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## SYNTHESIS OF SOME BENZO-14-CROWN-4 ETHERS SUBSTITUTED TO 7,8-DIHYDROXY-3-PHENYLCOUMARIN DERIVATIVES

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**Abstract** – 7,8-Dihydroxy-3-phenylcoumarin derivatives reacted with 1,2-bis-(3-tosyloxypropoxy)benzene in CH<sub>3</sub>CN/alkali carbonate to furnish 3-phenylchromenone-14-crown-4 ether derivatives, which were identified with elemental analysis, IR, <sup>1</sup>H NMR, <sup>13</sup>C NMR and MS spectroscopy.

### INTRODUCTION

Macrocyclic molecules have attracted much attention because of their potential use in a variety of chemical processes, e.g., selective complexing agents for alkaline-earth metal ions, photo-induced electron transfer bio-mimetic studies, etc.<sup>1-4</sup> The coumarin nucleus is a very interesting chromophore due to its photochemical and photophysical properties and has been used to convert crown ethers and cryptands into fluorescent probes for alkaline and alkaline-earth metal ions.<sup>5</sup> In addition, it has been shown that 3-phenylcoumarins exhibit strong fluorescence emission intensities due to their high quantum yields.<sup>6</sup> We have recently synthesised various crown ethers with different chromophore moieties. The crown ether derivatives of the 4-*H* and 4-methyl-6,7-dihydroxy- and 7,8-dihydroxycoumarins displayed the binding effect of alkaline cations on the fluorescence emission spectra.<sup>7</sup>

This paper deals with the synthesis and structure elucidation of novel 3-phenyl-2*H*-chromenone derivatives of 14-crown-4 macrocycles.

### RESULTS AND DISCUSSION

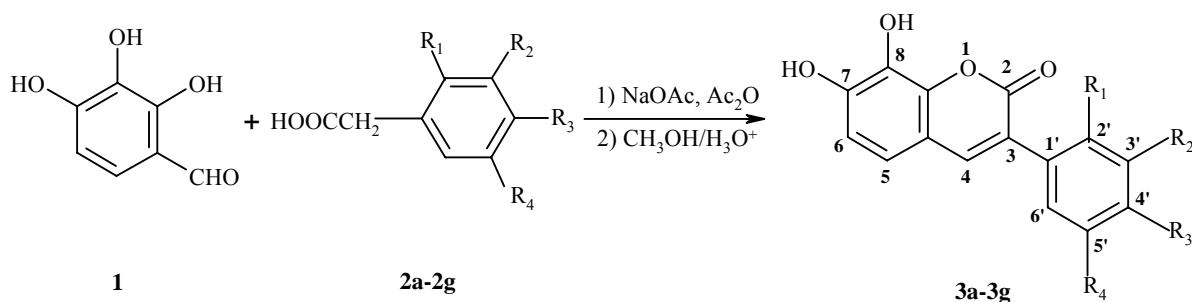
Benzo-14-crown-4 ethers (**6a-6g**) substituted to 7,8-dihydroxy-3-phenylcoumarin derivatives were synthesized from the 1,2-bis-(3-tosyloxypropoxy)benzene (**5**) and the corresponding 7,8-dihydroxy-3-phenylcoumarin derivatives (**3a-3g**) which were prepared from 2,3,4-trihydroxybenzaldehyde and corresponding phenylacetic acid in NaOAc/Ac<sub>2</sub>O mixture.

Compound (**5**) was reacted with **3a-3g** to give the crown ethers (**6a-6g**) respectively in the presence of  $\text{Na}_2\text{CO}_3$  in  $\text{CH}_3\text{CN}$ . The residue was chromatographed over a silica gel column eluting with  $\text{CHCl}_3$ . The fractions were further purified by preparative methods and the 3-phenylchromenone-14-crown-4 ethers (**6a-6g**) were obtained in 13-21% yield. The synthetic approach of benzo-14-crown-4 ethers (**6a-6g**) is shown in **Scheme 1**. The structures of the compounds (**6a-6g**) were characterized by elemental analysis,  $^1\text{H}$  and  $^{13}\text{C}$  NMR, MS and IR spectroscopies.

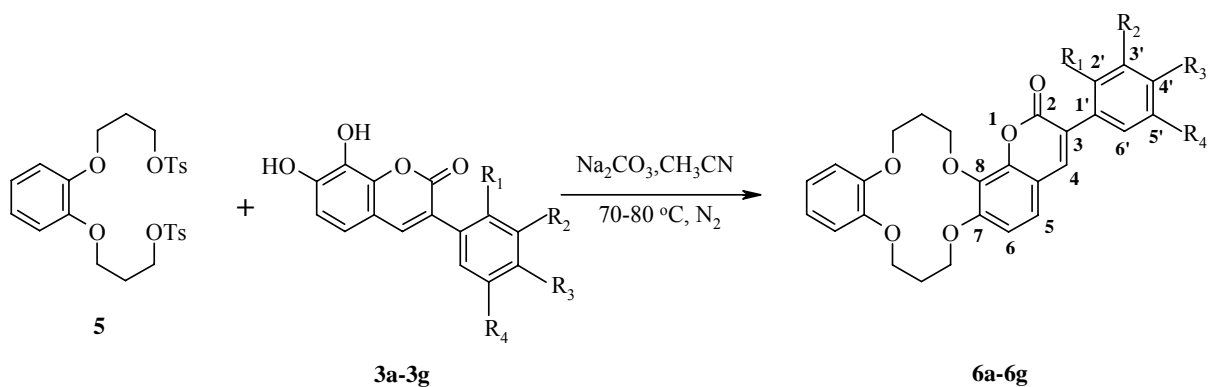
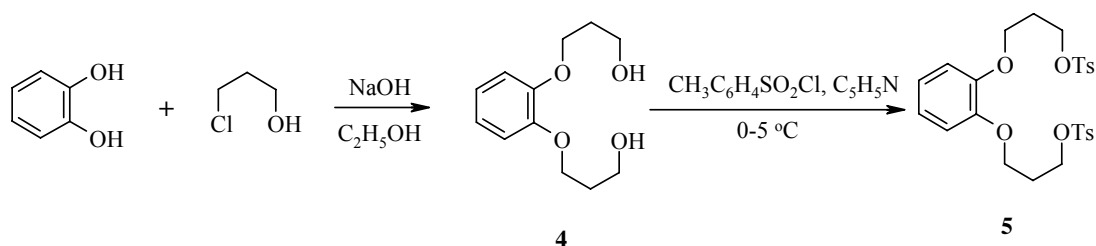
The IR spectra of **6a** showed two absorption at  $2876\text{-}2927\text{ cm}^{-1}$  for their C-H stretching frequency. The characteristic absorptions of the carbonyl group (C=O) and benzene ring were appeared at  $1720\text{ cm}^{-1}$  and  $1625\text{ cm}^{-1}$  respectively. The C-O-C ether chain of the crown ether was characterized by an absorption at  $1310\text{-}1090\text{ cm}^{-1}$ . A crown ether skeleton could also be deduced from analysis of the  $^1\text{H}$ NMR spectral data. The  $^1\text{H}$ NMR spectrum of compound (**6a**), which showed two triplets for the methylene protons  $[-\text{OCH}_2\text{CH}_2\text{CH}_2\text{O}-]$  at  $\delta$  4.31 ( $J = 6\text{ Hz}$ ) and  $\delta$  4.48 ( $J = 6\text{ Hz}$ ) and one quintet for the another methylene protons  $[-\text{OCH}_2\text{CH}_2\text{CH}_2\text{O}-]$  at  $\delta$  2.33 ppm ( $J = 6\text{ Hz}$ ), implied the presence of 14-crown-4 ether. In addition the chemical shifts of the aromatic protons were observed in the region of  $\delta$  6.88-7.72 ppm. The structure of newly synthesized **6a** was also checked by MS spectrometry. In the MS spectrum of **6a**, we observed a molecular ion at  $m/z$  444 $[\text{M}]^+$  (calcd for  $\text{C}_{27}\text{H}_{24}\text{O}_6$ ,  $M=444$ ).

The IR spectrum of **6b-6g** showed similar absorptions compared to the spectra of compound (**6a**). The  $^1\text{H}$  NMR spectra of compound (**6b**) showed a pair of doublet at  $\delta$  7.07 (br d,  $J = 8\text{ Hz}$ , H-3' and H-5') and  $\delta$  7.58 ppm (br d,  $J = 8\text{ Hz}$ , H-2' and H-6') for aromatic protons and a singlet at  $\delta$  2.40 ppm for methyl protons. As expected, the absorptions of the C-O-C ether chain, benzene ring and carbonyl group of compound (**6b**) were appeared at  $1300\text{-}1180\text{ cm}^{-1}$ ,  $1625\text{ cm}^{-1}$  and  $1720\text{ cm}^{-1}$  respectively.

The IR,  $^1\text{H}$  and  $^{13}\text{C}$  NMR spectral data of **6c** and **6d** were very similar to those of compound (**6b**). In the  $^1\text{H}$ NMR spectra of **6c**, signals were easily assigned to aromatic protons at  $\delta$  6.96 (br d,  $J = 8\text{ Hz}$ , H-3' and H-5') and  $\delta$  7.64 (br d,  $J = 8\text{ Hz}$ , H-2' and H-6') ppm. The methoxy protons showed a singlet at  $\delta$  3.85 ppm. In addition, the chemical shifts of the aromatic and ethereal carbons were observed around  $\delta$  114.09-160.76 ppm and  $\delta$  42.07-69.67 ppm respectively. Spectroscopic data of compound (**6d**) suggested the presence of *o*-methoxy substituted crown ethers. The structure of compound (**6e**) was elucidated by elemental analyses (C, H and N) and other spectroscopic data. The results of spectroscopy supported the structure of the new *p*-nitro substituted 14-crown-4 ether. The  $^1\text{H}$  NMR spectrum of **6f** and **6g** showed the expected resonances and integrals due to the protons of these two crown ether derivatives. Also MS spectra and elemental analysis data confirmed the formation of compounds (**6f**) and (**6g**).



|                      | 2a | 2b              | 2c               | 2d               | 2e              | 2f               | 2g               |
|----------------------|----|-----------------|------------------|------------------|-----------------|------------------|------------------|
| <b>R<sub>1</sub></b> | H  | H               | H                | OCH <sub>3</sub> | H               | H                | H                |
| <b>R<sub>2</sub></b> | H  | H               | H                | H                | H               | OCH <sub>3</sub> | OCH <sub>3</sub> |
| <b>R<sub>3</sub></b> | H  | CH <sub>3</sub> | OCH <sub>3</sub> | H                | NO <sub>2</sub> | OCH <sub>3</sub> | OCH <sub>3</sub> |
| <b>R<sub>4</sub></b> | H  | H               | H                | H                | H               | H                | OCH <sub>3</sub> |



|                      | 6a | 6b              | 6c               | 6d               | 6e              | 6f               | 6g               |
|----------------------|----|-----------------|------------------|------------------|-----------------|------------------|------------------|
| <b>R<sub>1</sub></b> | H  | H               | H                | OCH <sub>3</sub> | H               | H                | H                |
| <b>R<sub>2</sub></b> | H  | H               | H                | H                | H               | OCH <sub>3</sub> | OCH <sub>3</sub> |
| <b>R<sub>3</sub></b> | H  | CH <sub>3</sub> | OCH <sub>3</sub> | H                | NO <sub>2</sub> | OCH <sub>3</sub> | OCH <sub>3</sub> |
| <b>R<sub>4</sub></b> | H  | H               | H                | H                | H               | H                | OCH <sub>3</sub> |

Scheme 1. Synthesis of benzo-14-crown-4 ethers

## EXPERIMENTAL

Melting points were measured on a Gallenkamp apparatus uncorrected. IR spectra were recorded by using potassium bromide pellets on a Shimadzu FTIR-8300 spectrophotometer as KBr pellets. Elemental

analyses were performed by the Instrumental Analysis Laboratory of Tübitak-Ankara.  $^1\text{H}$  NMR and  $^{13}\text{C}$  NMR spectra were recorded in deuteriochloroform or methanol with an instrument Mercury-VX 400 MHz. EI MS spectra were recorded on a Fission UG-ZABSPEC.

The starting chemicals (**1**, **2a-2g**) were purchased from Aldrich or Merck. Initial compounds (**3a-3g**) and 3-phenylchromenone-14-crown-4 ether derivatives (**6a-6g**) have been prepared according to our early studies.<sup>8,9</sup>

### General procedure for the synthesis of 7,8-dihydroxy-3-phenylchromenones (**3a-3g**)

A mixture of (4.62 g, 30 mmol) of 2,3,4-trihydroxybenzaldehyde (**1**), (4.08 g, 30 mmol) of phenylacetic acid and (150 mmol) of sodium acetate in 25 mL of acetic anhydride was heated with stirring at 160 °C under  $\text{N}_2$  for 6 h. After removal of acetic acid by distillation, the resulting mixture was treated with  $\text{CH}_3\text{OH}/10\%$  HCl, and the precipitates were collected by filtration. The dried product was purified by recrystallisation from ethanol.

#### *7,8-Dihydroxy-3-phenyl-2H-chromen-2-one (3a)*

A mixture of compound (**1**) (4.62 g, 30 mmol), phenylacetic acid (**2a**) (4.08 g, 30 mmol), dry sodium acetate (12.30 g 150 mmol) in acetic anhydride (25 mL) was treated as described above to afford **3a**, 6.45 g (85%), mp 254 °C (ethanol); IR (KBr) 3361-3463, 3055, 1727, 1600-1625, 1200  $\text{cm}^{-1}$ ;  $^1\text{H-NMR}$  (MeOD, 400 MHz):  $\delta$  6.52 (d,  $J = 8.5$  Hz, 1H, H-6), 7.13 (d,  $J = 8.5$  Hz, 1H, H-5), 7.18 (br d,  $J = 8$  Hz, 2H, H-2' and 6'), 7.25 (t,  $J = 8$  Hz, 1H, H-4'), 7.38 (t,  $J = 8$  Hz, 2H, H-3' and 5'), 7.88 (s, 1H, H-4). Anal. Calcd for  $\text{C}_{15}\text{H}_{10}\text{O}_4$ : C 70.86; H 3.96. Found C 70.63; H 3.77.

#### *7,8-Dihydroxy-3-(4'-methylphenyl)-2H-chromen-2-one (3b)*

Compound (**1**) (4.62 g, 30 mmol), 4-methylphenylacetic acid (**2b**) (4.50 g, 30 mmol), dry sodium acetate (12.30 g 150 mmol) in acetic anhydride (25 mL) was treated as described above to afford **3b**, 4.58 g (57%), mp 220 °C (ethanol); IR (KBr) 3400, 3120, 2850-2953, 1710, 1590-1625, 1200  $\text{cm}^{-1}$ ;  $^1\text{H-NMR}$  (MeOD, 400 MHz):  $\delta$  2.41 (s, 3H,  $\text{CH}_3$ ), 6.70 (d,  $J = 8.5$  Hz, 1H, H-6), 6.83 (d,  $J = 8.5$  Hz, 1H, H-5), 6.95 (d,  $J = 8$  Hz, 2H, H-3' and H-5'), 7.18 (d,  $J = 8$  Hz, 2H, H-2' and H-6'), 7.85 (s, 1H, H-4). Anal. Calcd for  $\text{C}_{16}\text{H}_{12}\text{O}_4$ : C 71.64; H 4.47. Found C 71.03; H 4.32.

#### *7,8-Dihydroxy-3-(4'-methoxyphenyl)-2H-chromen-2-one (3c)*

Compound (**1**) (4.62 g, 30 mmol), 4-methoxyphenylacetic acid (**2c**) (4.98 g, 30 mmol), dry sodium acetate (12.30 g 150 mmol) in acetic anhydride (25 mL) was worked up as described above to give **3c**, 8.05 g (94%), mp 244 °C (ethanol); IR (KBr) 3400, 3100, 2870-2930, 1725, 1600-1630, 1150  $\text{cm}^{-1}$ ;  $^1\text{H-NMR}$  (MeOD, 400 MHz):  $\delta$  3.85 (s, 3H,  $\text{OCH}_3$ ), 6.70 (d,  $J = 8.5$  Hz, 1H, H-6), 6.83 (d,  $J = 8.5$  Hz, 1H,

H-5), 6.95 (d,  $J = 8$  Hz, 2H, H-3' and H-5'), 7.18 (d,  $J = 8$  Hz, 2H, H-2' and H-6'), 7.85 (s, 1H, H-4). Anal. Calcd for  $C_{16}H_{12}O_5$ : C 67.60; H 4.25. Found C 67.45; H 4.02.

*7,8-Dihydroxy-3-(2'-methoxyphenyl)-2H-chromen-2-one (3d)*

Compound **(1)** (4.62 g, 30 mmol), 2-methoxyphenylacetic acid (**2d**) (4.98 g, 30 mmol) dry sodium acetate (12.30 g, 150 mmol) in acetic anhydride (25 mL) was reacted as described above to give **3d**, 7.10 g (83%), mp 209 °C (ethanol); IR (KBr) 3400, 3100, 2850-2978, 1720, 1600-1620, 1180  $cm^{-1}$ ;  $^1H$ -NMR (MeOD, 400 MHz):  $\delta$  3.70 (s, 3H, OCH<sub>3</sub>), 6.70 (d,  $J = 8.5$  Hz, 1H, H-6), 6.83 (d,  $J = 8.5$  Hz, 1H, H-5), 6.93 (d,  $J = 8$  Hz, 1H, H-3'), 7.18 (dd,  $J = 8$  Hz,  $J = 2$  Hz, 1H, H-6'), 7.25 (t,  $J = 8$  Hz, 2H, H-5' and H-4'), 7.63 (s, 1H, H-4). Anal. Calcd for  $C_{16}H_{12}O_5$ : C 67.60; H 4.23. Found C 66.20; H 3.95.

*7,8-Dihydroxy-3-(4'-nitrophenyl)-2H-chromen-2-one (3e)*

Compound **(1)** (4.62 g, 30 mmol), 4-nitrophenylacetic acid (**2e**) (5.43 g, 30 mmol) dry sodium acetate (12.30 g, 150 mmol) in acetic anhydride (25 mL) was treated as described above to afford **3e**, 6.10 g (68%), mp 270 °C (ethanol); IR (KBr) 3400, 3050, 1700, 1600-1620, 1100  $cm^{-1}$ ;  $^1H$ -NMR (MeOD, 400 MHz):  $\delta$  6.70 (d,  $J = 8.5$  Hz, 1H, H-6), 6.83 (d,  $J = 8.5$  Hz, 1H, H-5), 6.85 (d,  $J = 8$  Hz, 2H, H-2' and H-6'), 7.46 (d,  $J = 8$  Hz, 2H, H-3' and H-5'), 7.90 (s, 1H, H-4). Anal. Calcd for  $C_{15}H_9NO_6$ : C 60.20; H 3.01. Found C 60.45; H 3.45.

*7,8-Dihydroxy-3-(3',4'-dimethoxyphenyl)-2H-chromen-2-one (3f)*

A mixture of compound **(1)** (4.62 g, 30 mmol), 3,4-dimethoxyphenylacetic acid (**2f**) (5.88 g, 30 mmol), dry sodium acetate (12.30 g, 150 mmol) in acetic anhydride (25 mL) was treated as described above to give **3f**, 7.72 g (82%), mp 220 °C (ethanol); IR (KBr) 3400, 3100, 2853-2950, 1700, 1580-1620, 1190  $cm^{-1}$ ;  $^1H$ -NMR (CD<sub>3</sub>OD, 400 MHz):  $\delta$  3.76 (s, 3H, OCH<sub>3</sub>), 3.78 (s, 3H, OCH<sub>3</sub>), 6.71 (d,  $J = 8.5$  Hz, 1H, H-6), 6.88 (d,  $J = 8.5$  Hz, 1H, H-5), 6.94 (d,  $J = 8.5$  Hz, 1H, H-5'), 7.15 (dd,  $J = 8.5$  and 2 Hz, 1H, H-6') 7.22 (d,  $J = 2$  Hz, 1H, H-2'), 7.80 (s, 1H, H-4). Anal. Calcd for  $C_{17}H_{14}O_6$ : C, 64.97; H, 4.49. Found: C 64.82; H 4.12.

*7,8-Dihydroxy-3-(3',4',5'-trimethoxyphenyl)-2H-chromen-2-one (3g)*

Compound **(1)** (4.62 g, 30 mmol), 3,4,5-trimethoxyphenylacetic acid (**2g**) (6.78 g, 30 mmol), dry sodium acetate (12.30 g, 150 mmol) in acetic anhydride (25 mL) was worked up as described above to give **3g**, 8.87g (86%), mp 238-239 °C (ethanol); IR (KBr) 3361-3463, 3055, 2851-2953, 1727, 1625, 1242  $cm^{-1}$ ;  $^1H$  NMR (CD<sub>3</sub>OD, 400 MHz):  $\delta$  3.69 (s, 3H, OCH<sub>3</sub>), 3.78 (s, 6H, OCH<sub>3</sub>), 6.72 (d,  $J = 8.5$  Hz, 1H, H-6), 6.90 (br s, 2H, H-2' and H-6'), 6.96 (d,  $J = 8.5$  Hz, 1H, H-5), 7.88 (s, 1H, H-4). Anal. Calcd. for  $C_{18}H_{16}O_7$ : C, 62.79; H, 4.68. Found: C, 61.60; H, 4.50.

### Synthesis of 1,2-bis(3-tosyloxypropoxy)benzene (5)

Catachol (24.25 g, 0.22 mol) was dissolved in 220 mL of C<sub>2</sub>H<sub>5</sub>OH and the mixture was stirred at 50 °C under N<sub>2</sub>. Finely ground NaOH (21.2 g, 0.53 mol) was added. After dissolving, 1-chloro-3-hydroxypropane (47 g, 0.50 mol) was added intervals over an ½ h period, the whole was stirred for 20 h at reflux temperature. The reaction mixture was worked up by filtering the undissolved salt then evaporated. The residue was dissolved in 750 mL of CHCl<sub>3</sub>. The CHCl<sub>3</sub> solution was washed with 1M 100 mL of NaOH and 2 x 200 mL of H<sub>2</sub>O. The CHCl<sub>3</sub> solution was dried over sodium sulfate and solvent stripped to yield 1,2-bis(3-hydroxypropoxy)benzene (4). 32 g (64%), mp 60 °C; IR (KBr) 3388, 3085, 2871-2900, 1600-1640, 1257-1053 cm<sup>-1</sup>; <sup>1</sup>H-NMR (CDCl<sub>3</sub>, 400 MHz): δ 2.05 (quintet, *J* = 5 Hz, 4H, -CH<sub>2</sub>CH<sub>2</sub>CH<sub>2</sub>-), 3.85 (t, *J* = 5 Hz, 4H, -CH<sub>2</sub>OH-), 4.16 (t, *J* = 5 Hz, 4H, -OCH<sub>2</sub>CH<sub>2</sub>CH<sub>2</sub>OH), 6.87-6.93 (m, 4H). Anal. Calcd for C<sub>12</sub>H<sub>18</sub>O<sub>4</sub>: C 63.70; H 7.96. Found C 63.40; H 7.80.

Compound (4) (5 g, 22 mmol) was dissolved in minimum amount of pyridine and cooled to -5 °C. Tosyl chloride (10.48 g, 55 mmol) in 10 mL of pyridine was added intervals over an ½ h. The reaction mixture was stirred for 2 days at 0-5 °C and then poured into the ice bath. The precipitate was filtered to yield 1,2-bis-(3-tosyloxypropoxy)benzene (5) 2 g (17%), mp 25 °C (ethanol); IR (KBr) 3056, 2873-2900, 1465-1590, 1386-1035 cm<sup>-1</sup>; <sup>1</sup>H-NMR (CDCl<sub>3</sub>, 400 MHz): δ 2.25 (s, 6H), 2.26 (quintet, *J* = 6 Hz, 4H, -CH<sub>2</sub>CH<sub>2</sub>CH<sub>2</sub>-), 3.77 (t, *J* = 6 Hz, 4H, -CH<sub>2</sub>OH-), 4.20 (t, *J* = 6 Hz, 4H, -OCH<sub>2</sub>CH<sub>2</sub>CH<sub>2</sub>-), 6.91-6.95 (m, 12H). Anal. Calcd for C<sub>26</sub>H<sub>30</sub>O<sub>8</sub>S<sub>2</sub>: C 58.43; H 5.62; S 12.00. Found C 58.30; H 5.25; S 12.44.

### General procedure for the synthesis of 3-phenylchromenone-crown ethers (6a-6g)

1,2-Bis-(3-tosyloxypropoxy)benzen (5) (1 g, 1.8 x 10<sup>-3</sup> mol), 7,8-dihydroxy-3-phenylcoumarin derivatives (3a-3g) (1.8 x 10<sup>-3</sup> mol) and sodium carbonate (0.38 g, 3.6 x 10<sup>-3</sup> mol) in 200 mL of acetonitrile were stirred for 5 days at 70-80 °C under N<sub>2</sub> atmosphere. The reaction mixture was acidified using 10% HCl and extracted with CHCl<sub>3</sub>. The crude product was chromatographed over a silica gel column eluting with chloroform, a gradient of chloroform-methanol up to 100% methanol.

#### 7,8-[6,7-Benzo-1,5,8,12-tetraoxadecylene]-3-phenyl-2H-1-benzopyran-2-one (6a)

Compound (5) (1 g, 1.8 x 10<sup>-3</sup> mol), 3a (0.457 g, 1.8 x 10<sup>-3</sup> mol) and sodium carbonate (0.38 g, 3.6 x 10<sup>-3</sup> mol) in acetonitrile (200 mL) was treated as describe above to afford 6a, (0.12 g, 15%), mp 110 °C (benzene); IR (KBr) 3100, 2927-2876, 1720, 1625, 1310, 1090; <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>): δ 2.33 (quintet, *J* = 6 Hz, 4H, H-3 and H-10), 4.31 (t, *J* = 6 Hz, 4H, H-4 and H-9), 4.48 (t, *J* = 6 Hz, 4H, H-2 and H-11), 6.88 (d, *J* = 8 Hz, 1H, coumarin H-6), 6.90-7.00 (m, 4H, benzo), 7.18 (d, *J* = 8 Hz, 1H, coumarin H-5), 7.40 (m, 3H, coumarin H-3', H-4' and H-5'), 7.68 (br d, *J* = 8 Hz, 2H, coumarin H-2' and H-6'), 7.72 (s, 1H, coumarin H-4); EIMS: *m/z* (rel. int.) 444 [M]<sup>+</sup> (27%). Anal. Calcd for C<sub>27</sub>H<sub>24</sub>O<sub>6</sub>: C 72.97; H 5.44. Found C 72.60; H 5.35.

*7,8-[6,7-Benzo-1,5,8,12-tetraoxadecylene]-3-[4'-methylphenyl]-2H-1-benzopyran-2-one (6b)*

Compound (**5**) (1 g,  $1.8 \times 10^{-3}$  mol), **3b** (0.482 g,  $1.8 \times 10^{-3}$  mol) and sodium carbonate (0.38 g,  $3.6 \times 10^{-3}$  mol) in acetonitrile (200 mL) was treated as describe above to afford **6b**, (0.14 g, 17%), mp 146-147 °C (benzene); IR (KBr) 3100, 2927-2876, 1720, 1625, 1300, 1180;  $^1\text{H}$  NMR (400 MHz,  $\text{CDCl}_3$ ):  $\delta$  2.32 (quintet,  $J = 6$  Hz, 4H, H-3 and H-10), 2.40 (s, 3H,  $-\text{CH}_3$ ), 4.31 (t,  $J = 6$  Hz, 4H, H-4 and H-9), 4.48 (t,  $J = 6$  Hz, 4H, H-2 and H-11), 6.88 (d,  $J = 8$  Hz, 1H, coumarin H-6), 6.88-6.95 (m, 4H, benzo), 7.07 (br d,  $J = 8$  Hz, 2H, coumarin H-3' and H-5'), 7.18 (d,  $J = 8$  Hz, 1H, coumarin H-5), 7.58 (br d,  $J = 8$  Hz, 2H, coumarin H-2' and H-6'), 7.70 (s, 1H, coumarin H-4); EIMS:  $m/z$  (*rel. int.*) 459  $[\text{M}+\text{H}]^+$  (13%). Anal. Calcd for  $\text{C}_{28}\text{H}_{26}\text{O}_6$ : C 73.35; H 5.67. Found C 73.58; H 5.88.

*7,8-[6,7-Benzo-1,5,8,12-tetraoxadecylene]-3-[4'-methoxyphenyl]-2H-1-benzopyran-2-one (6c)*

Compound (**5**) (1 g,  $1.8 \times 10^{-3}$  mol), **3c** (0.511 g,  $1.8 \times 10^{-3}$  mol) and sodium carbonate (0.38 g,  $3.6 \times 10^{-3}$  mol) in acetonitrile (200 mL) was treated as describe above to afford **6c**, (0.18 g, 21%), mp 180-181 °C (benzene); IR (KBr) 3070, 2927-2876, 1720, 1625, 1310, 1120;  $^1\text{H}$  NMR (400 MHz,  $\text{CDCl}_3$ ):  $\delta$  2.32 (quintet,  $J = 6$  Hz, 4H, H-3 and H-10), 3.85 (s, 3H,  $\text{OCH}_3$ ), 4.31 (t,  $J = 6$  Hz, 4H, H-4 and H-9), 4.48 (t,  $J = 6$  Hz, 4H, H-2 and H-11), 6.87 (d,  $J = 8$  Hz, 1H, coumarin H-6), 6.90-7.00 (m, 4H, benzo), 6.96 (br d,  $J = 8$  Hz, 2H, coumarin H-3' and H-5'), 7.18 (d,  $J = 8$  Hz, 1H, coumarin H-5), 7.64 (br d,  $J = 8$  Hz, 2H, coumarin H-2' and H-6'), 7.67 (s, 1H, coumarin H-4);  $^{13}\text{C}$  NMR (400 MHz,  $\text{CDCl}_3$ ):  $\delta$  42.07, 68.27, 69.67, 55.71, 112.94, 114.09, 122.25, 123.56, 123.97, 124.95, 139.02, 139.63, 147.22, 147.73, 148.30, 150.15, 159.94, 159.98, 160.76.; EIMS:  $m/z$  (*rel. int.*) 474  $[\text{M}]^+$  (20 %). Anal. Calcd for  $\text{C}_{28}\text{H}_{26}\text{O}_7$ : C 70.87; H 5.48. Found C 70.58; H 5.38.

*7,8-[6,7-Benzo-1,5,8,12-tetraoxadecylene]-3-[2'-methoxyphenyl]-2H-1-benzopyran-2-one (6d)*

Compound (**5**) (1 g,  $1.8 \times 10^{-3}$  mol), **3d** (0.511 g,  $1.8 \times 10^{-3}$  mol) and sodium carbonate (0.38 g,  $3.6 \times 10^{-3}$  mol) in acetonitrile (200 mL) was reacted as describe above to afford **6d**, (0.11 g, 13%), mp 175 °C (benzene); IR (KBr) 3055, 2927-2876, 1720, 1625, 1290, 1100;  $^1\text{H}$ -NMR(400 MHz,  $\text{CDCl}_3$ ): 2.20 (q,  $J = 5$  Hz, 4H, H-3 and H-10), 3.80 (s, 3H,  $\text{OCH}_3$ ), 4.13 (t,  $J = 5.5$  Hz, 4H, H-4 and H-9), 4.51 (t,  $J = 5.5$  Hz, 4H, H-2 and H-11), 6.86 (d,  $J = 8$  Hz, 1H, coumarin H-6), 6.90-7.00 (m, 4H, benzo), 6.94 (br d,  $J = 8$  Hz, 1H, coumarin H-6'), 7.05 (d,  $J = 8$  Hz, 1H, coumarin H-5), 7.10 (dd,  $J = 7$  Hz,  $J = 1.5$  Hz, 1H, coumarin H-3'), 7.22 (t,  $J = 7$  Hz, 1H, coumarin H-5'), 7.35 (t,  $J = 7$  Hz, 1H, coumarin H-4'), 7.70 (s, 1H, coumarin H-4). Anal. Calcd for  $\text{C}_{28}\text{H}_{26}\text{O}_7$ : C 70.87; H 5.48. Found C 71.05; H 5.35.

*7,8-[6,7-Benzo-1,5,8,12-tetraoxadecylene]-3-[4'-nitrophenyl]-2H-1-benzopyran-2-one (6e)*

Compound (**5**) (1 g,  $1.8 \times 10^{-3}$  mol), **3e** (0.538 g,  $1.8 \times 10^{-3}$  mol) and sodium carbonate (0.38 g,  $3.6 \times 10^{-3}$  mol) in acetonitrile (200 mL) was treated as describe above to afford **6e**, (0.18 g, 20%), mp 206-208 °C

(benzene); IR (KBr)  $\text{cm}^{-1}$  3100, 2927-2876, 1720, 1625, 1270, 1120;  $^1\text{H}$  NMR (400 MHz,  $\text{CDCl}_3$ ):  $\delta$  2.33 (quintet,  $J = 6$  Hz, 4H, H-3 and H-10), 4.31 (t,  $J = 6$  Hz, 4H, H-4 and H-9), 4.50 (t,  $J = 6$  Hz, 4H, H-2 and H-11), 6.90-7.00 (m, 4H, benzo), 6.93 (d,  $J = 8$  Hz, 1H, coumarin H-6), 7.20 (d,  $J = 8$  Hz, 1H, coumarin H-5), 7.81 (s, 1H, coumarin H-4), 7.83 (d,  $J = 8$  Hz, 2H, coumarin H-2' and H-6'), 8.22 (br d,  $J = 8$  Hz, 2H, H-3' and H-5'). Anal. Calcd for  $\text{C}_{27}\text{H}_{23}\text{NO}_8$ : C 66.25; H 4.70, N 2.86. Found C 65.75; H 4.38, N 2.83.

*7,8-[6,7-Benzo-1,5,8,12-tetraoxadecylene]-3-[3',4'-dimethoxyphenyl]-2H-1-benzopyran-2-one (6f)*

Compound (**5**) (1 g,  $1.8 \times 10^{-3}$  mol), **3f** (0.57 g,  $1.8 \times 10^{-3}$  mol) and sodium carbonate (0.38 g,  $3.6 \times 10^{-3}$  mol) in acetonitrile (200 mL) was treated as describe above to afford **6f**, (0.13 g, 14%), mp 182°C (benzene); IR (KBr) 3050, 2927-2876, 1720, 1625, 1350, 1150;  $^1\text{H}$ -NMR (400 MHz,  $\text{CDCl}_3$ ): 2.18 (q,  $J=6$  Hz, 4H, H-3 and H-10), 3.66 (t,  $J = 6$  Hz, 2H, H-9), 3.82 (s, 3H, 4'-OCH<sub>3</sub>), 3.86 (s, 3H, 3'-OCH<sub>3</sub>), 4.08 (t,  $J = 6$  Hz, 2H, H-4), 4.26 (t,  $J = 5.5$  Hz, 2H, H-11), 4.46 (t,  $J = 5.5$  Hz, 2H, H-2), 6.81 (d,  $J = 8$  Hz, 1H, coumarin H-6), 6.84-6.93 (m, 4H, benzo), 7.08 (d,  $J = 8$  Hz, 1H, coumarin H-5), 7.56 (dd,  $J = 9$  Hz,  $J=1.5$  Hz, 1H, coumarin H-6'), 7.63 (s, 1H, coumarin H-4), 7.64 (d,  $J = 1.5$  Hz, 1H, coumarin H-2'), 7.72 (d,  $J = 8.5$  Hz, 1H, coumarin H-5'); EIMS:  $m/z$  (*rel. int.*) 505  $[\text{M}+\text{H}]^+$  (10%), 504  $[\text{M}]^+$  (25%). Anal. Calcd for  $\text{C}_{29}\text{H}_{28}\text{O}_8$ : C 69.04; H 5.55. Found C 69.80; H 5.38.

*7,8-[6,7-Benzo-1,5,8,12-tetraoxadecylene]-3-[3',4',5'-trimethoxyphenyl]-2H-1-benzopyran-2-one (6g)*

Compound (**5**) (1 g,  $1.8 \times 10^{-3}$  mol), **3g** (0.62 g,  $1.8 \times 10^{-3}$  mol) and sodium carbonate (0.38 g,  $3.6 \times 10^{-3}$  mol) in acetonitrile (200 mL) was treated as describe above to afford **6g**, (0.17 g, 18%), mp 185°C (benzene); IR (KBr) 3100, 2927-2876, 1720, 1625, 1300, 1110;  $^1\text{H}$ -NMR (400 MHz,  $\text{CDCl}_3$ ): 2.08 (q,  $J = 6$  Hz, 4H, H-3 ve H-10), 3.82 (s, 3H, OCH<sub>3</sub>), 3.84 (s, 6H, OCH<sub>3</sub>), 4.14 (t,  $J = 6$  Hz, 4H, H-4 ve H-9), 4.23 (t,  $J = 6$  Hz, 2H, H-2), 4.40 (t,  $J = 6$  Hz, 2H, H-11), 6.80-6.90 (m, 4H, benzo), 7.42 (d,  $J = 6$  Hz, 1H, coumarin H-6), 7.44 (d,  $J = 7$  Hz, 1H, coumarin H-5), 7.60 (s, 1H, coumarin H-4), 7.62 (d,  $J = 2$  Hz, 2H, coumarin H-2' ve H-6'); EIMS:  $m/z$  (*rel. int.*) 535  $[\text{M}+\text{H}]^+$  (5 %), 534  $[\text{M}]^+$  (20 %). Anal. Calcd for  $\text{C}_{30}\text{H}_{30}\text{O}_9$ : C 67.41; H 5.62. Found C 67.15; H 5.48.

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## REFERENCES

1. B. Valeur and I. Leray, *Coordin. Chem. Rev.*, 2000, **3**, 205.
2. B. R. Pandya and Y. K. Agrawal, *Dyes and Pigments*, 2000, **52**, 161.
3. M. V. Alfimov and S. P. Gromov In: W. Rettig, B. Strehmel, S. Schrader, and H. Seifort, Ed.,



Applied fluorescence in chemistry, biology and medicine, Springer-Verlag, Berlin, Heidelberg, 1999, 161.

4. C. J. Pedersen, *Angew. Chem., Int. Ed. Engl.*, 1988, **100**, 1053.
5. Ç. Erk, A. Göçmen, and M. Bulut, *Supramol. Chem.*, 2000, **11**, 49; B. Raju and T. S. Vadarajan, *J. Phys. Chem.*, 1994, **98**, 8903.
6. M. Bulut and Ç. Erk, *J. Heterocycl. Chem.*, 2001, **38**, 1291.
7. Ç. Erk, A. Göçmen, and M. Bulut, *J. Incl. Phenom.*, 1998, **31**, 319; Ç. Erk, A. Göçmen, and M. Bulut, *J. Incl. Phenom.*, 2000, **37**, 441.
8. S. Abdurrahmanoğlu, C. Gündüz, Ü. Çakır, B. Çiçek, and M. Bulut, *Dyes and Pigments*, 2005, **65**, 197.
9. C. Gündüz, Ü. Salan, N. Özkul, İ. Başaran, Ü. Çakır, and M. Bulut, *Dyes and Pigments*, in press.