were recorded on a JASCO IR-810 spectrophotometer, and UV spectra were obtained in MeOH using a Beckman DU-530 spectrophotometer. MS spectra were obtained using a JEOL JMS-SX102A spectrometer. <sup>1</sup>H and <sup>13</sup>C NMR spectra were measured on a JEOL ALPHA-600 MHz spectrometer. The <sup>1</sup>H and <sup>13</sup>C NMR (Table 1) signals of the compounds (**1–3**) were assigned based on the <sup>1</sup>H-<sup>1</sup>H COSY, NOESY, HSQC and HMBC spectra. Column chromatography was performed using Merck silica gel (230–400 mesh). The procedure for the MIC and MBC measurement has been described in a previous publication.<sup>4</sup>

**Plant material.** The roots of *E. variegata* were collected in Kaohsiung, Taiwan, R.O.C. in November 2001. A voucher specimen (No. 011130) was deposited at the Department of Natural Product Chemistry in the Faculty of Pharmacy, Meijo University.

**Extraction and isolation.** The finely powdered roots (3.19 kg) were macerated with acetone (73 L) at 23 °C for 48 h (2 times) and the solvent was removed to give a residue that was divided into n-

hexane-, CH<sub>2</sub>Cl<sub>2</sub>-, and EtOAc-soluble fractions. The CH<sub>2</sub>Cl<sub>2</sub>-soluble fraction (228.4 g) was applied to silica gel column first eluted with CHCl<sub>3</sub>-acetone (40 : 1  $\rightarrow$  10 : 1.5  $\rightarrow$  3 : 1  $\rightarrow$  1 : 1) and acetone (each volume; 3 L, Column A) to afford 7 fractions. Fraction A3 (27.7 g) was subjected to silica gel column chromatography using CHCl<sub>3</sub>-acetone (40 : 1  $\rightarrow$  20 : 1) (each volume; 200 mL, Column B) to yield 24 fractions. Fraction B7 (1.93 g) was separated by silica gel column chromatography successively using *n*-hexane-acetone (5 : 1) and benzene-EtOAc (80 : 1) to furnish eryvarin U (3) (2.2) mg) and phaseollin (4.1 mg). Fraction B14 (1.51 g) was subjected to silica gel column chromatography successively using benzene-EtOAc (10:1) and *n*-hexane-acetone (2:1) to provide bidwillol A (314 mg), erythrabyssin II (264 mg) and eryvarin K (22 mg). Fraction B20 (845 mg) was purified by repeated silica gel column chromatography using *n*-hexane-acetone (2.5 : 1) to give auriculatin (33 mg). Fractions B22-24 (1.62 g) were separated by silica gel column chromatography successively using *n*-hexane-acetone (3:1) and benzene-EtOAc (10:1) to afford bidwillol B (51 mg), eryvarin S (1) (7.2 mg) and phaseollidin (83 mg). Fraction A4 (70.4 g) was applied to silica gel column chromatography using CHCl<sub>3</sub>-acetone (10 : 1.5  $\rightarrow$  1 : 1) and acetone (each volume; 500 mL, Column C) to yield 21 fractions. Fraction C5 (1.28 g) was separated by silica gel column chromatography successively using *n*-hexane-acetone (2.5 : 1  $\rightarrow$  1.5 : 1) and benzene-EtOAc (20 : 1  $\rightarrow$  3 : 1) to furnish bidwillol A (24 mg) and bidwillol B (4.7 mg). Fractions C6 and C7 (8.72 g) were applied to silica gel column chromatography successively using benzene-EtOAc (5 : 1  $\rightarrow$  3 : 1) (each volume; 40 mL, Column D) to provide 40 fractions. Fractions D5-10 (3.38 g) were separated by silica gel column chromatography successively using *n*-hexane-acetone (1.5:1) and benzene-EtOAc (10:1) $\rightarrow$  5 : 1) to give auriculatin (34 mg), erystagallin A (41 mg) and erysubin E (24 mg). Fractions D11-18 (2.58 g) were purified by silica gel column chromatography successively using *n*-hexane-acetone  $(3:1 \rightarrow 5:1)$  and benzene-EtOAc (10:1) to afford eryvarin T (2) (8.6 mg).

**Eryvarin S (1).** Amorphous powder; IR (film)  $v_{max}$  cm<sup>-1</sup>: 3380, 1620; UV  $\lambda_{max}$  nm (log  $\varepsilon$ ): 205 (4.62), 255 (4.46), 309 (3.97); EIMS *m/z* (rel. int.): 390 ([M]<sup>+</sup>, 100), 373 (24), 347 (16), 335 (47), 319 (72), 291 (21), 279 (29); HREIMS *m/z*: 390.1823 (M<sup>+</sup>, Calcd for C<sub>25</sub>H<sub>26</sub>O<sub>4</sub>, 390.1830); <sup>1</sup>H NMR (CDCl<sub>3</sub>):  $\delta$  1.77 (3H, s, H-5'''), 1.80 (6H, s, H-4'' and H-5''), 1.86 (3H, s, H-4'''), 3.45 (2H, d, *J* = 7.3 Hz, H-1''), 3.61 (2H, d, *J* = 7.3 Hz, H-1'''), 5.27 (1H, t, *J* = 7.3 Hz, H-2'''), 5.33 (1H, t, *J* = 7.3 Hz, H-1'''), 5.97 (1H, br s, OH), 6.18 (1H, s, OH), 6.84 (2H, d, *J* = 8.8 Hz, H-3' and H-5'), 7.37 (2H, d, *J* = 8.8 Hz, H-2' and H-6'), 7.96 (1H, s, H-5), 7.97 (1H, s, H-2); <sup>13</sup>C NMR: see Table 1.

**Eryvarin T (2).** Amorphous powder;  $[\alpha]_D \pm 0^\circ$ ; CD (MeOH; *c* 2.32 x 10<sup>-5</sup>): no Cotton effect; IR (film)  $v_{max}$  cm<sup>-1</sup>: 3420; UV  $\lambda_{max}$  nm (log  $\varepsilon$ ): 205 (4.66), 230 (sh, 4.11), 289 (3.87), 320 (3.43); EIMS

*m/z* (rel. int.): 302 ([M]<sup>+</sup>, 68), 180 (100), 167 (46), 165 (46), 137 (17), 122 (6); HREIMS *m/z*: 302.1161 (M<sup>+</sup>, Calcd for C<sub>17</sub>H<sub>18</sub>O<sub>5</sub>, 302.1153); <sup>1</sup>H NMR (acetone-*d*<sub>6</sub>):  $\delta$  2.77 (1H, ddd, *J* = 15.6, 5.4, 2.0 Hz, H-4b), 2.96 (1H, dd, *J* = 15.6, 11.2 Hz, H-4a), 3.47 (1H, m, H-3), 3.77 (3H, s, OMe-5'), 3.78 (3H, s, OMe-2'), 3.96 (1H, t-like, *J* = 10.3 Hz, H-2a), 4.17 (1H, ddd, *J* = 10.3, 3.4, 2.0 Hz, H-2b), 6.28 (1H, d, *J* = 2.4 Hz, H-8), 6.36 (1H, dd, *J* = 8.1, 2.4 Hz, H-6), 6.57 (1H, s, H-3'), 6.85 (1H, s, H-6'), 6.89 (1H, d, *J* = 8.1 Hz, H-5), 7.54 (1H, s, OH), 8.07 (1H, br s, OH); <sup>13</sup>C NMR: see Table 1.

**Eryvarin U (3).** Amorphous powder; IR (film)  $v_{max}$  cm<sup>-1</sup>: 3420, 1630; UV  $\lambda_{max}$  nm (log  $\varepsilon$ ): 214 (sh, 4.02), 236 (4.17), 273 (4.15), 319 (4.31), 329 (sh, 4.27); EIMS *m/z* (rel. int.): 322 ([M]<sup>+</sup>, 62), 307 (100), 292 (66); HREIMS *m/z*: 322.1214 (M<sup>+</sup>, Calcd for C<sub>20</sub>H<sub>18</sub>O<sub>4</sub>, 322.1204); <sup>1</sup>H NMR (CDCl<sub>3</sub>):  $\delta$  1.46 (6H, s, H-5" and H-6"), 3.78 (3H, s, OMe-2'), 5.70 (1H, d, *J* = 10.3 Hz, H-3"), 6.68 (1H, d, *J* = 10.3 Hz, H-4"), 6.69 (1H, d, *J* = 8.8 Hz, H-5'), 6.76 (1H, dd, *J* = 8.1, 2.2 Hz, H-5), 6.99 (1H, d, *J* = 2.2 Hz, H-7), 7.10 (1H, s, H-3), 7.40 (1H, d, *J* = 8.1 Hz, H-4), 7.71 (1H, d, *J* = 8.8 Hz, H-6'); <sup>13</sup>C NMR: see Table 1.

#### REFERENCES

- 'Dictionary of Chinese Herbal Medicine', ed. Jiangsu New Medical College, Shanghai People's Press, Shanghai, 1977, p. 1941.
- 2. L. A. Mitscher, S. Drake, S. R. Gollapudi, and S. K. Okwute, J. Nat. Prod., 1987, 50, 1025.
- 3. H. Tanaka, M. Sato, S. Fujiwara, M. Hirata, H. Etoh, and H. Takeuchi, *Lett. Appl. Microbiol.*, 2002, **35**, 494.
- 4. H. Tanaka, M. Hirata, H. Etoh, M. Sako, M. Sato, J. Murata, H. Murata, D. Darnaedi, and T. Fukai, *Chem. Biodiv.*, 2004, **1**, 1101.
- K. V. S. Raju, G. Srimannarayana, B. Ternai, R. Stanley, and K. R. Markham, *Tetrahedron*, 1981, 37, 957.
- 6. M. Iinuma, Y. Okawa, T. Tanaka, Y. Kobayashi, and K. Miyauchi, *Heterocycles*, 1994, **39**, 687.
- 7. H. Tanaka, T. Tanaka, and H. Etoh, *Phytochemistry*, 1997, 45, 835.
- 8. H. Tanaka, H. Etoh, N. Watanabe, H. Shimizu, M. Ahmad, and G. H. Rizwani, *Phytochemistry*, 2001, **56**, 769.
- 9. V. S. Kamat, F. Y. Chuo, I. Kubo, and K. Nakanishi, *Heterocycles*, 1981, 15, 1163.
- H. Tanaka, M. Hirata, H. Etoh, M. Sako, M. Sato, J. Murata, H. Murata, D. Darnaedi, and T. Fukai, *Heterocycles*, 2003, 60, 2767.
- 11. D. R. Perrin, C. P. Whittle, and T. J. Batterham, Tetrahedron Lett., 1972, 1673.
- 12. K. R. Markham and T. J. Mabry, Ultraviolet-visible and Proton Magnetic Resonance Spectroscopy of Flavonoids, in *The Flavonoids*. ed. by J. B. Harborne, T. J. Mabry, and H.

Mabry, Chapman and Hall, London, 1975, pp. 45-77.

- 13. H. Tanaka, T. Tanaka, H. Etoh, N. Watanabe, M. Ahmad, I. Qurashi, and M. R. Khan, *Heterocycles*, 1998, **48**, 2661.
- 14. H. Tanaka, T. Oh-Uchi, H. Etoh, M. Sako, F. Asai, T. Fukai, M. Sato, J. Murata, and Y. Tateishi, *Phytochemistry*, 2003, **64**, 753.
- 15. P. M. Dewick, Isoflavonoids, in *The Flavonoids: Advances in Research*. ed. by J. B. Harborne and T. J. Mabry, Chapman and Hall, London, 1982, p. 537.
- S. Demizu, K. Kajiyama, K. Takahashi, Y. Hiraga, S. Yamamoto, Y. Tamura, K. Okada, and T. Kinoshita, *Chem. Pharm. Bull.*, 1988, 36, 3474.
- 17. M. Takayama, T. Fukai, Y. Hano, and T. Nomura, *Heterocycles*, 1992, 33, 405.